

**CHARACTERIZATIONS AND MECHANISMS  
OF METHANE AND NITROUS OXIDE  
EMISSIONS IN SHALLOW LAKES**

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2021

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## ***Chapter 1: Introduction for greenhouse gases emission in shallow lakes***

### **1.1 Introduction**

Carbon dioxide (CO<sub>2</sub>), methane (CH<sub>4</sub>), and nitrous oxide (N<sub>2</sub>O) are well-known greenhouse gasses (GHGs) (Lashof and Ahuja, 1990). The rapid increase in GHGs in the atmosphere is one of the primary reasons for global warming. Recently, World Meteorological Organization (WMO) reported that CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O concentrations in the atmosphere have reached 408 ppm, 1869 ppb, and 331 ppb, steadily increasing by 147%, 259%, and 147% since the mid-18<sup>th</sup> century, respectively (WMO, 2019). The global warming potential of CH<sub>4</sub> is approximately 34-times greater than that of CO<sub>2</sub> (Beaulieu et al., 2019). N<sub>2</sub>O is an ozone-depleting and highly potent GHG with a too-long lifetime, causing serious global warming after CO<sub>2</sub> and CH<sub>4</sub> in the stratosphere (Ravishankara et al., 2009). Given that the increasing GHGs concentrations in the atmosphere, GHGs emissions from natural ecosystems and engineering systems are paying more attention.

It has been well documented that the CH<sub>4</sub> emission from aquatic ecosystems significantly contributes to the global GHGs balance (Bastviken, 2011). Although the total area of freshwater lakes worldwide does not exceed 2% of the Earth's surface area, lakes have been considered as hotspots for CO<sub>2</sub> and CH<sub>4</sub> emissions. Recent studies indicate that CH<sub>4</sub> emissions from lakes account for 6–16% of the non-anthropogenic emissions, particularly the small size water bodies are significantly contribute to CH<sub>4</sub> emissions (Bastviken et al., 2008; Holgerson and Raymond, 2016). There are understood to be great uncertainties regarding the production and emission of CH<sub>4</sub> from shallow lakes (Li et al., 2018; Palma-Silva et al., 2013). The global average CH<sub>4</sub> flux from freshwaters is estimated to be 225.7 mmol m<sup>-2</sup> yr<sup>-1</sup> (Bastviken, 2011); however, this potentially includes an underestimation of the emissions from eutrophic lakes. The average CH<sub>4</sub> flux from Chinese lakes is estimated to be 531.5 mmol m<sup>-2</sup> yr<sup>-1</sup>, and emissions of up to 2106.3

mmol m<sup>-2</sup> yr<sup>-1</sup> have been reported for Lake Taihu, which is a hyper-eutrophic shallow lake (Yang et al., 2011). The underestimation of CH<sub>4</sub> emissions from hypereutrophic systems has also been mentioned in other studies (DelSontro et al., 2019; Schilder et al., 2018).

On the other hand, the atmospheric N<sub>2</sub>O has increased by 20%, from 1750 to 2018, and a steadily increasing trend at a rate of 0.2% per year (Tian et al., 2020). The IPCC reported that approximately 10% of anthropogenic N<sub>2</sub>O sources were derived from freshwater and coastal marine systems (IPCC, 2013). Given that inland freshwater lakes are recipients of nutrients transported from terrestrial ecosystems to trigger uncertainty of GHGs emissions (Wang et al., 2009; Zhou et al., 2020), they function as N<sub>2</sub>O hot spots where the turnovers of nitrogen (N) compounds are high. Hence, the N<sub>2</sub>O emissions from aquatic ecosystems have received considerable attention (Beaulieu et al., 2011; Kortelainen et al., 2020). Although multiple factors, *e.g.*, nutrients compositions, eutrophication, temperature, and so on, likely regulate N<sub>2</sub>O emissions from freshwater lakes at regional and global scales, a dominant factor for N<sub>2</sub>O emissions in freshwater lakes remains poorly understood (Kortelainen et al., 2020; Li et al., 2018; Yan et al., 2017). Moreover, there is a growing need to better evaluate and quantify freshwater lake ecosystems as a globally crucial N<sub>2</sub>O source (Lauerwald et al., 2019).

Given that the dramatically uncertain of GHGs emissions in aquatic ecosystems, understanding the mechanisms that drive the differences and order of magnitude changes in CH<sub>4</sub> and N<sub>2</sub>O emissions from freshwater lakes is crucial when effective policies and measures to tackle global warming are designed.

## **1.2 Lake eutrophication**

In the 1920s, the concept of lake eutrophication is first proposed by Nauman (1919). Lake eutrophication refers to the natural phenomenon that occurs during the development or aging of the lake, that is, the large accumulation of nutrients such as nitrogen and phosphorus in the lake exceeds the lake environmental threshold. Subsequently, Carlson

(1977) first proposed the numerical trophic state index (TSI) to describe the difference of the trophic state in different water bodies. Of that the model incorporates most lakes in a scale of 0 to 100. Correspondingly, the different trophic state lakes are clearly classified into five states: oligotrophic, mesotrophic, eutrophic, middle-eutrophic, and hyper-eutrophic. With the development of industrialization, a large number of nutrients and organic matters were transported to the lake, the trophic state was shifted from oligotrophic to eutrophic. Previously study reported that more than 30% of lakes in the world display a state of eutrophication, and they are showing a rapid upward trend (Ortiz-Llorente and Alvarez-Cobelas, 2012). To date, eutrophication, which has become a major problem affecting the lake manager, brings many environmental problems of such as water crisis, biodiversity destruction, alga booms, etc. Importantly, the uncertainty of GHG emission in the different trophic state lakes has brought considerable attentions (Chen et al., 2016; Beaulieu et al., 2019; Xiao et al., 2019).

### **1.3 CH<sub>4</sub> emissions in the lake ecosystems**

As we know, organic matter degradation is one of the most important processes for dominated CH<sub>4</sub> production and emission in aquatic ecosystems. In lacustrine ecosystems, methanogenic archaea and methanotrophic bacteria function as a CH<sub>4</sub> source and sink, respectively (Bastviken et al., 2008; Borrel et al., 2011). It has been well documented that CH<sub>4</sub> production rates are positively correlated with the diversities of methanogenic archaea and methanotrophic bacteria (He et al., 2019), and that their abundances and compositions are sensitive to environmental factors such as temperature, dissolved oxygen (DO), and organic matter concentrations (He et al., 2019; Henckel et al., 2000; Yang et al., 2019). Studies have documented that methanogenesis in lakes mainly occurs in anaerobic zones, whereas methanotrophic bacteria are present in oxic zones (Sepulveda-Jauregui et al., 2018). Therefore, the level of oxygen is a key parameter for determining the production and consumption of CH<sub>4</sub> in various situations (Bastviken et al., 2008; Borrel et al., 2011; Holgerson and Raymond, 2016). Temperature is also a



crucial controlling factor for these microbial processes (Marotta et al., 2014; Yvon-Durocher et al., 2014). A study reported a linear response of methanotrophy with increasing temperature under substrate-saturated conditions (Lofton et al., 2007), while substrate availability has been found to positively affect the methanotrophy rate more strongly than temperature (Dunfield et al., 1993; Nguyen et al., 2010). The growth of methanogenic archaea, however, is significantly facilitated by increasing concentrations of organic matter (Musenze et al., 2016). Moreover, the composition of organic matter has a close relationship with the microbial community structure. A previous study observed that different sources of organic matter had a significant influence on CH<sub>4</sub> emissions (Zhou et al., 2018). In particular, an elevated terrestrial organic matter input may potentially enhance the level of dissolved CH<sub>4</sub> in eutrophic shallow lakes. Collectively, these environmental factors determine the microbial processes that occur in lakes, and yet the various combinations of environmental factors induce a greater uncertainty regarding the actual CH<sub>4</sub> emissions from shallow lakes (Yvon-Durocher et al., 2014).

Shallow lakes are considered to be one of the important sources of CH<sub>4</sub> released into the atmosphere. The eutrophication of lakes is exacerbating CH<sub>4</sub> emissions from shallow lakes. It has been reported that eutrophic lakes have a larger emission of CH<sub>4</sub> to the atmosphere than oligotrophic lakes (Chen et al., 2016; Juutinen et al., 2009; Ortiz-Llorente and Alvarez-Cobelas, 2012). For example, the excessive growth of aquatic plants and algae is a source of organic matter, and the decomposition of this material by microorganisms can lead to the development of strong anaerobic conditions. This redox transition provides a positive feedback for CH<sub>4</sub> emissions from eutrophic shallow lakes (Schilder et al., 2018; Yan et al., 2017). To date, the production and flux of CH<sub>4</sub> from lakes have been extensively investigated (Bastviken et al., 2004; Beaulieu et al., 2019; DelSontro et al., 2019; Li et al., 2018). It is estimated that global GHG emissions are strongly biased by traditional approaches that extrapolate average emissions (DelSontro

et al., 2019). On the other hand, the estimation of CH<sub>4</sub> emissions from the world's lakes without taking account of nutrients and organic matter also results in great uncertainty (Beaulieu et al., 2019). Many previous studies were largely based on individual lakes, and the biogeographic-scale patterns of different types of trophic lakes have not been systematically investigated. The driving factors that induce extensive CH<sub>4</sub> emissions are also yet to be clarified. It is important to establish the relationship between the driving factors and CH<sub>4</sub> emissions by studying lake emissions with this biogeographic scale in mind (Beaulieu et al., 2019; DelSontro et al., 2019).

#### **1.4 N<sub>2</sub>O production and consumption in nitrogen cycles**

As the Fig. 1 shown that N<sub>2</sub>O is mainly produced by a by-product from nitrification and an intermediate from denitrification (Wenk et al., 2016). Of these multiple sources, denitrification is deemed a main N<sub>2</sub>O source in aquatic ecosystems (Beaulieu et al., 2011; Li et al., 2019; Salk and Ostrom, 2019). Importantly, N<sub>2</sub>O reduction, as the final step of denitrification ( $\text{NO}_3^- \rightarrow \text{NO}_2^- \rightarrow \text{NO} \rightarrow \text{N}_2\text{O} \rightarrow \text{N}_2$ ), is catalyzed by N<sub>2</sub>O reductase encoded by *nosZ* gene (Yoon et al., 2016). Namely, denitrification exerts either an N<sub>2</sub>O source or sink to determine N<sub>2</sub>O emission fluxes. Reportedly, N<sub>2</sub>O production is regulated by microbial community structures in an aquatic ecosystem (Zhao et al., 2018; Zhao et al., 2019). Therefore, understanding the impacts of N<sub>2</sub>O production and consumption by the variations of environmental factors and its contribution to the changes of microbial community structure is crucial.

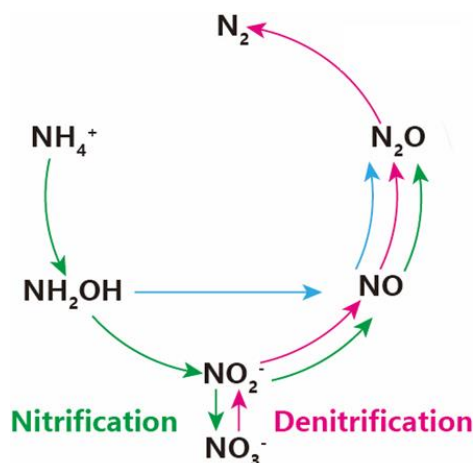


Fig. 1 N<sub>2</sub>O production and consumption pathway in nitrification and denitrification.

N<sub>2</sub>O-reducing bacteria (N<sub>2</sub>ORB), carried *nosZ* gene, is considered the N<sub>2</sub>O sink in N cycle (Conthe et al., 2019). Owing to differences in the N<sub>2</sub>O reductases (N<sub>2</sub>OR) amino acid sequences (similarity < 50%) and the protein-translocation pathways, N<sub>2</sub>O-reducing bacteria (N<sub>2</sub>ORB) harboring a functional *nosZ* gene can be classified into either clade I or clade II (Conthe et al., 2018; Jones et al., 2013; Sanford et al., 2012). Recently, Zhao et al. (2018) reported that the *nosZ* clade II-type N<sub>2</sub>ORB showed a higher diversity in the riparian zone than in other environments. Whereas some clade II type N<sub>2</sub>ORB display a higher affinity for N<sub>2</sub>O than clade I type N<sub>2</sub>ORB (Suenaga et al., 2018; Yoon et al., 2016), research on N<sub>2</sub>ORB biokinetics of N<sub>2</sub>O consumption toward an effective strategy to reduce N<sub>2</sub>O emissions is limited (Hallin et al., 2018; Yoon et al., 2016; Yoon et al., 2019). Therefore, more knowledge regarding the N<sub>2</sub>O sinks of N<sub>2</sub>ORB to understand the N<sub>2</sub>O sink activities in natural and engineering systems are urged.

The microbial community compositions are important for the amount of N<sub>2</sub>O emissions as some bacteria lack *nosZ* and nitrite reductase genes, significantly contributing to N<sub>2</sub>O consumption and production in natural ecosystems, respectively (Domeignoz-Horta et al., 2016). Among myriad environmental factors, nutrients, redox conditions, and temperature determine microbial guilds for N<sub>2</sub>O production and consumption (Hinshaw and Dahlgren, 2013; Xiao et al., 2019). Seasonal changes cover

multiple environmental fluctuations, probably affecting lacustrine microbial community structures and causing the uncertainty of N<sub>2</sub>O emissions (Song et al., 2012). Especially, the microbial N<sub>2</sub>O production and consumption processes carried different temperature dependencies. N conversion rates and eutrophication progress are crucial to regulate final N forms (N<sub>2</sub> or N<sub>2</sub>O) in aquatic ecosystems (Jiang et al., 2020; Salk and Ostrom, 2019; Zhu et al., 2020). Such N conversions are dynamic and dependent on eutrophication progress (Li et al., 2018; Liikanen et al., 2003).

### **1.5 N<sub>2</sub>O emission in the lake ecosystems**

N<sub>2</sub>O emissions from eutrophic shallow lakes have been also surveyed, providing the way how different N<sub>2</sub>O emission fluxes are among shallow lakes with different trophic states. For the survey, the commonly used default emission factor ( $EF_{5r}$ ) by the IPCC has been applied (Maavara et al., 2019), resulting in the underestimation or overestimation of the N<sub>2</sub>O budgets in freshwater ecosystems with different trophic states (Zhang et al., 2020). For instance, Xiao et al. (2019) indicated that the mean N<sub>2</sub>O emission fluxes in the East and West zones of Lake Taihu (eutrophic and oligotrophic, respectively) were substantially different, partially caused by the N-loadings flowing to the lake. Besides, N-loadings regulate the distributions of denitrifying bacteria, often indicated by functional genes for N<sub>2</sub>O production (*nirS* and *nirK*) and reduction (*nosZ*) (Huang et al., 2011; Zhao et al., 2018). Importantly, a high N-loading to a eutrophic lake accumulates algal growth, potentially enhancing N<sub>2</sub>O emissions and N turnover rates (Yan et al., 2017; Zhu et al., 2020). As the algal accumulation regarding decomposition for altering the redox condition favors denitrification (Yan et al., 2017), while further the anoxic condition limited nitrification for supplying nitrogen oxides to denitrification (Zhu et al., 2020), hence, these observations limit our understanding in the role of algae in N<sub>2</sub>O emission. Such correlation in eutrophic and hypereutrophic lakes have been intensively investigated (Lauerwald et al., 2019; Xiao et al., 2019); however, shallow lakes have not been subject to the systematic investigation, particularly comparing N<sub>2</sub>O emission fluxes among the

different trophic states (Salk and Ostrom, 2019). The trophic states of lakes brought the uncertainty of N<sub>2</sub>O emission estimates (Kortelainen et al., 2020), yet the relationship between N<sub>2</sub>O emission fluxes and trophic states of lakes is not holistically disentangled. Therefore, the investigation of N<sub>2</sub>O emissions associated with lacustrine trophic states on a biogeographic scale could help understand the potentials as N<sub>2</sub>O emission sources.

## **1.6 Objectives of this study**

Currently, numerous studies have well documented that the aquatic ecosystems are significantly contribute to CH<sub>4</sub> and N<sub>2</sub>O emission (Beaulieu et al., 2019; Beaulieu et al., 2010; DelSontro et al., 2019). However, the eutrophication of water bodies (*e.g.*, shallow lake, river, and reservoir) results to a large uncertainty in the CH<sub>4</sub> and N<sub>2</sub>O emission fluxes (Li et al., 2018). It is worth noted that the complex nitrogen cycles contained multiple pathways to produce N<sub>2</sub>O, while only one pathway can be the N<sub>2</sub>O sink. Therefore, understanding the denitrifier carried out the *nosZ* gene, encoding N<sub>2</sub>O reductase, to act the N<sub>2</sub>O sink is very important. Given that the shallow lakes carried different nutrients loadings, the carbon and nitrogen cycles associated the CH<sub>4</sub> and N<sub>2</sub>O emissions needs to be clarified. In this study, we investigated the CH<sub>4</sub> and N<sub>2</sub>O emission fluxes and environmental variables for different lakes along a trophic state gradient at the biogeographic scale of the Yangtze River basin, China, which includes a large number of shallow lakes. Based on the pure culture denitrifier, the key driving factors affecting N<sub>2</sub>O consumption were further analyzed. The key contents of research are listed in Fig. 2. The aims of this study were to (1) identify CH<sub>4</sub> and N<sub>2</sub>O emission patterns in the different trophic state shallow lakes; (2) analyze the relationship between the CH<sub>4</sub> and N<sub>2</sub>O emission flux and environmental variables as a means of exploring the driving factors that can be used to predict the CH<sub>4</sub> and N<sub>2</sub>O fluxes; (3) investigate the discrepancy of main microorganisms and functional genes for N<sub>2</sub>O emissions in the sediments of different trophic state shallow lakes; (4) understanding the N<sub>2</sub>O consumption process of N<sub>2</sub>ORB. This study is expected to expand our knowledge to accurately predict CH<sub>4</sub> and

N<sub>2</sub>O emission patterns from shallow lakes.

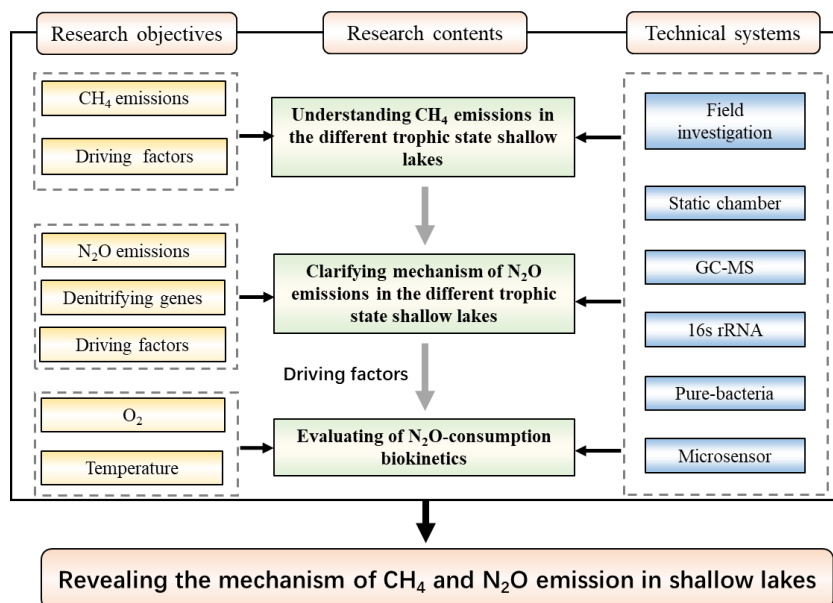


Fig. 2 Flow diagram of the research.

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## ***Chapter 2: Characterization of CH<sub>4</sub> emission and its microbial driven mechanisms in different trophic shallow lakes***

### **2.1 Introduction**

The rapid increase in greenhouse gases (GHGs) in the atmosphere is one of the primary reasons for global warming. The global warming potential of methane (CH<sub>4</sub>) is approximately 34-times greater than that of carbon dioxide (CO<sub>2</sub>) (IPCC, 2014). Bastviken et al. (2011) reported that the CH<sub>4</sub> fluxes from fresh waterbodies significantly contribute to the global GHG balance. Despite the fact that the total area of freshwater lakes worldwide does not exceed 2% of the Earth's surface area, lakes have been considered as hotspots for CO<sub>2</sub> and CH<sub>4</sub> emissions. Recent studies indicate that CH<sub>4</sub> emissions from lakes account for 6–16% of the non-anthropogenic missions, particularly the small size water bodies are significantly contribute to CH<sub>4</sub> emissions (Bastviken et al., 2008; Holgerson and Raymond, 2016). There are understood to be great uncertainties regarding the production and emission of CH<sub>4</sub> from shallow freshwater lakes (Zhuang et al., 2009; Palma-Silva et al., 2013; Li et al., 2018). The global average CH<sub>4</sub> flux from freshwaters is estimated to be 225.7 mmol m<sup>-2</sup> yr<sup>-1</sup> (Bastviken et al., 2011); however, this potentially includes an underestimation of the emissions from eutrophic lakes. The average CH<sub>4</sub> flux from Chinese lakes is estimated to be 531.5 mmol m<sup>-2</sup> yr<sup>-1</sup>, and emissions of up to 2106.3 mmol m<sup>-2</sup> yr<sup>-1</sup> have been reported for Lake Taihu, which is a hypereutrophic shallow lake (Yang et al., 2011). The underestimation of GHG emissions from hypereutrophic systems has also been mentioned in other studies (Schilder et al., 2017; DelSontro et al., 2018). Understanding the mechanisms that drive the order of magnitude differences in CH<sub>4</sub> emissions from lakes is crucial when designing effective measures to tackle global climate change.

Widespread evidence has revealed that the production and emission of CH<sub>4</sub> are

associated with microbial processes and are controlled by multiple environmental factors. In lacustrine ecosystems, methanogenic archaea and methanotrophic bacteria function as a CH<sub>4</sub> source and CH<sub>4</sub> sink, respectively (Borrel et al., 2011; Bastviken et al., 2015). It has been observed that CH<sub>4</sub> production rates are positively correlated with the diversities of methanogenic archaea and methanotrophic bacteria (He et al., 2019a), and that their abundances and compositions are sensitive to environmental factors such as temperature, dissolved oxygen (DO), and organic matter concentrations (Henckel et al., 2000; Yang et al., 2019; He et al., 2019a). Studies have documented that methanogenesis in lakes mainly occurs in anaerobic zones, whereas methanotrophic bacteria are present in oxic zones (Sepulveda-Jauregui et al., 2018). The level of oxygen is a key parameter for determining the production and consumption of CH<sub>4</sub> in various situations (Borrel et al., 2011; Bastviken et al., 2015; Holgerson and Raymond, 2016). Temperature is also a crucial controlling factor for these microbial processes (Yvon-Durocher et al., 2014; Marotta et al., 2014). Lofton et al. (2014) reported a linear response of methanotrophy with increasing temperature under substrate-saturated conditions, while substrate availability has been found to positively affect the methanotrophy rate more strongly than temperature (Dunfield et al., 1993; Duc et al., 2010). The growth of methanogenic archaea, however, is significantly facilitated by increasing concentrations of organic matter (Musenze et al., 2016). Moreover, the composition of organic matter has a close relationship with the microbial community structure. Zhou et al. (2018) observed that different sources of organic matter had a significant influence on CH<sub>4</sub> emissions. In particular, an elevated terrestrial organic matter input may potentially enhance the level of dissolved CH<sub>4</sub> in eutrophic shallow lakes. Collectively, these environmental factors determine the microbial processes that occur in lakes, and yet the various combinations of environmental factors induce a greater uncertainty regarding the actual CH<sub>4</sub> emissions from shallow lakes (Yvon-Durocher et al., 2014).

Shallow lakes receive and deposit nutrients and organic matter transported from the upstream regions of their catchments, which can lead to the transition from oligotrophic

to eutrophic states. It has been reported that eutrophic lakes have a larger emission of CH<sub>4</sub> to the atmosphere than oligotrophic lakes (Juutinen et al., 2009; Ortiz-Llorente and Alvarez-Cobelas, 2012; Chen et al., 2016). For example, the excessive growth of aquatic plants and algae is a source of organic matter, and the decomposition of this material by microorganisms can lead to the development of strong anaerobic conditions. This redox transition provides a positive feedback for CH<sub>4</sub> emissions from eutrophic shallow lakes (Schilder et al., 2017; Yan et al., 2017). To date, the production and flux of CH<sub>4</sub> from lakes have been extensively investigated (Bastviken et al., 2004; Rinta et al., 2015; DelSontro et al., 2018; Li et al., 2018; Beaulieu et al., 2019). It is estimated that global GHG emissions are strongly biased by traditional approaches that extrapolate average emissions (DelSontro et al. 2018). On the other hand, the estimation of CH<sub>4</sub> emissions from the world's lakes without taking account of nutrients and organic matter also results in great uncertainty (Beaulieu et al., 2019). Many previous studies were largely based on individual lakes, and the biogeographic-scale patterns of different types of trophic lakes have not been systematically investigated. The driving factors that induce extensive CH<sub>4</sub> emissions are also yet to be clarified. It is important to establish the relationship between the driving factors and CH<sub>4</sub> emissions by studying lake emissions with this biogeographic scale in mind (Rinta et al., 2015; DelSontro et al., 2018; Beaulieu et al., 2019).

In this study, we investigated the CH<sub>4</sub> emission fluxes and environmental variables for different lakes along a trophic state gradient at the biogeographic scale of the Yangtze River basin, China, which includes a large number of shallow lakes. We hypothesise that eutrophication, especially hyper-eutrophication, triggers a nonlinear increase in CH<sub>4</sub> emissions in shallow lakes. The aims of this study are to (1) elucidate the relationship between CH<sub>4</sub> emissions and the level of eutrophication in shallow lakes, and (2) analyse the relationship between the CH<sub>4</sub> emission flux and environmental variables as a means of exploring the driving factors that can be used to predict the CH<sub>4</sub> flux. The results of this study are intended to help guide the management of lakes and support the assessment of regional CH<sub>4</sub> budgets.



## **2.2 Material and methods**

### **2.2.1 Sampling sites and sample collection**

Overlying water and sediment samples from 17 shallow lakes along the middle and lower reaches of the Yangtze River (Fig. 1) were collected during August and September 2018. The lakes are located in Shanghai, Jiangsu, Anhui, Jiangxi, Hubei, and Hunan provinces, with a maximal distance between the lakes of > 1000 km. The selected shallow lakes had a mean depth of < 7 m and were selected based on their different trophic states, as determined from a literature review and government report. The physicochemical properties of the 17 sampling sites are listed in Table S1. Temperature, DO, and pH were measured *in-situ* at a water depth of ~20 cm using a HQ3d meter (Hach, USA). Triplicate water samples (20 cm below the water surface) and surface sediments samples (0–10 cm depth) were collected in the littoral zone of each lake. Gas samples were collected simultaneously using plastic, floating static chambers, which have been previously shown to provide unbiased measurements of water–gas exchange (Cole et al., 2010; Gålfalk et al., 2013). Measurements of GHG fluxes were made between 11:00 and 14:00 using a floating static chamber (38.5 cm × 30.5 cm × 18.5 cm), which was designed to collect gas samples every 10 minutes for 1 hour. Details of the sampling method can be found in Zhou et al. (2019).

### **2.2.2 Sample analysis**

The water samples were analysed for total nitrogen (TN), total phosphorus (TP),  $\text{NO}_3^-$ -N,  $\text{NH}_4^+$ -N, chlorophyll-*a* (chl-*a*), and dissolved organic carbon (DOC) contents. Triplicate water samples were collected using a tube sampler and mixed thoroughly. The ratio of C:N was determined by the DOC and TN concentrations in the overlying water. The analytical methods are described elsewhere (Liu et al., 2017; Xu et al., 2019a). The sediment samples were analysed for the sediment total nitrogen (STN), sediment total phosphorus (STP), and sediment total organic carbon (STOC) contents. The analytical

methods are given in Zhou et al. (2019). In addition, a conversion factor of 1.724 was used to estimate the organic matter (OM) content from  $\text{TOC:OM} = \text{TOC} \times 1.724$  (Zeng et al., 2018).

The  $\text{CO}_2$  and  $\text{CH}_4$  concentrations were analysed using a 7890B Agilent gas chromatography equipped with a flame ionisation detector (FID). The gas chromatography configurations described by Wu et al. (2018) were adopted for the gas concentration analysis.

### **2.2.3 DNA extraction and Illumina high-throughput sequencing**

The biomass for the analysis of the sediment microbial community was collected from four lakes with a different trophic state: Lake Taihu (TH), Lake Honghu (HH), Lake Poyanghu (PYH), and Lake Baidanghu (BDH). A MOBIO PowerSoil<sup>®</sup> DNA Isolation Kit (MOBIO, USA) was used to extract DNA from the sediment according to the manufacturer's instructions. The concentration and purity of DNA were measured using the NanoDrop One (Thermo Fisher Scientific, Ma, USA), and the extracted DNA was then stored at  $-20^\circ\text{C}$  awaiting further analysis. The 16S rRNA gene sequencing was analysed using an Illumina Miseq platform at Magigene Biotechnology Institute (Guangzhou, China). The primers used for the high-throughput sequencing were modified 515F (5'- GTGYCAGCMGCCGCGGTAA - 3') and 806R (5' - GGACTACHVGGGTWTCTAAT - 3'), thus targeting the V3 + V4 regions of both bacterial and archaeal 16S rRNA genes (Chen et al., 2017; Gong et al., 2019). Polymerase chain reaction (PCR) samples containing 25  $\mu\text{l}$  of  $2 \times$  Premix Taq (Takara Biotechnology, Dalian Co. Ltd., China), 1  $\mu\text{l}$  of each primer (10 mM), and 3  $\mu\text{l}$  of the DNA (20  $\text{ng } \mu\text{l}^{-1}$ ) template in a volume of 50  $\mu\text{l}$  were amplified by thermocycling. The amplification program was as follows: 5 min at  $94^\circ\text{C}$  for initialisation; 30 cycles of 30 s denaturation at  $94^\circ\text{C}$ , 30 s annealing at  $52^\circ\text{C}$ , and 30 s extension at  $72^\circ\text{C}$ ; 10 min final elongation at  $72^\circ\text{C}$ . The PCRs were conducted on a BioRad S1000 (Bio-Rad Laboratory, CA, USA).

### **2.2.4 Calculated trophic state index and greenhouse gas concentrations**

The trophic state of the 17 lake samples were evaluated using the trophic lake index (*TLI*) developed by Carlson (1977) and modified by Jin (1995) (Eq. 1), which is more suitable for Asian freshwater lakes. The lake trophic state was described by the *TLI* value, and the chl-*a* and TN concentration were used as criterion to calculate the lake trophic state (Zhang et al., 2011).

$$TLI(\Sigma) = \sum_{j=1}^m W_j \times TLI(j) \quad , \quad (1)$$

$$W_j = \frac{r_{ij}^2}{\sum_{j=1}^m r_{ij}^2}, \quad (2)$$

$$TLI(\text{chl-}a) = 10 \times [2.46 + \ln(\text{chl-}a)/\ln(2.5)], \quad (3)$$

$$TLI(\text{TN}) = 10 \times [5.453 + 1.694 \times \ln(\text{TN})], \quad (4)$$

where *TLI* ( $\Sigma$ ) is the integrated trophic level index; *TLI*(*j*) is the trophic level index of *j*; *W<sub>j</sub>* is the correlative constant for the *TLI* of *j*; *r<sub>ij</sub>* is the relative coefficient, which was described by Zhang et al. (2011); *j* and *m* represent the parameters (*i.e.* chl-*a* and TN) and the number of parameters, respectively. Subsequently, the calculated *W<sub>chl-a</sub>* and *W<sub>TN</sub>* were 0.5979 and 0.4021, respectively. The lake trophic state was further classified based on the *TLI* ( $\Sigma$ ) value as either oligotrophic (*TLI*( $\Sigma$ ) ≤ 30), mesotrophic (30 < *TLI*( $\Sigma$ ) ≤ 50), eutrophic (50 < *TLI*( $\Sigma$ ) ≤ 60), middle eutrophic (60 < *TLI*( $\Sigma$ ) ≤ 70), or hyper-eutrophic (*TLI* ( $\Sigma$ ) > 70).

The CH<sub>4</sub> and CO<sub>2</sub> emission fluxes estimated by the floating static chamber method were calculated using Eq. (5) (Zhou et al., 2019):

$$F = K \times \frac{V}{A} \times \frac{dc}{dt}, \quad (5)$$

where *F* is the emission flux (mg m<sup>-2</sup> h<sup>-1</sup>), *V* (m<sup>3</sup>) and *A* (m<sup>2</sup>) are the static chamber volume and surface area, respectively, *dc/dt* is the variation rate of the gas concentration (mg L<sup>-1</sup> min<sup>-1</sup>), and *K* is a scaling factor (*K* = 6 × 10<sup>4</sup>).

### 2.2.5 Statistical analysis

Statistical analysis was conducted using the SPSS 19.0 software (SPSS Inc., Chicago, USA). Any significant differences between the lakes were tested by conducting a one-

way analysis of variance (ANOVA). The criteria of  $p < 0.05$  and  $p < 0.01$  were used to determine the statistical significance at the 0.05 and 0.01 levels (two-tailed), respectively. The K-S test was carried out to check if the data was of a normal distribution. Multiple stepwise linear regression analysis was conducted to estimate the best predictors of the CH<sub>4</sub> and CO<sub>2</sub> emission fluxes.

## **2.3 Results**

### **2.3.1 Physicochemical parameters of the overlying water**

The environmental variables of the overlying water and sediments from the sampled lakes included field monitoring indicators and nutrient parameters. The characteristics of these indicators are shown in Table 1 and S1. During the study period, the temperature of the overlying water ranged from 27.2°C to 31.3°C; thus, all lakes had a temperature of > 25°C (Table S1). The physicochemical parameters exhibited large spatial variations and followed a trend corresponding to the trophic state (Table 1, S1). As shown in Table 1, the TN concentration displayed an increasing trend (from 0.6 to 3.4 mg L<sup>-1</sup>), whereby the eutrophic state shifted from mesotrophic to hypereutrophic. The concentrations of NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, TP, and chl-*a* also showed the same tendency. The TN and TP concentrations of the hyper-eutrophic lakes were approximately 5- and 3-times higher than those of the mesotrophic lakes, respectively. In particular, the chl-*a* concentration in the hyper-eutrophic lakes was approximately 13-times higher than that in the mesotrophic lakes (Table 1). Compared to overlying water parameters, the STN, STOC, and OM contents of the sampled sediments of eutrophic lakes were not significantly different from mesotrophic or hyper-eutrophic lakes ( $p > 0.05$ ) (Table 1). However, the STN, STOC, and OM concentrations of sediments in the mesotrophic lakes were significantly lower than those of the hyper-eutrophic lakes ( $p < 0.01$ ). In addition, the DO concentrations (8.2–5.6 mg L<sup>-1</sup>) were negatively correlated with the TN concentration ( $p < 0.05$ ) (Tables 1 and S2). There were no significant changes in the pH and DOC of the overlying water among the different trophic states. However, pH was significantly positively correlated with the

DO concentrations ( $p < 0.05$ ). The *TLI* and the C:N ratio displayed a negative correlation in the overlying water ( $R^2 = 0.75$ ,  $p < 0.01$ ) (Fig. 2). The C:N ratio decreased with the shift from a mesotrophic to hyper-eutrophic state.

### 2.3.2 CH<sub>4</sub> and CO<sub>2</sub> emissions fluxes in different trophic lakes

During the study period, the CH<sub>4</sub> (0.1–351.9 mg m<sup>-2</sup> h<sup>-1</sup>) and CO<sub>2</sub> emission fluxes (–19.1–352.2 mg m<sup>-2</sup> h<sup>-1</sup>) exhibited large inter-lake variations for the different trophic states (Table S3 and Fig. 3). The highest CH<sub>4</sub> and CO<sub>2</sub> emission fluxes were measured in Lake Chaohu (CH), which is hyper-eutrophic (Fig. 3). The mean CH<sub>4</sub> emission flux from the mesotrophic, eutrophic, middle-eutrophic, and hyper-eutrophic lakes was 0.1, 4.4, 12.0, and 130.4 mg m<sup>-2</sup> h<sup>-1</sup>, respectively (Table S3); thus, the mean CH<sub>4</sub> emission flux from hyper-eutrophic lakes was approximately 1000-, 30-, and 10-times that from mesotrophic, eutrophic and middle-eutrophic lakes, respectively. With a shift in the trophic state from mesotrophic to hyper-eutrophic, the studied lakes functioned as a CH<sub>4</sub> source, whereby eutrophication accelerated the emission of CH<sub>4</sub>. In addition, the mean CO<sub>2</sub> emission flux from hyper-eutrophic lakes was approximately 30-, 20-, and 4-times that of from mesotrophic, eutrophic, and middle-eutrophic lakes, respectively. It is interesting that this differed to the trend of the CH<sub>4</sub> emissions, as only the hyper-eutrophic lakes functioned as a complete source for CO<sub>2</sub> (Fig. 3b).

The CH<sub>4</sub> and CO<sub>2</sub> emission fluxes had positive correlations with the *TLI* (Fig. S1). The CH<sub>4</sub> emission flux displayed an exponential increase with the *TLI* in the studied lakes ( $R^2 = 0.67$ ,  $p < 0.01$ ) (Fig. S1a); meanwhile, the CO<sub>2</sub> emissions flux also had a significant exponential correlation ( $R^2 = 0.50$ ,  $p = 0.024$ ) (Fig. S1b). This was best fitted by a nonlinear exponential function to show the relationships between the mean CH<sub>4</sub> and CO<sub>2</sub> emission fluxes and *TLI* ( $R^2 > 0.99$ ) (Fig. 4). It is clear that the CH<sub>4</sub> and CO<sub>2</sub> emission fluxes increased with an increase in the *TLI*, and that the hyper-eutrophic lakes acted as a very large source of CH<sub>4</sub> and CO<sub>2</sub> emissions.

### 2.3.3 Relationship between CH<sub>4</sub> and CO<sub>2</sub> emissions and environmental parameters

The correlations between the CH<sub>4</sub> and CO<sub>2</sub> fluxes and physicochemical parameters of the overlying water were analysed (Table S2). The CO<sub>2</sub> emission flux was strongly and significantly correlated with the CH<sub>4</sub> emission flux ( $p < 0.01$ ). The CH<sub>4</sub> and CO<sub>2</sub> fluxes were significantly correlated with the TN, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, and chl-*a* concentrations of the overlying water ( $p < 0.05$ ). However, the DO and pH of the overlying water were both significantly negatively correlated with the CO<sub>2</sub> emission flux ( $p < 0.05$ ). In addition, the TP and DOC concentrations of the overlying water were not significantly correlated with the CH<sub>4</sub> and CO<sub>2</sub> emission fluxes ( $p > 0.05$ ). Further analysis revealed that ln(CH<sub>4</sub>) and CO<sub>2</sub> emission fluxes had significant, positive linear correlations with the TN ( $R^2 = 0.50$ ,  $p = 0.001$  for ln(CH<sub>4</sub>);  $R^2 = 0.33$ ,  $p = 0.015$  for CO<sub>2</sub>) and chl-*a* concentration ( $R^2 = 0.60$ ,  $p < 0.001$  for ln(CH<sub>4</sub>);  $R^2 = 0.48$ ,  $p = 0.002$  for CO<sub>2</sub>) (Fig. 5). The parameter of ln(CH<sub>4</sub>) exhibited a significant, negative linear correlation with the C:N ration of the overlying water ( $R^2 = 0.52$ ,  $p = 0.001$ ). However, there was no significant correlation between the CO<sub>2</sub> emission flux and the C:N ratio of the overlying water ( $p = 0.104$ ).

Multiple stepwise regression models incorporating the variables of the overlying water were established for the CH<sub>4</sub> and CO<sub>2</sub> emission fluxes (Table 2). The results showed that only NH<sub>4</sub><sup>+</sup>-N could be used to predict CH<sub>4</sub> emissions ( $R^2 = 0.74$ ,  $p < 0.01$ ). In addition, NH<sub>4</sub><sup>+</sup>-N and pH could jointly predict CO<sub>2</sub> emissions based on given samples ( $R^2 = 0.68$ ,  $p < 0.01$ ). Obviously, NH<sub>4</sub><sup>+</sup> ( $t_1 = 3.86$ ) was more positively correlated with CO<sub>2</sub> emissions than pH ( $t_2 = -2.66$ ), which suggests that pH was only a minor explanatory variable compared with NH<sub>4</sub><sup>+</sup>.

### **2.3.4 Abundance and community of methanogens and methanotrophs in different trophic lakes**

Methanogenic archaea and methanotrophic bacteria community structures at the genus level are summarised in Fig. 6. The relative abundance of methanogenic archaea increased between the mesotrophic and hyper-eutrophic lakes (Fig. 6a). The total relative abundance of methanogens in mesotrophic, eutrophic, mid-eutrophic, and hypereutrophic

states were 0.03%, 0.35%, 0.94%, and 1.17%, respectively. These methanogens were classified into five orders: *Methanosarcinales*, *Methanomicrobiales*, *Methanocelleales*, *Methanofastidiosales*, and *Methanobacteriales*. *Methanobacterium* (genus of the *Methanobacteriales* order) were found in lakes of different trophic states, and the mesotrophic lakes possessed *Methanobacterium* as the only genus. Clearly, the relative abundances of the orders *Methanosarcinales* and *Methanomicrobiales* in the hyper-eutrophic and middle-eutrophic lakes were higher than those of the eutrophic and mesotrophic lakes.

The relative abundance of methanotrophs did not show an obvious trend as observed for the methanogens. The relative abundance of methanotrophs in the middle-eutrophic lakes (2.27%) was higher than that at the hyper-eutrophic (1.28%), eutrophic (0.47%), and mesotrophic (0.75%) states. Aerobic methanotrophs dominated the assemblages, whereas anaerobic methanotrophs (e.g. the phylum *NC10*) had a very low total abundance. The relative abundances of different methanotrophs in the mesotrophic lakes were higher than those in the eutrophic lakes.

## **2.4 Discussion**

### **2.4.1 Nonlinear response of CH<sub>4</sub> emissions from lakes of different trophic states**

In recent years, studies have reported that freshwater lakes are a potential source of CH<sub>4</sub> emissions (Yang et al., 2011). The loadings of C and N in lakes gradually accumulate, and eventually, a hypereutrophic lake functions as a fermentation “tank”, thus resulting in increased CH<sub>4</sub> emissions. Our results showed that the mean CH<sub>4</sub> emission flux from each of the sampled lakes (except those classified as mesotrophic) was higher than the global mean CH<sub>4</sub> emission flux from freshwater ecosystems during summer (Table S3) (Bastviken et al., 2011). In particular, the mean CH<sub>4</sub> emission flux for the hyper-eutrophic lakes was approximately 300 times higher than the global mean CH<sub>4</sub> emission flux for freshwater ecosystems. Although numerous studies have demonstrated that lake eutrophication induces CH<sub>4</sub> emissions (Sepulveda-Jauregui et al., 2018; Beaulieu et al.,

2019), these studies did not summarise the CH<sub>4</sub> emission pattern across lakes with different trophic states. The pattern of the CH<sub>4</sub> fluxes from the lakes in the present study was consistent with the hypothesis that the CH<sub>4</sub> emission flux increases as the trophic state shifts from mesotrophic to hyper-eutrophic. Surprisingly, compared with lakes of a lower level of eutrophication in this study, the CH<sub>4</sub> emission flux from hyper-eutrophic lakes was extremely high, thus indicating a nonlinear increase of CH<sub>4</sub> release with increasing eutrophication (Fig. S1a). Moreover, the CO<sub>2</sub> flux exhibited a very similar pattern to that of the CH<sub>4</sub> flux (Fig. S1b). As the *TLI* is robust means for quantifying the trophic state of lakes, it is possible to propose a nonlinear model to estimate the CH<sub>4</sub> and CO<sub>2</sub> emissions using the *TLI* (Fig. 4). Beaulieu et al. (2019) proposed a simplistic upscaling method to describe a global CH<sub>4</sub> emission rate without considering the driving mechanisms, which may cause large uncertainties with respect to estimated global CH<sub>4</sub> emissions. Given that lakes can have similar characteristics (i.e. latitudinal, geographic, and morphological conditions), our results indicate that multiple driving factors co-regulate CH<sub>4</sub> emissions, and that the pattern of CH<sub>4</sub> emissions follows a nonlinear increase with the increasing trophic level of lakes.

#### **2.4.2 Environmental factors driving CH<sub>4</sub> emissions in shallow lakes**

Variations in the organic matter content and redox conditions in lakes are key aspects for regulating the biological release of CH<sub>4</sub>. First, it is necessary to emphasise that nutrient and organic matter overloading facilitates high CH<sub>4</sub> emissions. The lakes that were investigated in this study were classified into four trophic states, and the results showed that the mean CH<sub>4</sub> emission flux from mesotrophic lakes was significantly lower than that from the lakes of other trophic states (Table S3). This suggests that nutrients are the most basic substrate source for CH<sub>4</sub> production processes. Lake eutrophication is induced by the accumulation of excess nutrients in waterbodies, thereby leading to a large amount of nutrient and carbon fixation by excessive algal growth with subsequent contributions to the sedimentary organic pool after the algae collapse (Table 1 and S1) (Yan et al., 2017). This can also be confirmed by the decreased C:N ratio in the overlying water with the



increase of the *TLI* (Fig. 2) because autochthonous primary production in lakes has a very low C:N ratio (Meyers, 1994). Therefore, our results reveal the non-negligible contribution of autochthonous primary production (i.e. accumulated algae) to hyper-eutrophication in lakes and CH<sub>4</sub> emissions (Figs. 5b and 5c). This trend is consistent with previous studies that found that high aquatic primary productivity caused by the overloading of nutrients drives the increase in CH<sub>4</sub> emission fluxes (Huttunen et al., 2002; Bastviken et al., 2004; Furlanetto et al., 2012).

In several lakes, a positive correlation between  $\ln(\text{CH}_4)$  and chl-*a* (Fig. 5b) was observed. It is known that chl-*a* can act as a proxy for algal productivity (Liikanen and Martikainen, 2003), which is consistent with Beaulieu et al. (2019), who showed that CH<sub>4</sub> emissions were positively correlated with productivity variables in lakes. For instance, the continual accumulation of algae blooms have been found to be a crucial factor that triggers higher emission CH<sub>4</sub> fluxes in eutrophic lakes (Yan et al., 2017). Correspondingly, it has been widely observed that large algae blooms always occur in the hypereutrophic lakes during the summer (Xu, et al., 2004; Chen, et al., 2014; Yan, et al., 2017). In hypereutrophic lakes, algae are a crucial driver for the accumulation of organic matter. For example, Xu et al. (2019b) found that at least 50% of the accumulated organic matter in Lake Taihu was from algae. Therefore, the overloading of nutrients and the subsequent algae growth are crucial drivers for CH<sub>4</sub> emissions. Ultimately, these relationships reflect the link between an increase in the organic substrate and enhanced rates of methanogenesis in shallow lakes.

The accumulation of organic substrates in lakes is bound to change the microenvironment of the lake. Correspondingly, the alteration of factors such as DO, pH, and temperature in microenvironments can affect the CH<sub>4</sub> emissions from freshwater lakes. It has been well documented that low DO and high temperature conditions facilitate the emission of CH<sub>4</sub> (Xing et al., 2005; Duc et al., 2010). In the present study, the highest DO concentrations in surface waters were observed in the mesotrophic lakes, which had the lowest mean CH<sub>4</sub> emission flux. A close relationship between DO and pH was

observed in this study ( $p < 0.05$ ). Obviously, hypoxia and sufficient organic matter are prerequisites for fermentation. Subsequently, the processes of fermentation and acid production affect the pH levels in the water column, and accelerate oxygen consumption (Yan et al., 2017). These biotic processes involve methanogenesis. Therefore, the accumulated labile organic matter in eutrophic lakes is rapidly consumed with oxygen, which still leaves a large amount of substrate for methanogenesis.

In this study, the relative abundance of methanogens in the mesotrophic lakes was lower than that in the lakes with a higher level of eutrophication, which was in accordance with the CH<sub>4</sub> emissions (Fig. 6). The fermentation processes involved in methanogenesis are generally affected by the quality and quantity of organic matter that contribute to fermentation (Zhou et al., 2018). A recent study found that the quantity rather than the quality of organic matter was a limiting factor for CH<sub>4</sub> emissions, which suggests that autochthonous organic matter is a significant contributor to CH<sub>4</sub> emissions (Grasset et al., 2018). This is consistent with our study, whereby the lower CH<sub>4</sub> emissions were found in the mesotrophic lakes that contained a lower organic matter content than the lakes of a higher trophic state (Table 1).

The emission process of CH<sub>4</sub> is controlled by a combination of production by methanogens and oxidation by methanotrophs (Borrel et al., 2011). Aerobic and anaerobic methanotrophs had relatively high abundances in the middle-eutrophic lakes, especially in Lake Honghu (HH) (Fig. 6b). This is consistent with Yang et al. (2019) who found that high methanotroph abundances were induced by eutrophication. As a consumer of CH<sub>4</sub>, methanotrophs significantly reduced the CH<sub>4</sub> emissions in the middle-eutrophic lakes in comparison to the hyper-eutrophic lakes, which had substantial CH<sub>4</sub> emissions and comparatively higher methanogens and lower methanotrophs abundances (Fig. 6). A high relative abundance of methanotrophs is likely to occur under high DO concentrations. Previous studies have considered that oxygen availability is a major factor affecting the regulation of methanotrophic activity (Borrel et al., 2011; Yang et al., 2019). In the present investigation, the DO concentration of the middle-eutrophic lakes was significantly

higher than that of the hyper-eutrophic lakes (Tale 1 and S1). Therefore, the high relative abundance of methanotrophs in the middle-eutrophic lake contributed to the process of CH<sub>4</sub> consumption, which is also consistent with previous studies (Yang et al., 2019). Moreover, the present study also suggests that the overloading of NH<sub>4</sub><sup>+</sup>-N, which can easily occur under low oxygen conditions, contributed to the limitation of methanotrophic activity (Bosse et al., 1993). Therefore, the production and emission of CH<sub>4</sub> were determined by the abundances of methanogens and methanotrophs, which were affected by the surrounding redox conditions.

To explore the potential drivers of CH<sub>4</sub> and CO<sub>2</sub> emissions, a stepwise linear regression involving all environmental variables indicated that CH<sub>4</sub> and CO<sub>2</sub> emissions could be well-predicted by the NH<sub>4</sub><sup>+</sup>-N concentration trend for lakes of different trophic states, as evidenced by their strong correlations (Table 2). This finding agrees with previous studies that showed that NH<sub>4</sub><sup>+</sup>-N, CH<sub>4</sub>, and CO<sub>2</sub> are simultaneously produced during mineralisation processes (Furlanetto et al., 2014; Yan et al., 2017). It is worth mentioning that the process of organic matter mineralisation involves the formation of NH<sub>4</sub><sup>+</sup>-N, which is easily oxidised by methane monooxygenase and hence retards the CH<sub>4</sub> oxidation rate (Holmes et al., 1995; He et al., 2019b; Yan et al., 2019). During CH<sub>4</sub> oxidation, the accumulated and decomposed algae probably provide an important source of autochthonous organic matter in hyper-eutrophic lakes. Bastviken et al. (2015) suggested that anaerobic carbon mineralisation accounts for 20–60% of the overall mineralisation in aquatic ecosystems. Autochthonous organic matter, especially algae, are closely connected with CH<sub>4</sub> production (Chen et al., 2016). In addition, the decomposition, fermentation, and mineralisation processes also involve strong CO<sub>2</sub> emissions, which were significantly and positively correlated with the CH<sub>4</sub> emissions in this study ( $p < 0.01$ ) (Table S2). Therefore, the availability of organic matter produced from nutrients is a primary factor for the CH<sub>4</sub> production that is linked to biotic processes.

#### **2.4.3 Relationship between eutrophication and CH<sub>4</sub> emissions: Implications for**

## climate change

To date, many studies have discussed the impact and feedback of climate change on lake eutrophication processes. In particular, hypereutrophic lakes are known to play a crucial role as a sink for GHGs (Sepulveda-Jauregui et al., 2018; Beaulieu et al., 2019). Our findings evidence the change that occurs in CH<sub>4</sub> emissions with lake eutrophication, whereby a nonlinear model was proposed to describe the CH<sub>4</sub> emissions (Fig. 4) across a gradient from mesotrophic to hypertrophic lakes (and increasing *TLI*). The dramatic increase in CH<sub>4</sub> emissions from the hyper-eutrophic lakes is probably related to the simultaneous shift from a clear-water state to a turbid-water algae-dominated state. The results further emphasise that phytoplankton biomass, as induced by excess nutrients (Lau and Lane, 2002), is a potential facilitator driving increased CH<sub>4</sub> emissions in eutrophic lakes. The combination of our results with those of previous studies (Yan et al., 2017; Zhou et al., 2018; Yan et al., 2019) indicates that the algae decomposition process, which easily occurs in hyper-eutrophic lakes, provides abundant organic matter and suitable microenvironments for methanogenesis. The CH<sub>4</sub> emission flux from each lake of a different typical trophic state was estimated using our conceptual model based on the *TLI* index and CH<sub>4</sub> emission flux. Given that the mean *TLI* for the mesotrophic, eutrophic, middle-eutrophic, and hyper-eutrophic lakes were 45, 55, 65, and 75, respectively, the CH<sub>4</sub> emissions would increase by a factor of ~10 for each increase in the trophic level. Therefore, it is extremely important that an accurate model (incorporating key driving factors) can estimate the CH<sub>4</sub> emission flux from lakes with different trophic states. Further research regarding GHG emissions should consider a parameterised conceptual model for shallow lakes that includes potential drivers such as nutrients, sediment organic matter, and DO. Additionally, it should be noted that the model in the present study is only applicable during the summer and that CH<sub>4</sub> emissions most likely vary seasonally; hence, this aspect requires further study.

## 2.5 Conclusions

In this study, we investigated the CH<sub>4</sub> emission flux and environmental variables across different lakes along a trophic state gradient of the Yangtze River basin, China. This is the first study to elucidate the nonlinear response of the release of CH<sub>4</sub> from shallow lakes along a gradient of increasing eutrophication. An increase in the *TLI* of a lake by one level (*e.g.*, from mesotrophic to eutrophic) would correspond to a 10-fold increase in CH<sub>4</sub> emissions. This phenomenon is related to the fact that the accumulated nutrients and subsequent production of algae organic matter are the crucial drivers for the microbial community responsible for the CH<sub>4</sub> emissions from lakes. In addition, the decomposition of algae makes a significant contribution to the CH<sub>4</sub> emissions from eutrophic lakes, especially hyper-eutrophic lakes. Correspondingly, the NH<sub>4</sub><sup>+</sup> concentration in the water column can be used to predict the CH<sub>4</sub> emissions from lakes. Thus, our results strongly suggest that the eutrophication of lakes not only causes a deterioration in water quality, but also has a positive feedback effect on the global greenhouse effect.

## 2.6 References

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## Tables and figures

Table 1 Abiotic variables of the water column and sediment in each type of eutrophic lake. Values are presented as the mean  $\pm$  standard deviation.

Eutrophic state $TLI (\Sigma)$	DO (mg L <sup>-1</sup> )	pH	TN (mg L <sup>-1</sup> )	NO <sub>3</sub> <sup>-</sup> -N (mg L <sup>-1</sup> )	NH <sub>4</sub> <sup>+</sup> -N (mg L <sup>-1</sup> )	TP (mg L <sup>-1</sup> )	DOC (mg L <sup>-1</sup> )	chl- <i>a</i> ( $\mu$ g L <sup>-1</sup> )	STN (g kg <sup>-1</sup> )	STOC (g kg <sup>-1</sup> )	OM (g kg <sup>-1</sup> )
Mesotrophic ( $\leq 50$ )	8.2 $\pm$ 0.8	7.9 $\pm$ 0.6	0.6 $\pm$ 0.1	0.1 $\pm$ 0.1	0.3 $\pm$ 0.1	0.1 $\pm$ 0.1	5.6 $\pm$ 0.4	7.1 $\pm$ 1.8	0.9 $\pm$ 0.5	28.6 $\pm$ 13.5	49.2 $\pm$ 23.1
Eutrophic (50 – 60)	8.0 $\pm$ 1.1	8.3 $\pm$ 0.8	1.2 $\pm$ 0.2	0.2 $\pm$ 0.2	0.5 $\pm$ 0.2	0.2 $\pm$ 0.2	6.5 $\pm$ 2.2	15.6 $\pm$ 4.9	1.5 $\pm$ 0.4	36.0 $\pm$ 17.5	62.1 $\pm$ 30.1
Middle-eutrophic (60 – 70)	6.8 $\pm$ 2.3	7.9 $\pm$ 0.6	1.4 $\pm$ 0.5	0.5 $\pm$ 0.4	0.6 $\pm$ 0.2	0.2 $\pm$ 0.2	5.0 $\pm$ 1.5	41.9 $\pm$ 10.7	2.0 $\pm$ 0.5	31.5 $\pm$ 9.5	54.3 $\pm$ 16.3
Hyper-eutrophic ( $> 70$ )	5.6 $\pm$ 2.0	7.8 $\pm$ 0.7	3.4 $\pm$ 0.6	1.7 $\pm$ 0.5	1.1 $\pm$ 0.3	0.3 $\pm$ 0.1	6.4 $\pm$ 1.1	92.7 $\pm$ 19.4	2.4 $\pm$ 0.3	41.3 $\pm$ 18.0	71.1 $\pm$ 31.1

Table 2. Multiple stepwise linear regression model of CH<sub>4</sub> and CO<sub>2</sub> emission fluxes incorporating the variables of the overlying water

Parameters	Equations	Adj. R <sup>2</sup>	<i>p</i>	Significance level		
				F-test	T-test	
				F	t <sub>1</sub>	t <sub>2</sub>
CH <sub>4</sub>	$\ln(\text{CH}_4) = 7.181(\text{NH}_4^+\text{-N}) - 3.454$	0.74	$< 0.01$	43.37	6.59	-
CO <sub>2</sub>	$\text{CO}_2 = 174.66(\text{NH}_4^+\text{-N}) - 60.51(\text{pH}) + 417.23$	0.68	$< 0.01$	15.05	3.86	-2.66

CH<sub>4</sub> (mg m<sup>-2</sup> h<sup>-1</sup>), CO<sub>2</sub> (mg m<sup>-2</sup> h<sup>-1</sup>), and NH<sub>4</sub><sup>+</sup>-N (mg L<sup>-1</sup>).

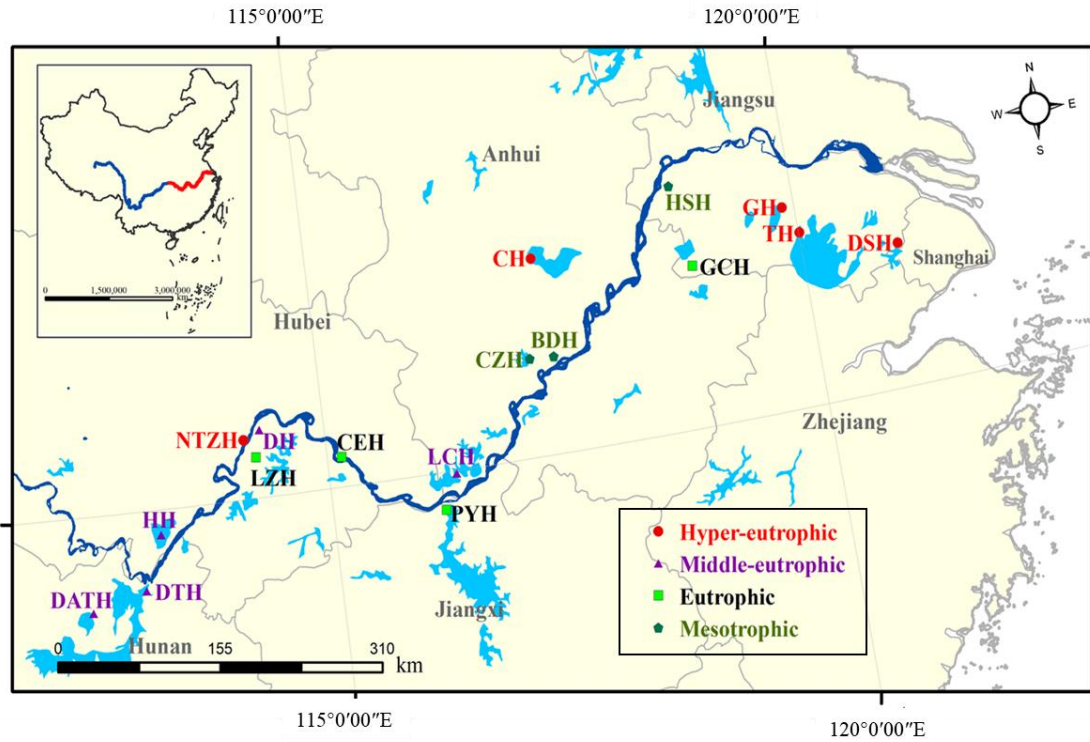


Fig. 1 Distribution of sampling sites in the middle and lower reaches of the Yangtze River basin, China. DSH: Lake Dianshang; TH: Lake Taihu; GH: Lake Gehu; GCH: Lake Gucheng; HSH: Lake Huashen; CH: Lake Chaohu; BDH: Lake Baidang; CZH: Lake Caizi; LGH: Lake Longgan; PYH: Lake Poyang; CEH: Lake Cehu; DH: Lake Donghu; LZH; Lake Liangzi; NTZH; Lake Nantaizi; HH: Lake Honghu; DTH; Lake Dongting; DATH: Lake Datong.

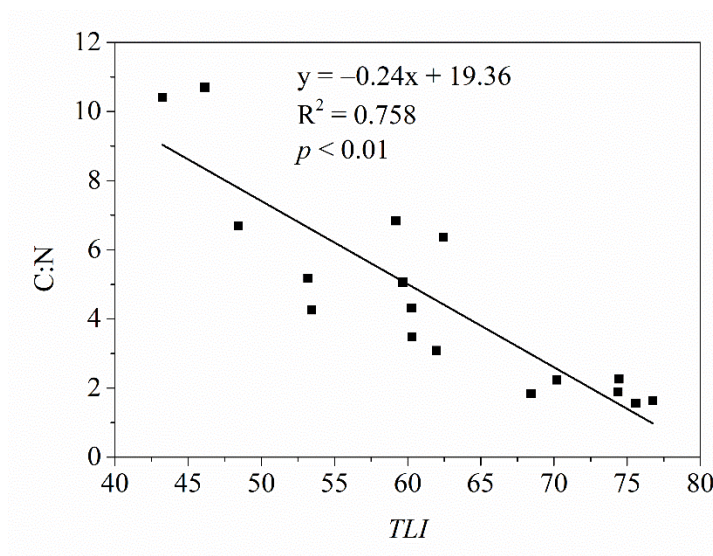


Fig. 2 Relationship between the *TLI* and C:N in the sampled lakes.

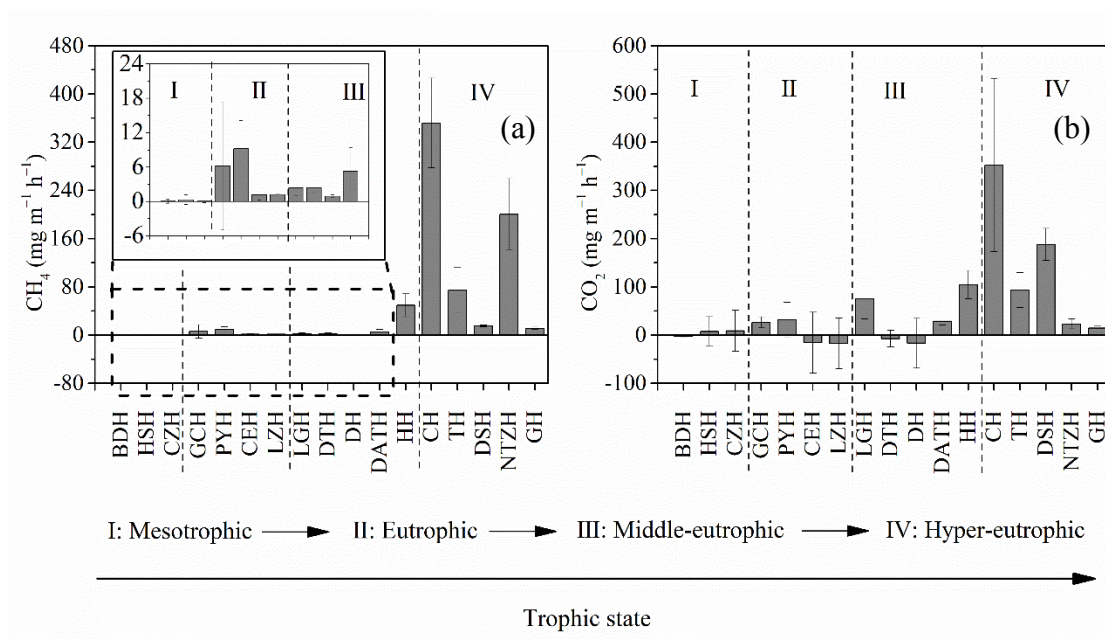


Fig. 3 Net  $\text{CH}_4$  (a) and  $\text{CO}_2$  (b) emissions at the water-air interface for the 17 sampled lakes. Values are as the mean  $\pm$  standard deviation.

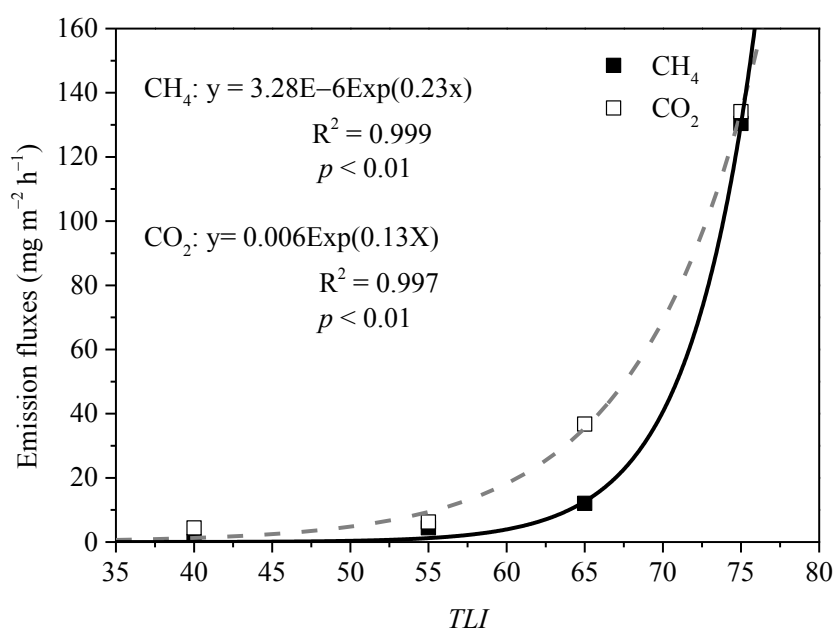


Fig. 4 Mean  $\text{CH}_4$  (solid line) and  $\text{CO}_2$  (dashed line) emissions for four categories of lakes (mesotrophic, eutrophic, middle-eutrophic, and hypereutrophic) based on their  $TLI$  values.

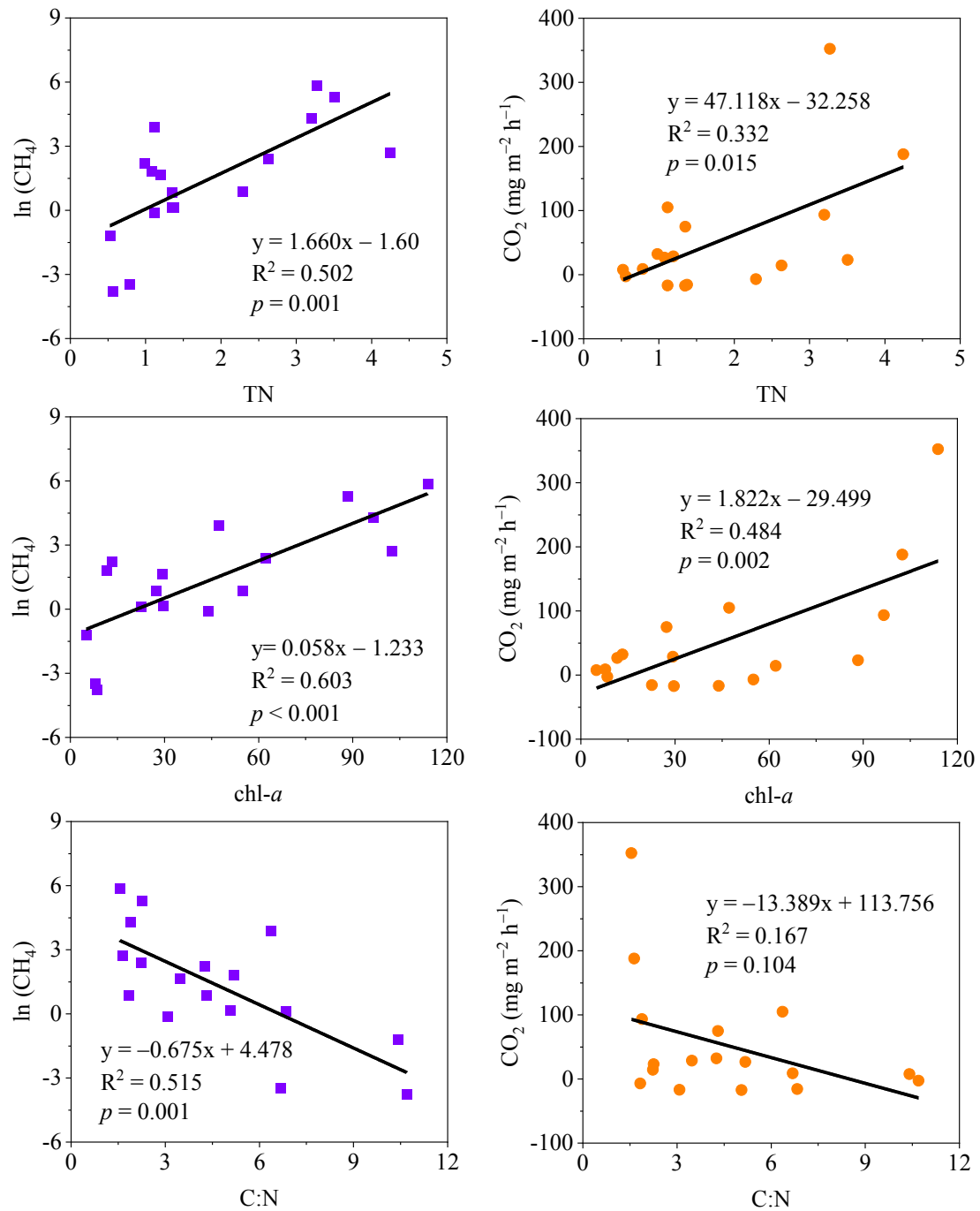


Fig. 5 Relationships between environmental factors and greenhouse gas ( $\text{CH}_4$  and  $\text{CO}_2$ ) emissions in shallow lakes.

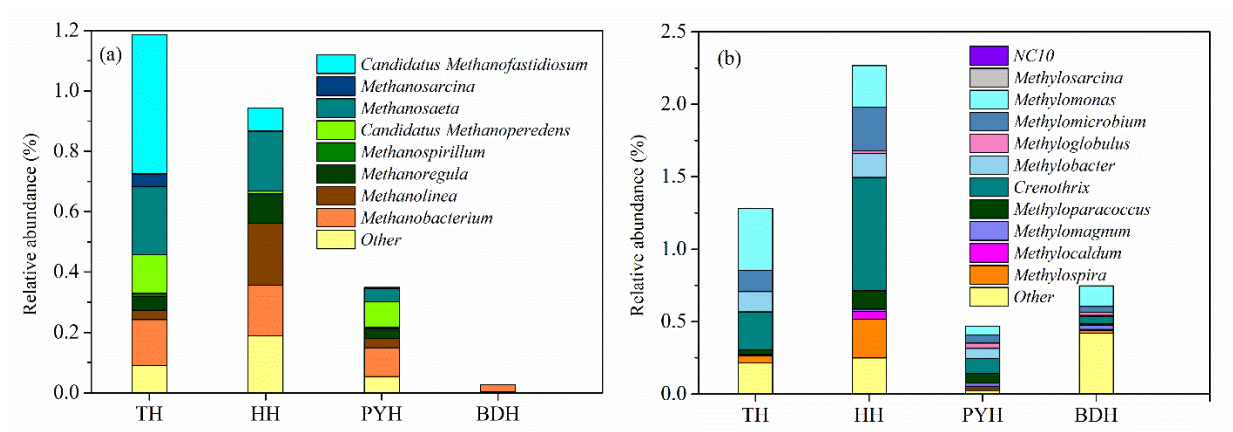


Fig. 6 Relative abundance of methanogen (a) and methanotroph (b) bacteria community compositions at the genus level in different trophic lakes. Lake Baidang (BDH), Lake Poyang (PYH), Lake Honghu (HH), and Lake Taihu (TH) represent mesotrophic, eutrophic, middle-eutrophic, and hyper-eutrophic lakes, respectively.



## Supporting materials

Table S1 Physiochemical parameters of each lake in the middle and lower reaches of the Yangtze River region, China.

Lakes	TN (mg L <sup>-1</sup> )	TP (mg L <sup>-1</sup> )	NO <sub>3</sub> <sup>-</sup> -N (mg L <sup>-1</sup> )	NH <sub>4</sub> <sup>+</sup> -N (mg L <sup>-1</sup> )	DOC (mg L <sup>-1</sup> )	chl- <i>a</i> (μg L <sup>-1</sup> )	Temperature (°C)	pH	DO (mg L <sup>-1</sup> )	<i>TLI</i>	Mean depth (m)	Eutrophic state
BDH	0.5	0.17	0.06	0.35	5.5	8.5	30.7	8.5	9.08	46.14	3.1	Mesotrophic lake
HSH	0.6	0.12	0.04	0.34	6.0	5.0	30.4	7.8	7.46	43.26	< 7	
CZH	0.8	0.04	0.26	0.32	5.3	7.8	29.4	7.3	8.00	48.43	3	
GCH	1.1	0.14	0.14	0.86	5.6	11.6	29.7	7.9	7.07	53.18	1.6	
PYH	1.0	0.13	0.52	0.45	4.2	13.3	28.6	7.6	6.95	53.43	5.1	Eutrophic lake
CEH	1.4	0.43	0.11	0.45	9.4	22.7	30.4	9.1	8.92	59.19	2	
LZH	1.2	0.04	0.10	0.35	6.8	29.7	29.6	8.9	8.91	59.66	3	
LGH	1.4	0.08	0.48	0.70	5.8	27.4	27.2	7.6	6.30	60.27	3.7	
DTH	2.3	0.13	1.18	0.45	4.2	55.0	28.2	8.0	9.20	68.43	5.5	Middle-eutrophic lake
DH	1.1	0.16	0.47	0.44	3.4	44.0	28.7	8.7	9.25	61.96	2.5	
DATH	1.3	0.46	0.39	0.69	4.2	29.4	27.4	8.1	4.72	60.30	2.6	
HH	1.1	0.16	0.18	0.77	7.1	47.3	27.1	7.2	4.59	62.44	1.3	
CH	3.3	0.31	1.52	1.34	5.1	113.9	31.3	7.4	5.41	75.59	2.9	Hyper-eutrophic lake
TH	3.2	0.32	1.77	0.93	6.0	96.6	29.6	7.3	3.93	74.38	1.9	
DSH	4.3	0.36	2.40	1.00	7.0	102.5	30.7	7.3	3.73	76.74	2.1	
NTZH	3.6	0.39	1.61	1.30	7.9	88.4	29.3	8.8	6.56	74.45	2	
GH	2.7	0.33	0.92	0.71	5.9	62.2	30.6	8.5	8.44	70.19	1.3	

Table S2 Correlations between variables and GHGs emission fluxes.

	CH <sub>4</sub>	CO <sub>2</sub>	TN	NO <sub>3</sub> <sup>-</sup>	NH <sub>4</sub> <sup>+</sup>	TP	chl- <i>a</i>	DOC	<i>TLI</i>	DO	pH
CH <sub>4</sub>	1										
CO <sub>2</sub>	0.749**	1									
TN	0.556*	0.582*	1								
NO <sub>3</sub> <sup>-</sup>	0.497*	0.591*	0.965**	1							
NH <sub>4</sub> <sup>+</sup>	0.789**	0.722**	0.786**	0.716**	1						
TP	0.344	0.269	0.587*	0.473	0.553*	1					
chl- <i>a</i>	0.689**	0.696**	0.941**	0.810**	0.533*	0.405	1				
DOC	0.096	0.000	0.236	0.115	0.219	0.377	0.143	1			
<i>TLI</i>	0.546*	0.545*	0.921**	0.843**	0.755**	0.566*	0.944**	0.169	1		
DO	-0.349	-0.657**	-0.486*	-0.549*	-0.663**	-0.425	-0.585*	-0.081	-0.408	1	
pH	-0.165	-0.587*	-0.158	-0.344	-0.280	0.158	-0.256	0.275	-0.067	0.691**	1

Significant correlation at the \* $p < 0.05$  and \*\* $p < 0.01$  levels.

Table S3 Mean CH<sub>4</sub> and CO<sub>2</sub> emission fluxes in each type of eutrophic lake

Types	Mesotrophic			Eutrophic			Middle-eutrophic			Hyper-eutrophic		
	Range	Mean	SD	Range	Mean	SD	Range	Mean	SD	Range	Mean	SD
CH <sub>4</sub> (mg m <sup>-2</sup> h <sup>-1</sup> )	0.1-0.3	0.1	0.2	1.1-9.2	4.4	4.0	0.9-49.3	12.0	20.9	11.2-351.9	130.4	145.5
CO <sub>2</sub> (mg m <sup>-2</sup> h <sup>-1</sup> )	-2.9-8.7	4.4	6.4	-17.5-31.9	6.2	26.6	-16.8-104.6	36.7	52.2	14.4-352.2	134.1	140.4

)Exp(0.14x)

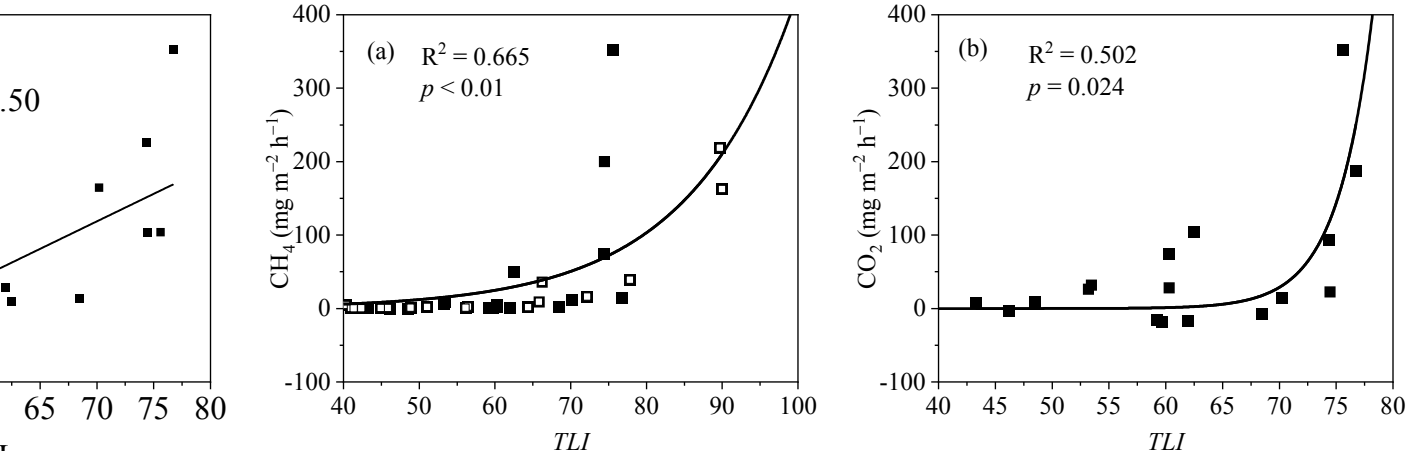


Fig. S1 Relationship between the *TLI* and CH<sub>4</sub> (a) and CO<sub>2</sub> (b) emission fluxes in shallow lakes. The solid points represent the data from this study, the hollow points represent the data from other studies<sup>1</sup>.

<sup>1</sup>Reference:

- Gonzalez-Valencia, R., Sepulveda-Jauregui, A., Martinez-Cruz, K., Hoyos-Santillan, J., Dendooven, L., Thalasso, F., 2014. Methane emissions from Mexican freshwater bodies: correlations with water pollution. *Hydrobiologia* 721(1), 9-22.
- Martinez-Cruz, K., Gonzalez-Valencia, R., Sepulveda-Jauregui, A., Plascencia-Hernandez, F., Belmonte-Izquierdo, Y., Thalasso, F., 2017. Methane emission from aquatic ecosystems of Mexico City. *Aquat. Sci.* 79(1), 159-169.
- Sepulveda-Jauregui, A., Walter Anthony, K. M., Martinez-Cruz, K., Greene, S., Thalasso, F., 2015. Methane and carbon dioxide emissions from 40 lakes along a north-south latitudinal transect in Alaska. *Biogeosciences* 12(11), 3197-3223.
- Sepulveda-Jauregui, A., Hoyos-Santillan, J., Martinez-Cruz, K., Walter Anthony, K.M., Casper, P., Belmonte-Izquierdo, Y., Thalasso, F., 2018. Eutrophication exacerbates the impact of climate warming on lake methane emission. *Sci. Total Environ.* 636, 411-419.

### ***Chapter 3: Suspended particles potentially enhance nitrous oxide (N<sub>2</sub>O) emissions in the oxic estuarine waters of eutrophic lakes***

#### **3.1 Introduction**

Rivers and lakes, which play the role of linking Earth's various layers, contain a large amount of nitrogen (N) from terrestrial ecosystems, and are potential to be a major storage reservoir and a short- to long-term sink within the N cascade (Galloway et al., 2004; Schmadel et al., 2018). Estuaries bridge the riverine and lacustrine ecosystems while carrying large amounts of land-based organic or inorganic materials into lakes, particularly particulate organic matter (Howarth et al., 2011; Zhou et al., 2018). The sources of suspended particles (SPS) in inland river-lake system are complex. A large amount of terrestrial particulate matter enters water bodies and is affected by various factors, such as human activity, soil erosion, and rainfall processes (Walling, 2005; Wetz et al., 2017; Woodward and Walling, 2007; Yao et al., 2016). Meanwhile, sediment resuspension occurs due to hydraulic disturbances and biological activities, causing particles to re-enter water columns (Chen et al., 2018; Huang and Liu, 2009; Maa et al., 1998). These SPS have recently been found to affect the physicochemical properties and biological processes of N in estuaries (Turner and Millward, 2002; Yao et al., 2016). A number of studies have focused on the N cycles of oceans and adjoining estuary ecosystems (Dalsgaard et al., 2003; Howarth et al., 2011; Lebreton et al., 2016). However, the potential effects of transporting matter in river estuaries of lake ecosystems with high SPS concentrations on the N cycle are still poorly understood.

SPS can adhere a large amount of aerobic heterotrophic microbes and inorganic ions (NH<sub>4</sub><sup>+</sup>, Fe<sup>2+</sup>, Mn<sup>2+</sup>, and HS<sup>-</sup>) to its surface, serving as the main attachment interface for micro-organic materials (Jeris and Owens, 1975; Ochs et al., 2010; Xia et al., 2004; Xia et al., 2009). The organic matter and reducing inorganic ions in the SPS are oxidised by micro-organisms, resulting in significantly lower DO concentrations in the SPS than

those of the surrounding water (Bianchi et al., 1992; Bonin et al., 1981; Falkowski et al., 2008; Jørgensen, 1977). To date, surface sediment is considered to be a hotspot for the formation of gaseous-N (including  $N_2$  and  $N_2O$ ) by denitrification in common shallow-lake ecosystems. However, recent studies have highlighted that a large fraction of the gaseous-N released from aquatic ecosystems can be produced in oxygenated water columns containing a high quantity of SPS (Garnier et al., 2006; Venkiteswaran et al., 2014; Yao et al., 2016). In the estuaries of the Yangtze River or Yellow River, the large amounts of SPS serve as a significant interface and an alternative zone for denitrification, and accelerate the N cycle of the aquatic ecosystem (Xia et al., 2017; Yao et al., 2016). The particular microenvironment at the SPS-interface provides a new  $N_2O$  production source (Liu et al., 2013; Xia et al., 2017). The release flux of  $N_2O$  in aquatic ecosystems may, therefore, be underestimated without consideration of the presence of SPS. Although the N cycle at the SPS-interface has received great attention (Lebreton et al., 2016; Ma et al., 2016), there is little understanding of the influence of SPS on  $N_2O$  emissions and their potential mechanisms in the estuaries of eutrophic lakes.

Unlike the aforementioned high turbid inorganic materials in the studied river estuaries, the SPS in the estuaries of eutrophic lakes mainly consist of organic matter derived from terrestrial leaf litter and autochthonous phytoplankton (Teeter et al., 2001). The different physic-chemical properties of SPS, including particle size, source and composition, may alter the mechanisms of  $N_2O$  production and release with microbial processes (Xia et al., 2017). We hypothesize that these organic-enriched SPS in the estuaries of eutrophic lakes potentially enhance the  $N_2O$  emissions. Laboratory simulations were simultaneously conducted to study the effect of SPS concentrations on  $N_2O$  emission characteristics. The objectives of this study are to provide more direct evidence via a simulation experiment about the relationship between  $N_2O$  emissions and SPS concentrations interpreted by the potential microbial mechanisms.

## **3.2 Material and methods**

### **3.2.1 Microcosm system**

To investigate whether SPS affected N<sub>2</sub>O emissions, a set of double-Perspex closed chambers (inner diameter and height of 7 and 40 cm, respectively) containing water and SPS were designed. In order to ensure high microbial activities, these chambers were heated in a water bath and maintained at 25 °C, an optimized temperature for comparing N<sub>2</sub>O emissions from different chambers in the same level. A stirring device with a polytetrafluoroethylene paddle was used to adjust the speed and ensure that the particles remained suspended in the water column. The chamber could maintain its atmospheric pressure at 0.5 atm for 24 h by applying pressure in a closed experiment. After removing particulates by a 200-mesh nylon sieve, 1 L of lake water from the lake region of Shatang River was added to each chamber. Then 0, 0.5, 1.5, 3, and 5 g of the homogenized sediments from the same place with lake water were added to different chambers. The NO<sub>3</sub><sup>-</sup>-N concentration of the water was adjusted to 4 mg L<sup>-1</sup>, in consideration of sufficient substances for denitrification, by KNO<sub>3</sub> without NH<sub>4</sub><sup>+</sup>-N. The total organic carbon (TOC) and TN contents of the sediments were 44.31 and 2.19 mg g<sup>-1</sup>, respectively, and the NH<sub>4</sub><sup>+</sup>-N and NO<sub>3</sub><sup>-</sup>-N contents were 94.58 and 15.70 µg g<sup>-1</sup>, respectively. The headspace of 538 mL, reserved in each chamber, was open to the atmosphere for 30 min to purge the N<sub>2</sub>O from the previous incubation period and maintain the oxygen saturation level during the experiment. At predetermined intervals, gas samples were collected using a gastight syringe to detect the N<sub>2</sub>O and CO<sub>2</sub> concentrations during the incubation periods. The average N<sub>2</sub>O or CO<sub>2</sub> emission rate = (C<sub>t</sub>-C<sub>a</sub>)/t (C<sub>t</sub>: the N<sub>2</sub>O or CO<sub>2</sub> concentration at t day; C<sub>a</sub>: the N<sub>2</sub>O or CO<sub>2</sub> concentration in atmosphere; t: sampling days). Ten millilitres of water was collected using a syphon for NO<sub>3</sub><sup>-</sup>-N analysis. Water and gas samples were collected by sampling, and stored at 4 °C for less than 24 h before analysis.

### **3.2.2 Chemical analysis**

### **3.2.2.1 Physicochemical properties of water and sediment**

The water samples for chemical analysis, including TN,  $\text{NO}_3^-$ -N,  $\text{NH}_4^+$ -N, and dissolved organic carbon (DOC), were discontinuously collected using a tube sampler and mixed thoroughly, with three replicates. The detailed methodology for testing the parameters can be found in Chapter 2. The SPS concentrations of overlying water were measured by using a Whatman GF/C 0.45- $\mu\text{m}$  filter to weigh the dried weight that kept at 100 °C for 4 h. In the simulation system, the SPS concentrations of the chamber were determined by the moisture content via calculating weight difference between wet and dried samples.

### **3.2.2.2 $\text{N}_2\text{O}$ and $\text{CO}_2$ concentration analysis**

A syringe was used to extract gases for  $\text{N}_2\text{O}$  analysis using a 7890B Agilent gas chromatograph equipped with ECD (Electron Capture Detector) and a nickel catalyst methaniser. Greenhouse gas production was expressed as the cumulative production. The  $\text{CO}_2$  produced in microcosm system was analysed using a 7890B Agilent gas chromatograph equipped with FID (Flame Ionization Detector). The gas chromatograph configurations detailedly described by Shaaban et al. (Shaaban et al., 2016) were adopted for the gas concentration analysis.

### **3.2.3 Biological analysis**

The microbial population analysis of SPS from each treatment was collected at the end of incubation experiments. DNA was extracted from the collected SPS samples using the MOBIO PowerSoil<sup>®</sup> DNA Isolation Kit (MOBIO, USA) according to the manufacturer's instructions. The DNA was stored at -20°C until analysis by PCR. The qPCR assay was performed using an ABI 7900HT real-time qPCR detection system (Applied Biosystems, Inc., USA). PCR amplification of *nirK* fragment, which encode for nitrite reductase of bacteria, was conducted using the F1aCu (5'-ATCATGGTScTGCCGCG-3') and R3Cu (5'-GCCTCGATCAGRTTGTGGTT-3') primers set. Each 20  $\mu\text{L}$  reaction system consisted of 10  $\mu\text{L}$  of SYBR Premix (Takara, China), 1  $\mu\text{L}$  of DNA template or total DNA, 0.5  $\mu\text{L}$  each of forward and reverse

primers, and 8  $\mu$ L of sterile water. After an initial denaturation step at 95 °C for 30 s, 30 cycles of 95 °C denaturation (10 s), 55 °C annealing (40 s), 72 °C elongation (50 s) were followed by a final elongation step of 7 min at 72 °C. All plasmids containing the target genes were provided by Tyhygene Biotech Co., Ltd. and subjected to  $10 \times$  serial dilution to serve as DNA templates for the construction of standard curves ( $R^2 > 0.999$ ).

#### **3.2.4 Statistical analysis**

Statistical analysis was conducted using the SPSS 19.0 software package (SPSS Inc., Chicago). Significant differences between treatments were tested by conducting a one-way ANOVA. The criteria of  $p < 0.05$  and  $p < 0.01$  were used to determine statistical significances at the 0.05 and 0.01 levels (two-tailed), respectively. The statistical significance of the results was determined by conducting a Duncan test at the 5% level.

### **3.3 Results**

#### **3.3.1 Physicochemical properties of water in microcosm system**

The average DO concentration in water exceeded 7.0 mg L<sup>-1</sup> in each treatment during the experimental periods, suggesting oxic conditions in the system (Fig. 1a). The DO concentrations decreased as the SPS concentrations increased when the SPS concentration increased from 0.5 to 5 g L<sup>-1</sup>, the average DO concentration decreased from 8.15 to 7.60 mg L<sup>-1</sup>. In contrast, the pH values exhibited a decreasing trend during the initial incubation periods. The pH values significantly decreased with 0 g L<sup>-1</sup> SPS compared to the treatment with 5 g L<sup>-1</sup> SPS ( $p < 0.01$ ). The variations in the NO<sub>3</sub><sup>-</sup>-N concentrations of the microcosm system are shown in Fig. 1(c). The initial NO<sub>3</sub><sup>-</sup>-N concentration of the overlying water in all treatments was 4 mg L<sup>-1</sup>. There was no significant change in the concentration of NO<sub>3</sub><sup>-</sup>-N during the first nine days under different SPS concentrations. In the treatments with 3 and 5 g L<sup>-1</sup> of SPS, the NO<sub>3</sub><sup>-</sup>-N concentration significantly increased, with the highest concentrations reaching 5.22 and 7.12 mg L<sup>-1</sup>, respectively. However, in the treatments with 0.5 and 1.5 g L<sup>-1</sup> SPS, the



$\text{NO}_3^-$ -N concentrations first increased and then decreased, with minimum values of 3.58 and 3.34  $\text{mg L}^{-1}$ , respectively. In the treatment with without SPS, the  $\text{NO}_3^-$ -N concentration in the overlying water remained relatively stable at approximately 4.0  $\text{mg L}^{-1}$  during the experimental periods.

### 3.3.2 $\text{N}_2\text{O}$ emissions and denitrifying bacteria population in microcosm system

For all treatments with different SPS concentrations, the accumulated  $\text{N}_2\text{O}$  emission rates increased with the SPS concentrations (Fig. 2). When the SPS concentration increased from 0.5 to 5  $\text{g L}^{-1}$ , the average  $\text{N}_2\text{O}$  emission rate increased from 4.78 to 10.89  $\mu\text{g m}^{-3} \text{ d}^{-1}$ . In the experimental system without SPS, the accumulated  $\text{N}_2\text{O}$  emission and average  $\text{N}_2\text{O}$  emission rate were 7.01  $\mu\text{g m}^{-3}$  and 1.28  $\mu\text{g m}^{-3} \text{ d}^{-1}$ , respectively. For the experimental system with 3  $\text{g L}^{-1}$  of SPS, the accumulated  $\text{N}_2\text{O}$  emission and average  $\text{N}_2\text{O}$  emission rate were higher than those of the other system, at 407.38  $\mu\text{g m}^{-3}$  and 10.89  $\mu\text{g m}^{-3} \text{ d}^{-1}$ , respectively. As shown in Fig. 2b, the average  $\text{N}_2\text{O}$  emission rate increased linearly with the SPS concentrations ( $R^2 = 0.7018$ ,  $p < 0.05$ ), suggesting that the SPS concentration significantly affected the  $\text{N}_2\text{O}$  emissions. Similarly, with the increase of SPS concentrations, the positively correlated average  $\text{CO}_2$  emission rate ( $R^2 = 0.855$ ,  $p < 0.05$ ) further indicated the strong microorganism activities in the microcosm system (Fig. 2c).

The characteristics of the denitrifying bacteria population in the SPS of all treatments with different SPS concentrations at the end of incubation was summarized in Fig. 3. The highest and lowest denitrifying bacteria populations were occurred in the treatments of 3 and 0.5  $\text{g L}^{-1}$  SPS, respectively (Fig. 3a). Although the treatment of 5  $\text{g L}^{-1}$  SPS was significantly lower than the treatment of 3  $\text{g L}^{-1}$  SPS, the denitrifying bacteria population in the SPS displayed an increased tendency with the SPS concentrations. In addition, the liner response of average  $\text{N}_2\text{O}$  emission rates to the  $\ln$  (denitrifying bacteria) ( $R^2 = 0.973$ ,  $p < 0.01$ ) (Fig. 3b), further indicated that the  $\text{N}_2\text{O}$  emissions was significantly affected by the denitrifying bacteria in the SPS.

### 3.4 Discussion

Our results revealed that high SPS concentrations can accelerate the emissions of  $\text{N}_2\text{O}$  from oxic water, particularly in estuarine regions with high SPS concentrations and N loadings. Coupled nitrification-denitrification occurs in wastewater treatment and natural micro-ecological systems, as aerobic microorganisms consume oxygen to generate a hypoxic microenvironment, and  $\text{NO}_3^-$ -N is produced by nitrification (Walter et al., 2005; Xia et al., 2017). Correspondingly, the simulation experiment exhibited increasing  $\text{NO}_3^-$ -N concentrations in the water after 12 days. Xia et al. (2017) reported that nitrifying bacteria appeared on the SPS-interface, as evidenced by the positive correlation between their abundance and the SPS concentration. The organic N in the SPS is the main form of N, and the main form of inorganic N is  $\text{NH}_4^+$ -N. Therefore, the nitrification of  $\text{NH}_4^+$ -N to  $\text{NO}_3^-$ -N is an important process to supplement the  $\text{NO}_3^-$ -N concentration in an oxic overlying water. In the chambers with SPS concentrations of 3 and 5  $\text{g L}^{-1}$ , the concentrations of  $\text{NO}_3^-$ -N increased from 4  $\text{mg L}^{-1}$  to maximum concentrations of 5 and 7  $\text{mg L}^{-1}$ , respectively (Fig. 1c). In addition, a decrease in the  $\text{NO}_3^-$ -N concentration of the water column was observed with SPS concentrations of 0.5 and 1.5  $\text{g L}^{-1}$ , and the concentrations of  $\text{NO}_3^-$ -N in the water column increased with SPS concentrations of 3 and 5  $\text{g L}^{-1}$ . These results indicated that the nitrification and denitrification processes occurred in the microcosm system. However, the denitrification was hardly observed in the chamber without SPS as there was almost no change in the level of  $\text{NO}_3^-$ -N in the water column. Based on these observations, we speculate that the denitrification occurring on the SPS-interface in the water column is another important source of  $\text{N}_2\text{O}$ . It is confirmed by the findings that the denitrifying bacteria population in the SPS increased with the SPS concentrations (Fig. 3a). Additionally, the denitrifying bacteria population had a significantly positive correlation with  $\text{N}_2\text{O}$  emissions (Fig. 3b). Although  $\text{N}_2\text{O}$  emissions are controlled by multiple pathways, many studies suggest that they are primarily controlled by denitrification (Galloway et al., 2004; Yan et al., 2012; Yan et al., 2017). In addition,

the coupling of nitrification-denitrification on SPS accelerates N transfer and the emissions of  $\text{N}_2\text{O}$  in naturally oxic water. Hence, the SPS facilitates the conversion of N and also accelerates the emissions of  $\text{N}_2\text{O}$  in oxic water columns, which suggests that  $\text{N}_2\text{O}$  emissions were underestimated when the SPS concentrations and N loadings in aquatic ecosystems were high.

Currently, many studies have discussed the impact or feedback of climate change on cyanobacterial blooms (Yan et al., 2017) and eutrophic processes (Cole and Caraco, 2001; Wang et al., 2009) in shallow lakes, particularly those that are hypereutrophic. There are large external particle sources and waves accelerate the suspension of sediments in the estuaries of shallow lakes, resulting in high SPS concentrations and N loadings. Based on our experimental model to estimate the  $\text{N}_2\text{O}$  emissions fluxes in the estuaries of the hypereutrophic Taihu Lake,  $\text{N}_2\text{O}$  emissions are easily to be underestimated in estuarine regions, particularly in those of hypereutrophic lakes. Given the average TN concentration of  $5.9 \text{ mg L}^{-1}$ , SPS concentration of  $0.344 \text{ g L}^{-1}$  and DO concentration of  $5.19 \text{ mg L}^{-1}$  (Zhao et al., 2015), the estimated average  $\text{N}_2\text{O}$  emission flux is  $114.37 \mu\text{g m}^{-2} \text{ h}^{-1}$  in the estuarine region of shallow eutrophic lakes. In the case of Taihu Lake, the  $\text{N}_2\text{O}$  emissions are approximately  $36.20 \text{ t yr}^{-1}$  from its estuaries that are considered as a semicircle with a radius of 1 km and the total 23 main inflow rivers, which constitute an area of  $36.13 \text{ km}^2$  (Zhao et al., 2015). However, the influences of anthropologic activities, climate change, N concentrations, hydrodynamic conditions, and the particle size and organic composition of SPS will significantly affect the N cycle and  $\text{N}_2\text{O}$  emissions, and thus require further study.

### **3.5 Conclusions**

This study was the first to report  $\text{N}_2\text{O}$  emissions from a lacustrine estuary particularly that of a hypereutrophic shallow lake. The relationship between environmental factors and  $\text{N}_2\text{O}$  emissions was analysed and indicated that the concentrations of both N and SPS would accelerate the N cycle and facilitate the emission of  $\text{N}_2\text{O}$  from an oxic water column. Both N and SPS concentrations played an

important role in controlling N<sub>2</sub>O emissions from the oxic water column, which should be considered in the future estimations of N<sub>2</sub>O emissions from eutrophic lakes.

### 3.6 References

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## Figures

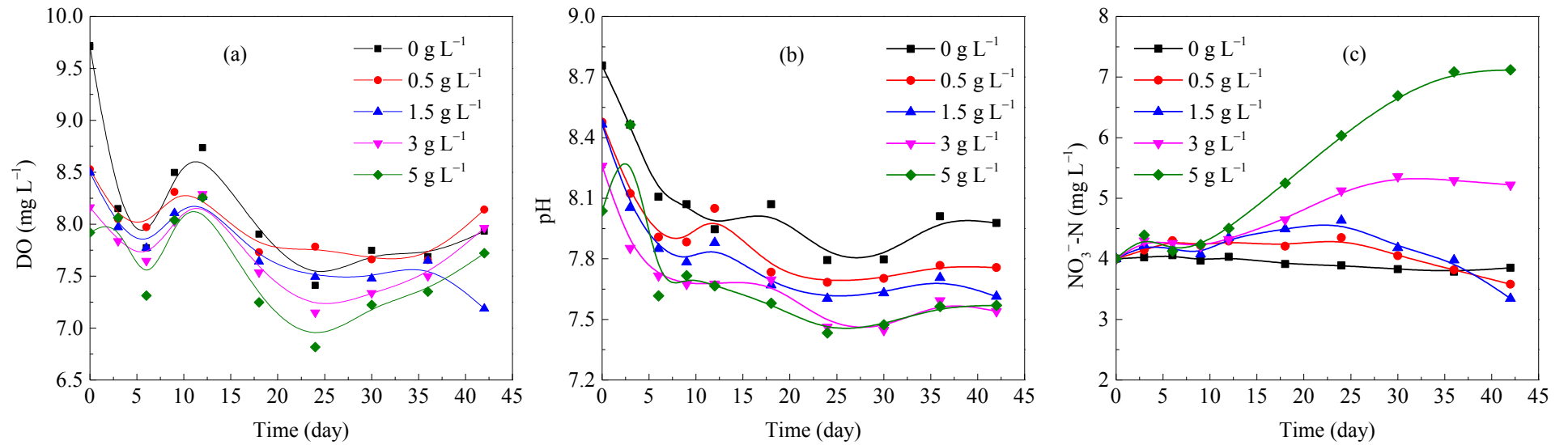


Fig. 1. Variations of physiochemical characteristics, (a) DO, (b) pH, and (c) NO<sub>3</sub><sup>-</sup>-N, of the overlying water during the incubation periods.



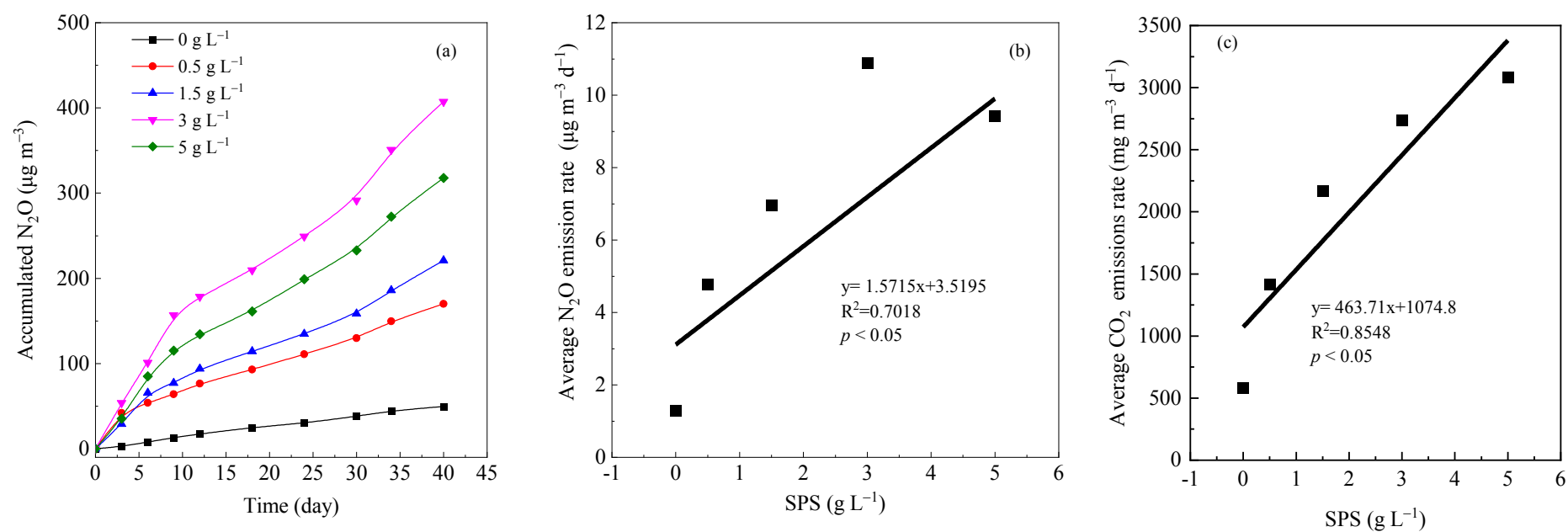


Fig. 2. Effect of the SPS concentration on the (a)  $\text{N}_2\text{O}$  emissions, (b) average  $\text{N}_2\text{O}$  emission rates, and (c) average  $\text{CO}_2$  emission rates during the 42 days of incubation.

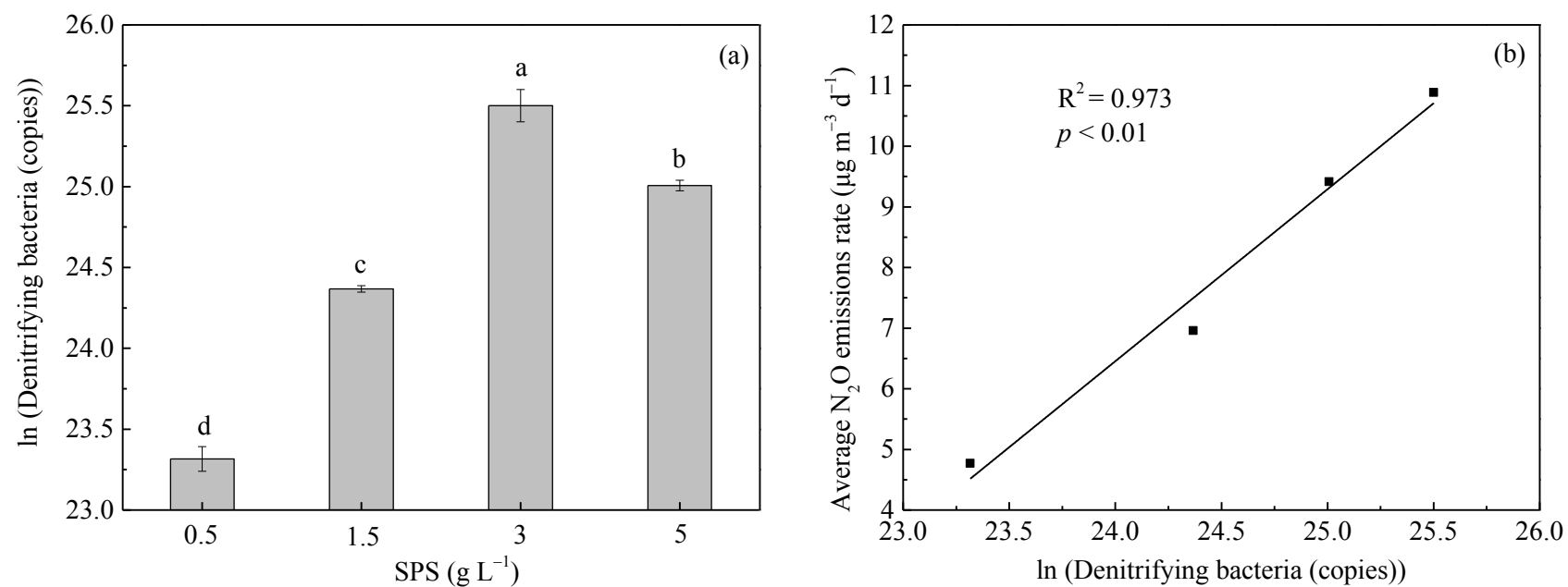


Fig. 3. The denitrifying bacteria population in SPS in incubation systems (a) and the relationship between the denitrifying bacteria population in SPS and N<sub>2</sub>O emissions rate (b).

## ***Chapter: 4 Characterization of N<sub>2</sub>O emission in different trophic shallow lakes***

### **4.1 Introduction**

Nitrous oxide (N<sub>2</sub>O) is an ozone-depleting and highly potent greenhouse gas (GHG) with a long half-life that contributes to global warming, in addition to carbon dioxide (CO<sub>2</sub>) and methane (CH<sub>4</sub>), in the stratosphere (Ravishankara et al., 2009). Atmospheric N<sub>2</sub>O has increased by 20% from 1750 to 2018 and is steadily increasing at a rate of 0.2% per year (Tian et al., 2020). The IPCC reported that approximately 10% of anthropogenic N<sub>2</sub>O sources are derived from freshwater and coastal marine systems (IPCC, 2013). Given that inland freshwater lakes are recipients of nutrients transported from terrestrial ecosystems to trigger uncertainty of GHG emissions (Wang et al., 2009; Zhou et al., 2020a), they function as N<sub>2</sub>O hot spots where the high turnover of nitrogen (N) compounds. Hence, the N<sub>2</sub>O emissions from aquatic ecosystems have received considerable attention (Beaulieu et al., 2011; Kortelainen et al., 2020). Although multiple factors such as nutrient composition, eutrophication, and temperature likely regulate N<sub>2</sub>O emissions from freshwater lakes at regional and global scales, the dominant factors affecting N<sub>2</sub>O emissions in freshwater lakes remain poorly understood (Kortelainen et al., 2020; Li et al., 2018; Yan et al., 2017). There is also a need to evaluate and quantify the N<sub>2</sub>O emissions of freshwater lake ecosystems given that they are globally significant sources of N<sub>2</sub>O (Lauerwald et al., 2019). Generally, understanding the mechanisms underlying variation in N<sub>2</sub>O emissions from freshwater lakes can aid the development of policies to address global warming.

N<sub>2</sub>O is mainly produced by a by-product from nitrification and an intermediate from denitrification (Wenk et al., 2016). Of these multiple sources, denitrification is thought to

be a main source of N<sub>2</sub>O in aquatic ecosystems (Beaulieu et al., 2011; Li et al., 2019a; Salk and Ostrom, 2019). N<sub>2</sub>O reduction, the final step of denitrification ( $\text{NO}_3^- \rightarrow \text{NO}_2^- \rightarrow \text{NO} \rightarrow \text{N}_2\text{O} \rightarrow \text{N}_2$ ), is catalyzed by N<sub>2</sub>O reductase, which is encoded by the *nosZ* gene (Yoon et al., 2016). Denitrification plays a critical role in determining N<sub>2</sub>O emission fluxes, including whether aquatic ecosystems are N<sub>2</sub>O sources or sinks. N<sub>2</sub>O production is regulated by microbial community structure in aquatic ecosystems (Zhao et al., 2018; Zhao et al., 2019). Microbial community composition affects the amount of N<sub>2</sub>O emissions, as some bacteria lack *nosZ* and nitrite reductase genes, which significantly contribute to N<sub>2</sub>O consumption and production in natural ecosystems, respectively (Domeignoz-Horta et al., 2016). Among several environmental factors, nutrients, redox conditions, and temperature determine the microbial guilds involved in N<sub>2</sub>O production and consumption (Hinshaw and Dahlgren, 2013; Xiao et al., 2019). Seasonal changes involve multiple environmental fluctuations that affect lacustrine microbial community structure (Song et al., 2012) and lead to uncertainty in the magnitude of N<sub>2</sub>O emissions. N conversion rates and eutrophication progress are crucial for regulating final N forms (N<sub>2</sub> or N<sub>2</sub>O) in aquatic ecosystems (Jiang et al., 2020; Salk and Ostrom, 2019; Zhu et al., 2020). Such N conversions are dynamic and dependent on eutrophication progress (Li et al., 2018; Liikanen et al., 2003). There is thus a need to understand the microbial processes that regulate N<sub>2</sub>O sources or sinks in lakes of different trophic state and determine spatial heterogeneity in N<sub>2</sub>O emissions.

Shallow lakes receive massive amounts of nutrients from anthropogenic activities, which potentially lead to changes in lake trophic state (Zhou et al., 2020a; Zhou et al., 2019). GHG emissions from eutrophic shallow lakes have also been surveyed, and this work has shed light on differences in N<sub>2</sub>O emission fluxes among shallow lakes of different trophic state. In these surveys, the commonly used default emission factor ( $EF_{5t}$ ) by the IPCC has been used (Maavara et al., 2019), which results in either an underestimation or overestimation of the N<sub>2</sub>O budgets in freshwater ecosystems of

different trophic state (Zhang et al., 2020). For example, Xiao et al.(2019) indicated that the mean N<sub>2</sub>O emission fluxes in the East and West zones of Lake Taihu (eutrophic and oligotrophic, respectively) were substantially different, which is partially caused by N flowing to the lake. In addition, N loadings regulate the distribution of denitrifying bacteria, which is often indicated by functional genes for N<sub>2</sub>O production (*nirS* and *nirK*) and consumption (*nosZ*) (Huang et al., 2011; Zhao et al., 2018). High N flowing to eutrophic lakes increases algal growth, potentially enhancing N<sub>2</sub>O emissions and N turnover rates (Yan et al., 2017; Zhu et al., 2020). However, the accumulation of algae alters the redox conditions favoring denitrification (Yan et al., 2017), and more severe anoxic conditions limit the supply of nitrogen oxides generated by nitrification required for denitrification (Zhu et al., 2020); these observations impede our understanding of the role of algae in N<sub>2</sub>O emissions. Such correlations in eutrophic and hypereutrophic lakes have been extensively investigated (Lauerwald et al., 2019; Xiao et al., 2019); however, shallow lakes have been poorly studied. In particular, no studies have compared N<sub>2</sub>O emission fluxes among lakes of different trophic state (Salk and Ostrom, 2019). The trophic state of lakes leads to uncertainty in N<sub>2</sub>O emission estimates (Kortelainen et al., 2020), yet the relationship between N<sub>2</sub>O emission fluxes and the trophic state of lakes is not entirely decoupled. Studies of the N<sub>2</sub>O emissions of lacustrine trophic states on a biogeographic scale could help enhance our understanding of their potential to act as sources of N<sub>2</sub>O emissions.

Approximately 0.9% of China is covered with lakes. There are a total of 2,693 lakes (> 1.0 km<sup>2</sup>), about one-third of which are shallow lakes located in the middle and lower reaches of the Yangtze River basin (Ma et al., 2011). To enhance our understanding of the relationship between N<sub>2</sub>O emissions and eutrophication in shallow lakes in the Yangtze River basin, we characterized spatiotemporal variation in N<sub>2</sub>O emission fluxes and its underlying mechanisms in 17 lakes along a trophic state gradient at a biogeographic scale. We constructed a model to predict the N<sub>2</sub>O emission patterns in shallow lakes of different

trophic state. The aims of this study were to (i) identify N<sub>2</sub>O emission patterns in shallow lakes of different trophic state; (ii) characterize differences in the main microorganisms and functional genes for N<sub>2</sub>O emissions in the sediments in shallow lakes of different trophic state; (iii) evaluate the relationship between environmental variables and N<sub>2</sub>O emissions to reveal the main drivers of N<sub>2</sub>O emissions; and (iv) elucidate the role of algae on N<sub>2</sub>O emissions in shallow lakes. The results of this study enhance our ability to accurately predict N<sub>2</sub>O emission patterns from shallow lakes.

## **4.2 Material and methods**

### **4.2.1 Field survey**

#### **4.2.1.1 Lakes of different trophic state**

This study designated 17 sampling shallow lakes (< 7 m deep) in the middle and lower reaches of the Yangtze River basin. Lakes were sampled in the winter (November) of 2017 and summer (August and September) of 2018. Lake Taihu, Lake Guchenghu, Lake Chaohu, and Lake Donghu were sampled in winter 2017. Because river inflow affects the environmental conditions of lakes (Zhou et al., 2019), all sampling sites were located more than 1 km away from the mouth of inflow rivers. Based on the trophic level index (*TLI*) (see Chapter 2.2.4 for a description of how *TLI* was calculated), these lakes were classified into four trophic states: mesotrophic ( $30 < TLI \leq 50$ ), eutrophic ( $50 < TLI \leq 60$ ), middle-eutrophic ( $60 < TLI \leq 70$ ), and hyper-eutrophic ( $TLI > 70$ ) (Fig. 1) (Zhou et al., 2020b).

#### **4.2.1.2 Heavy algae-accumulated and light algae-accumulated zones in Lake Taihu**

Lake Taihu is a eutrophic lake that has experienced frequent and intensive cyanobacteria blooms since the 1980s (Qin et al., 2010). To characterize N<sub>2</sub>O emission fluxes with or without algae accumulation, three typical zones (from west to east) in Lake Taihu, heavy algae-accumulated (Zone A and B), transitional (Zone C), and light algae-

accumulated zones (Zone D), were studied (Fig. 1c). Sampling was conducted in summer (July) and winter (November) in 2019. The physicochemical parameters of the surface water and N<sub>2</sub>O emission fluxes were investigated.

#### **4.2.2 Sample collection and analysis**

At each sampling event, vertical samples (*i.e.*, overlying water (20 cm below the water level), surface sediment (0–10 cm), and gas samples) were collected in triplicate. The *in situ* dissolved oxygen (DO), temperature, and pH were measured with DO, temperature, and pH probes (HQ3d, HACH, USA) on-site, respectively. To measure dissolved N<sub>2</sub> concentrations in summer, a water sample from a glass water sampler (1 L) was slowly drained from the bottom and transferred to a sample vial (12 mL) through a silicone tube with minimal turbulence. The silicone tube was placed in the bottom of the vial to avoid the ingress of atmospheric N<sub>2</sub>. Next, 60 µL of saturated HgCl<sub>2</sub> solution (0.5% v/v final concentration) was added to the sample vial to inhibit microbial activity. These samples were stored in an ice cooler on-site and immediately transported to the laboratory in a cooler at 4°C. Water samples for chemical analyses, including total nitrogen (TN), total phosphorus (TP), NO<sub>3</sub><sup>-</sup>-N, NH<sub>4</sub><sup>+</sup>-N, dissolved organic carbon (DOC), and chlorophyll-*a* (chl-*a*), were tested using previously described procedures (Zhou et al., 2019). Briefly, TN and TP were measured using an ultraviolet spectrophotometry method and an ammonium molybdate spectrophotometric method, respectively. NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup> concentrations were measured by a water flow analyzer (Auto Analyzer 3, Seal, Germany), and the DOC concentration was determined using an elemental analyzer (Flash EA 1112, CE Instruments, Italy). Chl-*a* was quantified by extraction in 95% ethanol and measuring the absorbance at 630, 645, 663, and 750 nm using a UV-vis spectrophotometer (UV-6100, Mapada, China).

#### **4.2.3 Tested and calculated N<sub>2</sub>O emission fluxes**

N<sub>2</sub>O emission flux was estimated by a floating static chamber (Cole et al., 2010; Gålfalk et al., 2013). The headspace gas was collected between 11:00 and 14:00 using three floating static chambers (size: 38.5 cm × 30.5 cm × 18.5 cm) following a previously described procedure (Zhou et al., 2019). During each gas sampling event, six gas samples were collected at 10-min intervals for 1 h via a static chamber. The gas chromatography (7890B Agilent) configuration described by Shaaban et al. (2018) was used to measure the N<sub>2</sub>O concentration. The detailed methods for calculating N<sub>2</sub>O emission fluxes are described in our previous study (Zhou et al., 2019). N<sub>2</sub>O emission flux estimated by the floating static chamber method was calculated using Eq. (1):

$$F = \frac{V}{A} \times \frac{dC}{dt}, \quad (1)$$

where  $F$  is the N<sub>2</sub>O emission flux ( $\mu\text{g m}^{-2} \text{h}^{-1}$ );  $V$  ( $\text{m}^3$ ) and  $A$  ( $\text{m}^2$ ) are the static chamber volume and surface area, respectively; and  $dC/dt$  is the time derivative of the N<sub>2</sub>O concentration ( $\mu\text{g m}^{-3} \text{h}^{-1}$ ).

#### 4.2.4 Dissolved N<sub>2</sub> concentration and excess dissolved N<sub>2</sub> concentration

Dissolved N<sub>2</sub> was measured by a membrane inlet mass spectrometer system (MIMSS) with a probe inlet (HPR-40, Hiden Analytical Co.) using the N<sub>2</sub>:Ar method described in a previous study (Chen et al., 2014). N<sub>2</sub>:Ar ratios were calculated based on the quadrupole instrument signal (N<sub>2</sub> and Ar pressures at a detector) and calibrated using air-equilibrated water standards (Weiss, 1970). The dissolved N<sub>2</sub> concentrations of triplicate water samples were analyzed, and excess dissolved N<sub>2</sub> concentrations ( $\Delta\text{N}_2$ ) were calculated following previously described methods (Chen et al., 2014).  $\Delta\text{N}_2$  ( $\mu\text{mol L}^{-1}$ ) was calculated using Eq. (2):

$$\Delta\text{N}_2 = \text{N}_{2(\text{water})} - \text{N}_{2(\text{eq})}, \quad (2)$$

where  $\text{N}_{2(\text{water})}$  is the dissolved N<sub>2</sub> concentration in water measured by MIMSS, and  $\text{N}_{2(\text{eq})}$  is the concentration expected if the water were in equilibrium with the atmosphere. Both



were estimated following previously described methods (Weiss, 1970; Weiss and Price, 1980).

#### 4.2.5 Predication of N<sub>2</sub>O emission fluxes based on the IPCC model

A predictive model was used to determine the N<sub>2</sub>O emission factor ( $EF_{5r}$ ) as recommended in the IPCC-2019 guidelines. The dissolved N<sub>2</sub>O concentration ( $\mu\text{g-N L}^{-1}$ ) was estimated using Eq. (3):

$$\text{N}_2\text{O-N} = \text{NO}_3^- \text{-N} \times EF_{5r}, \quad (3)$$

where  $EF_{5r}$  is 0.26% according to the IPCC-2019 default value (IPCC, 2019), and  $\text{NO}_3^- \text{-N}$  ( $\mu\text{g-N L}^{-1}$ ) represents the concentration measured in a water column. N<sub>2</sub>O emission fluxes ( $F'$ ,  $\mu\text{g m}^{-2} \text{h}^{-1}$ ) were calculated by the dissolved N<sub>2</sub>O concentration using the two-layer model of diffusive gas exchange, which is given as Eq. (4)

$$F' = k \times (C_w - C_{eq}), \quad (4)$$

where  $C_w$ , is obtained from Eq. 3 and is the dissolved N<sub>2</sub>O concentration in water estimated by the  $EF_{5r}$ ;  $C_{eq}$  is the N<sub>2</sub>O concentration in water that is in equilibrium with the atmosphere at the *in situ* air pressure and temperature;  $k$  is the gas transfer coefficient ( $\text{m d}^{-1}$ ) and was normalized to the Schmidt number of 600, as described in the Supporting Materials (Cole and Caraco, 1998).

#### 4.2.6 DNA extraction, high-throughput sequencing, and real-Time qPCR analysis

Biomass for the sediment microbial community analysis was collected from lakes of different trophic state. DNA was extracted from the collected biomass using the DNA Isolation Kit (MOBIO, USA) per the manufacturer's instructions. The concentration and purity of DNA were measured using a microvolume UV-VIS spectrophotometer (NanoDrop<sup>TM</sup> One<sup>C</sup>, Thermo Fisher Scientific, USA), and the extracted DNA was stored at  $-20^\circ\text{C}$  before further analysis. 16S rRNA gene high-throughput sequencing was conducted using an Illumina MiSeq platform (Magigene Biotechnology Co. Ltd.,

Guangzhou, China). The primers used for high-throughput sequencing were modified 515F (5'-GTGYCAGCMGCCGCGGTAA-3') and 806R (5'-GGACTACHVGGGTWTCTAAT-3') targeting the V3 and V4 hypervariable regions of both bacterial and archaeal 16S rRNA genes (Zhou et al., 2020b). Given that the *nirK* and *nirS* genes and the *nosZ* gene encode enzymes for N<sub>2</sub>O production and consumption in denitrification, respectively (Zhao et al., 2018), the abundances of these three genes were analyzed. 16S rRNA gene abundances were quantified by real-time quantitative PCR (qPCR) with reported primer sets (Table S3). The PCR conditions for the amplification of *nirK*, *nirS*, and *nosZ* were described in a previous study (Chen et al., 2017).

#### 4.2.7 Statistical analysis

Statistical analyses were conducted using SPSS 19.0 (SPSS Inc., Chicago, USA). Significant differences among lakes were determined by one-way analysis of variance (ANOVA). The thresholds for statistically significant and highly statistically significant were  $p < 0.05$  and  $p < 0.01$  (two-tailed), respectively. The Kolmogorov-Smirnov test was conducted to determine if the data were normally distributed. The built-in exponential model of Origin 2019 software (OriginLab Inc., USA) was carried out to assess the relationship between N<sub>2</sub>O emission fluxes and *TLI*.

### 4.3 Results

#### 4.3.1 *In situ* N<sub>2</sub>O emission fluxes in lakes of different trophic state

The *TLI* of the examined lakes ranged from 44.0 to 69.7 and from 43.3 to 76.7 in the summer and winter, respectively (Fig. S1). The hyper-eutrophic sampling sites were not included in the winter dataset. Among these examined lakes, the ranges of N<sub>2</sub>O emission fluxes were  $-1.0$ – $53.0 \mu\text{g m}^{-2} \text{h}^{-1}$  and  $0.4$ – $102.9 \mu\text{g m}^{-2} \text{h}^{-1}$  in summer and winter, respectively, indicating a high degree of variation in N<sub>2</sub>O emission fluxes (Table S1 and Fig. 2). In addition, all examined lakes were N<sub>2</sub>O sources, with the exception of

mesotrophic lakes in summer. The mean N<sub>2</sub>O emission fluxes of the middle-eutrophic lakes in winter (50.4  $\mu\text{g m}^{-2} \text{h}^{-1}$ ) were higher than those of the hyper-eutrophic (39.1  $\mu\text{g m}^{-2} \text{h}^{-1}$ ) and middle-eutrophic (7.4  $\mu\text{g m}^{-2} \text{h}^{-1}$ ) lakes in summer (Table S1). Overall, the N<sub>2</sub>O emission fluxes gradually increased as *TLI* increased (Fig. 2a).

The lacustrine N<sub>2</sub>O emission fluxes in both winter and summer were positively correlated with *TLI* (Fig. 2a). The N<sub>2</sub>O emission fluxes exponentially increased with the *TLIs* of the sampled lakes (adj.  $R^2 = 0.85$ ,  $p < 0.01$  in summer; adj.  $R^2 = 0.63$ ,  $p < 0.01$  in winter). There was a non-linear relationship between the net N<sub>2</sub>O emission fluxes and *TLI* (adj.  $R^2 = 0.36$ ,  $p < 0.01$ ) (Fig. 2b). The N<sub>2</sub>O emission fluxes increased as the *TLI* of lakes increased, and the increase was more pronounced in hypertrophic lakes. In addition, N<sub>2</sub>O emission fluxes were predicted based on the  $EF_{5r}$  [Equation (3)]. There was a nonlinear exponential relationship between the predicted N<sub>2</sub>O emission fluxes and *TLI* (adj.  $R^2 = 0.80$ ,  $p < 0.01$ ) (Fig. 2b). These fluxes were higher than the observed values, especially in the hyper-eutrophic lakes in summer ( $TLI > 70$ ) (Fig. S2).

#### 4.3.2 Dissolved N<sub>2</sub> concentration in shallow lakes

Mean dissolved N<sub>2</sub> concentrations across all lakes in summer ranged from 403.3  $\mu\text{mol L}^{-1}$  to 443.8  $\mu\text{mol L}^{-1}$  and exhibited a unimodal relationship with *TLI* (Fig. 3a). Among these examined lakes, the lowest dissolved N<sub>2</sub> concentration was observed in hyper-eutrophic lakes. There was a significant unimodal relationship between the dissolved N<sub>2</sub> concentration and *TLI* ( $p < 0.001$ ). The excess dissolved N<sub>2</sub> ( $\Delta\text{N}_2$ ), which was obtained by subtracting dissolved N<sub>2</sub> from the saturated concentration, was consistently positive (8.1–16.1  $\mu\text{mol L}^{-1}$ ) in summer, indicating N<sub>2</sub> oversaturation. The trend of N<sub>2</sub> oversaturation as a function of *TLI* was consistent with dissolved N<sub>2</sub> (Fig. 3).

#### 4.3.3 Relationship between environmental factors and N<sub>2</sub>O emission fluxes

The Pearson correlations between N<sub>2</sub>O emission fluxes and environmental

parameters of the overlying water were analyzed in different seasons (Table S4). The N<sub>2</sub>O emission fluxes were significantly and positively correlated with TN, NO<sub>3</sub><sup>-</sup>-N, and *TLI* ( $p < 0.01$ ). The correlation between chl-*a* and N<sub>2</sub>O emission fluxes varied seasonally, and the correlation was stronger in summer. ΔN<sub>2</sub> was negatively correlated with N<sub>2</sub>O emission fluxes, TN, NO<sub>3</sub><sup>-</sup>-N, and chl-*a* in summer ( $p < 0.05$ ). There was a negative correlation between DO concentrations and N<sub>2</sub>O emission fluxes ( $p < 0.05$ ) in summer, and this correlation was not observed in winter and over the entire year ( $p > 0.05$ ) (Table S4). The overlying water DO was higher in winter ( $> 7 \text{ mg L}^{-1}$ ) than in summer (Fig. S1b). During the survey period, the temperature of the overlying water ranged from 9.5°C to 15.1°C and 27.1°C to 31.3°C in winter and summer, respectively (Fig. S1c). The temperature was negatively correlated with N<sub>2</sub>O emission fluxes in both summer and winter ( $p < 0.05$ , Table S5). N<sub>2</sub>O emission fluxes showed significant and positive linear correlations with TN (adj.  $R^2 = 0.797$ ,  $p < 0.01$  in winter; adj.  $R^2 = 0.908$ ,  $p < 0.01$  in summer), NO<sub>3</sub><sup>-</sup>-N (adj.  $R^2 = 0.787$ ,  $p < 0.01$  in winter; adj.  $R^2 = 0.826$ ,  $p < 0.01$  in summer), and NH<sub>4</sub><sup>+</sup>-N (adj.  $R^2 = 0.39$ ,  $p = 0.039$  in winter; adj.  $R^2 = 0.484$ ,  $p = 0.01$  in summer) (Fig. 4). In addition, there was a significant negative linear correlation between *TLI* and the C:N ratio (adj.  $R^2 = 0.649$ ,  $p < 0.001$ ) (Fig. S3). There was a strong nonlinear correlation between N<sub>2</sub>O emission fluxes and the C:N ratio (adj.  $R^2 = 0.414$ ,  $p < 0.001$ ) (Fig. S5).

A multiple stepwise regression model incorporating the physicochemical variables of the overlying water was established for N<sub>2</sub>O emission fluxes (Table 1). The results showed that TN and TP concentrations can predict N<sub>2</sub>O emission fluxes in summer (adj.  $R^2 = 0.94$ ,  $p < 0.001$ ). TN ( $t_1 = 10.80$ ) was more strongly positively correlated with N<sub>2</sub>O emission fluxes than TP ( $t_2 = 2.93$ ), which indicated that TN was an important parameter determining N<sub>2</sub>O emission fluxes in summer. In addition, TN, chl-*a*, and NO<sub>3</sub><sup>-</sup>-N could jointly predict N<sub>2</sub>O emission fluxes (adj.  $R^2 = 0.96$ ,  $p < 0.001$ ) in winter, demonstrating that both TN (6.11) and chl-*a* (−6.56) are important parameters determining N<sub>2</sub>O emission fluxes in winter. TN (12.83) and chl-*a* (−4.07) were important variables predicting N<sub>2</sub>O

emission flux in lakes (adj.  $R^2 = 0.86$ ,  $p < 0.01$ ).

#### 4.3.4 Microbial community structure and denitrifier abundances in lakes of different trophic state

Relative abundances of microbes were obtained at the phylum level in summer and winter (Fig. 5). Overall, the top 15 phyla made up more than 80% of microbial communities in all lake sediments. The following six phyla accounted for over 60% of the total population in summer: *Proteobacteria*, *Chloroflexi*, *Bacteroidetes*, *Acidobacteria*, *Verrucomicrobia*, and *Planctomycetes*; those in the winter were *Proteobacteria*, *Bacteroidetes*, *Acidobacteria*, *Nitrospirae*, *Planctomycetes*, and *Verrucomicrobia*. The most abundant phylum was *Proteobacteria*, which had relative abundances in hyper-eutrophic, middle-eutrophic, eutrophic, and mesotrophic lakes of 30.6–44.2%, 32.3–42.4%, 31.8–39.9%, and 32.0–46.4%, respectively. *Proteobacteria* (26.4–43.4%) was also common in all lakes in winter. Among the six dominant phyla, the relative abundances of *Chloroflexi* were 2.8–5.7% in winter and 4.7–15.9% in summer; the relative abundances of *Nitrospirae* were 2.3–9.7% in winter and 0.1–4.1% in summer.

The *nirS* gene abundances in winter ( $0.66 \times 10^7$ – $3.67 \times 10^7$  copies  $\text{g}^{-1}$ -sediment) were lower than those in summer ( $2.70 \times 10^7$ – $7.62 \times 10^8$  copies  $\text{g}^{-1}$ -sediment) (Fig. S5). By contrast, no significant difference was observed for *nirK* gene abundances in winter ( $0.36 \times 10^7$ – $2.29 \times 10^7$  copies  $\text{g}^{-1}$ -sediment) and summer ( $0.06 \times 10^7$ – $2.06 \times 10^7$  copies  $\text{g}^{-1}$ -sediment). The *nirS/nirK* ratio was greater than 1, indicating that the *nirS*-type denitrifiers were consistently more abundant than *nirK*-type denitrifiers in these shallow lakes (Fig. 6a). The ratios were 9–48 times higher in summer than in winter. In addition, the *nirS/nirK* ratio increased as *TLI* increased in summer and winter ( $p < 0.05$ ). *nirS/nirK* and  $\text{N}_2\text{O}$  emission fluxes were also positively correlated (Fig. S6a). The abundance of *nosZ*, which encodes an enzyme for  $\text{N}_2\text{O}$  consumption, varied from  $0.31 \times 10^7$  copies  $\text{g}^{-1}$ -sediment to  $2.88 \times 10^7$  copies  $\text{g}^{-1}$ -sediment in winter, which was lower than that in

summer ( $0.53 \times 10^7$ – $7.02 \times 10^7$  copies g<sup>-1</sup>-sediment) (Fig. S5). Variation in the ratios of (*nirK* + *nirS*)/*nosZ* was lower among the shallow lakes in winter (0.8–9.5) than in summer (2.9–12.3) (Fig. 6b). The ratio of (*nirK* + *nirS*)/*nosZ* was positively correlated with *TLI* ( $p = 0.022$ ) and N<sub>2</sub>O emission flux (Fig. S6b). Further analysis revealed the relationship between the spatial distribution of these denitrifying genes and multiple environmental factors (Fig. 7). A redundancy analysis (RDA) showed that the first two axes explained 60.38% of the variation in the denitrifying genes. The samples of the examined lakes were well separated among the different seasons. The RDA between the denitrifying gene abundances and environmental parameters indicated that temperature was an important factor affecting gene abundances. Among the tested parameters, *nirS* and *nosZ* abundances were sensitive to temperature compared with *nirK*. The dissimilarity in the abundance of denitrifying genes in lakes of different trophic state was greater in summer than in winter. The abundance of *nirS* was positively correlated with *TLI*, and the abundance of *nosZ* was marginally correlated with *TLI*.

#### **4.3.5 Characterization of N<sub>2</sub>O emissions in the heavy algae-accumulated and light algae-accumulated zones in Lake Taihu**

The N loading and chl-*a* concentrations were higher in the heavy algae-accumulated zones (Zones A and B) than in the light algae-accumulated zone (Zone D) (Table S2). The N<sub>2</sub>O emissions fluxes were characterized in summer and winter in the two typical zones in Lake Taihu. The N<sub>2</sub>O emission fluxes were significantly higher in the heavy algae-accumulated zones (Zones A and B) and transitional zone (Zone C) than in the light algae-accumulated zone (Zone D). The N<sub>2</sub>O emission fluxes were location-dependent and varied from 42.16–136.63 µg m<sup>-2</sup> h<sup>-1</sup> in the heavy algae-accumulated zones (Zones A and B), 21.35–31.89 µg m<sup>-2</sup> h<sup>-1</sup> in the transitional zone (Zone C), and 3.7–4.71 µg m<sup>-2</sup> h<sup>-1</sup> in the light algae-accumulated zone (Zone D) (Fig. 8). These fluxes in the algae-accumulated zones were significantly different in summer and winter ( $p < 0.05$ ), and

differences were not significant in Zone C ( $p = 0.076$ ) and Zone D ( $p = 0.677$ ). There was an exponential relationship between *TLI* and  $\text{N}_2\text{O}$  emission fluxes (Fig. S7) (adj.  $R^2 = 0.55$ ,  $p < 0.05$ ).

## 4.4 Discussion

### 4.4.1 Nonlinear $\text{N}_2\text{O}$ emission patterns

Shallow lakes are potential sources of  $\text{N}_2\text{O}$  emissions and have been extensively studied (Kortelainen et al., 2020; Lauerwald et al., 2019; McCrackin and Elser, 2011). Previous studies have documented variation in  $\text{N}_2\text{O}$  emissions across lakes of different trophic state (Kortelainen et al., 2020; Salk and Ostrom, 2019); nevertheless, predicting  $\text{N}_2\text{O}$  emission fluxes from shallow lakes remains a challenge. This study showed that the net  $\text{N}_2\text{O}$  emission fluxes in lakes located in the Yangtze River basin, which spans 1000 km, displayed spatial and temporal heterogeneity determined by lake trophic state (Fig. 2). Consistent with previous studies (Salk and Ostrom, 2019; Xiao et al., 2019; Zhou et al., 2020a), these findings indicate that eutrophic lakes in the Yangtze River basin are sources of  $\text{N}_2\text{O}$  emissions. The patterns of  $\text{N}_2\text{O}$  emission fluxes in the shallow lakes were not completely consistent with the results of previous studies and depended on eutrophication progress. The  $\text{N}_2\text{O}$  emission fluxes in the eutrophic and middle-eutrophic lakes (Table S1) were similar to the global median value of  $\text{N}_2\text{O}$  emission flux in lakes, whereas the mean  $\text{N}_2\text{O}$  emission flux in the hyper-eutrophic lakes was 8.7–11.2 times higher than the global median value (Hu et al., 2016). Shallow lakes in a mesotrophic state in summer were  $\text{N}_2\text{O}$  sinks (Fig. 2 and Table S1); these findings expand our knowledge regarding the prerequisites for freshwater lakes to act as either  $\text{N}_2\text{O}$  sinks or sources (Lauerwald et al., 2019). Our study underscores the significance of lake trophic state in determining  $\text{N}_2\text{O}$  emission fluxes, which is supported by the correlation between net  $\text{N}_2\text{O}$  emission flux and trophic state (Fig. 2a). Whether lakes of different trophic state act as  $\text{N}_2\text{O}$  sources or sinks can be predicted based on this correlation.

The major contribution of this study is the exponential model based on *TLI*, which could provide a robust means for quantifying lake trophic state; this model can be used to predict the N<sub>2</sub>O emissions from lakes. This model represents an improvement over previous approaches for estimating N<sub>2</sub>O emission fluxes because previous approaches do not consider differences in lake trophic state (Lauerwald et al., 2019). This model can be used to assess N<sub>2</sub>O emission fluxes in shallow lakes of different trophic state. The N<sub>2</sub>O emission fluxes and *TLI* were well fitted in summer (adj. R<sup>2</sup> = 0.85) and winter (adj. R<sup>2</sup> = 0.63). However, the coefficient of determination was low (adj. R<sup>2</sup> = 0.36) when all data (*i.e.*, summer and winter) were incorporated into the model, which is likely explained by the large differences in N<sub>2</sub>O emission fluxes in winter and summer (Fig. 2). Therefore, seasonal differences in N<sub>2</sub>O emission fluxes in shallow lakes should also receive consideration (Kortelainen et al., 2019; Miao et al., 2020). Given the limited data on seasonal differences and the limited number of shallow lakes investigated, more data on N<sub>2</sub>O emission fluxes are required to verify the credibility of the model.

#### **4.4.2 Potential drivers of N<sub>2</sub>O emissions**

Our results revealed that reactive N accumulation plays a major role in regulating lake trophic state levels and the biological N cycle and promotes N<sub>2</sub>O emissions via denitrification in shallow lakes. Degradation and metabolism were similar among shallow lakes despite variation in trophic state. *Proteobacteria* was identified as a predominant phylum based on the 16S rRNA gene analysis, and it was commonly detected in lakes of different trophic state (Fig. 5) (Li et al., 2019b); *Proteobacteria* might potentially contribute to degradation and metabolism (Huang et al., 2019). Previous studies have indicated that copiotrophic groups such as *Proteobacteria* and *Bacteroidetes* with high growth rates tend to thrive in nutrient-rich conditions (Fierer et al., 2012). The overlying water and sediments likely stored abundant nutrients that could be used by microorganisms in shallow lakes. In addition, nutrient abundance is an important factor



affecting the microorganisms responsible for N conversion (Saarenheimo et al., 2015; Zhang et al., 2019). The denitrifying genes varied greatly among the different trophic lakes and seasons (Figs. 6 and S5). These results suggest that denitrification is an important source of N<sub>2</sub>O emissions in shallow eutrophic lakes, which is consistent with the results of previous studies (Beaulieu et al., 2011; Zhang et al., 2020). This is also confirmed by the stronger correlation of N<sub>2</sub>O emission flux with NO<sub>3</sub><sup>-</sup> than with NH<sub>4</sub><sup>+</sup> (Fig. 4) and the negative correlation between DO and N<sub>2</sub>O emission fluxes (Table S4). The significant relationship between *TLI* and  $(nirK + nirS)/nosZ$  ( $p = 0.022$ ), which indicates the relative abundance of N<sub>2</sub>O producers relative to N<sub>2</sub>O consumers, suggests that N<sub>2</sub>O production may be greater than N<sub>2</sub>O consumption in hyper-eutrophic lakes (Fig. 6b) (Zhao et al., 2018). Higher net N<sub>2</sub>O emission fluxes were observed in hyper-eutrophic lakes because of their higher nutrient availability and  $(nirK + nirS)/nosZ$  (Kortelainen et al., 2020). These findings might explain the close relationship between lake trophic state and N<sub>2</sub>O emission flux.

N availability drives eutrophication, and the subsequent accumulation of algae alters the redox conditions favoring denitrification (Yan et al., 2017; Zhu et al., 2020) and increases N<sub>2</sub>O emissions in shallow lakes. In freshwater ecosystems, N loadings significantly contribute to N<sub>2</sub>O emissions via denitrification (Kortelainen et al., 2020; Mulholland et al., 2008), which explains the high N<sub>2</sub>O emission fluxes observed in eutrophic lakes when N loading was high (Figs. 2 and 4). A stepwise linear regression model indicated that TN and chl-*a* (algal density) are important parameters explaining N<sub>2</sub>O emissions (Table 1). Following algal accumulation, algal decay alters redox conditions and releases organic matter *in-situ* (Yan et al., 2017; Zhu et al., 2020). Moreover, algal blooms result in low DO concentrations and the accumulation of organic matter in hyper-eutrophic lakes (Yan et al., 2017; Zhou et al., 2020b; Zhu et al., 2020). For example, algal accumulation in Lake Taihu accounts for >50% of the organic matter (Xu et al., 2019). A C:N ratio lower than 8 indicates that organic matter is mainly derived

from autochthonous inputs (Meyers, 1994; Yan et al., 2017). The negative correlation between C:N ratio and *TLI* (Fig. S3) indicates that algal accumulation and decomposition alter the physicochemical conditions in shallow lakes. Therefore, the contribution of algal decomposition should receive increased consideration when exploring the relationship between N<sub>2</sub>O emission fluxes and chl-*a*. Our results indicated that N<sub>2</sub>O emission fluxes and the C:N ratio were negatively correlated ( $p < 0.001$ ) (Fig. S5). Algal decomposition results in oxygen consumption and thus a low DO concentration, which is favorable for denitrification (Zhu et al., 2020). This observation is consistent with the negative correlation between DO and chl-*a* (Tables S4 and S5). These findings might potentially explain the roles of decomposed algae in stimulating N<sub>2</sub>O emissions via consumed oxygen in eutrophic shallow lakes.

Temperature is an important variable determining denitrifier abundance and structure in both seasons (Figs. 6 and 7); the effect of temperature was also manifested by the differences in N<sub>2</sub>O emissions among seasons (Fig. 2). This pattern is similar to a previous study of 87 boreal lakes in Finland showing that N<sub>2</sub>O emissions peaked in winter (Kortelainen et al., 2020). Among enzymes responsible for denitrification, N<sub>2</sub>O reductase is most sensitive to changes in temperature (Kortelainen et al., 2020; Veraart et al., 2011; Zhou et al., 2020c). Our previous study indicated that *nosZ* in N<sub>2</sub>O-reducing bacteria is strongly related to temperature (Zhou et al., 2020c). In addition, N<sub>2</sub>O reduction activity is inhibited by oxygen exposure, which increases N<sub>2</sub>O emissions (Song et al., 2019). Low temperature increased the oxygen concentration (Fig. S1), and the relationship between the ratio of *nirS* to *nirK* with *TLI* varied in winter (slope: 0.08) and summer (slope: 1.11) ( $p < 0.05$ ) (Fig. 6a). Given that the algae in lakes gradually declined in winter concomitant with increasing DO concentrations (Fig. S1 and Table S1), N<sub>2</sub>O emission fluxes in lakes increased in winter (Fig. 2a) (Miao et al., 2020). In summer, algal decay further decreased the oxygen concentration. Previous studies have shown that *nirK* only achieves high abundances in conditionally oxygen-exposed environment (Huang et al., 2011), whereas

*nirS* genes have been more commonly detected in anoxic locations (Knapp et al., 2009). The observation that the N<sub>2</sub>O emission fluxes, *nirS/nirK*, and (*nirS+nirK*)/*nosZ* showed more significant positive relationships in winter than in summer suggests that the N<sub>2</sub>O emission fluxes in summer were also affected by other factors (Fig. S6). The dramatic difference between hyper-eutrophic lakes in winter and summer was in the frequency of algal blooms in hyper-eutrophic lakes in summer. The algae regulating N<sub>2</sub>O emission fluxes in shallow lakes are discussed in the subsequent section. In sum, low temperature associated with abundant N favored N<sub>2</sub>O accumulation via control oxygen concentrations and limit N<sub>2</sub>O reduction activity.

#### **4.4.3 The dual-impact of algae in N<sub>2</sub>O emissions**

N<sub>2</sub>O emissions were highest, and the chl-*a* concentration low, in Zone B in contrast to Zone A in a heavy algae-accumulated zone of Lake Taihu (Fig. 8 and Table S2). This result is inconsistent with the finding that algal accumulation, reflected by *TLI*, stimulated N<sub>2</sub>O emissions in a non-linear exponential manner in the other tested lakes (Fig. S7). In heavy algae-accumulated zones, lower N<sub>2</sub>O emissions, indicated by the high chl-*a* concentration, likely stem from algal accumulation, which suppresses denitrification activities caused by the decrease in N<sub>2</sub>O production. Recently, Zhu et al. (2020) reported that algal accumulation may inhibit denitrification during algal blooms in summer. Consistent with this finding, a unimodal relationship between *TLI* and excess dissolved N<sub>2</sub> ( $\Delta N_2$ ) in summer was observed (Fig. 3b), which indicated higher complete denitrification (*i.e.*, including N<sub>2</sub>O consumption) rates in the eutrophic lakes (Chen et al., 2014; Wang et al., 2018). In summer, the highest gene abundances of *nirK*, *nirS*, and *nosZ* were observed in eutrophic and middle-eutrophic lakes rather than in hyper-eutrophic lakes (Fig. S5). Although more abundant N is available in hyper-eutrophic lakes, the complete denitrification rates may be lower in hyper-eutrophic lakes than in eutrophic lakes in summer. This pattern potentially stems from the abundance of algal biomass in

the hyper-eutrophic lakes, as release of algal debris eventually leads to reductive conditions (Table S2). The decay of excess algal biomass could create favorable conditions for denitrification where organic carbon is present under hypoxic or anoxic conditions. However, hypoxia limits nitrification, which leads to a deficiency in the supply of  $\text{NO}_3^-$  for denitrification (Small et al., 2014; Zhu et al., 2020). This effect is supported by previous work in zones with accumulated algae in Lake Taihu showing that the  $\text{NH}_4^+$  concentration in sediments is two or three orders of magnitude higher than the  $\text{NO}_3^-$  concentration (Yan et al., 2019). Our results are also consistent with this observation (Table S2). In addition, chl-*a* (as a negative factor) and TN together predicted the  $\text{N}_2\text{O}$  emission fluxes (Table 1). Algal accumulation and decomposition create hypoxic conditions that limit nitrification by converting  $\text{NH}_4^+$  into  $\text{NO}_3^-$ , eventually suppressing the ensuing denitrification. In addition, hypoxic conditions favor complete denitrifying bacteria for mitigating  $\text{N}_2\text{O}$  emissions. These results further explain why higher  $\text{N}_2\text{O}$  emission fluxes of shallow lakes were observed in winter rather than in summer (Figs. 2a and 8). Therefore, these evidences indicated that algal accumulation played a dual role in stimulating and impeding  $\text{N}_2\text{O}$  emissions, especially in hyper-eutrophic lakes.

#### **4.4.4 Implications from eutrophic progress**

In aquatic ecosystems, excessive N loadings drive eutrophication and promote  $\text{N}_2\text{O}$  emissions in water bodies (Zhao et al., 2015). Meanwhile, the nutrient overload induces algae growth in shallow lakes, forming an anoxic or microaerobic micro-environment favoring  $\text{N}_2\text{O}$  production by algae accumulation (Zhu et al., 2020). In this study, a nonlinear exponential increase in  $\text{N}_2\text{O}$  emission flux as a function of *TLI* in shallow lakes of different trophic state was observed (Fig. 2). Eutrophic lakes had high  $\text{N}_2\text{O}$  emission fluxes, which is consistent with the relationship between lake trophic state and  $\text{CH}_4$  emission fluxes (Zhou et al., 2020b). The common trends indicate that increases in GHG emissions stem from eutrophication. The predicted  $\text{N}_2\text{O}$  emission fluxes based on the

IPCC  $EF_{5r}$  overestimated the observed fluxes but also exhibited a non-linear exponential increase with  $TLI$  (Fig. 2b). Xiao et al. (2019) reported that the  $N_2O$  emission factor in Lake Taihu was 0.18%, which is lower than the value of  $EF_{5r}$  based on the reported IPCC value. The overestimation by the IPCC default value indicates the need to calibrate an  $N_2O$  emission factor in shallow lakes depending on the eutrophic state. Pronounced differences between the predicted and observed  $N_2O$  emission fluxes were observed in hyper-eutrophic lakes where the abundance of algae impeded  $N_2O$  emissions. The high abundance of algae in summer compared with winter indicates that the degree to which  $N_2O$  emission flux was overestimated in summer was different from that in winter (Fig. S2). Algal decomposition made the redox conditions favorable for denitrification but unfavorable for nitrification because of an insufficient supply of  $NO_3^-$ . The retained nitrogen is absorbed by the newly grown algae (Zhu et al., 2020). In non-limited N, algal decay leads to low-oxygen conditions, which enhances denitrification and further stimulates  $N_2O$  emissions. Therefore, algae should be considered a nitrogen “pool” that maintains nitrogen in lakes. In such situations where algae pools N, lower  $N_2O$  emissions stemming from the suppression of nitrification do not contribute to the reduction in global  $N_2O$  emissions from lakes but potentially leads to substantial  $N_2O$  emissions from hyper-eutrophic state lakes when conditions for nitrification are suitable. We suggest that overestimation was possibly caused by the “dual role” of the algae because they help regulate denitrification to mitigate  $N_2O$  emissions. Estimation of flux by the two-layer model was one order of magnitude lower than that estimated by static chamber methods (Duchemin et al., 1999). Given that the two different methods plausibly overestimated or underestimated the  $N_2O$  emission fluxes in shallow lakes, an intensive survey of an  $N_2O$  emission factor in hyper-eutrophic lakes will be conducted in a follow-up study.

## 4.5 Conclusions

We performed a series of field measurements and characterized the  $N_2O$  emissions

in shallow lakes of different trophic state in the Yangtze River basin. The results of this study are detailed below.

- The N<sub>2</sub>O emission fluxes of shallow lakes were most strongly affected by lake trophic state, suggesting that estimation of N<sub>2</sub>O emission fluxes should consider lake trophic state.
- The nonlinear model incorporating trophic state levels can describe the N<sub>2</sub>O emissions from a shallow lake.
- The predicted N<sub>2</sub>O emission fluxes based on the IPCC *EF<sub>5r</sub>* overestimated the observed fluxes, particularly those in hyper-eutrophic lakes.
- Nutrient-rich conditions and algal accumulation were key factors determining N<sub>2</sub>O emission fluxes in shallow lakes, and algal accumulation played a dual role in stimulating and impeding N<sub>2</sub>O emissions, especially in hyper-eutrophic lakes.
- Changes in season accompanied the appearance and disappearance of algae and altered N<sub>2</sub>O emission fluxes, especially in hyper-eutrophic lakes.

#### 4.6 References

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## Table and Figures

Table 1. Validity of the multiple stepwise regression model for N<sub>2</sub>O emission fluxes incorporating variables of the overlying water

Season	Parameters	Equations	Variables	Adj. R <sup>2</sup>	<i>p</i>	Significance level			
						F-test	T-test		
						F	t <sub>1</sub>	t <sub>2</sub>	t <sub>3</sub>
Summer	N <sub>2</sub> O	N <sub>2</sub> O = 14.33(TN) – 10.45	TN	0.91	<i>p</i> < 0.001	158.86	12.60		
		N <sub>2</sub> O = 12.36(TN) + 27.99(TP) – 13.13	TN, TP	0.94	<i>p</i> < 0.001	123.95	10.80	2.93	
Winter	N <sub>2</sub> O	N <sub>2</sub> O = 21.04(TN) – 13.44	TN	0.80	<i>p</i> < 0.001	56.11	7.49		
		N <sub>2</sub> O = 32.06(TN) – 1.79(chl- <i>a</i> ) – 1.09	TN, chl- <i>a</i>	0.90	<i>p</i> < 0.001	60.65	8.76	–3.62	
		N <sub>2</sub> O = 20.68(TN) – 2.04(chl- <i>a</i> ) +	TN, chl- <i>a</i> ,	0.96	<i>p</i> < 0.001	112.69	6.11	–6.56	0.61
		22.55(NO <sub>3</sub> <sup>–</sup> -N) + 3.13	NO <sub>3</sub> <sup>–</sup> -N						
Summer + Winter	N <sub>2</sub> O	N <sub>2</sub> O = 18.93(TN) – 13.80	TN	0.79	<i>p</i> < 0.001	110.56	10.52		
		N <sub>2</sub> O = 23.45(TN) – 0.33(chl- <i>a</i> ) – 12.03	TN, chl- <i>a</i>	0.86	<i>p</i> < 0.001	74.08	12.83	–4.07	

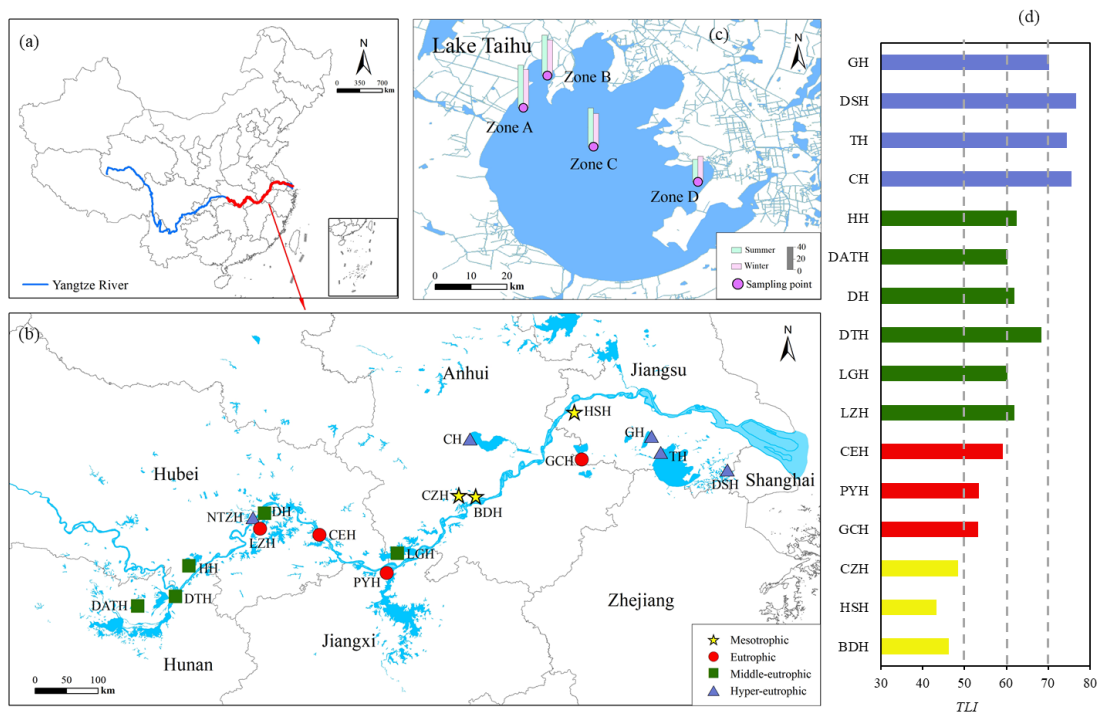


Fig. 1 Distribution of sampling sites (b) in the middle and lower reaches of the Yangtze River basin, China. Sampling sites (c) in the heavy algae-accumulated zone (Zone A and B), transitional zone (Zone C), and light algae-accumulated zone (Zone D) of Lake Taihu. The graph on the right (d) shows the *TLI* of each lake in summer. DSH: Lake Dianshang; TH: Lake Taihu; GH: Lake Gehu; GCH: Lake Gucheng; HSH: Lake Huashen; CH: Lake Chaohu; BDH: Lake Baidang; CZH: Lake Caizi; LGH: Lake Longgan; PYH: Lake Poyang; CEH: Lake Cehu; DH: Lake Donghu; LZH: Lake Liangzi; NTZH: Lake Nantaizi; HH: Lake Honghu; DTH: Lake Dongting; DATH: Lake Datong.

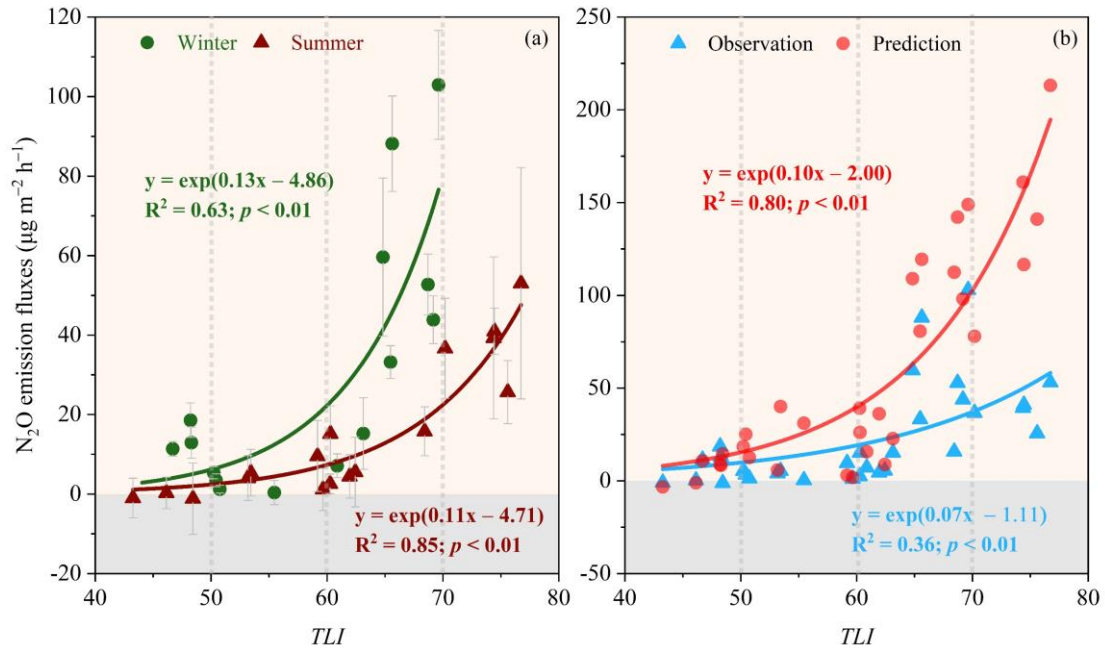


Fig. 2. Net  $N_2O$  emissions (a) in lakes of different trophic status during the winter (2017) and summer (2018). The plots (b) show the predicted  $N_2O$  emission fluxes based on the  $EF_{5r}$  of the IPCC default value (0.26%). Note that the  $TLI$  ranges for mesotrophic, eutrophic, middle-eutrophic, and hypereutrophic states are 40–50, 50–60, 60–70, and > 70, respectively.

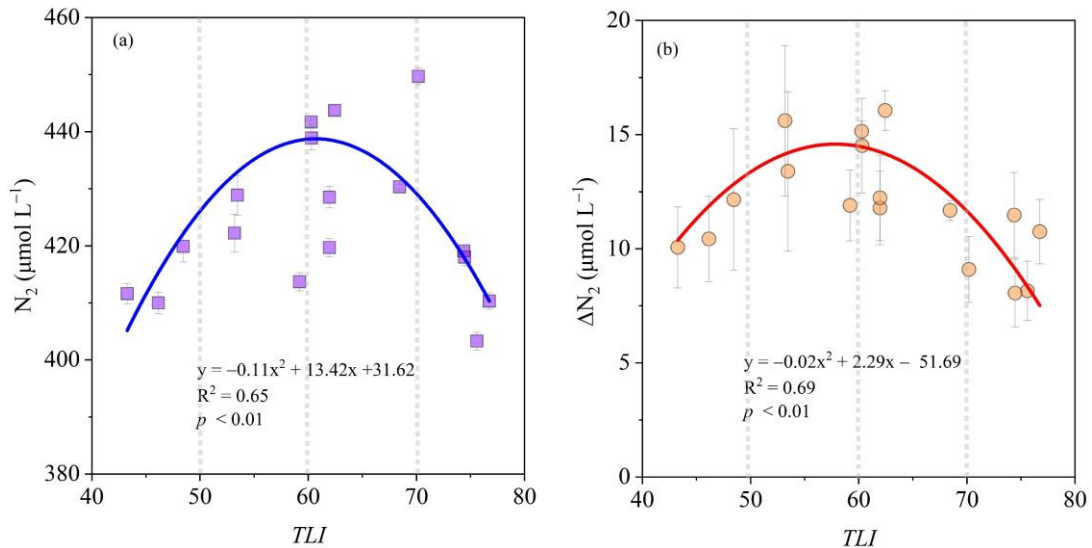


Fig. 3. The relationship between  $TLI$  and (a) dissolved  $N_2$  concentration as well as (b) excess dissolved  $N_2$  concentration ( $\Delta N_2$ ) in shallow lakes in summer (2018). Note that



the *TLI* ranges for mesotrophic, eutrophic, middle-eutrophic, and hypereutrophic states are 40–50, 50–60, 60–70, and > 70, respectively.

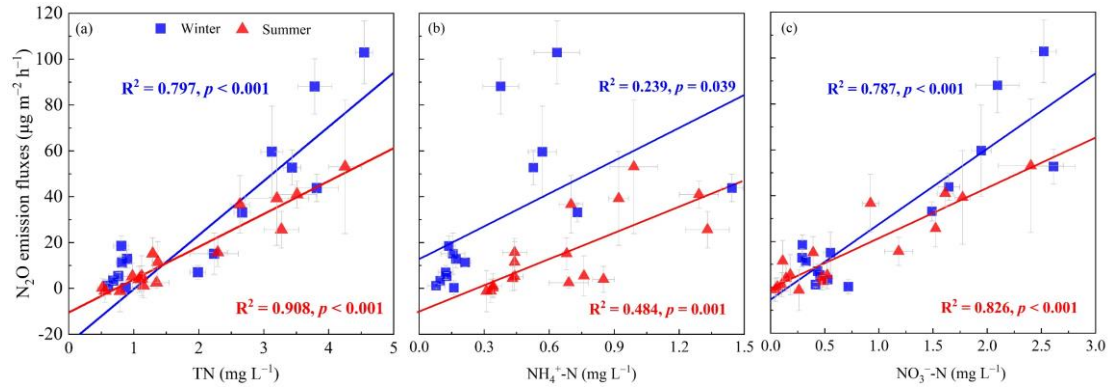


Fig. 4. Relationship between net  $\text{N}_2\text{O}$  emission fluxes and TN (a),  $\text{NH}_4^+\text{-N}$  (b), and  $\text{NO}_3^-$ -N (c) in winter (2017) and summer (2018) of shallow lakes.

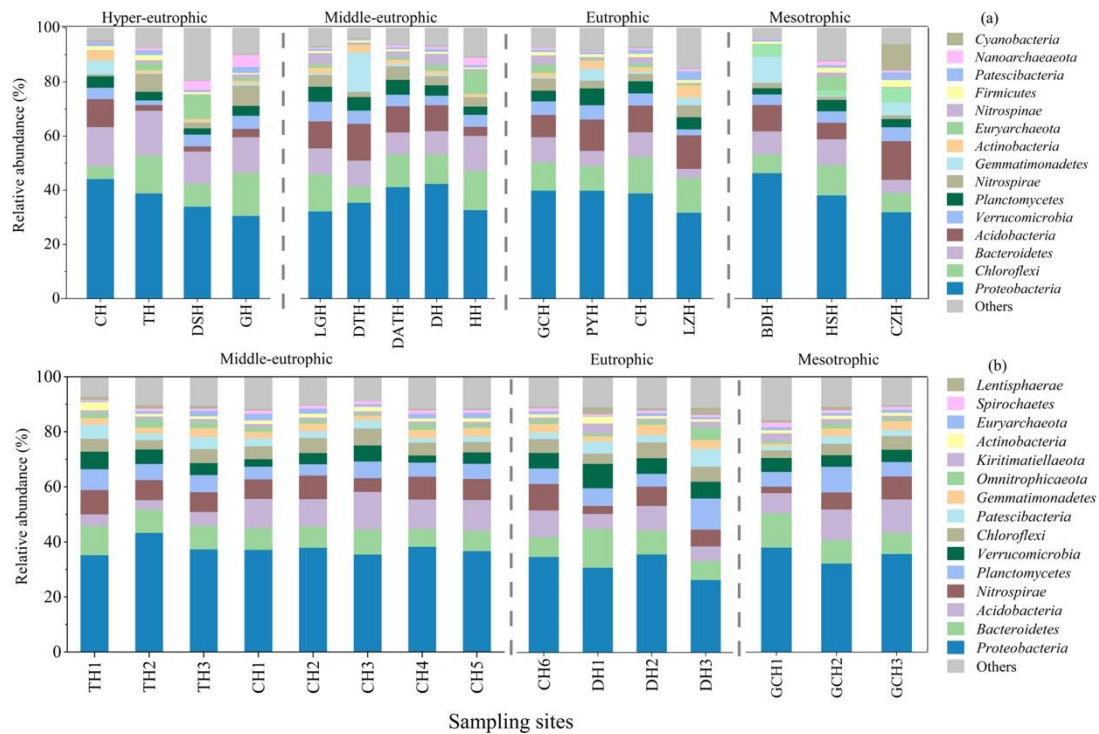


Fig. 5. Relative abundances of 16S rRNA gene-based microbial taxa for each sample at the phylum level in (a) summer (2018) and (b) winter (2017). The 15 most abundant phyla are shown.

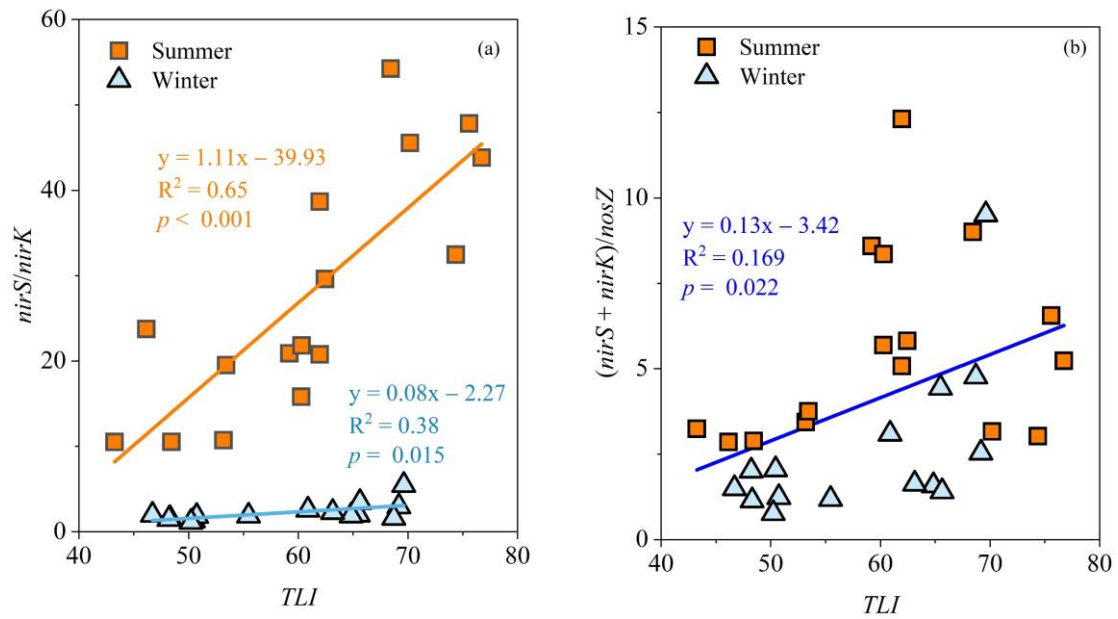


Fig. 6. The relationship between TLI and ratio of (a) *nirS/nirK* as well as (b)  $(nirS + nirK)/nosZ$ .

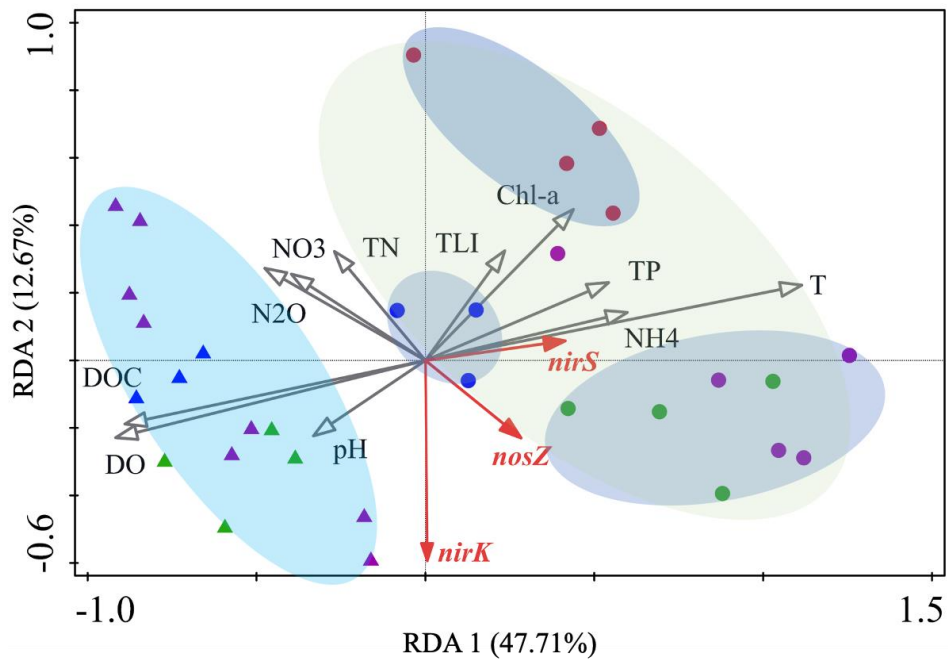


Fig. 7 Redundancy analysis (RDA) biplots of gene abundances and their relationship with environmental factors. The circle and triangle points correspond to samples taken in summer and winter, respectively. The trophic states of mesotrophic, eutrophic, middle-eutrophic, and hyper-eutrophic are represented by green, blue, violet, and red points,

respectively.

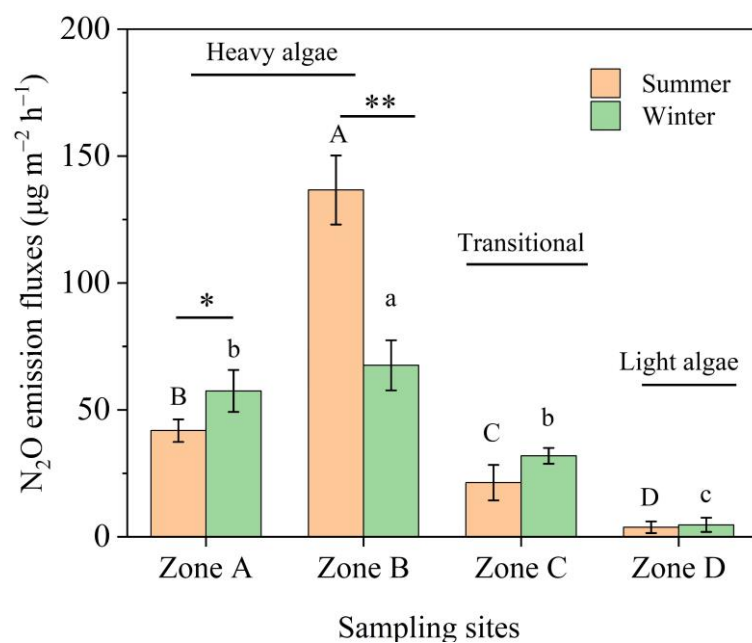


Fig. 8. Seasonal N<sub>2</sub>O emission fluxes in heavy algae-accumulated (Zones A and B), transitional (Zone C), and light algae-accumulated (Zone D) zones in Lake Taihu. Significant differences were performed at the \* $p < 0.05$  and \*\* $p < 0.01$  levels.

## Supporting information of material and methods

### S1. Calculation of the gas transfer coefficient ( $k$ )

$k$  is the gas transfer coefficient ( $\text{m d}^{-1}$ ) and was normalized to a Schmidt number of 600, which is given as Eq. (4):

$$k = k_{600} \times \left( \frac{S_c}{S_{c600}} \right)^{-n}, \quad (5)$$

where  $S_c$  is the Schmidt number of a given gas at a given temperature and water density;  $S_{c600}$  is the Schmidt number 600 at a temperature of 20°C;  $n$  is the Schmidt coefficient, which is either 2/3 for wind speeds  $< 3.6 \text{ m s}^{-1}$  or 1/2 for wind speeds  $> 3.6 \text{ m s}^{-1}$ ;  $k_{600}$  is the gas transfer coefficient adjusted to Schmidt number 600, which can be calculated by the wind speed (Cole and Caraco 1998):

$$k_{600} = 2.07 + 0.215 \times U_{10}^{1.7}, \quad (6)$$

where  $U_{10}$  is the wind speed at a 10-m height. The wind speed data ( $U_{10}$ ) were obtained from nearby weather stations, which are available at the China Meteo

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Table S1. Summary of the parameters of the overlying water and net N<sub>2</sub>O emission fluxes in the different seasons. Data in parentheses indicate the mean values of each parameter.

Season	Eutrophic state <i>TLI</i>	TN (mg L <sup>-1</sup> )	NO <sub>3</sub> <sup>-</sup> -N (mg L <sup>-1</sup> )	NH <sub>4</sub> <sup>+</sup> -N (mg L <sup>-1</sup> )	TP (mg L <sup>-1</sup> )	DOC (mg L <sup>-1</sup> )	chl- <i>a</i> (μg L <sup>-1</sup> )	N <sub>2</sub> O emission fluxes (μg m <sup>-2</sup> h <sup>-1</sup> )
Summer	Mesotrophic (≤50)	0.5-0.6 (0.6)	0.04-0.26 (0.1)	0.32-0.35 (0.3)	0.04-0.17 (0.1)	5.3-6.0 (5.6)	5.0-8.5 (7.1)	-1.0-0.3 (-0.6)
	Eutrophic (50-60)	1.0-1.4 (1.2)	0.10-0.52 (0.2)	0.35-0.86 (0.5)	0.04-0.43 (0.2)	4.2-9.4 (6.5)	11.6-29.7 (15.6)	4.0-9.5 (7.0)
	Middle-eutrophic (60-70)	1.1-2.4 (1.4)	0.39-1.18 (0.5)	0.44-0.70 (0.6)	0.08-0.46 (0.2)	3.4-7.1 (5.0)	27.4-44.0 (41.9)	1.6-15.8 (7.4)
	Hyper-eutrophic (≥70)	2.7-4.3 (3.4)	0.92-2.4 (1.7)	0.71-1.34 (1.1)	0.31-0.39 (0.3)	5.1-7.9 (6.4)	62.2-113.9 (92.7)	25.7-53.0 (39.1)
Winter	Mesotrophic (≤50)	0.8-0.9 (0.9)	0.29-0.33 (0.3)	0.14-0.21 (0.2)	0.02-0.03 (0.02)	12.8-14.0 (13.5)	5.6-7.2 (6.4)	11.7-18.6 (14.3)
	Eutrophic (50-60)	0.6-0.9 (0.7)	0.32-0.72 (0.5)	0.08-0.16 (0.1)	0.05-0.15 (0.09)	6.3-13.4 (8.5)	10.4-20.4 (14.4)	0.4-5.4 (2.6)
	Middle-eutrophic (60-70)	2.0-4.6 (3.2)	0.44-2.62 (1.7)	0.12-1.44 (0.6)	0.07-0.18 (0.13)	8.4-14.6 (10.6)	19.5-36.7 (27.4)	7.1-102.9 (50.4)

Table S2. Average constituent values of the overlying water in the different zones in Lake Taihu. The abbreviations Hy, Mi, and Me represent the hypereutrophic, middle-eutrophic, and mesotrophic states, respectively.

Sampling site	Classification	Season	<i>TLI</i>	Trophic state	TN (mg L <sup>-1</sup> )	NO <sub>3</sub> <sup>-</sup> -N (mg L <sup>-1</sup> )	NH <sub>4</sub> <sup>+</sup> -N (mg L <sup>-1</sup> )	TP (mg L <sup>-1</sup> )	DOC (mg L <sup>-1</sup> )	chl- <i>a</i> (μg L <sup>-1</sup> )	DO (mg L <sup>-1</sup> )	pH	T (°C)
Zone A	Heavy algae-accumulated zone	Summer	77.72	Hy	4.2 ± 0.2	0.13 ± 0.04	4.20 ± 0.12	0.47 ± 0.02	11.2 ± 1.3	120.0 ± 23.2	1.6 ± 0.2	7.46 ± 0.18	31.2
		Winter	69.69	Mi	3.4 ± 0.2	0.05 ± 0.01	1.00 ± 0.03	0.13 ± 0.01	13.8 ± 0.7	43.9 ± 11.7	7.1 ± 0.2	7.81 ± 0.11	11.1
Zone B	Heavy algae-accumulated zone	Summer	73.39	Hy	4.9 ± 0.8	0.36 ± 0.04	0.13 ± 0.01	0.48 ± 0.13	7.5 ± 1.4	51.8 ± 13.5	6.5 ± 0.5	8.38 ± 0.08	32.1
		Winter	64.51	Mi	3.1 ± 0.2	1.88 ± 0.13	0.52 ± 0.03	0.14 ± 0.03	12.2 ± 1.0	21.5 ± 3.7	8.5 ± 0.5	8.10 ± 0.05	9.8
Zone C	Transitional zone (Lake center)	Summer	70.65	Hy	3.7 ± 0.7	0.04 ± 0.01	0.02 ± 0.01	0.13 ± 0.08	5.6 ± 0.7	45.9 ± 11.6	7.3 ± 0.2	8.65 ± 0.11	30.2
		Winter	60.23	Mi	2.4 ± 0.1	1.95 ± 0.35	0.48 ± 0.09	0.21 ± 0.02	10.5 ± 0.6	14.6 ± 3.5	11.0 ± 0.3	8.25 ± 0.21	9.7
Zone D	Light algae-accumulated zone	Summer	41.08	Me	0.4 ± 0.2	0.04 ± 0.01	0.01 ± 0.01	0.13 ± 0.10	5.8 ± 1.2	4.9 ± 4.2	10.4 ± 0.4	8.62 ± 0.08	31.5
		Winter	46.34	Me	0.7 ± 0.2	0.29 ± 0.20	0.07 ± 0.11	0.03 ± 0.03	12.1 ± 1.3	6.3 ± 4.4	11.5 ± 0.5	8.53 ± 0.12	10.9

Table S3 Primer sequences used for sequencing and qPCR analysis

Primer	Target	Sequence (5'- 3')	Annealing temperature (°C)	Amplification size (bp)	References
16SrRNA-F	16S rRNA	CGGTGAATACGTTTCYCGG	An initial denaturation at 95°C for 5min, followed by 40 cycles of 95°C for 30s, 55°C 30s and 72°C 30s.	142	He et al., 2017
16SrRNA-R		GGHTACCTTGTTACGACTT			
cd3aF	<i>nirS</i>	GTSAACGTSAAGGARACSGG	An initial denaturation at 95°C for 5min, followed by 40 cycles of 95°C for 30s, 55°C 30s and 72°C 30s.	426	Zhang et al., 2018
R3cd		GASTTCGGRTGSGTCTTGA			
nirKFlaCu	<i>nirK</i>	ATCATGGTCTGCCGCG	An initial denaturation at 95°C for 5min, followed by 40 cycles of 95°C for 30s, 55 °C 30s and 72°C 1min	473	Henry et al., 2006
nirKR3Cu		GCCTCGATCAGRTTGTTGGTT			
nosZ2F	<i>nosZ</i>	CGCRACGGCAASAAGGTSMSSGT	An initial denaturation at 95°C for 5min, followed by 40 cycles of 95°C for 30s, 60°C 30s and 72°C 1min.	276	Henry et al., 2006
nosZ2R		CAKRTGCAKSGCRTGGCAGAA			

[References]

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- Zhang, H., Zhao, Z., Chen, S., Kang, P., Wang, Y., Feng, J., Jia, Y., Yan, M., Wang, Y., Xu, L., 218. *Paracoccus versutus* KS293 adaptation to aerobic and anaerobic denitrification: insights from nitrogen removal, functional gene abundance, and proteomic profiling analysis. *Bioresource. Technology* 260, 321-328.

Table S4 Pearson correlations between variables and N<sub>2</sub>O emission fluxes in summer and winter, respectively. Text in blue and black represent winter and summer, respectively. Note that N<sub>2</sub>O denotes N<sub>2</sub>O emission flux.

	N <sub>2</sub> O	N <sub>2</sub>	ΔN <sub>2</sub>	<i>TLI</i>	TN	NO <sub>3</sub> <sup>-</sup>	NH <sub>4</sub> <sup>+</sup>	TP	DOC	chl- <i>a</i>	C:N	DO	pH	T
N <sub>2</sub> O	1	-0.100	-0.511*	0.740**	0.901**	0.897**	0.545*	0.625*	-0.037	0.578*	-0.675**	-0.184	-0.396	-0.478
<i>TLI</i>	0.848**	0.044	-0.354	1	0.944**	0.859**	0.705**	0.737**	-0.022	0.942*	-0.901**	-0.221	-0.434	-0.535*
TN	0.956**	-0.190	-0.527*	0.915**	1	0.914*	0.733**	0.697**	-0.005	0.833**	-0.857**	-0.280	-0.507	-0.552*
NO <sub>3</sub> <sup>-</sup>	0.915**	-0.223	-0.487*	0.847**	0.967**	1	0.647**	0.811*	-0.051	0.801**	-0.757	-0.220	-0.389	-0.373
NH <sub>4</sub> <sup>+</sup>	0.719**	-0.141	-0.283	0.742**	0.786**	0.710**	1	0.483	0.208	0.730**	-0.555*	-0.593*	-0.652**	-0.251
TP	0.708**	-0.081	-0.330	0.548*	0.587*	0.587*	0.485*	0.553*1	-0.278	0.712**	-0.735**	0.165	-0.001	-0.128
DOC	0.266	-0.238	-0.164	0.178	0.236	0.236	0.066	0.219	0.3371	0.014	0.391	-0.660**	-0.722**	-0.368
Chl- <i>a</i>	0.865**	-0.178	-0.503*	0.939**	0.941**	0.908**	0.810**	0.533*	0.143	1	-0.800**	-0.257	-0.401	-0.359
C:N	-0.707**	-0.232	0.146	-0.870**	-0.770**	-0.761**	-0.630**	-0.425	0.195	-0.761**	1	-0.006	0.144	0.330
DO	-0.515*	-0.011	-0.174	-0.446	-0.502*	-0.537*	-0.662*	-0.406	-0.101	-0.535*	0.382	1	0.908**	0.287
pH	-0.109	-0.017	-0.281	-0.064	-0.175	-0.307	-0.278	0.184	0.248	-0.214	0.152	0.691**	1	0.595*
T	0.322	-0.711**	-0.734**	0.071	0.311	0.253	0.115	0.229	0.274	0.248	0.128	0.166	0.168	1

Significant correlation at the \* $p < 0.05$  and \*\* $p < 0.01$  levels.

Table S5 Pearson correlations between variables and N<sub>2</sub>O emission fluxes in two seasons. Note that N<sub>2</sub>O denotes N<sub>2</sub>O emission flux.

	N <sub>2</sub> O	<i>TLI</i>	TN	NO <sub>3</sub> <sup>-</sup>	NH <sub>4</sub> <sup>+</sup>	TP	DOC	chl- <i>a</i>	C:N	DO	pH	T
N <sub>2</sub> O	1											
<i>TLI</i>	0.627**	1										
TN	0.890**	0.865**	1									
NO <sub>3</sub> <sup>-</sup>	0.888**	0.757**	0.931**	1								
NH <sub>4</sub> <sup>+</sup>	0.375*	0.687**	0.611**	0.483**	1							
TP	0.252	0.525**	0.362**	0.278	0.527**	1						
DOC	0.246	-0.090	0.143	0.170	-0.193	-0.357*	1					
Chl- <i>a</i>	0.320	0.831**	0.607**	0.528**	0.694**	0.599**	-0.294	1				
C:N	-0.485**	-0.810**	-0.691**	-0.570**	-0.625**	-0.511**	0.498**	-0.606**	1			
DO	0.026	-0.350	-0.163	-0.075	-0.692**	-0.511**	0.356*	-0.586**	0.349	1		
pH	-0.159	-0.287	-0.313	-0.287	-0.597**	-0.087	0.064	-0.419*	0.255	0.52**	1	
T	-0.356*	0.091	-0.196	-0.284	0.361*	0.49**	-0.765**	0.410*	-0.312	-0.667**	-0.247	1

Significant correlation at the \* $p < 0.05$  and \*\* $p < 0.01$  levels.



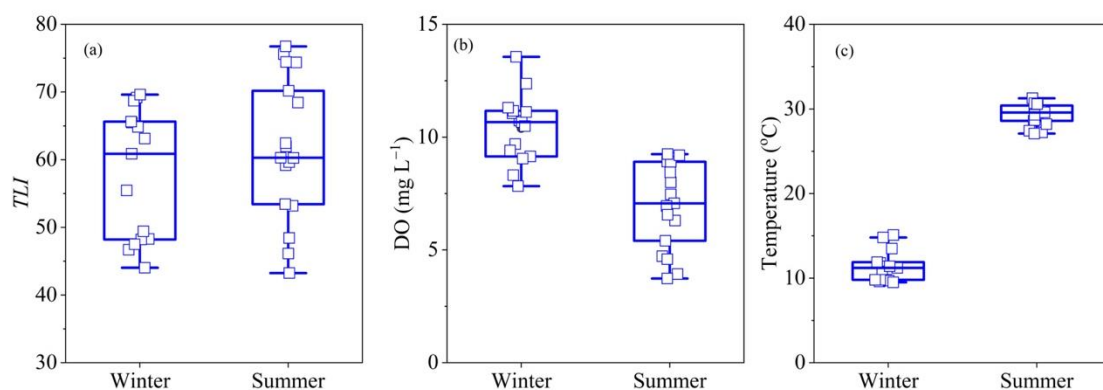


Fig. S1. (a) *TLI*, (b) *DO*, and (c) temperature of examined lakes in winter (2017) and summer (2018).

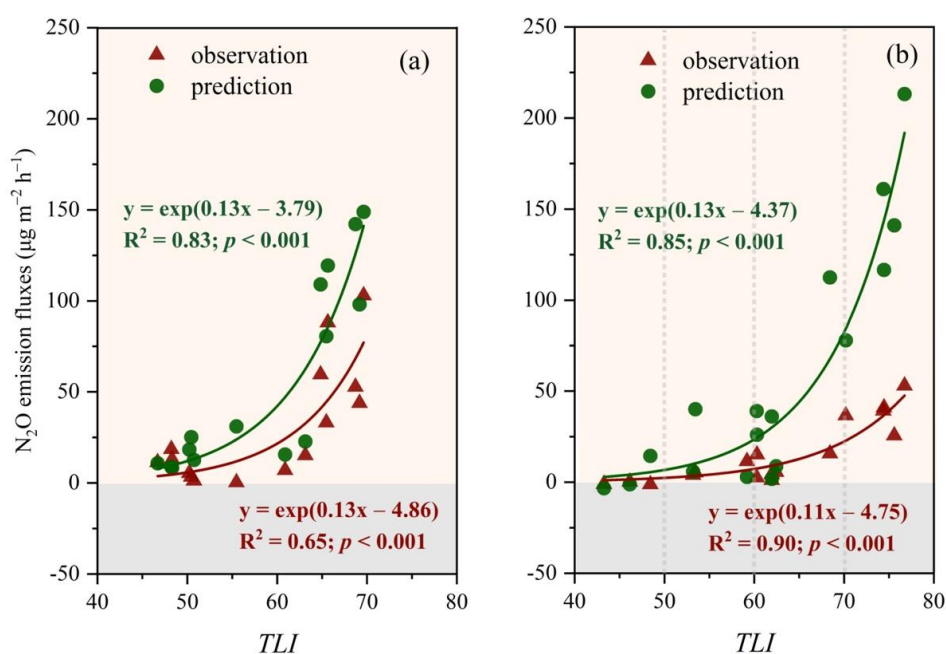


Fig. S2. Observation (red triangle) and prediction (green circle) of N<sub>2</sub>O emission fluxes in (a) winter (2017) and (b) summer (2018). Note that the *TLI* ranges for mesotrophic, eutrophic, middle-eutrophic, and hypereutrophic states are 40–50, 50–60, 60–70, and > 70, respectively.

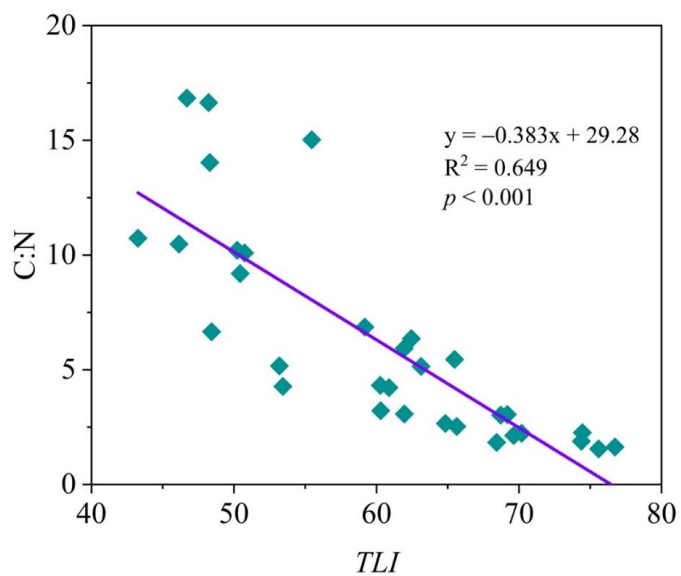


Fig. S3. The relationship between  $TLI$  and ratio of  $C:N$  in shallow lakes in winter (2017) and summer (2018).

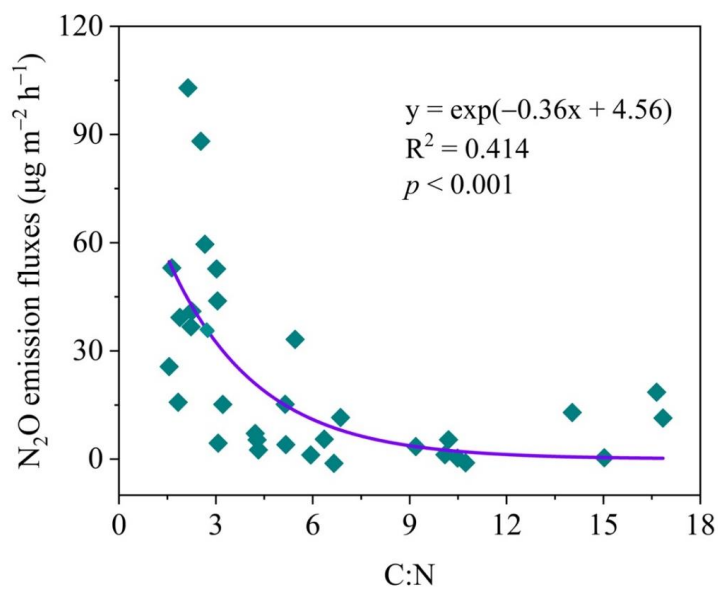


Fig. S4. The relationship between the ratio of  $C:N$  and  $N_2O$  emission fluxes in shallow lakes in winter (2017) and summer (2018).

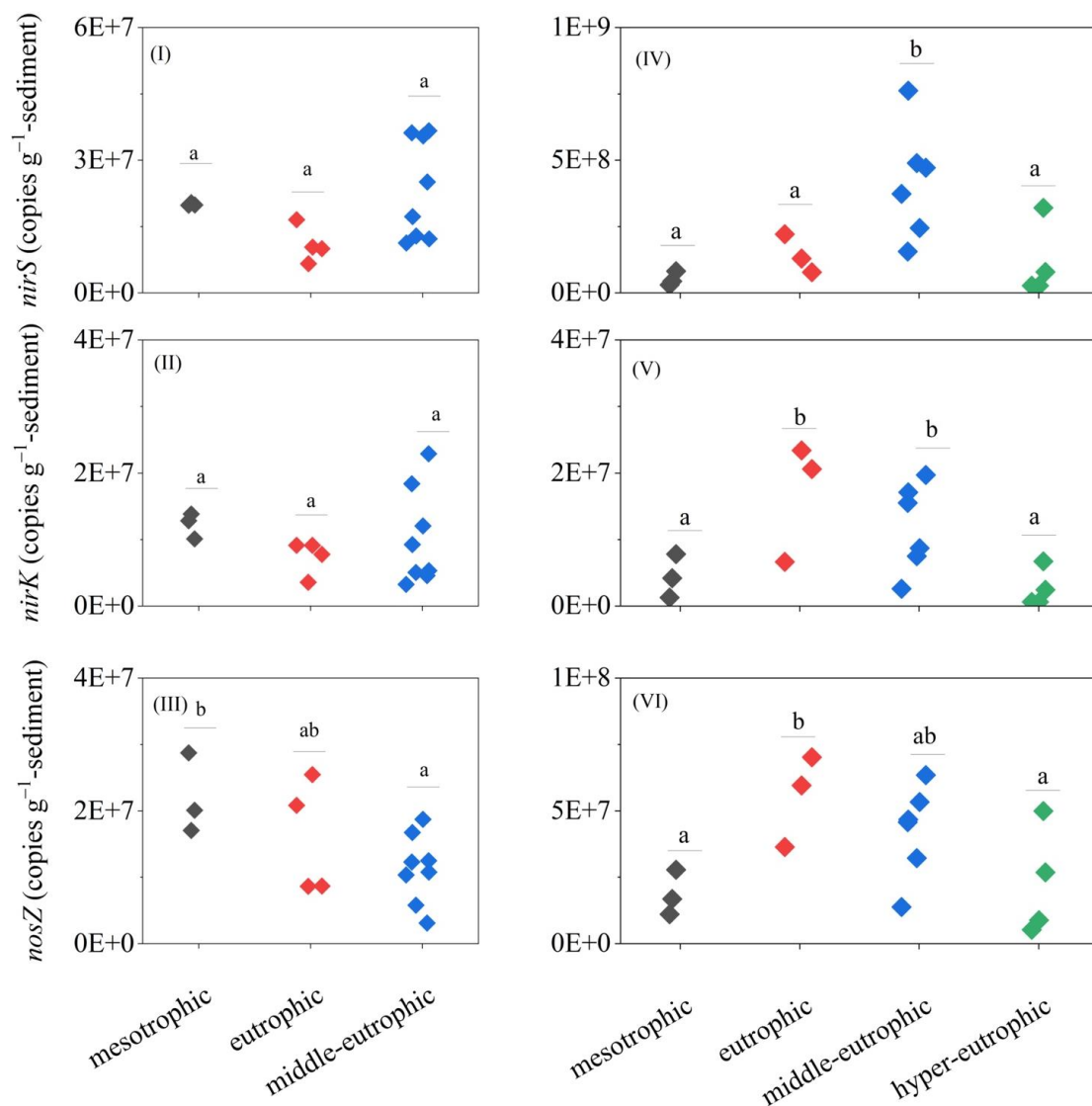


Fig. S5 Seasonal distribution (I–III: winter; IV–VI: summer) of denitrifying gene abundances, including N<sub>2</sub>O producers (*nirS* and *nirK* genes) and N<sub>2</sub>O reducers (*nosZ*), in different trophic levels. Lowercase letters indicate significant differences. The statistical analyses were performed using the Duncan test at the 5% level.

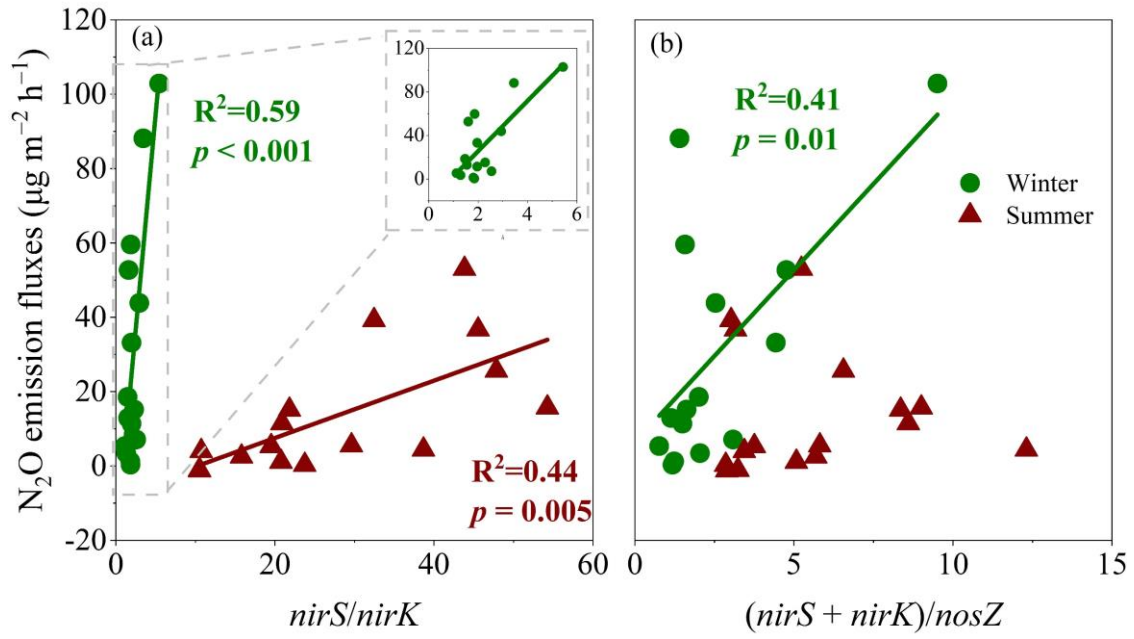


Fig. S6 Relationship between  $N_2O$  emission fluxes and (a)  $nirS/nirK$  as well as (b)  $(nirK + nirS)/nosZ$ .

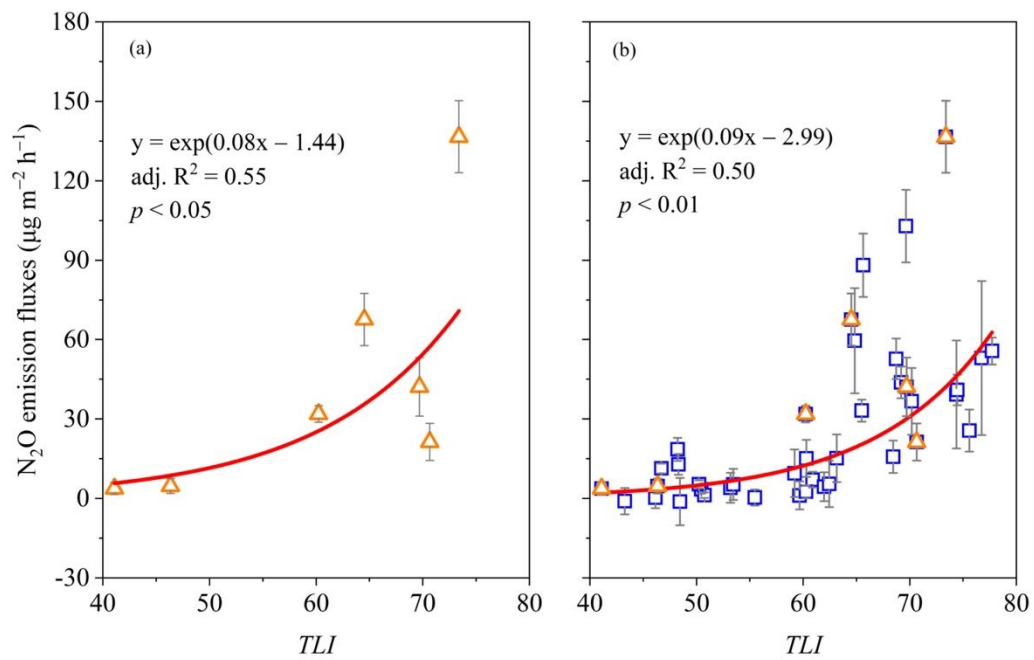


Fig. S7 Relationship between the  $N_2O$  emission fluxes and  $TLI$  in (a) Lake Taihu and (b) all shallow lakes. The triangles and squares correspond to Lake Taihu and other lakes, respectively.

## ***Chapter 5: Effects of temperature and oxygen on N<sub>2</sub>O respiration activities of heterotrophic N<sub>2</sub>O-reducing bacteria***

### **5.1 Introduction**

Nitrous oxide (N<sub>2</sub>O) emission into the atmosphere is one cause of ozone depletion and climate change (IPCC, 2014). The atmospheric N<sub>2</sub>O concentration has increased by 20%, from 270 ppb in 1750 to 331 ppb in 2018, and this increasing trend has continued by 0.2% per year (Tian et al., 2020). Wastewater treatment plants (WWTPs), which receive and remove dissolved nitrogen constituents, are a source of N<sub>2</sub>O emissions (Peng et al., 2014). The amount of N<sub>2</sub>O emitted from WWTPs was as high as 108 Mt carbon dioxide (CO<sub>2</sub>) equivalents in 2010, accounting for 3.4% of the estimated annual global N<sub>2</sub>O emissions (IPCC, 2014). A recent report indicated that N<sub>2</sub>O emissions from WWTPs have increased by 44% from 1990 to 2014 (US EPA, 2016). It is therefore essential to reduce the N<sub>2</sub>O-footprint of WWTPs by developing a strategy to mitigate N<sub>2</sub>O emissions.

N<sub>2</sub>O is a byproduct of nitrification and an intermediate of heterotrophic denitrification, which act as an N<sub>2</sub>O source and sink, respectively (Jones et al., 2013; Read-Daily et al., 2016). There are multiple pathways of N<sub>2</sub>O production, including biological and abiotic processes (Wunderlin et al., 2012; Terada et al., 2017). N<sub>2</sub>O reduction to N<sub>2</sub> gas is mediated by N<sub>2</sub>O reductase (N<sub>2</sub>OR), harbored by N<sub>2</sub>O-reducing bacteria (N<sub>2</sub>ORB) (Zumft and Kroneck, 2006). This pathway represents a pivotal role in wastewater treatment to mitigate N<sub>2</sub>O emissions (Kim et al., 2020), strongly dependent on the intrinsic N<sub>2</sub>O reduction capacity of N<sub>2</sub>ORB. Therefore, understanding the physiologies of N<sub>2</sub>ORB and developing N<sub>2</sub>O mitigation strategies has attracted more attention in recent years (Yoon et al., 2016; Conthe et al., 2018a; Hallin et al., 2018; Suenaga et al., 2019).

Owing to differences in the N<sub>2</sub>OR amino acid sequences (similarity <50%) and the protein-translocation pathways, N<sub>2</sub>ORB harboring a functional *nosZ* gene can be

classified into either clade I or clade II (Jones et al., 2013; Sanford et al., 2012; Yoon et al., 2016). Whereas some clade II type N<sub>2</sub>ORB display a higher affinity for N<sub>2</sub>O than clade I type N<sub>2</sub>ORB (Yoon et al., 2016; Suenaga et al., 2018a), research on N<sub>2</sub>ORB biokinetics of N<sub>2</sub>O consumption toward an effective strategy to reduce N<sub>2</sub>O emissions is limited (Yoon et al., 2016; Suenaga et al., 2018a; Yoon et al., 2019). Therefore, more knowledge of N<sub>2</sub>ORB physiologies concerning N<sub>2</sub>O reduction is required.

Oxygen level is a crucial factor to determine the rates of N<sub>2</sub>O production and consumption (Kinh et al., 2017; Massara et al., 2017). Under an anoxic environment, heterotrophic N<sub>2</sub>ORB switch their dominant metabolism from O<sub>2</sub>- to N<sub>2</sub>O-dependent respiration (Miyahara et al., 2010; Suenaga et al., 2018a), but favor O<sub>2</sub> respiration under aerobic conditions (Qu et al., 2016). The transition from anoxic to oxic conditions triggers N<sub>2</sub>O emissions due to the inhibition of N<sub>2</sub>OR in the presence of oxygen (Read-Daily et al., 2016); however, returning from oxic to anoxic conditions reactivates N<sub>2</sub>OR. N<sub>2</sub>OR activity recovery and oxygen tolerance levels likely differ among N<sub>2</sub>ORB. Previous work demonstrated that the betaproteobacterium *Azospira* sp. strain I13 displays a rapid recovery of N<sub>2</sub>O reduction activity and is tolerant of O<sub>2</sub> inhibition (Suenaga et al., 2018a). A mechanistic enzymatic model has been proposed to interpret N<sub>2</sub>OR activity after O<sub>2</sub> exposure (Zheng and Doskey, 2015). Nevertheless, the mechanism of the recovery of N<sub>2</sub>O reduction activity after O<sub>2</sub> exposure has yet to be systematically studied.

Seasonal temperature variations change the microbial activities in WWTPs (Poh et al., 2015; Massara et al., 2017; Li et al., 2018), potentially associated with N<sub>2</sub>O production and emission (Abdalla et al., 2009; Vasilaki et al., 2019). Lower temperatures retard N<sub>2</sub>OR activities, causing higher N<sub>2</sub>O emissions via denitrification from 20°C to 5°C (Adouani et al., 2015). Therefore, the temperature effect on N<sub>2</sub>O consumption by N<sub>2</sub>ORB is essential but has not yet been investigated. Particularly, whether or not the temperature dependence differs with *nosZ* clade types or bacterial species remains unclear.

To expand our understanding of N<sub>2</sub>ORB biokinetics at various temperatures, a pure culture-based investigation will allow us to elucidate N<sub>2</sub>ORB physiologies without the interference of other bacteria. Systematic analysis of pure culture biokinetics after an O<sub>2</sub> exposure event at various temperatures will demonstrate the potential of N<sub>2</sub>ORB as an N<sub>2</sub>O sink with changing environmental conditions. The objectives of this study are to (1) investigate the characteristics of O<sub>2</sub> and N<sub>2</sub>O consumption kinetics at different temperatures; (2) determine O<sub>2</sub> inhibition constants at different temperatures; and (3) elucidate the recovery of N<sub>2</sub>O reduction activity after O<sub>2</sub> exposure at different temperatures. Microrespirometric analysis of N<sub>2</sub>O dynamics by a microrespiration system and mathematical modeling are employed in this study. The outcomes will indicate how N<sub>2</sub>ORB exert an N<sub>2</sub>O sink, leading to the development of a strategy to mitigate N<sub>2</sub>O emissions.

## **5.2 Materials and methods**

### **5.2.1 Targeted N<sub>2</sub>ORB and culture conditions**

Four different heterotrophic bacteria harboring either clade I or II type *nosZ* (listed in Table S1) were subjected to biokinetic investigations. The two isolates affiliated to the genus *Azospira* are denitrifiers possessing clade II type *nosZ*, whereas those of genera *Pseudomonas* and *Paracoccus* are canonical clade I type *nosZ* denitrifying bacteria. These *Azospira* isolates, previously acquired from a bioreactor, were chosen as they have a proven high affinity for N<sub>2</sub>O (Suenaga et al., 2019). Contrarily, *Ps. stutzeri* and *Pa. denitrificans* were chosen as they are often used as model denitrifying bacteria (Jiang et al., 2020; Kim et al., 2020) and also are detected in WWTPs (Wang et al., 2014a; Huang et al., 2020; Han et al., 2020). All strains harbor a full set of denitrifying functional genes, according to their genomes on NCBI ([www.ncbi.nlm.nih.gov/](http://www.ncbi.nlm.nih.gov/); Details can be found in Table S1). These strains, stored in a deep freezer (−80°C), were aerobically pre-grown in the nutrient medium (pH 7.0) consisting of 5 g L<sup>−1</sup> NaCl, 5 g L<sup>−1</sup> Bacto Peptone, and 3 g L<sup>−1</sup> Oxoid™ Lab-Lemco

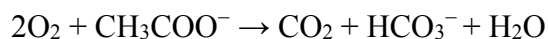
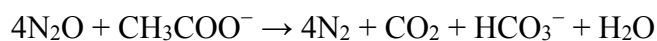
meat extract, at 30°C with a shaking speed 150 rpm. This medium favors the resuscitation of stored bacteria for subsequent biokinetic experiments and avoids bacterial cell aggregation (Suenaga et al., 2018a and 2019). The incubation time was preliminarily optimized by the growth curves of the tested bacteria, as shown in Fig. S1. After reaching the respective exponential growth and early stationary phases, the pre-cultured bacterial suspension was collected and washed twice with  $0.05 \times$  PBS by centrifugation at 5,000 rpm for 5 min and then resuspended in a medium composed of  $100 \text{ mg L}^{-1} \text{ KH}_2\text{PO}_4$ ,  $6.6 \text{ mg L}^{-1} \text{ NaCl}$ ,  $8.20 \text{ mg L}^{-1} \text{ MgSO}_4 \cdot 7\text{H}_2\text{O}$ ,  $13.4 \text{ mg L}^{-1} \text{ KCl}$ ,  $115 \text{ mg L}^{-1} \text{ NH}_4\text{Cl}$ ,  $188 \text{ mg L}^{-1} \text{ NaHCO}_3$ , and  $1 \text{ ml L}^{-1}$  trace element solution (Suenaga et al., 2018a). The  $\text{OD}_{600}$  value of the cell suspension was measured by a UV-vis spectrophotometer to confirm if the cell density was in an appropriate range. Prior to the experiment summarized in the next section, the pH of the experimental medium was adjusted to 7.5 with  $1 \text{ mol L}^{-1} \text{ HCl}$ . All the media were sterilized before use.

### 5.2.2 N<sub>2</sub>O respiration experiment

Whole-cell O<sub>2</sub> and N<sub>2</sub>O consumption kinetics of the tested bacteria were estimated with a laboratory-scale microrespiration system equipped with O<sub>2</sub> and N<sub>2</sub>O microsensors (Unisense, Aarhus, Denmark). The system, consisting of Clark-type N<sub>2</sub>O and O<sub>2</sub> microsensors, a double-port chamber (10 ml), and stirrers installed in a sensor stand (Fig. S2), allowed the online monitoring of dissolved O<sub>2</sub> and N<sub>2</sub>O concentrations (Suenaga et al., 2018a). The chamber was filled with the bacterial cell suspension and then injected with approximately 20  $\mu\text{L}$  of a concentrated N<sub>2</sub>O solution ( $25 \text{ mmol L}^{-1}$ ) and 25  $\mu\text{L}$  acetate ( $250 \text{ mmol L}^{-1}$ ) as the sole carbon source in the experiment (Suenaga et al., 2018a). The excess acetate supply allowed N<sub>2</sub>O to be the sole limiting substrate. Acetate was used because it cannot be fermented, excluding the possibility of fermentation. Prior to the experiment, the  $\text{OD}_{600}$  values of cell suspension were adjusted in the chamber to ensure the same cell density was used at each temperature. The N<sub>2</sub>O and acetate addition ensured the working concentrations of  $50 \mu\text{mol L}^{-1}$  and  $625 \mu\text{mol L}^{-1}$ . For all experiments, oxygen was not intentionally scavenged but present in the



experimental medium. The acetate concentration was in excess for the following stoichiometric relations:



This condition ensures preferential  $\text{O}_2$  consumption, followed by  $\text{N}_2\text{O}$  consumption by each assigned  $\text{N}_2\text{ORB}$  (an example shown in Fig. S3). The addition of the concentrated  $\text{N}_2\text{O}$  solution was repeated until a comparable profile of a time-lapse  $\text{N}_2\text{O}$  concentration was attained (Fig. S3). To achieve a completely mixed condition in the chamber, a glass-coated stirrer bar was stirred at 600 rpm for 5 min immediately after the injection of both  $\text{N}_2\text{O}$  and acetate, and then the rotation speed of the stirrer bar was adjusted to 400 rpm for the duration of each experiment. The  $\text{N}_2\text{O}$  solution was spiked to adjust the  $\text{N}_2\text{O}$  concentration to approximately  $50 \mu\text{mol L}^{-1}$  after the  $\text{N}_2\text{O}$  was depleted in the chamber. Subsequently, each bacterial suspension was applied for batch assays in triplicate. The incubation time was 6–8 h for the two *Azospira* strains and *Ps. stutzeri*, and 12–16 h for *Pa. denitrificans*. After each experiment, the cell suspension (total volume of 1 ml) was immediately taken from the chamber and fixed with a 2% glutaraldehyde solution. The suspension was filtered, and the captured cells on the filter were enumerated under microscopy (BZ-8100, Keyence, Osaka, Japan) as described elsewhere (Lunau et al., 2005). More than 500 cells (20–25 view fields), higher than the recommended cell numbers (400 cells) by Lanau et al. (2005), were counted. The batch experiments for each bacterium were conducted at 15°C, 25°C, 30°C, and 35°C in a water bath to evaluate the temperature effects on  $\text{N}_2\text{O}$  consumption dynamics.

### 5.2.3 Analysis of $\text{N}_2\text{O}$ respiration by $\text{N}_2\text{ORB}$

By using the microrespiration system, changes in  $\text{O}_2$  and  $\text{N}_2\text{O}$  concentrations over time, as shown in Fig. S3, are expressed as Eq. (1). The maximum cell-specific  $\text{O}_2$  and  $\text{N}_2\text{O}$  consumption rates ( $V_{\text{max, O}_2}$  and  $V_{\text{max, N}_2\text{O}}$ ) and the whole-cell half-saturation constants of  $\text{O}_2$  and  $\text{N}_2\text{O}$  ( $K_{\text{m, O}_2}$  and  $K_{\text{m, N}_2\text{O}}$ ) were determined by the equation of the whole-cell Michaelis-Menten kinetics in Eq. (2) and shown in Fig. S4:

$$V = \frac{dS}{dt} = \frac{S_t - S_{t+\Delta t}}{\Delta t} \quad (1)$$

$$V = \frac{V_{\max} \times S}{K_m + S} \times X \quad (2)$$

where  $S_t$  and  $S_{t+\Delta t}$  are the O<sub>2</sub> or N<sub>2</sub>O concentrations (μmol L<sup>-1</sup>) at the time points  $t$  and  $t+\Delta t$  ( $\Delta t$  represents the time interval between two points;  $\Delta t = 1.39\text{--}2.78 \times 10^{-3}$  h), respectively (Suenaga et al., 2018a).  $V_{\max}$  and  $K_m$  are the cell-specific maximum consumption rates [μmol (cell h)<sup>-1</sup>] and whole-cell half-saturation constants (μmol L<sup>-1</sup>), respectively.  $X$  is the bacterial cell density (cell L<sup>-1</sup>). For the comparison of the O<sub>2</sub> and N<sub>2</sub>O respiration activities among N<sub>2</sub>ORB, the specific affinities for O<sub>2</sub> and N<sub>2</sub>O ( $a_0$  [L (cell h)<sup>-1</sup>]) were independently calculated according to Eq. (3) (Kits et al., 2017; Suenaga et al., 2018a):

$$a_0 = \frac{V_{\max}}{K_m} \quad (3)$$

The activation energies for O<sub>2</sub> and N<sub>2</sub>O were calculated by the Arrhenius equation described in Eq. (4) (Holtan-Hartwig et al., 2002):

$$\ln V_{\max} = \left( -\frac{E_a}{R} \right) \left( \frac{1}{T} \right) + \ln A \quad (4)$$

where  $E_a$  is the activation energy (J mol<sup>-1</sup>);  $R$  is the gas constant of 8.314 J (mol K)<sup>-1</sup>; and  $A$  is the pre-exponential factor. An Arrhenius plot was drawn to calculate  $E_a$  for the temperature dependence on O<sub>2</sub> and N<sub>2</sub>O respiration rates, as previously reported (Rahman et al., 2019). The temperature coefficient  $Q_{10}$  for O<sub>2</sub> and N<sub>2</sub>O, defined as the factor by which the maximum respiration rate increases with a temperature increase in 10°C, was calculated with Eq. (5):

$$Q_{10} = \left( \frac{k_2}{k_1} \right)^{\frac{10}{(T_2 - T_1)}} \quad (5)$$

where  $k_2$  and  $k_1$  are the rate constants of O<sub>2</sub> or N<sub>2</sub>O consumption,  $T_2$  and  $T_1$ , respectively.

The effect of O<sub>2</sub> concentration on N<sub>2</sub>O consumption rate was parameterized according to previous research (Ni et al., 2011). The non-competitive inhibition heterotrophic denitrification model was applied to estimate O<sub>2</sub> inhibition constants [ $K_i$  (μmol L<sup>-1</sup>)] and incorporated into Eq. (6) (von Schulthess et al., 1994):

$$\frac{dS}{dt} = \frac{V_{\max, N_2O} \times S_{N_2O}}{K_m, N_2O + S_{N_2O}} \times \frac{K_I}{K_I + S_{O_2}} \quad (6)$$

$K_I$  was determined by minimizing the coefficient of determination as follows:

$$R^2 = 1 - \frac{\sum (S_{\exp} - S_{\text{model}})^2}{\sum (S_{\exp} - \overline{S_{\exp}})^2} \quad (7)$$

where  $S_{\exp}$  and  $S_{\text{model}}$  are the experimental and simulated  $N_2O$  concentrations, respectively; and  $\overline{S_{\exp}}$  is the average value.

#### 5.2.4 Analysis of the activity recovery of $N_2O$ respiration by $N_2ORB$

To quantitatively describe the activity recovery of  $N_2O$  reduction, a recovery model was developed to simulate the recovery of  $N_2O$  consumption activity after  $O_2$  depletion ( $V_{N_2O \text{ recovery}}$ ) as in Eq. (8):

$$V_{N_2O \text{ recovery}} = \frac{V_{\max, N_2O} \times S_{N_2O}}{K_m, N_2O + S_{N_2O}} \times v_{N_2O} \quad (8)$$

The relative activity of  $N_2O$  consumption rates ( $v_{N_2O}$ ) was defined by dividing an  $N_2O$  consumption rate by that under a completely anaerobic condition (*i.e.*,  $v_{N_2O} = V_{N_2O}/V_{\max, N_2O}$ ) (Suenaga et al., 2018a).  $v_{N_2O}$  can be broken down as Eq. (9):

$$v_{N_2O} = k' \times t + m \quad (9)$$

where  $k'$  represents the recovery rate constant ( $h^{-1}$ ), and  $m$  represents the initial  $v_{N_2O}$  when  $O_2$  was depleted.

#### 5.2.5 Data analysis

$O_2$  and  $N_2O$  concentration data were smoothed to reduce electrical noises by the SigmaPlot 14.0 software (Systat, CA) as in a previous report (Martens-Habbena et al., 2009). The  $K_I$  and  $k'$  values were estimated by fitting the  $N_2O$  profiles to Eqs. (6) and (9) based on the least-squares method using the solver function of Microsoft Excel (Microsoft, Redmond, WA, USA). All statistical analyses were conducted using the SPSS 19.0 software (IBM Corp., Armonk, NY, USA). The Spearman's rank correlation analysis was conducted to assess the relationship between  $V_{\max, N_2O}$  and recovery rate constants ( $k'$ ). The level of significance was  $p < 0.05$  for all tests. The statistical significance of those results was determined by a Duncan test at the 5% level.

## 5.3 Results

### 5.3.1 Biokinetic parameters of N<sub>2</sub>ORB on O<sub>2</sub> respiration

O<sub>2</sub> and N<sub>2</sub>O concentrations of N<sub>2</sub>ORB harboring clade I type *nosZ* (*Ps. stutzeri* JCM5965 and *Pa. denitrificans* NBRC102528) and clade II type *nosZ* (*Azospira* sp. strains I09 and I13) were monitored with a microrespiration system. The presence of N<sub>2</sub>O did not alter the biokinetics associated with oxygen (Fig. S5). The variation in O<sub>2</sub> and N<sub>2</sub>O biokinetic parameters at different temperatures is shown in Fig. 1. The maximum O<sub>2</sub> consumption rates ( $V_{\max, O_2}$ ) increased with an increase in temperature (Fig. 1a). The  $V_{\max, O_2}$  of *Pa. denitrificans* NBRC102528 [ $2.23 \pm 0.16 \times 10^{-9}$   $\mu\text{mol (cell h)}^{-1}$ ] harboring *nosZ* clade I was lower than that of the other tested isolates at 35 °C ( $p < 0.05$ ), which were statistically comparable. At 15°C, the  $V_{\max, O_2}$  values of *Azospira* sp. strain I09 [ $7.87 \pm 0.73 \times 10^{-10}$   $\mu\text{mol (cell h)}^{-1}$ ] and *Azospira* sp. strain I13 [ $2.60 \pm 0.69 \times 10^{-10}$   $\mu\text{mol (cell h)}^{-1}$ ] were lower than that of *Ps. stutzeri* JCM5965 [ $1.64 \pm 0.29 \times 10^{-9}$   $\mu\text{mol (cell h)}^{-1}$ ] ( $p < 0.05$ ). The highest  $V_{\max, O_2}$  values of *Azospira* sp. strain I09, *Azospira* sp. strain I13, *Ps. stutzeri* JCM5965, and *Pa. denitrificans* NBRC102528 were 5.26, 19.59, 2.64, and 2.04 times higher, respectively, than the lowest values, indicating that *Azospira* sp. strain I13 had the largest  $V_{\max, O_2}$  variation with temperature. The  $K_{m, O_2}$  values of the tested N<sub>2</sub>ORB consistently exhibited a marginal change with temperature ( $p > 0.05$ ) (Fig. 1b). In light of the classification by *nosZ* clade types, the  $K_{m, O_2}$  values of *Azospira* sp. strain I13 ( $0.51 \pm 0.03$  to  $0.75 \pm 0.25$   $\mu\text{mol L}^{-1}$  at 15–35°C) were significantly lower than those of clade I type *nosZ*, i.e., *Ps. stutzeri* JCM5965 ( $1.22 \pm 0.54$  to  $1.99 \pm 0.61$   $\mu\text{mol L}^{-1}$ ) and *Pa. denitrificans* NBRC102528 ( $0.86 \pm 0.29$  to  $1.50 \pm 0.37$   $\mu\text{mol L}^{-1}$ ) at each temperature ( $p < 0.05$ ). The  $K_{m, O_2}$  values of *Azospira* sp. strain I09 were also lower than those of *Ps. stutzeri* JCM5965 and *Pa. denitrificans* NBRC102528 except for the ones at 25°C and 35°C. The highest  $a_{0, O_2}$  values were usually observed at 35°C for the tested bacteria, while the value for *Azospira* sp. strain I13 was significantly higher than those of the other tested bacteria ( $p < 0.05$ ; Fig. 2a). Given the insignificant change in  $K_{m, O_2}$  with temperature (Fig. 1b), the  $V_{\max, O_2}$

determined the  $a_{0, O_2}$  values. Except for *Azospira* sp. strain I13, the  $a_{0, O_2}$  values of these tested bacteria displayed no significant difference with temperature (Fig. 2a).

### 5.3.2 Biokinetic parameters of N<sub>2</sub>ORB on N<sub>2</sub>O respiration

The decremental trends of N<sub>2</sub>O and O<sub>2</sub> concentrations were consistent irrespective of the tested bacterial species (an example in Fig. S3). The  $V_{\max, N_2O}$  of *Pa. denitrificans* NBRC102528 was one or two orders of magnitude lower than the other tested N<sub>2</sub>ORB at each temperature (Fig. 1c). Temperature increase concomitantly increased  $V_{\max, N_2O}$  in the tested N<sub>2</sub>ORB. The highest value for  $V_{\max, N_2O}$ ,  $7.12 \pm 1.26 \times 10^{-9} \mu\text{mol (cell h)}^{-1}$ , was attained at 35°C for *Azospira* sp. strain I13, 2.82- to 47.47-fold higher than the other species at 35°C ( $p < 0.05$ ). Over the tested temperature range, the highest  $V_{\max, N_2O}$  values for *Azospira* sp. strain I09, *Azospira* sp. strain I13, *Ps. stutzeri* JCM5965, and *Pa. denitrificans* NBRC102528 were approximately 4.85, 24.52, 2.42, and 3.60 times higher, respectively, than those of the lowest values. Irrespective of the tested temperature, the  $K_{m, N_2O}$  values for *Ps. stutzeri* JCM5965 and *Pa. denitrificans* NBRC102528 were higher than those for *Azospira* sp. strains I09 and I13 ( $p < 0.05$ ; Fig. 1d). For all tested N<sub>2</sub>ORB except *Azospira* sp. strain I13, the  $K_{m, N_2O}$  values did not statistically differ over the tested temperature range ( $p > 0.05$ ). Contrarily, the  $K_{m, N_2O}$  values for *Azospira* sp. strain I13 had a temperature dependence: the highest value ( $0.89 \pm 0.09 \mu\text{mol L}^{-1}$ ) at 35°C was 1.56 times higher than the lowest value ( $0.57 \pm 0.09 \mu\text{mol L}^{-1}$ ) at 15°C. As shown in Figs. 1d and 2b, the lower  $K_{m, N_2O}$  values for *Azospira* sp. strains I09 and I13 than for *Ps. stutzeri* JCM5965 and *Pa. denitrificans* NBRC102528 result in a higher  $a_{0, N_2O}$ , especially at 30°C and 35°C ( $p < 0.05$ ). In contrast,  $a_{0, N_2O}$  of *Ps. stutzeri* JCM5965 was comparable to the  $a_{0, N_2O}$  of *Azospira* sp. strain I13 at 25°C ( $p = 0.22$ ). The  $a_{0, N_2O}$  values of *Pa. denitrificans* NBRC102528 were two to three orders of magnitude lower than the other tested N<sub>2</sub>ORB. The low activity with an unknown reason was similar to previous reports (Suenaga et al., 2018a). Regardless of the temperature range and tested bacterial species, the positive linear correlations between  $V_{\max, O_2}$  and  $V_{\max, N_2O}$  ( $R^2 > 0.8$ ;  $p < 0.05$ ) were obtained (Fig. S6).

Among the tested N<sub>2</sub>ORB, the slope of the correlation for *Azospira* sp. strain I13 (slope, 1.41) was significantly higher than for the other N<sub>2</sub>ORB, while the slopes for *Azospira* sp. strain I09 (slope, 0.59) and *Ps. stutzeri* JCM5965 (slope, 0.51) were comparable (Fig. S6). These slopes were lower than 2, that is the theoretical value assuming the same electron donor consumption rate coupled with N<sub>2</sub>O ( $\text{N}_2\text{O} + 2\text{H}^+ + 2\text{e}^- \rightarrow \text{N}_2 + \text{H}_2\text{O}$ ) and O<sub>2</sub> ( $\text{O}_2 + 4\text{H}^+ + 4\text{e}^- \rightarrow 2\text{H}_2\text{O}$ ). A slope lower than 2 indicates that oxygen is a more favorable electron acceptor than N<sub>2</sub>O for the electron donor consumption.

### 5.3.3 Activation energy and sensitivity to temperature

The activation energies of O<sub>2</sub> and N<sub>2</sub>O for the N<sub>2</sub>ORB were estimated (Fig. 3). From the linear correlations of  $\ln V_{\max}$  and  $1/T$  in Fig. 3 ( $R^2 > 0.8$ ), the highest  $E_{a, \text{O}_2}$  and  $E_{a, \text{N}_2\text{O}}$  of  $122.2 \pm 21.6$  and  $114.0 \pm 22.6$  kJ mol<sup>-1</sup> were obtained for *Azospira* sp. strain I13 (Table S2), which had the largest temperature dependence. The ranges of  $E_{a, \text{O}_2}$  and  $E_{a, \text{N}_2\text{O}}$  for the other tested N<sub>2</sub>ORB were from  $36.0 \pm 12.6$  to  $97.6 \pm 20.2$  kJ mol<sup>-1</sup> and  $38.3 \pm 5.9$  to  $60.1 \pm 13.0$  kJ mol<sup>-1</sup>, respectively. In addition, the Q<sub>10</sub> value at 25–35°C for *Azospira* sp. strain I13 (9.04) was significantly higher than for the other N<sub>2</sub>ORB (Table S2). Except for *Pa. denitrificans* NBRC102528, the relative activation energies ( $E_{a, \text{O}_2}/E_{a, \text{N}_2\text{O}}$ ) were close to 1. *Ps. stutzeri* JCM5965 and *Pa. denitrificans* NBRC102528, harboring clade I type *nosZ*, possessed lower  $E_{a, \text{N}_2\text{O}}$  than *Azospira* sp. strains I09 and I13 (Table S2). The  $E_{a, \text{O}_2}$  and  $E_{a, \text{N}_2\text{O}}$  of *Ps. stutzeri* JCM5965 were the lowest.

### 5.3.4 Assessment of the O<sub>2</sub> inhibition constants

According to the  $V_{\text{N}_2\text{O}}$  profile in the presence of O<sub>2</sub> and the fitting to Eq. (6),  $K_I$  values at different temperatures were estimated (Fig. 4). Because of the marginal N<sub>2</sub>O consumption activity of *Pa. denitrificans* NBRC102528, the  $K_I$  value could not be numerically attained [Eq. (6)]; therefore, the data for *Pa. denitrificans* NBRC102528 were excluded hereafter. The highest  $K_I$  of *Azospira* sp. strain I09 ( $0.92 \pm 0.24$  μmol L<sup>-1</sup>), *Azospira* sp. strain I13 ( $0.19 \pm 0.04$  μmol L<sup>-1</sup>), and *Ps. stutzeri* JCM5965 ( $0.84 \pm$

0.31  $\mu\text{mol L}^{-1}$ ) were consistently attained at 35°C. The  $K_I$  values of the three bacteria were statistically comparable at 15°C ( $p = 0.243$ ), while the  $K_I$  value for *Azospira* sp. strain I09 was significantly higher than for *Azospira* sp. strain I13 at the other tested temperatures ( $p < 0.05$ ). For *Azospira* sp. strain I09 and *Ps. stutzeri* JCM5965, the  $K_I$  values increased with temperature, suggesting a lower  $\text{O}_2$  susceptibility of  $\text{N}_2\text{O}$  reduction activity. In contrast,  $K_I$  was independent of temperature from 15°C to 35°C for *Azospira* sp. strain I13 ( $p = 0.106$ ).

### 5.3.5 Recovery of $\text{N}_2\text{O}$ consumption activity after $\text{O}_2$ depletion

The  $\text{N}_2\text{O}$  consumption by  $\text{N}_2\text{ORB}$  after  $\text{O}_2$  depletion was evaluated using the relative activity of  $\text{N}_2\text{O}$  consumption rates ( $v_{\text{N}_2\text{O}} = V_{\text{N}_2\text{O}}/V_{\text{max, N}_2\text{O}}$ ). As shown in Fig. S7,  $v_{\text{N}_2\text{O}}$  at 30°C linearly increased ( $p < 0.01$ ) with time, but the degree of  $v_{\text{N}_2\text{O}}$  recovery was distinct among the tested bacteria. The  $\text{N}_2\text{O}$  consumption rate after  $\text{O}_2$  exposure ( $V_{\text{N}_2\text{O recovery}}$ ) at 30°C was fitted to the developed model in Eq. (8) (Fig. 5). By fitting Eq. (8) to the experimental results (Figs. 5, S8 and S9), the recovery rate constants ( $k'$ ) at different temperatures were acquired (Fig. 6). The highest  $k'$  was attained at different temperatures: *Azospira* sp. strain I09, *Azospira* sp. strain I13, and *Ps. stutzeri* JCM5965 displayed the highest recovery at 35°C ( $0.75 \pm 0.08 \text{ h}^{-1}$ ), 30°C ( $1.36 \pm 0.21 \text{ h}^{-1}$ ), and 35°C ( $0.37 \pm 0.11 \text{ h}^{-1}$ ), respectively. For *Azospira* sp. strain I09 and *Ps. stutzeri* JCM5965, these temperatures coincide with those resulting in the highest  $\text{N}_2\text{O}$  consumption rates (Fig. 1). Among the tested temperature conditions, *Azospira* sp. strain I13 had a faster recovery after  $\text{O}_2$  exposure compared with *Azospira* sp. strain I09 and *Ps. stutzeri* JCM5965 ( $p < 0.05$ ). The Spearman's rank correlation analysis indicated medium-strong positive correlations between  $k'$  and  $V_{\text{max, N}_2\text{O}}$  in *Azospira* sp. strains I09 and I13, and *Ps. stutzeri* JCM5965 ( $p < 0.05$ ) (Table S3).

## 5.4 Discussion

### 4.4.1 Temperature effect on $\text{O}_2$ and $\text{N}_2\text{O}$ respiration biokinetics of $\text{N}_2\text{ORB}$

Harnessing  $\text{N}_2\text{ORB}$  in engineered systems for reducing  $\text{N}_2\text{O}$  emissions entails

more knowledge of their ecophysologies (Hallin et al., 2018; Suenaga et al., 2019). In addition to typical *nosZ* clade I type canonical denitrifying bacteria serving as a N<sub>2</sub>O sink (Table S1), the applicability of newly isolated *nosZ* clade II type denitrifying bacteria has demanded further investigation (Suenaga et al., 2019). This study was undertaken to comprehensively evaluate the temperature effect on the whole-cell biokinetics of N<sub>2</sub>ORB in the two distinct *nosZ* clade types, including the *nosZ* clade II type *Azospira* spp., which has been considered promising as an efficient N<sub>2</sub>O consumer in WWTPs (Suenaga et al., 2019; Kim et al., 2020). The four tested N<sub>2</sub>ORB from the two different *nosZ* types have distinct biokinetic traits at different temperatures (Fig. 1), indicating that temperature is one of the potential factors governing the abundance and composition of the N<sub>2</sub>ORB communities. Seasonal temperature variations are often observed in WWTPs, where N<sub>2</sub>O emissions are correspondingly changed (Chen et al., 2019). Similarly, the temperature dependence of bioreactor performances is also reported (Watanabe et al., 2017), with lower observed N<sub>2</sub>O consumption activity at a low temperature (Wang et al., 2018). Although the temperature effect on N<sub>2</sub>O emissions has been investigated at the bioreactor level (Adouani et al., 2015; Chen et al., 2016), an in-depth analysis of N<sub>2</sub>O consumption activity in pure cultures has not been reported. Our results demonstrate that the effect of temperature on N<sub>2</sub>O consumption activity is N<sub>2</sub>ORB species-dependent (Fig. 3 and Table S2), supporting the significance of enriching and leveraging efficient N<sub>2</sub>ORB as an N<sub>2</sub>O sink.

This study also highlights the positive correlation between inherent O<sub>2</sub> and N<sub>2</sub>O respiration rates irrespective of temperature (Fig. S6). The relationship is somewhat expected as both N<sub>2</sub>O and O<sub>2</sub> are preferable electron acceptors for facultative denitrifying bacteria possessing the N<sub>2</sub>OR operon (Chen and Strous, 2013). More importantly, the relative  $V_{\max, \text{N}_2\text{O}}$  over  $V_{\max, \text{O}_2}$  (*i.e.*, the slopes in Fig. S6) differs among the four tested N<sub>2</sub>ORB, with *Azospira* sp. strain I13 having the most preferential use of N<sub>2</sub>O as an electron acceptor. The high relative  $V_{\max, \text{N}_2\text{O}}$  over  $V_{\max, \text{O}_2}$  by *Azospira* sp. strain I13 correlates with the behavior of N<sub>2</sub>ORB-enriched biomass in a chemostat



(Conthe et al., 2018b). The higher slope indicates that *Azospira* sp. strain I13 is effective as a potential N<sub>2</sub>O sink; however, *Azospira* sp. strain I09 does not show this potential as an N<sub>2</sub>O sink, even though they are within the same species of clade II type *nosZ* N<sub>2</sub>ORB (Fig. 1). Because N<sub>2</sub>ORB with high N<sub>2</sub>O affinities are affiliated with both *nosZ* clade I and clade II (Yoon et al., 2016; Conthe et al., 2018a; Suenaga et al., 2019), the classification of a physiological trait by *nosZ* clade types may be too simple (Kim et al., 2020), and the taxonomic classification may be explainable (Conthe et al., 2018c). Further holistic analysis on N<sub>2</sub>ORB physiologies is therefore required.

The specific affinities ( $a_0$ ) describes the bacterial abilities to acquire substrate from liquid (Button, 1991). Our respirometric analysis demonstrated that the values of  $a_{0, O_2}$  and  $a_{0, N_2O}$  and their temperature dependencies were different in the four N<sub>2</sub>ORB, affiliated with either *nosZ* clade type (Fig. 2). Meanwhile, the  $K_m$  values did not significantly change with increasing temperature, except for *Azospira* sp. strain I13 at 35°C, indicating a drastic increase in  $V_{max, O_2}$  and  $V_{max, N_2O}$ . *Azospira* sp. strains I09 and I13 had a higher  $a_{0, N_2O}$  than *Ps. stutzeri* JCM5965 and *Pa. denitrificans* NBRC102528 at 30°C and 35°C (Fig. 2b). This physiological trait rationally explains the predominance of the genus *Azospira* at 30°C in a bioreactor that is dedicated to the enrichment of N<sub>2</sub>ORB (Suenaga et al., 2019). In contrast, the two *Azospira* sp. strains did not demonstrate an advantage over the other N<sub>2</sub>ORB to consume N<sub>2</sub>O at 15°C (Fig. 2b). This suggests that the specific affinity ( $a_{0, N_2O}$ ), dependent on a temperature in WWTPs, determines the predominant N<sub>2</sub>ORB. According to the available N<sub>2</sub>O biokinetics summarized in Table S4, the genera *Dechloromonas* and *Azospira* possess higher specific affinities for N<sub>2</sub>O than other N<sub>2</sub>ORB. Recent reports have indicated the presence of the family *Rhodocyclaceae* including the genera *Dechloromonas* and *Azospira* as N<sub>2</sub>ORB in WWTPs (Kim et al., 2020), implying that *Azospira* spp. in this study may exert a N<sub>2</sub>O reduction potential.

Analysis of the Arrhenius plot and  $E_a$  helps understand the reaction rates and temperature dependency of targeted bacteria. The differences in  $E_a$  values by different

bacteria in the same guilds, for example, aerobic and anaerobic ammonia-oxidizing bacteria, likely result in different reaction rates (Jiang and Bakken, 1999; Oshiki et al., 2011). To the best of our knowledge, this study is the first to showcase the  $E_a$  comparisons among pure culture N<sub>2</sub>ORB. The calculation of the apparent  $E_a$  (Fig. 3) indicated a higher  $E_a$  for N<sub>2</sub>O by *Azospira* sp. strain I13 ( $114.0 \pm 22.6$  kJ mol<sup>-1</sup>) compared with the other N<sub>2</sub>ORB (38.3–60.1 kJ mol<sup>-1</sup>) as summarized in Table S2. The value is also higher than results from previous work in an arable field (28–76 kJ mol<sup>-1</sup>) and denitrifying activated sludge (44.1 kJ mol<sup>-1</sup>) (Holtan-Hartwig et al., 2002; Wang et al., 2014b). This trait has a stark contrast with the  $E_{a, N_2O}$  values of *Ps. stutzeri* JCM5965 and *Pa. denitrificans* NBRC102528, analogous to those for denitrification obtained in previous studies (Table S2). The Q<sub>10</sub> values for N<sub>2</sub>O consumption, except for *Azospira* sp. strain I13 [Q<sub>10</sub> (25–35): 9.04], were comparable to those in the previous studies for denitrification, ranging from 2 to 3 (Heinen, 2006; Saleh-Lakha et al., 2009; Wang et al., 2018). The higher  $E_{a, N_2O}$  and Q<sub>10</sub> for N<sub>2</sub>O by *Azospira* sp. strain I13 indicate that this species functions as a better N<sub>2</sub>O sink under mesophilic conditions than at the lower temperatures, such as 15°C, often found in WWTPs. Therefore, our results highlight the significance of biokinetics and temperature dependence in determining the predominant N<sub>2</sub>ORB in a nitrogen removal system. The trend can indicate the intrinsic capacity for N<sub>2</sub>O reduction to N<sub>2</sub> in microbial ecosystems, which is supported by a previous study (Holtan-Hartwig et al., 2002). To harness the N<sub>2</sub>O consumption potential of *Azospira* sp. strain I13 at a relatively lower temperature, immobilization in a biofilm or entrapped gel would allow a high density of the N<sub>2</sub>ORB (Suenaga et al., 2018b). In parallel with an immobilization strategy, exploring N<sub>2</sub>ORB that display a high activity of N<sub>2</sub>O consumption at a lower temperature warrants additional investigation.

#### **4.4.2 Change in N<sub>2</sub>O consumption dynamics from aerobic to anoxic conditions at various temperatures**

This study displays that the change in N<sub>2</sub>O consumption activity during the transition from aerobic to anoxic conditions is both temperature- and species-

dependent. The use of a microrespiration system allowed the direct observation of N<sub>2</sub>O consumption dynamics at different temperatures, while the O<sub>2</sub> concentration was consumed (Fig. 5). N<sub>2</sub>O reduction, the final step of denitrification mediated by the N<sub>2</sub>OR, is reportedly inactivated in the presence of O<sub>2</sub> (Pomowski et al., 2011). Although *nosZ* gene transcription dynamics has been monitored under suboxic or fluctuating oxygen conditions (Qu et al., 2016; Marchant et al., 2017), monitoring the gene transcripts does not directly provide the context of N<sub>2</sub>O consumption activities. We investigated the combined effect of O<sub>2</sub> concentration and temperature on N<sub>2</sub>ORB activity in the transition of redox changes. Our results demonstrated that *Azospira* sp. strain I09 and *Ps. stutzeri* JCM5965 had increased oxygen resistance as a response to increased temperature, as indicated by higher  $K_I$  values with an increase in temperature (Fig. 4). This property corroborates the hypothesis that the N<sub>2</sub>OR activity is affected by both O<sub>2</sub> and temperature. In contrast to the dependence on temperature in the other tested N<sub>2</sub>ORB, the oxygen resistance potential of *Azospira* sp. strain I13 ( $K_I$ ) did not change with temperature (Fig. 4). These distinct features of O<sub>2</sub> resistance correspond with the previous finding of the different O<sub>2</sub> sensitivities of various N<sub>2</sub>OR (Cavigelli and Robertson, 2001). Interestingly, different O<sub>2</sub> tolerance was observed within N<sub>2</sub>ORB belonging to the same species (Fig. 4). The O<sub>2</sub> tolerance mechanism of N<sub>2</sub>OR is unclear; therefore, holistic evaluations are needed into the N<sub>2</sub>OR activity of different N<sub>2</sub>ORB species and the clarification of the underlying mechanism.

The parameterization of the N<sub>2</sub>O activity recovery by N<sub>2</sub>ORB after O<sub>2</sub> exposure is important for the use of N<sub>2</sub>ORB in WWTPs for N<sub>2</sub>O mitigation. Contrary to the previous study, which models the N<sub>2</sub>O consumption activity in the presence of O<sub>2</sub> (Zheng and Doskey, 2015), our study highlighted the recovery process of N<sub>2</sub>O consumption activity after O<sub>2</sub> depletion (Fig. 5). The N<sub>2</sub>O reduction activity of N<sub>2</sub>ORB linearly increased over time ( $p < 0.001$ ; Fig. S7), supported by a previous study (Suenaga et al., 2018a). On the basis of the results, the recovery model incorporating the recovery of N<sub>2</sub>O reduction activity ( $v_{N_2O}$ ) of N<sub>2</sub>ORB into the whole-cell Michaelis-

Menten kinetics was successfully employed (Figs. 5, S8, and S9). The simple retrofitted model including  $v_{N_2O}$  allowed the parametrization of the N<sub>2</sub>OR response time against O<sub>2</sub> spikes. This parameter, capable of tracking a decrease in N<sub>2</sub>O reduction activity after O<sub>2</sub> exposure, could predict N<sub>2</sub>O emissions from WWTPs where aerobic and anoxic conditions spatiotemporally change.

The recovery rates ( $k'$ ) of N<sub>2</sub>O reduction activity, estimated by the model, indicated a considerable N<sub>2</sub>OR recovery variation by the tested N<sub>2</sub>ORB at different temperatures (Fig. 6), indicating that temperature affected the recovery rates of N<sub>2</sub>O consumption activity. Interestingly, the highest recovery rate was always in keeping with the higher N<sub>2</sub>O consumption rate by these N<sub>2</sub>ORB (Table S3), suggesting an interdependence of the maximum N<sub>2</sub>OR turnover and recovery of N<sub>2</sub>O consumption activity. Our results also demonstrated that the two *Azospira* spp., harboring *nosZ* clade II type enzymes, quickly recover N<sub>2</sub>O consumption activities after O<sub>2</sub> exposure in the tested temperature range (Fig. 6). It is worth noting that the rapid recovery of N<sub>2</sub>O consumption ability is conducive to mitigation of N<sub>2</sub>O emission. Although more study is needed, *nosZ* clade II type N<sub>2</sub>ORB may function as better N<sub>2</sub>O sinks than *nosZ* clade I type N<sub>2</sub>ORB in environments where the redox condition is spatiotemporally altered. Such condition provides another advantage for *nosZ* clade II type N<sub>2</sub>ORB, in addition to those found in previous studies indicating that *nosZ* clade II type N<sub>2</sub>ORB are better N<sub>2</sub>O sinks than clade I type (Yoon et al., 2016; Suenaga et al., 2019). However, this trend may not be consistent with all N<sub>2</sub>ORB belonging to either *nosZ* clade type, and more data are required to support the reduction of N<sub>2</sub>O emissions in natural and engineered systems (Conthe et al., 2018b, 2018c; Hallin et al., 2018).

In conclusion, the N<sub>2</sub>ORB with this high recovery ability are beneficial to rapidly reduce N<sub>2</sub>O emissions after O<sub>2</sub> exposure events, which is required to develop an efficient N<sub>2</sub>O sink in the WWTPs. This application of N<sub>2</sub>ORB can be expanded to wetlands, woodchip bioreactors, and agricultural soil, where redox conditions dynamically change (Moorman, et al., 2010; Morse and Bernhardt, 2013; Song et al.,

2019). Therefore, exploring N<sub>2</sub>ORB with high resilience after O<sub>2</sub> exposure and high N<sub>2</sub>O consumption activity allows the development of N<sub>2</sub>O mitigation technology, and our biokinetic study indicates that *Azospira* sp. strain I13 is a promising potential N<sub>2</sub>O consumer that exhibits a high rate of N<sub>2</sub>O consumption (high  $V_{\max, \text{N}_2\text{O}}$ ) and rapid N<sub>2</sub>OR recovery after O<sub>2</sub> exposure (high  $k'$ ), especially at a mesophilic temperature.

## 5.5 Conclusions

This study presents the temperature effect on O<sub>2</sub> and N<sub>2</sub>O consumption biokinetics of pure culture N<sub>2</sub>ORB, which is deemed promising a N<sub>2</sub>O consumer. The results of this study are summarized below:

- The effect of temperature on O<sub>2</sub> and N<sub>2</sub>O consumption rates and the specific affinities are distinct among the tested N<sub>2</sub>ORB species, which are not merely categorized by *nosZ* clade types.
- N<sub>2</sub>O-reducing activities of *Azospira* sp. strain I09 and *Ps. stutzeri* JCM5965 are sensitive to temperatures in the presence of O<sub>2</sub>, suggesting that the N<sub>2</sub>ORB in response to oxygen tolerance is altered by temperature.
- The model incorporating the N<sub>2</sub>O activity recovery after oxygen exposure can describe N<sub>2</sub>ORB physiologies in the transient anoxia.
- The Arrhenius plot indicates that *Azospira* sp. strain I13 is a specialist as an N<sub>2</sub>O sink at a temperature above 30°C, suggesting that dominant N<sub>2</sub>OBR species dynamically change with different temperatures.

Taken together, our microrespiration analysis illuminates drastic changes in N<sub>2</sub>O consumption capacities of N<sub>2</sub>ORB at different temperatures and oxygen-level conditions.

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## Figures

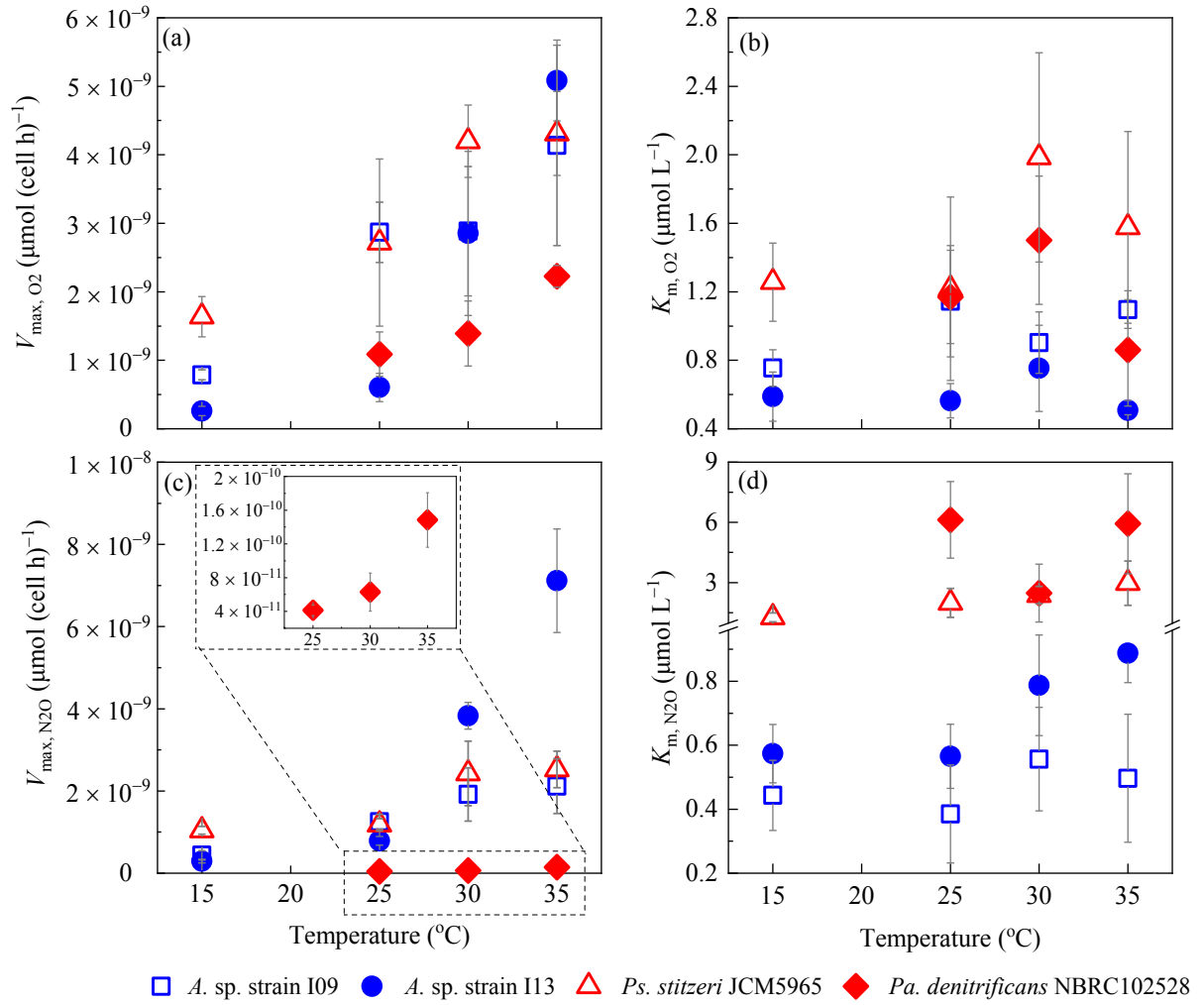


Fig. 1 The temperature dependence of biokinetic parameters (a: maximum O<sub>2</sub> consumption rate; b: O<sub>2</sub> half-saturation constant; c: maximum N<sub>2</sub>O consumption rate; d: N<sub>2</sub>O half-saturation constant). Bars denote standard deviations (n = 3). The clade I and II type of N<sub>2</sub>ORB are represented in red and blue plots, respectively. The statistical analyses were performed by the Duncan test at the 5% level.

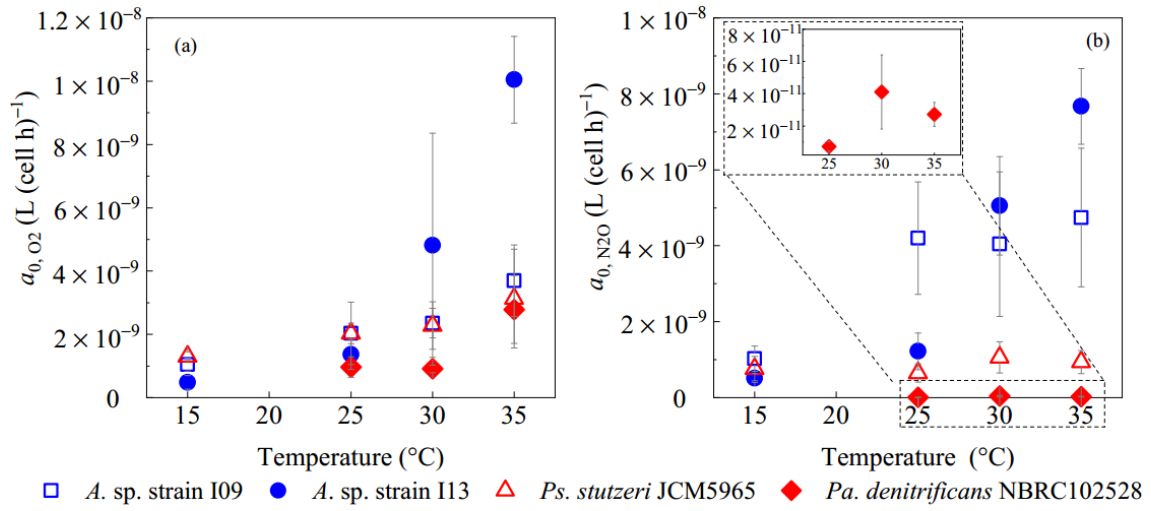


Fig. 2 The effect of temperature on the specific affinity constants for  $O_2$  (a) and  $N_2O$  (b). Bars denote standard deviations (n = 3). The clade I and II type of  $N_2ORB$  are represented in red and blue plots, respectively. The statistical analyses were performed by the Duncan test at the 5% level.

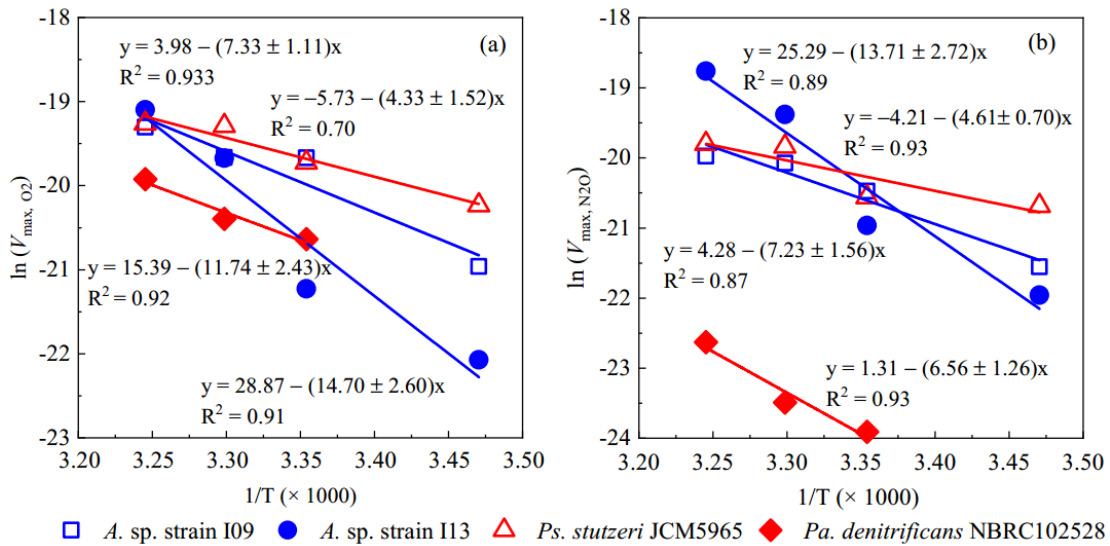


Fig. 3 Arrhenius plot for the calculation of activation energy for  $O_2$  (a) and  $N_2O$  (b) showing the correlation between the maximum  $O_2$  and  $N_2O$  respiration rates and the inverse temperature. Each point represents the average value from triplicate measurements. The clade I and II type of  $N_2ORB$  are represented by the red and blue, respectively.

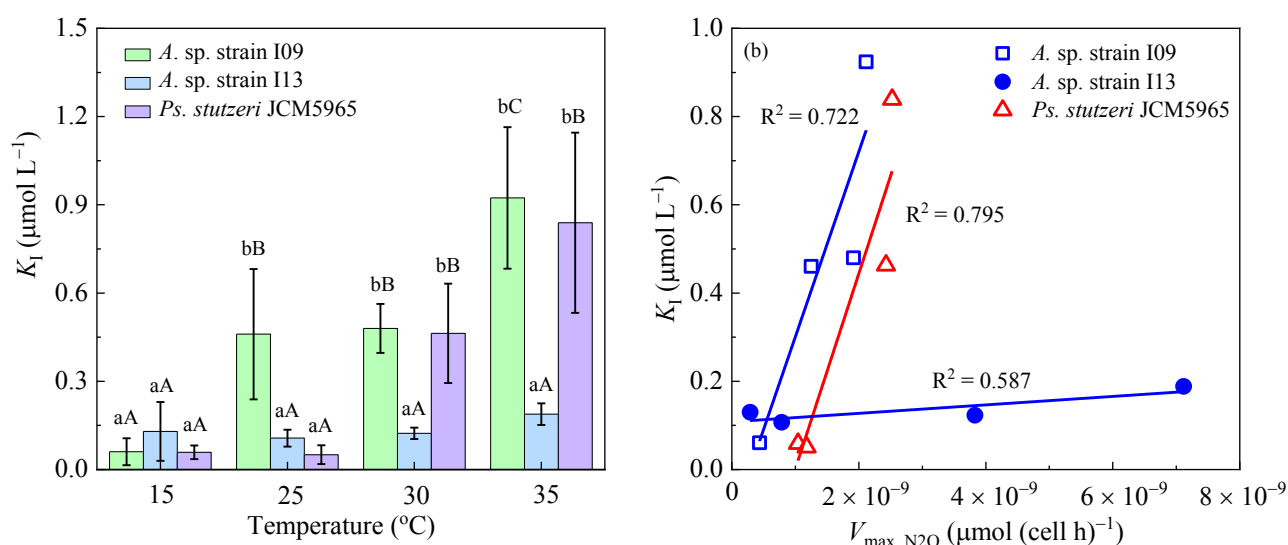


Fig. 4 Effects of temperature on the oxygen inhibition constants of *Azospira. sp.* strain I09, *Azospira. sp.* strain I13, and *Ps. stutzeri* JCM5965. Bars denote standard deviations ( $n = 3$ ). The lowercase and uppercase letters indicate the significant differences among the bacteria at each temperature and those among the temperatures within each bacterium, respectively. The statistical analyses were performed by the Duncan test at the 5% level.

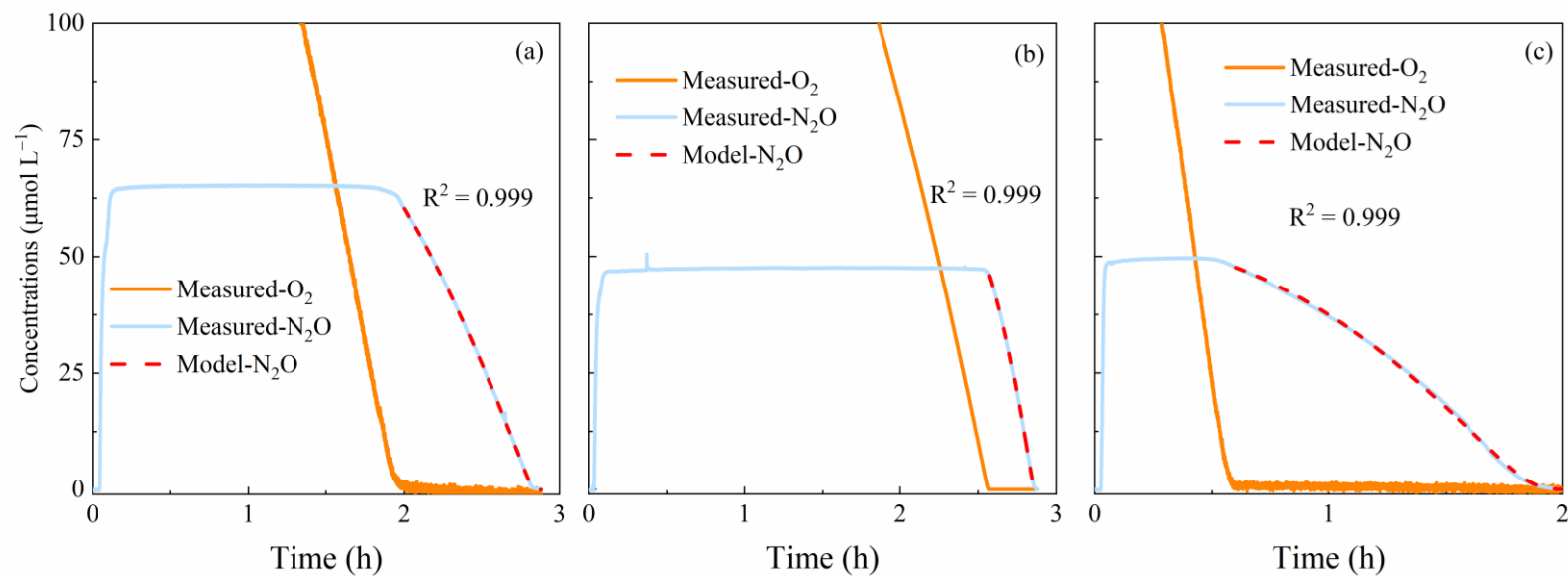


Fig. 5 Fitting the dynamics of  $N_2O$  consumption of *Azospira* sp. strain I09 (a), *Azospira* sp. strain I13 (b), and *Ps. stutzeri* JCM5965 (c), respectively, by the recovery model (Eq. 8) at 30°C. The least square method was used to estimate the recovery rates. The replicates are shown in Fig. S6.



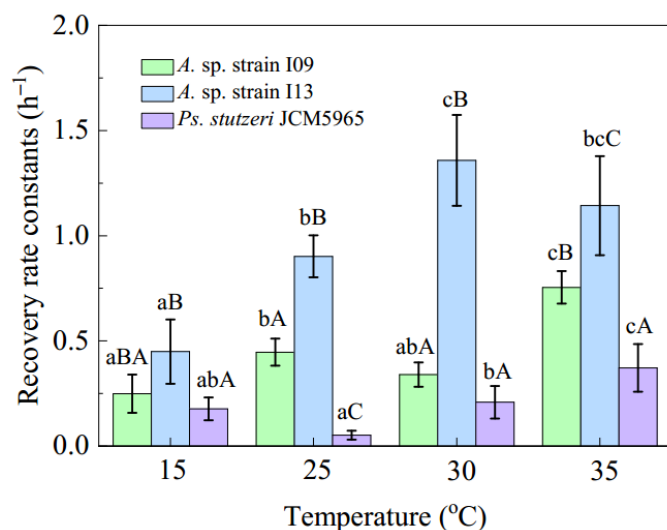


Fig. 6 The recovery rate constants of N<sub>2</sub>O consumption activity by *Azospira* sp. strain I09, *Azospira* sp. strain I13, and *Ps. stutzeri* JCM5965. Bars denote standard errors (n = 3). The lowercase and uppercase letters indicate the significant differences among the temperatures for each bacterium and those among the bacteria at each temperature, respectively. The statistical analyses were performed by the Duncan test at the 5% level.

## Supporting materials

Table S1 Targeted bacteria harboring *nosZ* used in this study

Organisms	Strain name	Type culture collection accession No.	<i>nosZ</i> clade	Genomic information (NCBI accession No.)	Source
<i>Azospira</i> sp. strain	I09	-	II	NZ_AP021844- NZ_AP021845	Bioreactor <sup>1,2</sup>
<i>Azospira</i> sp. strain	I13	-	II	BFBP01000001- BFBP01000026	Bioreactor <sup>1,2</sup>
<i>Pseudomonas stutzeri</i>	JCM5965	ATCC17588	I	CP002881	RIKEN BRC, Japan
<i>Paracoccus denitrificans</i>	NBRC102528	ATCC17741	I	BJUR01000001- BJUR01000161	NBRC, Japan

Table S2 Activation energy ( $E_a$ ) and temperature coefficient ( $Q_{10}$ ) for all tested bacteria compared to literature values. Values of  $Q_{10}$  were calculated between 15°C and 25°C; and between 25°C and 35°C.

Types	O <sub>2</sub> -respiration			N <sub>2</sub> O-respirattion			$E_{a, O_2}/E_{a, N_2O}$	Temperature range (°C)	References
	$E_{a, O_2}$	Q <sub>10</sub>		$E_{a, N_2O}$	Q <sub>10</sub>				
		Q <sub>10</sub> (15–25°C)	Q <sub>10</sub> (25–35°C)		Q <sub>10</sub> (15–25°C)	Q <sub>10</sub> (25–35°C)			
<i>Azospira</i> sp. strain I09	60.9 ± 9.2	3.65	1.44	60.1 ± 13.0	2.87	1.69	1.01	15-35	This study
<i>Azospira</i> sp. strain I13	122.2 ± 21.6	2.32	8.43	114.0 ± 22.6	2.71	9.04	1.07	15-35	This study
<i>Pseudomonas. stutzeri</i> JCM5965	36.0 ± 12.6	1.66	1.58	38.3 ± 5.9	1.13	2.14	0.94	15-35	This study
<i>Paracoccus denitrificans</i> NBRC102528	97.6 ± 20.2	Null	2.04	54.5 ± 10.5	Null	3.60	1.79	15-35	This study
Denitrifying soil (N <sub>2</sub> O reduction)	-	-	-	28-76	-	-	-	5-20	3
Denitrifying activity sludge	-	-	-	48 <sup>a</sup>	2.0	-	-	15-35	4
<i>Pseudomonas aeruginosa</i>	-	-	-	-	1.35	-	-	15-30	5
Denitrifying activity sludge (N <sub>2</sub> O reduction)	-	-	-	44.1	-	-	-	4-34	6
River biofilm (denitrification)	-	-	-	136.9 <sup>a</sup>	7.0	-	-	1-31	7
Constructed urban stormwater wetlands (denitrification)	-	-	-	41-64 <sup>a</sup>	1.3-2.4	-	-	7.5-35	8
Lab-scale SBR (denitrification)	-	-	-	39.6 <sup>a</sup>	-	-	-	15-30	9
Mixed bacterial culture (denitrification)	-	-	-	50.1 <sup>a</sup>	2.3	-	-	15-35	10

<sup>a</sup> The data indicate the activation energy of denitrification.

Table S3 The Spearman's rank correlation between the recovery rate constants ( $k'$ ) and  $V_{max, N_2O}$  at different temperatures

	<i>A. sp.</i> strain I09	<i>A. sp.</i> strain I13	<i>Ps. stutzeri</i> JCM5965
Spearman correlation constants	0.556*	0.716**	0.596*

Significant correlation at the \* $p < 0.05$  and \*\* $p < 0.01$  levels.

Table S4 Summary of the biokinetics of N<sub>2</sub>O-reducing bacteria.

Types	Type of <i>nosZ</i>	$V_{\max, \text{N}_2\text{O}}$ (10 <sup>-9</sup> μmol (cell h) <sup>-1</sup> )	$K_m, \text{N}_2\text{O}$ (μmol L <sup>-1</sup> )	$a_0 (V_{\max, \text{N}_2\text{O}}/K_m, \text{N}_2\text{O})$ (10 <sup>-9</sup> L (cell h) <sup>-1</sup> )	Culture conditions			References
					Temperature (°C)	pH	Carbon source	
<i>Dechloromonas aromatica</i> RCB	Clade II	7.75 ± 0.57 <sup>a</sup>	0.32	24.22	30	Null	acetate	11
<i>Dechloromonas denitrificans</i> ED-1	Clade II	12.87 ± 1.08 <sup>a</sup>	1.01	12.74	30	Null	acetate	11
<i>Anaeromyxobacter dehalogenans</i> 2CP-C	Clade II	0.29 ± 0.04 <sup>a</sup>	1.34	0.21	30	Null	acetate	11
<i>Dechloromona</i> I20	Clade II	18.0 ± 4.6	2.04	8.82	30	7.5	acetate	12
<i>Gemmatimonas aurantiaca</i> T-27	Clade II	0.07 ± 0.01	Null	Null	30	7.0	glucose	13
<i>Azospira</i> sp. strain I09	Clade II	2.12 ± 0.67	0.50	4.74	35	7.5	acetate	This study
<i>Azospira</i> sp. strain I13	Clade II	7.12 ± 1.26	0.89	7.67	35	7.5	acetate	This study
<i>Shewanella loihica</i> PV-4	Clade I	7.49 ± 0.40 <sup>a</sup>	7.07	1.06	30	Null	acetate	11
<i>Pseudomonas stutzeri</i> DCP-Ps1	Clade I	69.9 ± 7.4 <sup>a</sup>	35.5	1.97	30	Null	acetate	11
<i>Alycyclophilus denitrificans</i> I51	Clade I	3.78 ± 2.29	8.98	0.42	30	7.5	acetate	12
<i>Bradyrhizobium japonicum</i>	Clade I	0.085	Null	Null	30	Null	yeast extract	14
<i>Ps. stutzeri</i> JCM5965	Clade I	2.52 ± 0.42	2.97	0.94	35	7.5	acetate	This study
<i>Pa. denitrificans</i> NBRC 102528	Clade I	0.15 ± 0.03	5.94	0.03	35	7.5	acetate	This study
Microbial enrichment bioreactor (mixed culture)	Null	Null	6.98	Null	20	7.0	acetate	15

<sup>a</sup> Converted using the 0.28 pg dry-weight per bacterial cell<sup>1</sup>. Null represents that the data are not reported

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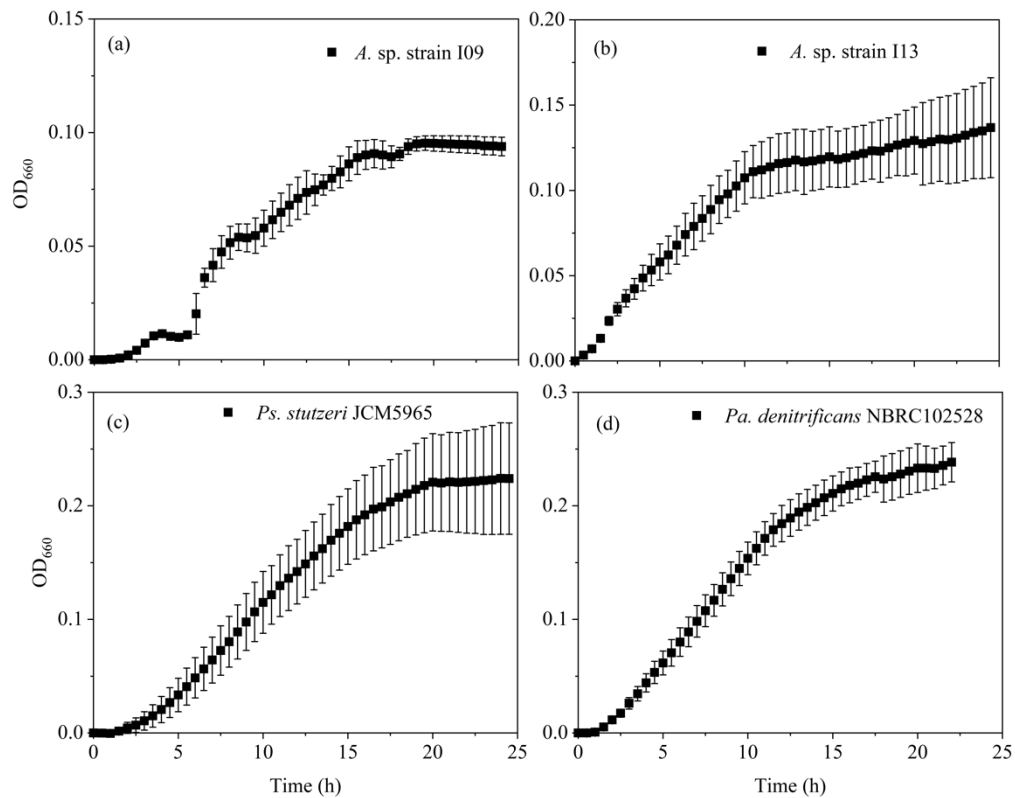


Fig. S1 The growth curves of (a) *Azospira* sp. strain I09, (b) *Azospira* sp. strain I13, (c) *Ps. stutzeri* JCM5965, and (d) *Pa. denitrificans* NBRC102528, respectively, at 30°C.

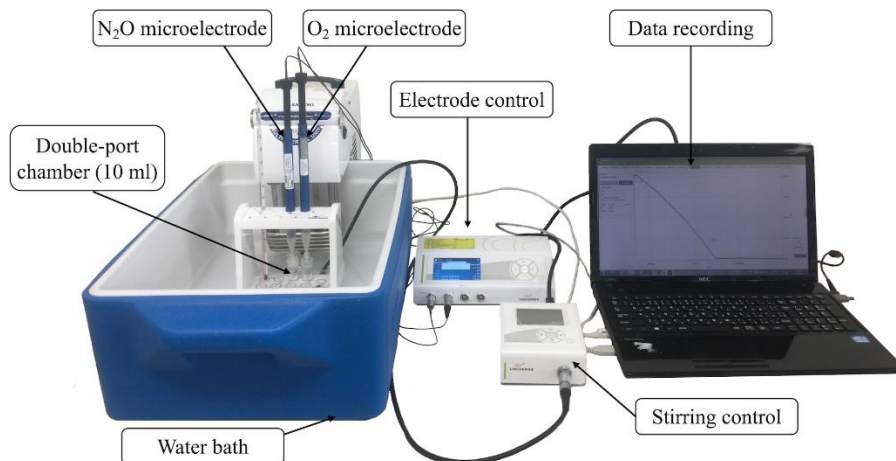


Fig. S2 Schematic diagram of a micro-respiration system using O<sub>2</sub> and N<sub>2</sub>O sensors.

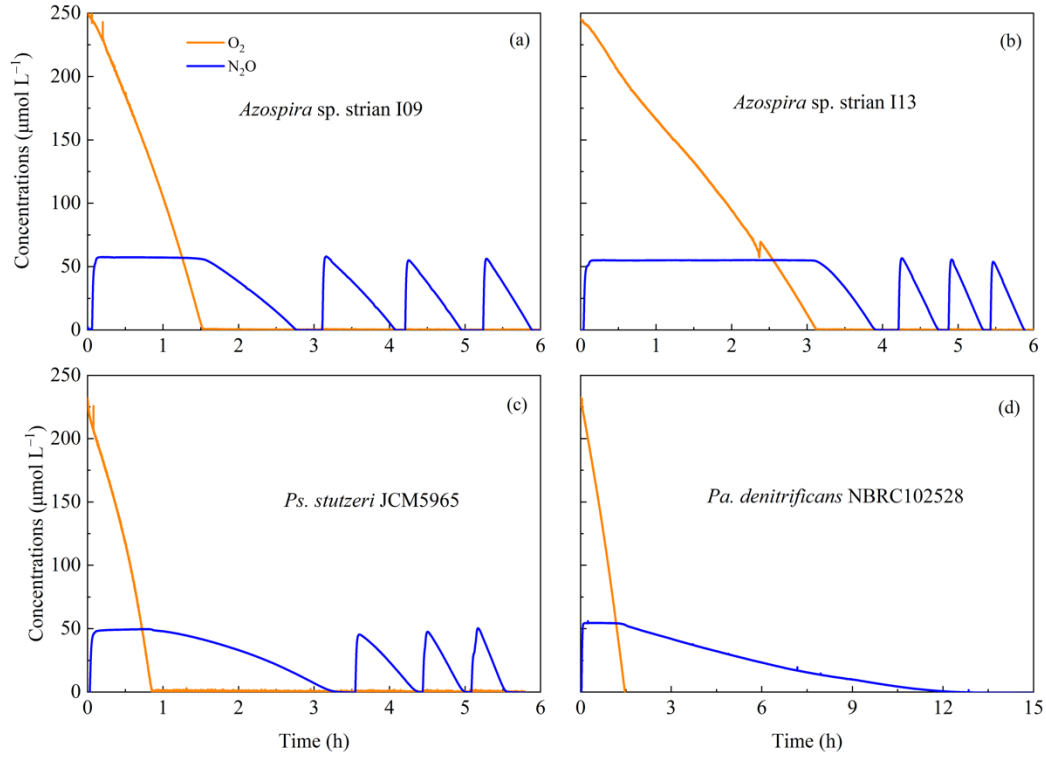


Fig. S3 Time courses of  $O_2$  and  $N_2O$  concentrations consumed by (a) *Azospira* sp. strains I09, (b) *Azospira* sp. strain I13, (c) *Ps. stutzeri* JCM5965, and (d) *Pa. denitrificans* NBRC102528 at 25°C. The orange and green lines represent  $O_2$  and  $N_2O$ , respectively.

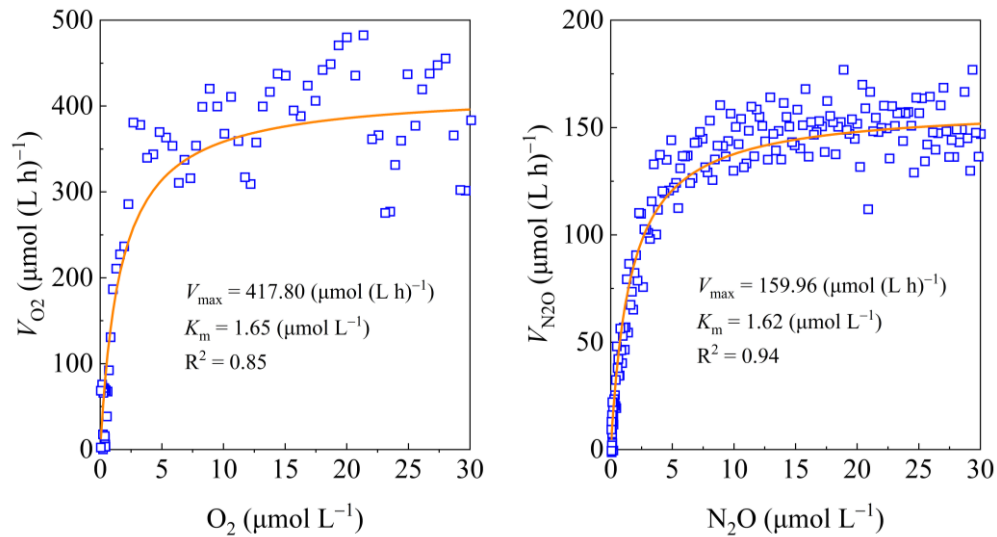


Fig. S4 Experimental and modeled  $V_{O_2}$  and  $V_{N_2O}$  as a function of (a)  $O_2$  and (b)  $N_2O$  concentrations by *Ps. stutzeri* JCM5965 at 25°C.

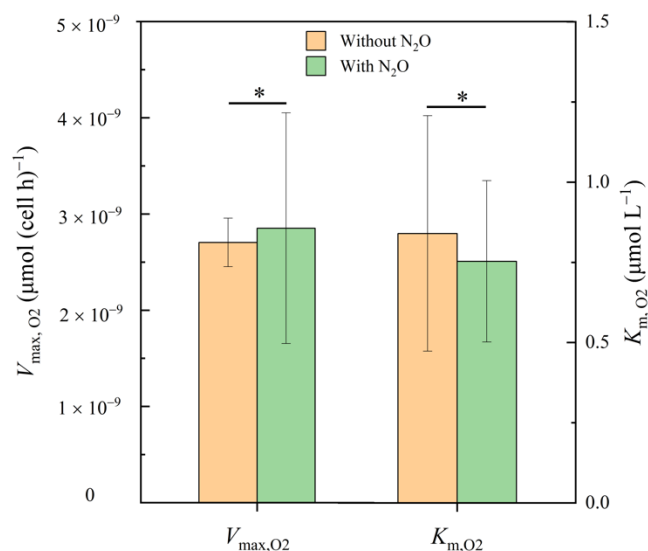


Fig. S5 The effect of N<sub>2</sub>O on oxygen respiration kinetics of *Azospira* sp. strain I13 at 30°C. The result indicates that the presence of N<sub>2</sub>O did not affect  $V_{\max, O_2}$  and  $K_{m, O_2}$  (\* $p > 0.05$ ).

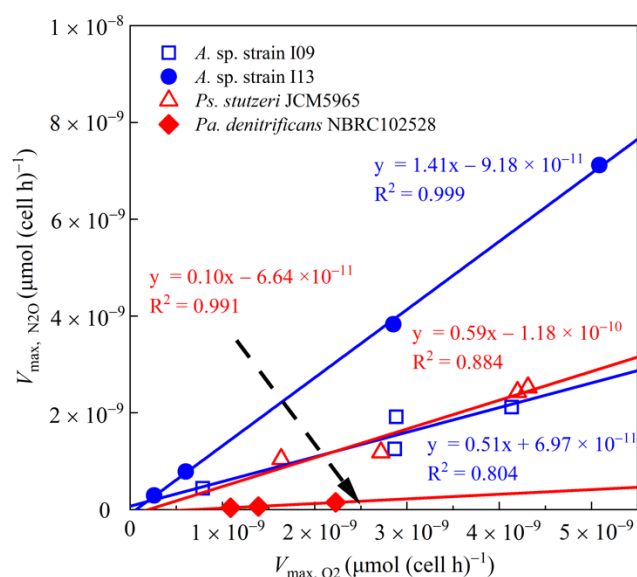


Fig. S6 The relationship between maximum O<sub>2</sub> consumption rates and maximum N<sub>2</sub>O consumption rates among tested N<sub>2</sub>ORB at different temperatures. The clade I and II type N<sub>2</sub>ORB are plotted in red and blue, respectively. Note that the plots are derived from the tests at different temperatures.



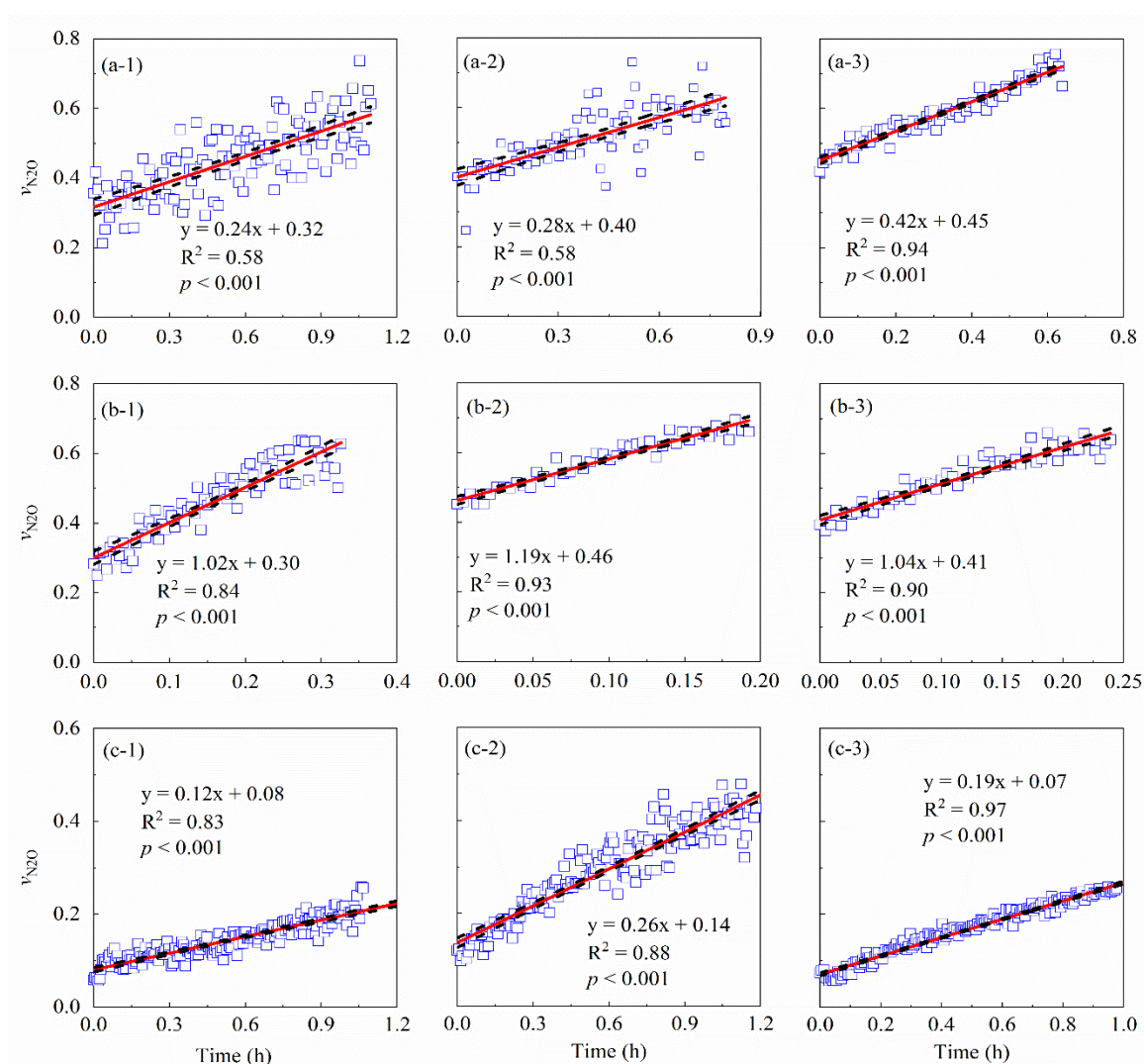


Fig. S7 The variations in relative  $N_2O$  consumption rates ( $v_{N_2O}$ ) of (a) *Azospira* sp. strain I09, (b) *Azospira* sp. strain I13, and (c) *Ps. stutzeri* JCM5965, respectively, at 30°C. Experiments were performed in triplicate. Solid and broken lines represent fitting  $v_{N_2O}$  and the 95% confidence interval of each approximated line, respectively.

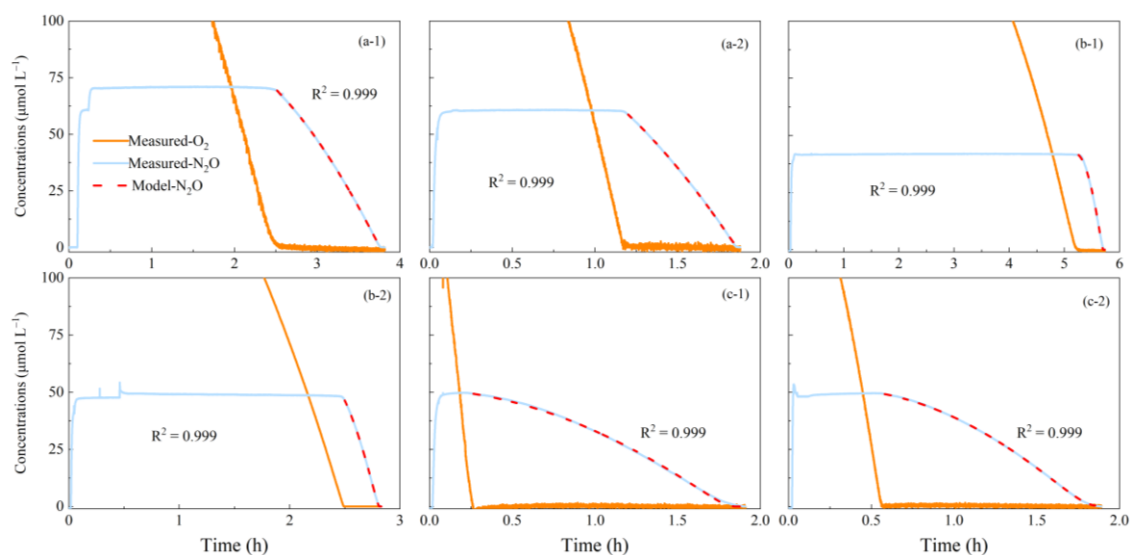


Fig. S8 Simulating  $\text{N}_2\text{O}$  concentration dynamics of (a) *Azospira* sp. strain I09, (b) *Azospira* sp. strain I13, and (c) *Ps. stutzeri* JCM5965 by the recovery model at 30°C.

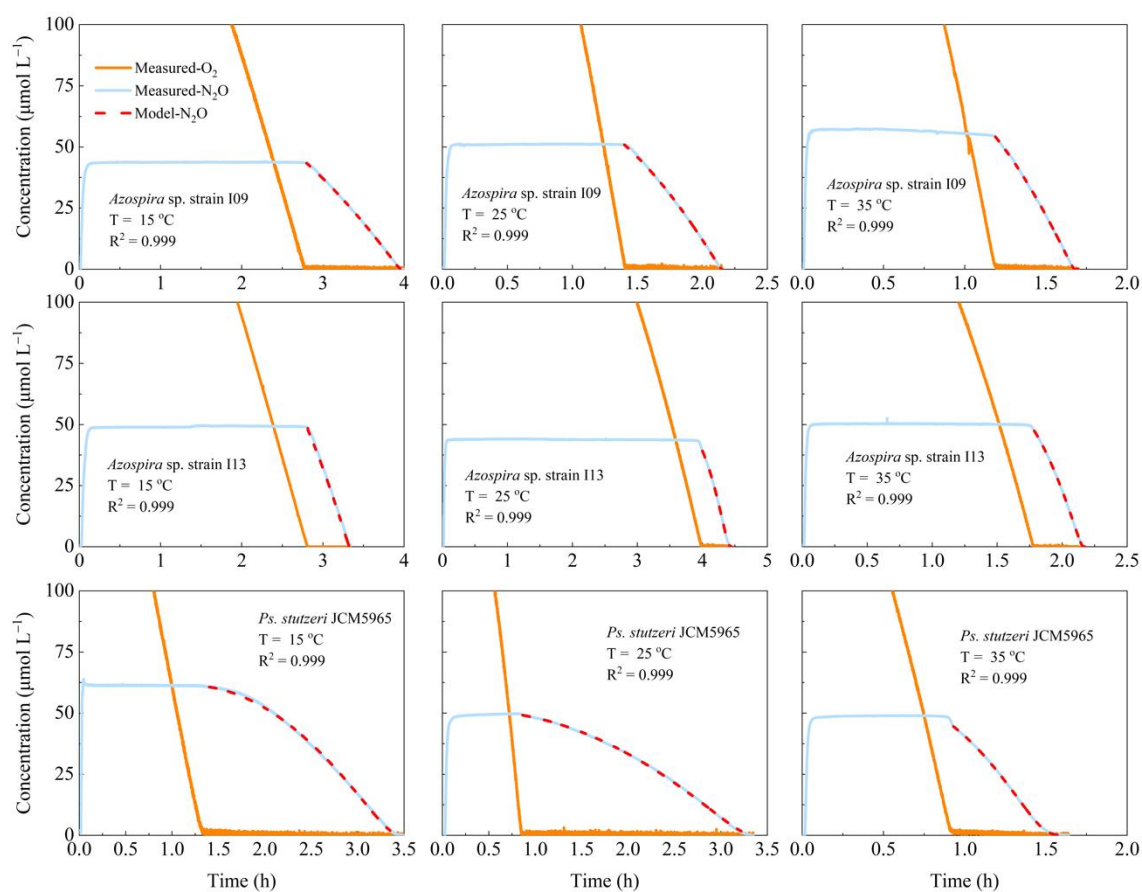


Fig. S9 Simulating  $\text{N}_2\text{O}$  concentration dynamics of *Azospira* sp. strain I09, *Azospira* sp. strain I13, and *Ps. stutzeri* JCM5965 by the recovery model at 15, 25, and 35°C, respectively.

## ***Chapter 6: Summary***

### **6.1 Main conclusions**

Lake plays an important role in global carbon and nitrogen cycles, and there is still great uncertainty about the CH<sub>4</sub> and N<sub>2</sub>O emissions in the increasing requirements of nutrients. In this study, 17 shallow lakes across the different trophic states in the Yangtze River basin, China, were selected to understand the characterizations and microbial driving mechanisms of CH<sub>4</sub> and N<sub>2</sub>O emission in shallow lakes across different trophic states. Meanwhile, the biokinetic characterizations of N<sub>2</sub>O consumption were elucidated for understanding the effects of temperature and oxygen on N<sub>2</sub>O consumption. The conclusions of this study are summarized below:

(1) We performed a series of field measurements and characterized the CH<sub>4</sub> emissions in shallow lakes of different trophic state in the Yangtze River basin. This is the first study to elucidate the nonlinear response of the release of CH<sub>4</sub> from shallow lakes along a gradient of increasing eutrophication. An increase in the *TLI* of a lake by one level (*e.g.*, from mesotrophic to eutrophic) would correspond to a 10-fold increase in CH<sub>4</sub> emissions. This phenomenon is related to the fact that the accumulated nutrients and subsequent production of algae organic matter are the crucial drivers for the microbial community responsible for the CH<sub>4</sub> emissions from lakes. In addition, the decomposition of algae makes a significant contribution to the CH<sub>4</sub> emissions from eutrophic lakes, especially hyper-eutrophic lakes. Correspondingly, the NH<sub>4</sub><sup>+</sup> concentration in the water column can be used to predict the CH<sub>4</sub> emissions from lakes. Thus, our results strongly suggest that the eutrophication of lakes not only causes a deterioration in water quality, but also has a positive feedback effect on the global greenhouse effect.

(2) In addition, the N<sub>2</sub>O emission fluxes of shallow lakes were most strongly affected by lake trophic state, suggesting that estimation of N<sub>2</sub>O emission fluxes should consider lake trophic state; The nonlinear model incorporating trophic state levels can

describe the N<sub>2</sub>O emissions from a shallow lake; The predicted N<sub>2</sub>O emission fluxes based on the IPCC *EF<sub>5r</sub>* overestimated the observed fluxes, particularly those in hyper-eutrophic lakes; Nutrient-rich conditions and algal accumulation were key factors determining N<sub>2</sub>O emission fluxes in shallow lakes, and algal accumulation played a dual role in stimulating and impeding N<sub>2</sub>O emissions, especially in hyper-eutrophic lakes; Changes in season accompanied the appearance and disappearance of algae and altered N<sub>2</sub>O emission fluxes, especially in hyper-eutrophic lakes.

(3) Further elucidated the two key parameters *i.e.*, temperature and oxygen, to regulate N<sub>2</sub>O consumption process of N<sub>2</sub>ORB by biokinetic analyzation. The effect of temperature on O<sub>2</sub> and N<sub>2</sub>O consumption rates and the specific affinities are distinct among the tested N<sub>2</sub>ORB species, which are not merely categorized by *nosZ* clade types; N<sub>2</sub>O-reducing activities of *Azospira* sp. strain I09 and *Ps. stutzeri* JCM5965 are sensitive to temperatures in the presence of O<sub>2</sub>, suggesting that the N<sub>2</sub>ORB in response to oxygen tolerance is altered by temperature; The model incorporating the N<sub>2</sub>O activity recovery after oxygen exposure can describe N<sub>2</sub>ORB physiologies in the transient anoxia; The Arrhenius plot indicates that *Azospira* sp. strain I13 is a specialist as an N<sub>2</sub>O sink at a temperature above 30°C, suggesting that dominant N<sub>2</sub>OBR species dynamically change with different temperatures; Taken together, our microrespiration analysis illuminates drastic changes in N<sub>2</sub>O consumption capacities of N<sub>2</sub>ORB at different temperatures and oxygen-level conditions.

## 6.2 Innovations and outlooks

This study major concerned that the eutrophication and GHGs emission in shallow lakes. Interestingly, the non-linear model is firstly proposed for quantifying the differences of CH<sub>4</sub> and N<sub>2</sub>O emission fluxes in the shallow lakes across the different trophic states. This finding will deeply our knowledges regarding the eutrophication progress impacts on the CH<sub>4</sub> and N<sub>2</sub>O emissions. It is implicated that the estimation of CH<sub>4</sub> and N<sub>2</sub>O emission fluxes should be considered the trophic state of lakes.

Although we have clear that the eutrophication of lakes can significantly stimulate

the GHGs emission from lakes, the relationship between the eutrophication processes and global warming is still unclear. Further research should pay attention to the relationship between global climate changes, lake eutrophication, and GHGs emissions.

In addition, this study also evaluates the characterizations of N<sub>2</sub>ORB to consume N<sub>2</sub>O in different temperature and oxygen conditions, mitigations of N<sub>2</sub>O by the N<sub>2</sub>ORB in the shallow lakes are still unclear. In particular, how the N<sub>2</sub>ORB regulates the reduction of N<sub>2</sub>O in the heavy algae-accumulated zones needs further research.

## ***List of Publications***

1. **Zhou, Y.**, Xu, X., Han, R., Li, L., Feng, Y., Yeerken, S., Song, K. and Wang, Q., **2019**. Suspended particles potentially enhance nitrous oxide (N<sub>2</sub>O) emissions in the oxic estuarine waters of eutrophic lakes: Field and experimental evidence. *Environmental Pollution*, 252, 1225-1234. (Published) (Chapter 3 and 4)
2. **Zhou, Y.**, Song, K., Han, R., Riya, S., Xu, X., Yeerken, S., Geng, S., Ma, Y., Terada, A., **2020**. Nonlinear response of methane release to increased trophic state levels coupled with microbial processes in shallow lakes. *Environmental Pollution*, 265, 114919. (Published) (Chapter 2)
3. **Zhou, Y.**, Suenaga, T., Qi, C., Riya, S., Hosomi, M., Terada, A., **2021**. Temperature and oxygen level determine N<sub>2</sub>O respiration activities of heterotrophic N<sub>2</sub>O-reducing bacteria: Biokinetic study. *Biotechnology and Bioengineering*, 118, 1330-1341. (Published) (Chapter 5)
4. **Zhou, Y.**, Xu, X., Song, K., Yeerken, S., Deng, M., Li, Lu., Riya, S., Wang, Q., Terada, A., **2021**. Nonlinear pattern and algal dual-impact in N<sub>2</sub>O emission with increasing trophic levels in shallow lakes. *Water Research*. Doi.org/10.1016/j.watres.2021.117489 (Published) (Chapter 4)

## *Acknowledgments*

Time flies, the three-years doctoral program is coming to an end. I am very fortunate to be able to come to Tokyo University of Agriculture and Technology (TUAT), Japan, and join Terada-Riya laboratory (Hosomi-Terada laboratory) to study and live. In this period, I have gained a lot of knowledge and also encountered many challenges.

I would like to appreciate my supervisor, Prof. Akihiko Terada for his kind support and continuity encouragement in my study. He's very serious and rigorous working attitude has benefited me a lot.

I would like to express my deeply respect for Prof. Masaaki Hosomi. His attitude towards life, work, and students are worthy of our lifelong learning.

Many thanks to my colleagues at the Terada-Riya laboratory (Hosomi-Terada laboratory), Department of Applied Chemistry, TUAT for their friendship help and kindly supports in my study and life. It is because of their friendly help and support in my experiment that I was able to carry out my experimental work smoothly. Especially, I would like to express my thanks to Riya-sensei, Suenaga-sensei, and all other laboratory members for giving me many meaningful suggestions in my study.

Grateful acknowledgment is for the China Scholarship Council (grant number CSC201808420224) to support my study in Japan.

Finally, I would like to express my many thanks to my friends and family members for all their love and encouragement. Too much gratitude is indescribable and will always be in my heart.