

**Influence of gypsum application on methane  
production and emission from saline paddy soils in  
relation to growth and root exudation of rice plants**

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**Influence of gypsum application on methane production and  
emission from saline paddy soils in relation to growth and  
root exudation of rice plants**

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## **DECLARATION OF ORIGINALITY**

I hereby declare that this thesis and the work reported herein was composed by and originated entirely from me. Information derived from the published and unpublished work of the others has been acknowledged in the text and references are given in the list of sources.

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Date 09.03.2015

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## Abstract

Paddy rice fields are identified as one of the major sources of global warming methane (CH<sub>4</sub>) gas. However, the extent of CH<sub>4</sub> emission is largely influenced by the soil condition to which rice is planted and management practices associated with rice cultivation. Among the world total rice areas, about 30% contains too high levels of salts for normal rice growth. In general, soil salinity is caused by occasional or periodic intrusion of sea water, or by surface evaporation of soil water initially high in salt content. Thus, CH<sub>4</sub> emission from rice cultivation was evaluated under irrigation water salinity and saline soil. To improve the rice growth in Na<sup>+</sup> saturated soil or water, gypsum fertilizer application is one of the common practices. Because of current increasing rate (about 1.0% each year) of global CH<sub>4</sub> concentration in the atmosphere (IPCC, 2007) and its strong infrared absorption characteristic, it also needs to evaluate the impact of paddy field management practices on CH<sub>4</sub> emission to avoid climate change impact (Rogner et al., 2007). Therefore, the effect of gypsum fertilizer addition to saline condition on CH<sub>4</sub> emission was also studied in relation to rice growth.

In chapter (1), CH<sub>4</sub> production potential of soil was evaluated under five different irrigation water salinity levels; 0, 10, 30, 60 and 90 mmol L<sup>-1</sup> NaCl for 3 weeks. The addition of NaCl up to 30 mmol L<sup>-1</sup> increased in CH<sub>4</sub> production. The increase of CH<sub>4</sub> production in salinity 10 and 30 mmol L<sup>-1</sup> NaCl was 1.2 times and 1.5 times higher than control, respectively. This addition of saline water up to 30 mmol L<sup>-1</sup> to soil might fulfils the

requirement of  $\text{Na}^+$  by methanogenic bacteria because Kalmokoff (1988) reported that sodium is required for amino acid transport, growth, methanogenesis, and internal pH regulation in methanogenic bacteria. At higher salinity levels (S60 and S90),  $\text{CH}_4$  production was 19 to 33% lower than control. It might be due to the  $\text{Cl}^-$  toxicity at higher salinity levels.

The salinity levels such as 30 and 90  $\text{mmol L}^{-1}$   $\text{NaCl}$  which showed maximum and minimum  $\text{CH}_4$  production were selected and studied in rice cultivation with or without gypsum fertilizer addition. It was conducted as pot experiment. The continuous application of saline water significantly suppressed the rice growth at both salinity levels. The rice plants in salinity 90  $\text{mmol L}^{-1}$  could survive until 7 weeks after transplanting. The higher total numbers of dead leaves were observed in saline condition with or without gypsum fertilizer addition. The  $\text{CH}_4$  emission in salinity level 30  $\text{mmol L}^{-1}$   $\text{NaCl}$  without gypsum fertilizer (296  $\text{kg CH}_4 \text{ ha}^{-1}$ ) was not significant different with that of control (316.213  $\text{kg CH}_4 \text{ ha}^{-1}$ ) but the amount of emission was numerically lower. It might be due to its lower above plant biomass yield. The lowest  $\text{CH}_4$  emission was observed in salinity 90  $\text{mmol L}^{-1}$   $\text{NaCl}$  (44.2-55.5  $\text{kg CH}_4 \text{ ha}^{-1}$ ). It might be due to its lower  $\text{CH}_4$  production potential and lower above biomass yield and the shortest growth duration period. Gypsum fertilizer addition reduced  $\text{CH}_4$  emission either in saline or non-saline condition. The reduction rate due to gypsum fertilizer addition was about 53-83% in saline condition and about 56% in non-saline condition. It might be due to the competition of  $\text{SO}_4^{2-}$  reducing bacteria and methanogenic bacteria for the organic carbon substrate.

Thus, the effect of different rates of gypsum fertilizer addition in to rice cultivation on CH<sub>4</sub> emission was evaluated in chapter with pot experiment. The gypsum application rates were 0, 1, 2.5 and 5 ton ha<sup>-1</sup> and irrigation water salinity levels were 0 and 25 mmol L<sup>-1</sup> NaCl. The results showed that the irrigation of water with salinity level 25 mmol L<sup>-1</sup>NaCl could not suppress CH<sub>4</sub> emission. Higher application of gypsum resulted in higher reduction in CH<sub>4</sub> emission. However, the reduction of CH<sub>4</sub> emission due to gypsum was higher in non-saline condition than saline condition. It might be due to the higher carbon availability that was contributed from dead plant materials in saline condition compared to non-saline condition.

As saline soil is composed of a range of dissolved salts such as NaCl, Na<sub>2</sub>SO<sub>4</sub>, MgSO<sub>4</sub>, CaSO<sub>4</sub>, MgCl<sub>2</sub>, KCl, and Na<sub>2</sub>CO<sub>3</sub>, CH<sub>4</sub> emission was also studied in coastal saline soil in chapter (4). The coastal saline soil was collected from the tsunami flooded paddy field in Sendai city, Tokyo, Japan. Three different rates of gypsum fertilizer such as gypsum 0.5, 1 and 2 ton ha<sup>-1</sup> were applied to the coastal saline soil to provide the appropriate Na/Ca ratios in the soil. Methane emission was evaluated during vegetative stage for 30 days. Gypsum fertilizer addition significantly reduced the soil pH and increased soil EC. However, the increase in K<sup>+</sup> concentration of plants and decrease in Na<sup>+</sup> concentration of plants were observed in gypsum 0.5 and 1 ton ha<sup>-1</sup>. The total organic carbon concentration in gypsum 0.5 ton ha<sup>-1</sup> was the highest among the treatments at 30 days after transplanting. There was no significant difference in the shoot dry weight except gypsum 2 ton ha<sup>-1</sup>. The significant highest CH<sub>4</sub> emission was observed in gypsum 0.5 ton ha<sup>-1</sup> and it might relate with its

highest total carbon concentration, the added amount of  $\text{SO}_4^{2-}$  that that did not reach the threshold limit necessary for a successful competition of carbon substrates between sulphate reducers with methanogens. The lowest  $\text{CH}_4$  emission was observed in gypsum 2 ton  $\text{ha}^{-1}$ . This lowest  $\text{CH}_4$  emission might relate with its lowest pH value (about 5.82), highest input of  $\text{SO}_4^{2-}$  concentration from gypsum fertilizer and lowest above dry matter yield.

Rice plants supply carbon substrate to methanogens via root exudates and dead plant materials. Plants release about 40% of net carbon fixed during photosynthesis as root exudates. As high salinity reduces photosynthetic efficiency and plant growth, it may limit the extent of carbon contribution to the soil via root exudation. Thus, root exudates were collected from plants which are grown in chapter (4) at maximum tillering stage. After that, organic acids concentrations of root exudates were analyzed in chapter (5). Six organic acids species such as Citric, Tartaric, Malic, Lactic, Formic and Acetic acids were observed in the root exudates of control, gypsum 0.5, and 1 ton  $\text{ha}^{-1}$ . Only four organic acids species, Citric, Tartaric, Malic, and Lactic acids were observed in gypsum 2 ton  $\text{ha}^{-1}$ . It might relate with its lower shoot dry weight. Furthermore, the higher values of organic acid exudation potential were observed in gypsum fertilizer treatments over that control. It might relate with higher EC values in gypsum fertilizer treatments over that of control. Under the salt stress, most of the plants accumulate low molecular weight organic solutes (Jouyban, 2012), and plants can secrete more metabolites and some unknown substances under environmental stress Bucio et al. (2000).

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## List of Abbreviations

C: Carbon

Ca: Calcium

CH<sub>4</sub>: Methane

CO<sub>2</sub>: Carbon dioxide

DAT: Days After Transplanting

EC: Electrical conductivity

Eh: Redox potential

K: Potassium

LSD: Least Significant Difference

N: Nitrogen

NH<sub>4</sub>: Ammonium

NO<sub>3</sub>: Nitrate

SO<sub>4</sub>: Sulphate

SOC: Soil Organic Carbon

TOC: Total Organic Carbon

## **Chapter 1**

### **General Introduction**

# **1. General Introduction**

## **1.1 Rice production around the world**

Rice is the most loved cereal of Asia. It feeds more than half of the world's population (IRRI, 2006). In 2007, the average annual consumption per capita was about 197 kg (FAOSTAT, 2012) and provided 49% of the calories and 39% of the protein in the diet (FAOSTAT, 2012). According to the FAOSTAT online data base, approximately 164.72 million hectares were harvested worldwide in 2013. Among them, about 88.9% (146.46 million ha) were in Asia and 34.9% (51 million ha) of the cultivation area were harvested in Southeast Asia alone.

## **1.2 Salinity in rice growing regions of the world**

In the present, salinity is the second most widespread soil problem in rice growing countries next to drought and considered as a serious constraint to increased rice production (Amaranatha et al., 2014). About 30% of the 130 million hectares where rice is grown contain salt that is too high to allow normal rice yield (Mishra, 2004). Rice is cultivated in more than 115 countries, of which Asia's share is more than 91% of the world total. Majority of the Asian rice production zone is confined to South and South East Asia wherein about 48 Mha of potentially useful agricultural land is unusable for growing rice in Southern Asia and South East Asia due to saline soils (Vinod et al., 2013). Although rice is considered as a sensitive crop to salinity, it is one of the most widely grown crops in coastal areas

(Amaranatha et al., 2014). Redfern et al. (2012) reported that in the humid regions of Southeast Asia, there are many hectares that are technically appropriate for rice production but are left uncultivated or are grown with very low yields because of salinity and soil problem. In general, soil salinity is caused by occasional or periodic intrusion of sea water, or by surface evaporation of soil water initially high in salt content. Thus, some saline land is found along every coast, but the extent of affected areas varies greatly. Generally, salt accumulation increases in the drier climates and diminishes strongly in an equatorial climate without a pronounced dry season. Extremely saline deltas are found in arid areas (Datta et al., 1981). In view of current levels of the growing world population, it is estimated that there will be a need to increase food production up to 38 % by 2025 and 57 % by 2050 to maintain the food supply. Green revolution helped to solve the world's demand for food, but is not enough to meet the 21st century's exploding population (Barus et al., 2013). Therefore, it is necessary to promote rice production not only to increase yield per unit of fertile land but also to expand the area on uncultivated land efficiently.

### **1.2.1 Impact of salinity on rice growth**

Soil salinity is caused by an increased activity of soluble salts. Generally, soils with electrical conductivity higher than  $4 \text{ dS m}^{-1}$  are generally considered as saline soils (Munns, 2005). According to the USDA (1954), soils can be classified as non-saline/non-sodic soils ( $\text{ESP} \leq 15\%$ ;  $\text{EC} \leq 4 \text{ mmhos cm}^{-1}$ ), saline soil ( $\text{ESP} \leq 15\%$ ;  $\text{EC} > 4 \text{ mmhos cm}^{-1}$ ), sodic soil ( $\text{ESP} > 15\%$ ;  $\text{EC} \leq 4 \text{ mmhos cm}^{-1}$ ) and as saline sodic soil ( $\text{ESP} > 15\%$ ;  $\text{EC} > 4 \text{ mmhos cm}^{-1}$ ).

cm<sup>-1</sup>). The pH of saline soils is generally less than 8.5, of saline sodic soils about 8.5 and of sodic soils more than 8.5.

Salinity reveals adverse effects on plants. Elevated levels of salt ions in soil solution surrounding plant roots induce an imbalance in water potential between plant root cells and ambient soil solution and result in cellular dehydration (Kosová et al., 2013). It is known as osmotic stress. In addition, elevated salts lead to a passive salt ion penetration via plasma membrane and to an accumulation of salt ions in cell cytoplasm which can lead to inhibition of intracellular enzyme activity (Munns, 2008). If excessive amount of salt enter the plant, the concentration of salt will eventually rise to a toxic level in older transpiring leaves causing premature senescence and reduced the photosynthetic leaf area of a plant to a level that can not sustain growth (Shereen et al., 2005). Thus, salinity appears to affect two plant processes such as water relations and ionic relations. During initial exposure to salinity, plants experience water stress, which in turn reduces leaf expansion. During long-term exposure to salinity, plants experience ionic stress, which can lead to premature senescence of adult leaves (Amirjani, 2011). Therefore, it can be said that salinity has three potential effects on plants:

- i. Lowering of the water potential
- ii. Direct toxicity of any Na<sup>+</sup> and Cl<sup>-</sup> absorbed
- iii. Interference with the uptake of essential nutrients (Flowers and Flowers, 2005)

Rice is relatively salt-sensitive amongst cereals (Kavitha et al., 2012). According to the classification of salt tolerance to salinity, the rice crop is within the sensitive division from 0  $\text{dsm}^{-1}$  to 8  $\text{dsm}^{-1}$  (Mass and Hoffman, 1986). There are two essential parameters sufficient for expressing salt tolerance; threshold and slope values. Threshold value means the maximum allowable salinity without yield reduction and slope value means the percent of yield reduction per unit increase in salinity beyond the threshold. The threshold and slope values of rice (*Oryza sativa*) are 3.0  $\text{dsm}^{-1}$  and 12% per  $\text{dsm}^{-1}$  of saturated soil extract (ECe), respectively (Mass and Hoffman, 1977). Relative salt tolerance of rice at 50 % yield and at 50 % emergence are 3.6  $\text{dsm}^{-1}$  and 18  $\text{dsm}^{-1}$  of ECe, respectively (Wahhab, 1961).

Although, salinity affects all stages of the growth and development of rice plant and the crop responses to salinity varies with growth stages, concentration and duration of exposure to salt (Shereen et al., 2005). The symptoms of salt injury in rice are stunted growth, rolling of leaves, white tips, drying of older leaves and grain sterility. Soil salinity limits the rice plant's growth and development, resulting in yield losses of more than 50% (Zeng and Shanon, 2000). Sensitivity of rice to salinity stress varies with the growth stage. Though salinity affects all stages of the growth and development of the rice plant, when the rice is at the young seedling stage it becomes even more sensitive to salinity (Shereen et al., 2005; Deepa Sankar, Saleh and Selvaraj, 2011). Rad et al. (2012) also reported that the

primary stages of rice growth such as tillering and panicle initiation stages are more sensitive to salinity stress.

### **1.2.2 Amelioration practices used in salt affected area**

As salinity is one of the most serious abiotic stresses on rice growth of the main rice growing area especially in saline deltas of arid areas, it needs to find out low-cost, efficient treatment strategies to reduce the salt toxicity of soils and to improve the soil properties. There are many procedures that can be used to improve salt affected land, such as, water leaching, chemical remediation and phytoremediation (Ghafoor et al., 2008; Sharma and Minhas, 2005; Qadir et al., 2007; Feizi et al., 2010). The remediation of saline soil using Ca amendments such as ( $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ ), calcite ( $\text{CaCO}_3$ ), calcium chloride ( $\text{CaCl}_2$ ) and organic matter (farmyard manure, green manure, organic amendment and municipal solid waste), is a fruitful topic of investigation and can be applied worldwide, being low cost, effective and simple to implement (Mitchell et al., 2000; Hanay et al., 2004; Sharma and Minhas, 2005; Tejada et al., 2006; Makoi and Verplancke, 2010). Among those Ca amendments, gypsum is typically used as a source of calcium (Ca) (Shaaban et al., 2013). High  $\text{Ca}^{2+}$  concentrations can reduce the permeability of plasma membrane to  $\text{Na}^+$ . The reduction in membrane permeability to  $\text{Na}^+$  by  $\text{Ca}^{2+}$  reduces the accumulation of  $\text{Na}^+$  by passive influx (Cramer et al., 1985). The  $\text{Ca}^{2+}$  ions in the gypsum fertilizer will replace  $\text{Na}^+$  ions that adsorbed to the soil particles and  $\text{NaSO}_4$  will be reached away from the root zone.

In addition, the application of  $\text{Ca}^{2+}$  amendment can improve various soil properties and act as soil modifiers that prevent the development of sodicity which is directly related to plant growth, crop productivity and crop yields (Muhammad and Khattak, 2011). Many studies such as Ghafoor et al. (2001), Choudhary et al. (2004), and Wong et al. (2009) approved that the physical, chemical and biological properties of salt affected soil are improved by the application of gypsum and/or farmyard manure (FYM) as remediation for sustainable land usage and crop productivity, leading to enhanced plant growth and development. Since organic materials improve the soil physicochemical properties that accelerate exchange of cations on soil solids and leaching of salts from the root zone (Clark et al., 2007), hence preventing root from salt injuries and roots can grow more smoothly, gypsum is usually applied in combination to organic materials such as farmyard manure and humid acids etc. Ullah and Bhatti (2007) found that the conjunctive uses of farm manure with gypsum significantly improve soil physicochemical properties of sodic soils as compared to their alone application (Ullah and Bhatti 2007). Cha-um et al. (2011) observed that the application of gypsum and FYM to saline soil had 79.6% spikelet fertility, while rice grown in soil without the gypsum and FYM treatment had only 46.4%. Thus, this author suggested that gypsum and FYM treatment should effectively remedy the saline soil problem, resulting in yield improvement from paddy fields.

### 1.3 Global warming and methane gas emission

Methane ( $\text{CH}_4$ ) is the second most important greenhouse gas after carbon dioxide ( $\text{CO}_2$ ). Its global warming potential is 25 times greater than  $\text{CO}_2$ . It contributes to the greenhouse effect is about  $1.7 \text{ W m}^{-2}$  (Purkait et al., 2006) that accounts for 15 to 20% of global warming (IPCC, 1996). Moreover, global concentration of greenhouse gas  $\text{CH}_4$  has been increasing at the rate of  $\sim 1\%$  each year (IPCC, 2007). If the current rate of increase is maintained, in the next 50 years, it is estimated to contribute an additional  $0.5 \text{ W m}^{-2}$  in radiative heating (Purkait et al., 2006). Thus, global warming due to increasing greenhouse gases emission is current great environmental concern.

Methane is produced under anaerobic environments by obligate anaerobic microorganisms through either  $\text{CO}_2$  reduction or transmethylation processes (Hou et al., 2000). There are two main sources for  $\text{CH}_4$  production; anthropogenic sources and natural sources. More than 50% of global  $\text{CH}_4$  is related to human activities (US EPA, 2006). Anthropogenic sources include fossil fuel production, animal husbandry (enteric fermentation in livestock and manure management), rice cultivation, biomass burning, and waste management, etc. Among the anthropogenic sources, rice fields occupied 10% of global anthropogenic  $\text{CH}_4$  emission (IPCC, 2007) because of the flooding condition required for the rice cultivation. However, there are several factors influencing upon  $\text{CH}_4$  emission in paddy fields such as soil type, climatic conditions, agronomic practices including water and fertilizer

management, organic matter amendment, and application of pesticides, etc. (Zou et al., 2005; Xiong et al., 2007; Yan et al., 2005; Aulakh et al., 2001; Liesack et al., 2000).

### 1.3.1 Factors affecting methane emission

Methane emission from rice fields is affected by various factors such as prevalent weather conditions, water regime, various cultural practices, and soil properties (Jian et al., 2004). Among those factors, soil factor is largely influenced on CH<sub>4</sub> emission from paddy fields where rice plants are grown associated with specific management practices. The nature of paddy rice fields require for flooded condition. When soil is saturated with water, pH drops at first due to organic acid produced from fermentation of easily decomposable organic materials. Then, pH gradually starts to rise as a result of the buffering capacity of the soil (Kyuma, 2004). Methane production in flooded rice soil is very sensitive to pH and Setyanto et al. (2002) reported that the optimum pH of paddy soils required for methanogenic bacteria to produce CH<sub>4</sub> is around 6 to 6.6. At the same time, the redox potential (Eh) of flooded soils also decrease as a result of the production of reduced forms. The sequence of the redox reaction and approximate values for redox potentials associated with specific oxidation-reduction process are as follows (Kyuma, 2004);

<u>Eh values</u>	<u>Reactions</u>
+330 mV	$\frac{1}{2} O_2 + 2e^- + 2H^+ \rightarrow H_2O$ (by facultative anaerobes and aerobes)
+220 mV	$2NO_3^- + 12 H^+ + 10e^- \rightarrow N_2 + 6H_2O$ (by denitrifiers)

+200 mV                       $\text{MnO}_2 + 4\text{H}^+ + 2\text{e}^- \rightarrow \text{Mn}^{2+} + 2\text{H}_2\text{O}$  (by manganese reducing bacteria)

+120 mV                       $\text{Fe}(\text{OH})_3 + 3\text{H}^+ + 2\text{e}^- \rightarrow \text{Fe}^{2+} + 2\text{H}_2\text{O}$  (by iron reducing bacteria)

-150 mV                       $\text{SO}_4^{2-} + 10\text{H}^+ + 8\text{e}^- \rightarrow \text{H}_2\text{S} + 4\text{H}_2\text{O}$  (by sulfate reducing bacteria)

-250 mV                       $\text{CO}_2 + 8\text{H}^+ + 8\text{e}^- \rightarrow \text{CH}_4 + 2\text{H}_2\text{O}$  (by methanogens)

Based on this explanation,  $\text{CH}_4$  production can be occurred below the redox potential value (-150 mV) (Jain et al., 2004). As  $\text{CH}_4$  is an end product of microbial decomposition organic materials under anaerobic condition, the availability of soil organic carbon (SOC) is an important factor for  $\text{CH}_4$  production in rice fields. Jiao et al. (2005) proved that there was significant positive correlation between SOC and  $\text{CH}_4$  emission in rice soil. In addition, the decomposition rate of organic matter from which methanogenic substrates are produced is affected by soil temperature. In Japanese paddy fields, Hattori et al. (2001) recorded that the optimum temperature for  $\text{CH}_4$  production was 40 °C due to dominance of methanogenic population at this temperature. Methane formation was very slow at temperature below 20 °C (Yamane and Sato, 1961). Furthermore, availability of electron acceptors such as Fe (III),  $\text{SO}_4^{2-}$  is also an important indicator in  $\text{CH}_4$  emission in rice fields. Seasonal  $\text{CH}_4$  emission in Fe-amended pots was lower than the control treatment (Huang et al., 2009). Van der gon and Neue (1995) also proved that the application of  $\text{SO}_4^{2-}$  containing fertilizer to paddy fields reduced in  $\text{CH}_4$  emission. Beside these above soil factors, salinity is considered

as one of the soil factors to influence upon CH<sub>4</sub> emission by affecting soil microbial activities including methanogens (Pattnaik et al., 2000).

### **1.3.2 Impact of salinity on methane emission**

Biswas et al. (2006) have studied CH<sub>4</sub> emission from the saline rice fields of Sunderban mangroves of east coast of India and reported significant reduction of CH<sub>4</sub> emission from the rice fields reclaimed from mangrove swamp compared to upland rice fields and mangrove forest area. Supprattanapan et al. (2009) have studied the effect of organic amendment on CH<sub>4</sub> emission from saline rice soil and reported no significant difference in CH<sub>4</sub> emission from soils amended with organic matter as compared to unamended soils. Generally, salinity is known to inhibit CH<sub>4</sub> production. However, saline soils are composed of different levels of carbonates/bicarbonates, sulphates and chlorides of sodium (NaCl), potassium (KCl), calcium (CaCl<sub>2</sub>), and magnesium (MgCl<sub>2</sub>). These different components of salinity inhibit CH<sub>4</sub> production followed the order of sulphate > bicarbonate > chloride (Mishra et al., 2003). Thus, among these anion species, the addition of NaCl showed the least inhibition on CH<sub>4</sub> production and the reduction rate was 38% compared with control at pore water EC 8 dSm<sup>-1</sup> (Mishra et al., 2003). Some of the methanogenic strains showed inhibition only at high concentrations (>0.17 M), well above the concentrations normally found in saline rice soils (0.04-0.1 M) (Van der gon and Neue, 1995) because Na is required for amino acid transport, growth, methanogenesis, and internal pH regulation in methanogenic bacteria. The amount of sodium requirement is different among methanogens

including very high values for halophilic methanogens (Jarrell and Kalmokoff, 1988). The mechanism might be osmotic imbalance by affecting the cation-anion equilibrium that in turn may adversely affect the activity of methanogens and/or microorganisms producing substrates for methanogens.

### **1.3.3 Impact of gypsum amendment on methane emission**

For Na-saturated soil or water, gypsum fertilizer is a commonly suggested fertilizer as the nutrient  $\text{Ca}^{2+}$  source (Davitt et al., 1981; Aslam et al., 2001; Mahmood et al., 2009). The chemical composition of gypsum fertilizer ( $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ ) is 23.8%  $\text{Ca}^{2+}$  and 18% S. Regarding with the  $\text{CH}_4$  emission, sulphate ( $\text{SO}_4^{2-}$ ) containing fertilizers including gypsum fertilizer is frequently suggested as  $\text{CH}_4$  mitigation option. Van der gon and Neue (1994) proved that the application of gypsum ( $6.66 \text{ ton ha}^{-1}$ ) to rice paddy soil could reduce  $\text{CH}_4$  emission in the wet season. At the successive growing period, Van der gon and Neue (1995) studied the salinity effect on  $\text{CH}_4$  emission by adding NaCl to the same paddy fields which is amended with or without gypsum fertilizer. The authors observed that addition of NaCl reduced in  $\text{CH}_4$  emission. The reduction value of  $\text{CH}_4$  emission was considerably smaller in the rice paddies where the soils have low in  $\text{SO}_4^{2-}$  than that of  $\text{CH}_4$  emission where the soils have high in  $\text{SO}_4^{2-}$ . Nevertheless, there are still contradictory report on rice fields fertilized with  $\text{SO}_4^{2-}$  containing fertilizers such as ammonium sulphate ( $\text{NH}_4 (\text{SO}_4)_2$ ), calcium sulphate ( $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ ) etc. on  $\text{CH}_4$  emission;  $\text{CH}_4$  emission either increased (Cicerone and Shetter, 1981), stayed constant (Wassmann et al., 1993) or decreased (Schutz et al., 1989a; Van der

gon, 1995; Lindau et al., 1993). These contradicting results may be due to differences in C substrate and  $\text{SO}_4^{2-}$  availability at various field sites.

#### **1.4 The role of root exudates in methane emission**

Plant roots have ability to synthesize, accumulate and secrete a diverse array of compounds. More than 200 C compounds released from plant roots in the form of exudates (Bowen, 1999). On the basis of molecular weight, they can be grouped into high molecular-weight compounds such as mucilage, gelatinous material, covering root surfaces, ectoenzymes and low molecular-weight compounds such as organic acids, sugars, phenolics, amino acids, phytosiderophores, flavonoids and vitamins. Secretion of compounds into the rhizosphere is one of the most remarkable metabolic features of plant roots (Kumar et al., 2006). Cereals transfer 20–30% of total assimilated C into the soil, and half of this amount is subsequently found in the roots and about one-third is lost in the form of  $\text{CO}_2$  by root respiration and microbial utilization of root-borne organic substances (Gregory and Atwell, 1991).

The remaining part of underground translocated C is incorporated into the soil microorganisms and SOC. Researchers argue that photosynthetically fixed C in cereals and grasses is transported rapidly towards the root and can reach the external environment of the root within an hour (Kuzyakov, 1999). During one vegetative period, cereals and grasses allocated below the ground are about 1500 and 2200 kg C/ha respectively (Bowen, 1999).

This accounted for nearly 5 to 21% of all photosynthetically fixed C transferred to the rhizosphere through root exudates and ranged from 20 to 50% of plant biomass.

Rice plants play an important role in different levels of the CH<sub>4</sub> budget of rice fields. Various studies have reported about root exudates in regulating the CH<sub>4</sub> budget of rice field (Nakićenović, 2000). Wassmann and Aulakh (2000) reported that CH<sub>4</sub> production in rice fields largely depends on plant-borne material that can be either decaying tissue or root exudates. Holzapfel-Pschorn et al. (1986) observed that methanogenesis in submerged soil when rice plants were stimulated and concluded that a second peak observed in CH<sub>4</sub> emission at the end of tillering stage was due to the release of root exudates or products of root autolysis. By contrast, high in situ fluxes of CH<sub>4</sub> during the reproductive period of rice plants have been attributed to the supply of organic C from rice root exudation (Lindau et al., 1991; Minoda and Kimura, 1994; Neue and Sass, 1994). By using a <sup>13</sup>C-trace technique, Chidthaisong and Watanabe (1997) indicated that the contribution of plant-supplied organic material to the formation and emission of CH<sub>4</sub> sharply increase after heading. However, the amount of knowledge available on exudation from rice plants is minute and the literatures about the impact of quality and quantity of root exudation on CH<sub>4</sub> production and emission are also limited.

#### **1.4.1 Factors affecting root exudation**

The qualitative and quantitative composition of root exudates is affected by environmental factors such as pH, soil type, oxygen status, light intensity, soil temperature, plant growth, nutrient availability and microorganisms (Mimmo et al., 2011). Generally, it was reported that plant roots would excrete more metabolites and some unknown substances under environmental stress (López-Bucio et al., 2000). However, there is limited information about all possible factors influencing on root exudation of rice crop. Furthermore, there is also limited knowledge to explain how physiological process involved in exudation process. The possible factors influencing on quantity and composition of root exudates in different crops and different situation are summarized and described as follows.

##### **(a) Soil types**

Neumann et al., (2014) studied that the effect of three soil types on root exudation of the lettuce plant (*Lactucasativa* L. cv.Tizian (BBCH19)). Apart from variability in root development, huge quantitative differences particularly for sugars and amino acids were detected in the root exudates samples of lettuce plants. Mutual interactions between plant roots and soil- specific microbes seem to be important determinants for these quantitative differences of root exudation.

### **(b) Soil moisture**

Katznelson et al. (1954) found that temporarily wilting of plants greatly increased the release of amino acids from roots, which would be important under field conditions.

### **(c) Temperature**

Husain and McKeen (1963) found more amino acids in exudates of strawberry plants (*Fragaria vesca* L.) grown at 5 to 10 °C than at 20 to 30 °C; this markedly influenced the pathogenicity of *Rhizoctonia fragariae* which attacks strawberries at low soil temperature.

### **(d) Nutrient deficiency**

Lilia et al. (2012) compared the root exudates from axenically grown maize plants exposed to nitrogen (N), potassium (K), phosphate (P) or iron (Fe) deficiency. The results showed that a higher release of glutamate, glucose, ribitol, and citrate from Fe-deficient plants. Phosphate deficiency stimulated the release of C-amionobutyric acid and carbohydrate. Potassium-starved plants released less sugar, in particular glycerol, ribitol, fructose, and maltose, while under N deficiency lower amounts of amino acids were found in root exudates.

### **(e) Ion toxicity**

According to Ma and Furukawa (2003), plant roots under Al stress might exude various organic acids, including citric and Malic acids, and these organic acids could play a very important role in alleviating aluminium toxicity in soil.

### **(f) Plant species**

Within same rice species but among 4 different cultivars including Supanburi 1, Supanburi 60, Supanburi 90 and Chainat 1, Pusatjapong (2003) observed higher quantity of sugars and organic acids in the root exudates of Supanburi 60 and Supanburi 90 compared to the rest 2 rice cultivars.

Between legume and grass species, the use of <sup>15</sup>N-labelled amino acids has shown that efflux of glycine and serine from roots of legumes is higher than from roots of grasses among amino acids found in root exudates (Lesuffleur et al., 2007).

### **(g) Microorganisms**

Shi et al. (2011) studied the effects of artificial root exudates solutions, based on those collected *in situ*, on the community composition and diversity of metabolically active soil bacteria in a soil microcosm experiment. This study showed that organic acids caused significantly greater increases than sugars in the detectable richness of the soil bacterial community. Therefore, organic acids play a significant role in shaping soil bacterial communities.

Understanding the factors that affect on root exudation in various crops may insight an effective approach for increasing crop production and mitigating CH<sub>4</sub> emission from paddy rice soil since root exudation is known as the significant C cost for crop production and C pool for CH<sub>4</sub> production in paddy rice cultivation.

#### **1.4.2 Function and mechanism of organic acids in plant**

Based on the reference of (López-Bucio, 2000), organic acid metabolism is of fundamental importance at the cellular level for several biochemical pathways, including energy production, formation of precursors for amino-acid biosynthesis and at the whole plant level in modulating adaptation to the environment. A striking feature of plant tissues is that the total content of organic acids is higher than in other organisms. The high accumulation of organic acids in plant tissues is most probably due to their important role as photosynthetic intermediates. However, organic acids have a potential role as metabolically active solutes for the osmotic adjustment and the balance of cation excess. Organic acids also participate as key components in the mechanisms that some plants use to cope with nutrient deficiencies, metal tolerance and plant–microbe interactions operating at the root–soil interface (López-Bucio, 2000). Organic acids are mainly produced in mitochondria through the tricarboxylic acid or Krebs cycle and to a lesser extent in the glyoxysome as part of the glyoxylate cycle. Because of the catalytic nature of the Krebs cycle, organic acids are present only in very small pools in the mitochondria and are preferentially stored in the vacuole (López-Bucio, 2000).

## **Chapter 2**

### **Influence of different levels of saline irrigation water on methane production and emission in paddy rice soil with or without addition of gypsum fertilizer**

#### **The objectives of this study are;**

1. To clarify the influence of irrigation water salinity levels on CH<sub>4</sub> emission from paddy rice soil
2. To evaluate the effect of gypsum fertilizer as Ca<sup>2+</sup> on CH<sub>4</sub> emission in relation to plant growth.

## **2.1 Materials and methods**

### **2.1.1 Laboratory Incubation Experiment**

#### **(a) Effect of salinity levels on methane production potential of soil**

A composite soil sample was collected from a depth of 0–15 cm at FM Fuchu Honmachi, Field Science Center of the Tokyo University of Agriculture and Technology. The soil was kept in a refrigerator for one week until the incubation experiment. The background salinity level of soil (EC 1:1) was  $0.23 \text{ d Sm}^{-1}$ . The methane production potential of this soil was studied at 5 salinity levels: control, 10 (S10), 30 (S30), 60 (S60) and 90 (S90)  $\text{mmol L}^{-1}$  NaCl. The objective of this study was to evaluate the  $\text{CH}_4$  production potential of soil by the addition of different concentration of NaCl solution. Twenty gram of wet soil was put into 100-ml conical flasks and then flooded with 20 ml of tap water for the control or with different levels of saline water for the salinity treatments. The conical flasks were fitted with rubber stoppers possessing two tubes to facilitate flushing with nitrogen gas ( $\text{N}_2$ ) and the collection of gas samples. These flasks were kept in an incubator at  $30^\circ\text{C}$  for 21 days. There were 4 replications for each salinity level.

Three replications were used to collect the  $\text{CH}_4$  gas samples and another one replication was used to monitor the changes of soil pH under each salinity level during the incubation period by using Beckman,  $\Phi$  260 pH/ Temp/ mV meter. To check the redox potential value (Eh) of incubated soil whether it could support for anaerobic condition or not, the extra one control sample was prepared and platinum Eh probe was permanently inserted

into that soil. The value of Eh was recorded by using SWC-201RP, Sanyo water checker on each sampling day. Gas samples were collected on 2, 4, 7, 14 and 21 days after incubation.

The flushing with N<sub>2</sub> gas was carried out one day before and immediately after the sampling at 250 ml min<sup>-1</sup> for 3 minutes. The conical flasks were shaken just before sampling to release the CH<sub>4</sub> entrapped within the soil. Gas samples were drawn with 20 ml plastic disposable syringe fitted with a needle and a three-way valve to prevent diffusion of the gas sample. During sampling, the air in headspace of each conical flask was thoroughly mixed by pushing the plunger of the syringe up and down for ten times at first, and then about 5 ml of gas sample was collected. After collecting the samples, the beakers were flushed with nitrogen gas immediately at 250 ml per minute for 3 seconds. Methane gas samples were analyzed by using a gas chromatograph (GC-8A, Shimadzu, Kyoto, Japan) with a flame ionization detector (FID). The GC machine was equipped with a Porapak N (80/100 mesh) column and used He gas as carrier gas. The temperature of column, and injector were maintained at 60 °C and 180 °C. The amount of CH<sub>4</sub> flux was calculated by using the following equation;

$$C_i = C_a \times V_{hs} \times MW \times MV / W_S \times d \times 1000 \quad (1)$$

Where C<sub>i</sub>: CH<sub>4</sub> production (µgCH<sub>4</sub>/g soil /day)

C<sub>a</sub>: CH<sub>4</sub> concentration in the head space (ppm)

$V_{hs}$ : the volume of head space (ml)

MW: the molecular mass of  $CH_4 = 16.123 \text{ mg mmol}^{-1}$

MV: the molar volume of  $CH_4$  at  $30^\circ\text{C} = \text{mmol mL}^{-1}$

$W_s$ : the wet weight of soil (g)

### **2.1.2 Pot experiment**

Pot experiment was conducted in the open field of FM Fuchu Honmachi, Field Science Center of the Tokyo University of Agriculture and Technology, Fuchu, Tokyo, Japan from 15<sup>th</sup> June 2010 to 30<sup>th</sup> September 2010.

#### **(a) Preparation of soil and cultivation of rice**

The soil used in pot experiment was also collected from paddy fields in FM Fuchu Honmachi, Field Science Center of the Tokyo University of Agriculture and Technology. The physico-chemical properties of each soil are shown in Table 1 (Tanaka et al., 2008). A salt tolerant Indica rice variety, Dorfak cultivar which is originated from Iran, was used in this study for both experiments.

About 8 kg of soil was placed in  $30 \times 20$  cm plastic pots. About one week before transplanting, puddling was conducted by irrigating the pots twice on alternate days with specific salinity levels of NaCl solutions or tap water (control). The chemical fertilizers such as  $40 \text{ kg P ha}^{-1}$  and  $70 \text{ kg K ha}^{-1}$  were applied one day before transplanting.  $70 \text{ kg N ha}^{-1}$  was

applied in 3 equal splits at basal (one day before transplanting), active tillering stage (2 weeks after transplanting) and panicle initiation stage (4 weeks after transplanting). Urea, ammonium phosphate and potassium chloride were used as a source of N, P and K, respectively.

Two salinity levels 30 mmol L<sup>-1</sup> NaCl (S30), and 90 mmol L<sup>-1</sup> NaCl (S90) were tested in this pot experiment. The treatments comprised 1) control (tap water), 2) 30 mmol L<sup>-1</sup> NaCl (S30), 3) 90 mmol L<sup>-1</sup> NaCl (S90), 4) gypsum 1 ton ha<sup>-1</sup> (G1), 5) 30 mmol L<sup>-1</sup> NaCl plus gypsum 1 ton ha<sup>-1</sup>(S30-G1), and 6) 90 mmol L<sup>-1</sup> NaCl plus gypsum 1 mmol L<sup>-1</sup> ha<sup>-1</sup>(S90-G1). All treatments were laid out in a completely randomized design with 3 replications. Twenty-one-day-old seedlings were transplanted with 3 seedlings per pot on 23<sup>rd</sup> June, 2010 and harvested on 30<sup>th</sup> September, 2010. A water level of about 2–3 cm was maintained in pots throughout the growing seasons by irrigating regularly with the assigned NaCl concentration until crop maturity. The purposes of this experiment was to examine the effect of those saline levels (S30 and S90) upon CH<sub>4</sub> emission related with rice growth and to evaluate the CH<sub>4</sub> emission by applying gypsum amendment that is commonly used into saline soil, to Na-salinized paddy soil under rice cultivation.

#### **(b) Measured parameters and analytical methods**

Data regarding agronomic characters such as plant height, tiller number, number of dead leaves, and soil environment were collected at weekly interval. The pH and EC of the flooded water was directly measured by portable meters (Beckman, Φ 260 pH/ Temp/ mV

meter, and ES-51 COND METER, Horiba, Japan, respectively). The Eh value of soil was monitored by inserting platinum electrodes permanently into the soil during the rice growing season at a depth of 5 cm in each plot using Eh probes (SWC-201RP, Sanyo, Japan). Soil temperature was also measured at 5cm depth by inserting OPTEX Thermometer TBW-3 at each sampling time. Flowers and Flowers (2005) pointed that salinity affects on plants by lowering water potentials and interfering the uptake of essential nutrients including N. Furthermore, as there is an interactive effect between  $\text{NH}_4^+$  and  $\text{Na}^+$  and/or between  $\text{Cl}^-$  and  $\text{NO}_3^-$ , surface water samples were also collected at weekly intervals for the analysis of ammonium ( $\text{NH}_4^+$ ) and nitrate ( $\text{NO}_3^-$ ) ions concentrations (Hu and Schmidhalter, 2005). Both  $\text{NH}_4^+$  and  $\text{NO}_3^-$  concentrations were determined by using a UV spectrophotometer (UV-VI Mini 1240, Shimadzu Corporation, Kyoto, Japan). Before analysis of these ions, water samples were filtered with 0.45- $\mu\text{m}$  filter paper. For  $\text{NH}_4^+$  analysis, 5 ml of water samples, 2 ml of solution A and 3 ml of solution B were mixed and allowed to stand for about 45 minutes. After 45 minutes,  $\text{NH}_4^+$  was determined at the wavelength 630 nm. Solution A contains the chemicals such as 30g of  $\text{Na}_2\text{HPO}_4 \cdot \text{H}_2\text{O}$ , 30g  $\text{C}_6\text{H}_5\text{Na}_3\text{O}_7 \cdot 2\text{H}_2\text{O}$ , 3g of  $\text{EDTA} \cdot 2\text{Na}$ , 60 g of Phenol 60 g and 0.02 g of  $\text{Na}_2\text{Fe}(\text{CN})_5\text{NO} \cdot 2\text{H}_2\text{O}$  which is prepared to get the final volume 1 Liter by adding distilled water. Solution B contains 16 g of  $\text{NaOH}$ , 20 ml of  $\text{NaClO}$  which is prepared to get the final volume 1Liter by adding distilled water. For  $\text{NO}_3^-$  analysis, no reagent was added to the 5-ml water samples and measurements were made directly at the wavelength 230 nm.

### **(c) Gas sampling, analysis and calculation**

Gas sampling was taken at 09:00 – 12:00 am by a closed-chamber method (Lu et al., 1999). The chambers used were 100 cm in height, 30 cm in length, 30 cm in width, and made of acrylic transparent sheets. A plastic tray with a length of 40 cm, width of 40 cm, and height of 5 cm was filled with 3 cm of water and placed under the pot. The chamber was put into the tray by covering the pot, and the tray water sealed the surrounding area of the chamber to form an airtight chamber. A battery-operated fan and Tedlar bag were installed at the chamber to mix the air inside the chamber and regulate the pressure, respectively. The temperature inside the chamber was recorded using a micro-temperature thermometer (PC-9125, AS ONE Co., Tokyo, Japan) fitted with rubber septum inserted into the small hole of the chamber. To assess the linear rate increase of gas concentration emitted from the surface area of soil inside the pot with time, gas samples were drawn from the chambers through a three-way stop cock using a 50-ml airtight syringe at intervals of 15 min (0, 15 and 30 min). The air inside the chamber was thoroughly mixed before collecting gas samples by flushing the syringe 3 times. Approximately 45 ml of gas samples were then taken with the 50-ml plastic syringe, adjusted to 40 ml and then transferred into a 20-ml pre-vacuumed glass vial. The amount of CH<sub>4</sub> flux was calculated using the following equation:

$$Q = (V/A) \times (\Delta c/\Delta t) \times (M/22.4) \times (273/K) \quad (2)$$

Where Q: flux of the CH<sub>4</sub> gas (mg m<sup>-2</sup> min<sup>-1</sup>)

V: volume of gas chamber

A: gas emitted surface area (m<sup>2</sup>)

Δc: increased or decreased change of the gas concentration (mg m<sup>-3</sup>)

Δt: unit of time intervals (min)

M: molar weight of the gas

K: Kelvin temperature of air inside the chamber

The cumulative amount of CH<sub>4</sub> gas measured during one crop season was calculated by multiplying the daily gas flux at each measurement with the time interval and then adding all of these values. All the data were evaluated by an analysis of variance (ANOVA) by using CropStat 7.0 statistical software. Comparison of treatment means was performed using least significant differences (LSD) at p=0.05. Correlation and regression analysis was performed using Sigma Plot 11.0 statistical software.

## **2.2 Results**

### **2.2.1 Laboratory incubation experiment**

#### **(a) Effect of salinity levels on methane production potential of soil**

The addition of NaCl kept the pH lower than that of the non-saline control (Table 2.2). The Eh values were in the range -320 to -435 mV. The addition of NaCl up to S30 increased CH<sub>4</sub> production. The increase of CH<sub>4</sub> production in salinity S10 and S30 was 1.2 and 2 times

higher than that of control, respectively, although they were not significantly different. At higher salinity levels (S60 and S90), CH<sub>4</sub> production was 19 to 33% lower than the control.

## **2.2.2 Pot Experiment**

### **(a) Soil environment during the rice growing season**

Soil temperature throughout the growing season ranged from 20 to 35°C (Fig. 2.1b). There was no large difference in temperature between the treatments. The desired EC level was maintained by adding saline water or tap water. However, the desired EC level during the first 3 weeks could not be achieved due to continuous rainy days (Fig. 2.1c). Soils of saline condition showed higher EC values compared to those of non-saline conditions (Fig. 2.1c). High fluctuation of pH was observed among the treatments until 5 weeks after transplanting (Fig. 2.1d). The pH was then between 6 and 7.4 throughout the subsequent growing period.

### **(b) Impact of different levels of irrigation water salinity on rice growth with or without addition of gypsum fertilizer**

Continuous irrigation with saline water severely suppressed rice growth for both S30 and S90 salinity levels. A significant difference in the level of growth suppression was observed between S90 and S30 from immediately after transplanting because rice growth was much more seriously affected by salinity level S90 compared to S30 and the non-saline control in terms of tillering pattern (Fig. 2.3a) and above-ground plant biomass yield (Table

2.3). The survival period of rice plants in S90 was until 7 weeks after transplanting (Fig. 2.3a) and the lowest number of dead leaves was observed in S90 (Table 2.3). Rice plants in S30 survived until the harvesting period, but the above-ground plant biomass yield in S30 was significantly lower and the total number of dead leaves was significantly higher compared to the non-saline control (Table 2.3). The addition of gypsum as a source of  $\text{Ca}^{2+}$  under continuously flooded conditions with saline water suppressed the above-ground plant biomass yield significantly compared with the non-saline condition with or without the addition of gypsum.

### **(c) Impact of different levels of irrigation water salinity on methane emission**

#### **with or without addition of gypsum fertilizer**

A distinct higher  $\text{CH}_4$  flux was observed during the reproductive stage of rice growth in both non-saline and saline conditions (Fig. 2.4a, b). Methane emissions gradually increased during the reproductive stage (Fig. 2.4a), while the rate of dead leaves also increased about 63 days after transplanting and the tiller numbers decreased after the maximum tillering period at about 56 days after transplanting. The daily  $\text{CH}_4$  fluxes under saline conditions were lower than those under non-saline conditions, especially during the vegetative stage of rice growth. Under saline conditions, the highest total  $\text{CH}_4$  emission (296  $\text{kg CH}_4 \text{ ha}^{-1}$ ) was found in S30 without gypsum amendment. However, the amount of  $\text{CH}_4$  emission in S30 without gypsum amendment was 6% lower than that of the non-saline control, although the difference was not significant. The lowest  $\text{CH}_4$  emission among the

treatments was observed in S90 with (55.5 kg CH<sub>4</sub> ha<sup>-1</sup>) or without gypsum amendment (44.2 kg CH<sub>4</sub> ha<sup>-1</sup>). The addition of gypsum suppressed CH<sub>4</sub> emission under saline and non-saline conditions throughout the rice-growing period. The reduced rates due to gypsum addition were about 56% for non-saline conditions and 53–83% for saline conditions (S30-G1, and S90-G1) compared to CH<sub>4</sub> emission in the non-saline control (Fig. 2.5a). Although rice plants are known as major CH<sub>4</sub> transport conduits, the number of tillers in S30-G1 was significantly lower than that in G1, even though the level of the reduced CH<sub>4</sub> emission rate did not differ significantly. However, the total number of dead leaves, which can contribute as a source of organic carbon (C) for methanogens, was higher in S30-G1 than G1 (Table 2.3).

## **2.3 Discussion**

### **2.3.1 Laboratory incubation experiment**

#### **(a) Effect of salinity levels on methane production potential of soil**

The higher CH<sub>4</sub> production in S10 and S30 compared with non-saline control might be due to the Na<sup>+</sup> ion concentration of those salinity levels which meet the Na<sup>+</sup> requirement of methanogenic bacteria. Ramakrishnan et al. (1998) also found that the addition of 27 mM NaCl to illuvial soil caused a two fold increase in CH<sub>4</sub> production over that of control and higher addition of NaCl (54, 135 and 274mM) causing about 50% reduction in CH<sub>4</sub> production. According to the report of Jarrell and Kalmokoff (1988), sodium is required for amino acid transport, growth, methanogenesis, and internal pH regulation in methanogenic

bacteria. However, the quantum of sodium requirement varies widely among methanogens.

The lower CH<sub>4</sub> production over S30 level (S60 and S90) might be due to Cl<sup>-</sup> toxicity under higher additions of NaCl.

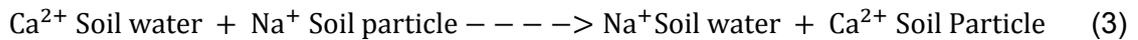
### **2.3.2 Pot experiment**

#### **(a) Impact of different levels of irrigation water salinity on rice growth with or without addition of gypsum fertilizer**

The concentration of salts in normal soil is generally lower than that in the cells of plants' roots. Water is absorbed by plant roots through a process called osmosis, which involves movement of water from soil with lower concentration of salts to a place that has a higher concentration of salts, inside the cells of plants' roots (Alex, 2006). In the pot experiment, the maintaining of continuous flooded condition with specific salinity level either S30 or S90 inhibited the rice growth by lowering the tiller numbers, above-ground plant biomass yield and enhancing the numbers of dead leaves (Table 2.3). The addition of saline water with salinity levels (S30) or (S90) may have disturbed the uptake of water by plants. Moreover, higher concentration of NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> in flooded water was observed in saline condition compared with non-saline condition especially in early stage of vegetative growth (3 weeks after transplanting) (Fig. 2.2a and b). Neves-Piestun and Bernstein (2001), and Homaei (2002) pointed that as the salinity and osmotic pressure increase, this process leads to less water and nutrient absorb including N and potassium by plants hence the plants growth will get stunted. Thus, this higher NO<sub>3</sub><sup>-</sup> concentration (especially in S90) and NH<sub>4</sub><sup>+</sup>

concentration of flooded water under saline condition might be due to less water and N nutrient uptake by the plants under osmotic stress. Although N fertilizer was applied as the  $\text{NH}_4^+$  form in this experiment, the existence of  $\text{NO}_3^-$  in flood water might be contributed from the rain water or the conversion of  $\text{NH}_4^+$  to  $\text{NO}_3^-$  through nitrification process in the flood water. In comparison of S30 and S90, higher  $\text{NH}_4^+$  and  $\text{NO}_3^-$  concentrations of S90 indicated that the osmotic stress in S90 levels was much more severe than that in S30. Thus, the shortest growth duration was observed in S90. Therefore, it can be assumed that the growth reduction under saline condition might be due to the osmotic stress, and interference with the uptake of essential nutrient such as N.

Although many studies showed that the addition of  $\text{Ca}^{2+}$  has beneficial effects either in saline soil (Cha-um et al., 2011) and saline solution with hydroponic culture (Song and Fujiyama, 1996), the present study did not show any improvement in rice growth in terms of tiller numbers or total number of dead leaves following irrigation with saline water (Table 2.3). When irrigation water with a high  $\text{Na}^+$  concentration relative to divalent cations is used, there will be a great tendency for adsorption of exchangeable  $\text{Na}^+$  around soil particles (Alobaidy et al., 2010). Naeem and Qureshi (2005) suggested that an adequate amount of  $\text{Ca}^{2+}$  in proportion to  $\text{Na}^+$  must be present in the system to improve the selectivity of ion. The purpose of adding gypsum was to replace the monovalent  $\text{Na}^+$  ion by increasing the  $\text{Ca}^{2+}$  ion concentration in the root zone and to maintain appropriate Na/Ca concentration in the root zone as shown in the following equation:



The use of gypsum ( $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ ) as a source of  $\text{Ca}^{2+}$  is a well-established practice for the amelioration and management of  $\text{Na}^+$ -saturated waters/soils (Mathad and Hiremath, 2010). Soil water laden with those Na ions is then drained or leached out of the soil. However, the present study was conducted as pot experiment and the depth of a pot was only 30 cm. The water outlets of the pots were also covered with rubber stoppers during the whole rice-growing period, and there was no space for regular leaching of the soil solution with  $\text{Na}^+$  ions as occurring in ordinary field conditions. Therefore, the regular addition of saline water could not provide sufficient  $\text{Na}^+/\text{Ca}^{2+}$  in the root zone, and a clear effect due to Ca amendment was not found.

**(b) Impact of different levels of irrigation water salinity on methane emission  
with or without addition of gypsum fertilizer**

In the pot experiment, the results showed that a salinity level of  $90 \text{ mmol L}^{-1} \text{ NaCl}$  (S90) suppressed  $\text{CH}_4$  emission (Fig. 2.5) through a lower  $\text{CH}_4$  production potential as shown in the laboratory incubation experiment (Table. 2), and a lower biomass yield (Table. 2.3) and shorter growth duration period (Fig. 3). Although the extent of above biomass in S30 was 49% lower than that of non-saline control (Table. 2.2), the amount of  $\text{CH}_4$  emission in S30 was only 6% lower than that of non-saline control that was also not statistically

significantly different (Fig. 5). Therefore, S30 did not suppress CH<sub>4</sub> production potential of soil. The addition of saline irrigation with salinity level S30 might provide the Na required by methanogens, thereby favouring CH<sub>4</sub> production as shown in laboratory incubation experiment (Table. 2.2). However, the lower value of CH<sub>4</sub> emission in S30 over that of the non-saline control under rice cultivation might relate with its lower above biomass yield. The observed reduction of CH<sub>4</sub> emission under gypsum application might be due to its high SO<sub>4</sub><sup>2-</sup> content (Fig. 2.5). In the presence of SO<sub>4</sub><sup>2-</sup>, SO<sub>4</sub><sup>2-</sup>-reducing bacteria will out compete methanogens for the same substrates such as hydrogen (H<sub>2</sub>) or carbon dioxide (CO<sub>2</sub>) or Acetate (CH<sub>3</sub>COO<sup>-</sup>) that methanogens use in methane production (Lindau et al., 1993; Van der gon and Neue, 1994; Epule et al., 2011). Furthermore, Gauci (2004) mentioned that SO<sub>4</sub><sup>2-</sup>-reducing bacteria have a higher affinity for both H<sub>2</sub> and CH<sub>3</sub>COO<sup>-</sup> than methanogens, which enables them to maintain the pool of these substrates at concentrations too low for methanogens to use. Therefore, the reduction of CH<sub>4</sub> emission following the addition of gypsum (Fig. 2.5) might be due to the competition between SO<sub>4</sub><sup>2-</sup>-reducing bacteria and CH<sub>4</sub>-producing bacteria for substrates as stated by Lindau et al. (1993), Van der gon and Neue (1994) and Epule et al. (2011).

**(c) Relationship between soil parameters, rice growth, and gypsum fertilizer addition and methane emission**

There was positive relationship between soil temperature and CH<sub>4</sub> emission (p=0.614\*\*) because temperature plays an important role in the rate of activity of soil

microorganisms including those involved in CH<sub>4</sub> production and consumption. According to the Van Hulzen et al. (1999), temperature influences CH<sub>4</sub> production by regulating anaerobic carbon mineralization, availability of alternative electron acceptors and methanogenic activity. At higher temperature, mineralization increases and more carbon substrate becomes available, result in faster depletion of the alternative electron acceptor pool. Moreover, Temperature also CH<sub>4</sub> transport through the rice plant (Hosono and Touchi, 1997)

As many studies pointed that the rice plants are the main conduit for CH<sub>4</sub> transport through well-developed system of inter cellular air space (aerenchyma), the above plant biomass yield may play a significant role in CH<sub>4</sub> emission (Schutz et al., 1989). Therefore, the positive relationship were observed between total CH<sub>4</sub> emission ( $\rho = 0.704^{**}$ ) and dry matter yield, and between total CH<sub>4</sub> emission and tiller numbers per pot ( $\rho = 0.819^{**}$ ). There was negative relationship between total CH<sub>4</sub> emission and gypsum fertilizer application although it was not significantly different.

## **2.4 Conclusion**

Our study showed that saline conditions do not suppress CH<sub>4</sub> emission, but favour CH<sub>4</sub> emission since Na<sup>+</sup> is required for methanogens in terms of the CH<sub>4</sub> production potential. However, excessive addition of NaCl (>30 mmol L<sup>-1</sup>NaCl) can inhibit CH<sub>4</sub> production. Although gypsum was added as Ca<sup>2+</sup> source to maintain appropriate Na<sup>+</sup>/Ca<sup>2+</sup> in the soil, no improvement of rice growth due to gypsum addition in saline condition might be due to the

high  $\text{Na}^+/\text{Ca}^{2+}$  ratio under addition of saline water into the limited space of pot. The amendment with gypsum resulted in a reduction of  $\text{CH}_4$  emission. Sulfate-containing fertilizer can be used as a  $\text{CH}_4$  mitigation option either in saline or non-saline conditions. In gypsum treatments, the extent of  $\text{CH}_4$  emission depended not only on the  $\text{SO}_4^{2-}$  content of the soil, but also on the availability of a C source from dead plant materials. Finally, it can be concluded that  $\text{CH}_4$  emission under saline conditions was influenced by salinity levels and rice growth in relation to nutrient absorption.

**Table 2.1** Soil properties of Honmachi field (Tanaka et al., 2008)

<b>Parameters</b>	<b>Values</b>
Soil Type	Inceptisols
Soil Texture	Sandy loam
Clay %	16%
Silt %	33%
Sand %	51%
pH	6.0
Total N	3.6 g kg <sup>-1</sup>
Available P <sub>2</sub> O <sub>5</sub>	51.8 g kg <sup>-1</sup>
Available K <sub>2</sub> O	198.3 mg kg <sup>-1</sup>
CEC	19.8 cmol kg <sup>-1</sup>

**Table 2.2** Results of laboratory incubation experiment

Treatments	Soil pH		Methane production potential of soil ( $\mu\text{g CH}_4/\text{g soil/day}$ )
	(2DAI)	(21DAI)	
Control	6.4	7.0	130 $\pm$ 26
S10	6.3	6.7	165 $\pm$ 57
S30	6.0	6.0	200 $\pm$ 57
S60	6.0	6.0	105 $\pm$ 32
S90	6.0	6.0	100 $\pm$ 21

Values are the means  $\pm$  standard deviation, n=3. ns stands for no statistically significant difference among the treatments within each column. DAI means days after incubation.

**Table 2.3** Effect of different salinity levels and gypsum fertilizer addition on rice growth

<b>Treatments</b>	<b>Maximum Tiller numbers (Numbers pot<sup>-1</sup>)</b>	<b>Numbers of dead leaves (Numbers pot<sup>-1</sup>)</b>	<b>Above ground plant biomass yield (g pot<sup>-1</sup>)</b>
Control	67 ± 1.7 <sup>a</sup>	143 ± 8.9 <sup>b</sup>	177± 8.0 <sup>a</sup>
G1	64 ± 0.6 <sup>b</sup>	141 ± 9.8 <sup>b</sup>	174± 15.0 <sup>a</sup>
S30	60 ± 0.6 <sup>c</sup>	223 ± 11.6 <sup>a</sup>	91 ± 2.6 <sup>b</sup>
S90	14 ± 1.7 <sup>f</sup>	54 ± 2.1 <sup>d</sup>	0 <sup>c</sup>
S30-G1	52 ± 4.0 <sup>d</sup>	208 ± 9.5 <sup>a</sup>	64± 2.6 <sup>b</sup>
S90-G1	19 ± 0.0 <sup>e</sup>	85 ± 11.0 <sup>c</sup>	0 <sup>c</sup>

Values are the means ± standard deviation (n= 3 replications). In each column, means followed by a common letter are not significantly different by using least significant difference (LSD) at p=0.05.

**Table 2.4** Results of ANOVA test for CH<sub>4</sub> emission and crop growth

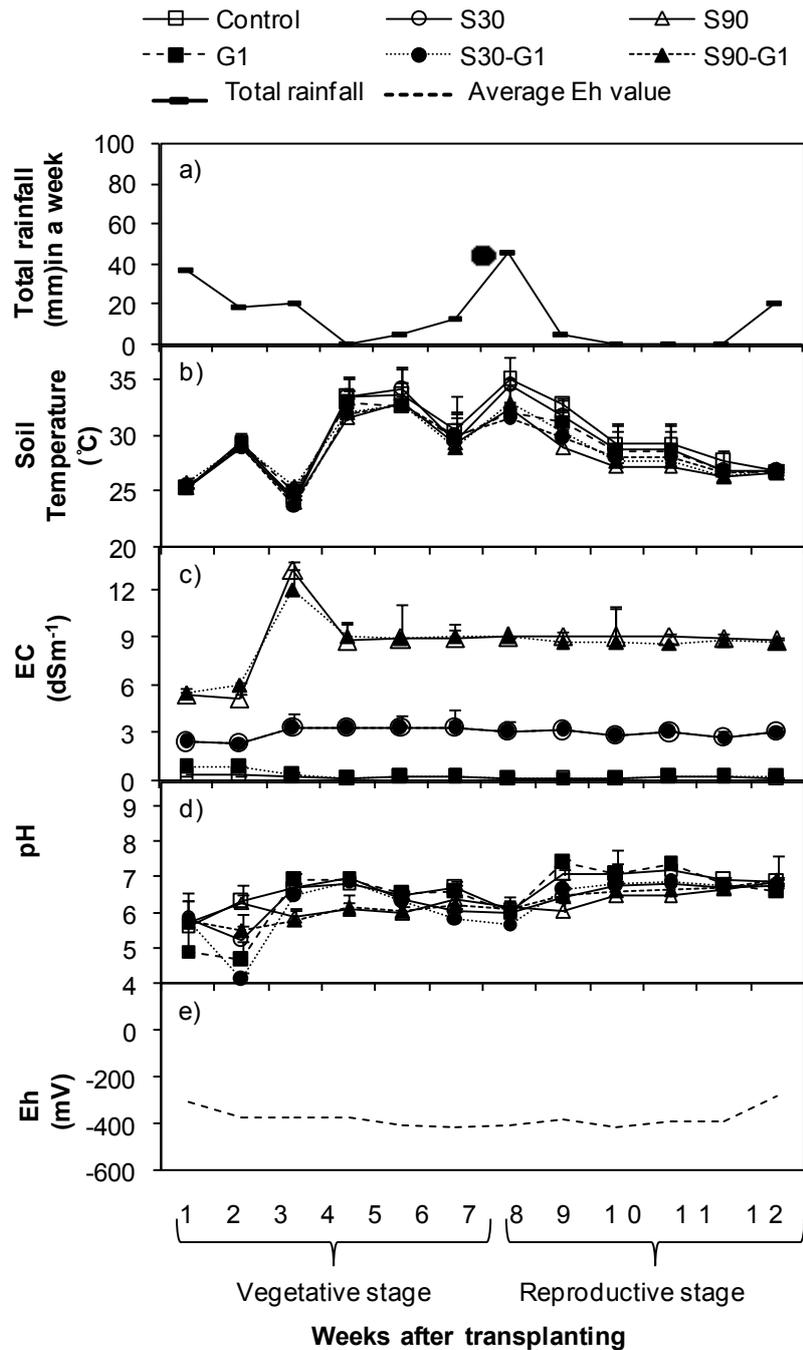
<b>Pot experiment</b>				
<b>Source of variation</b>	<b>Maximum tiller numbers (numbers pot<sup>-1</sup>)</b>	<b>Total numbers of dead leaves (numbers pot<sup>-1</sup>)</b>	<b>Above biomass yield (g pot<sup>-1</sup>)</b>	<b>CH<sub>4</sub> emission in one crop season (kg CH<sub>4</sub> ha<sup>-1</sup>)</b>
<b>Salinity</b>	**	**	**	**
<b>Gypsum application</b>	**	ns	ns	**
<b>Salinity* Gypsum application</b>	**	*	ns	**

\*\* , \* and ns stand for significance at 1%, 5% and non-significance respectively.

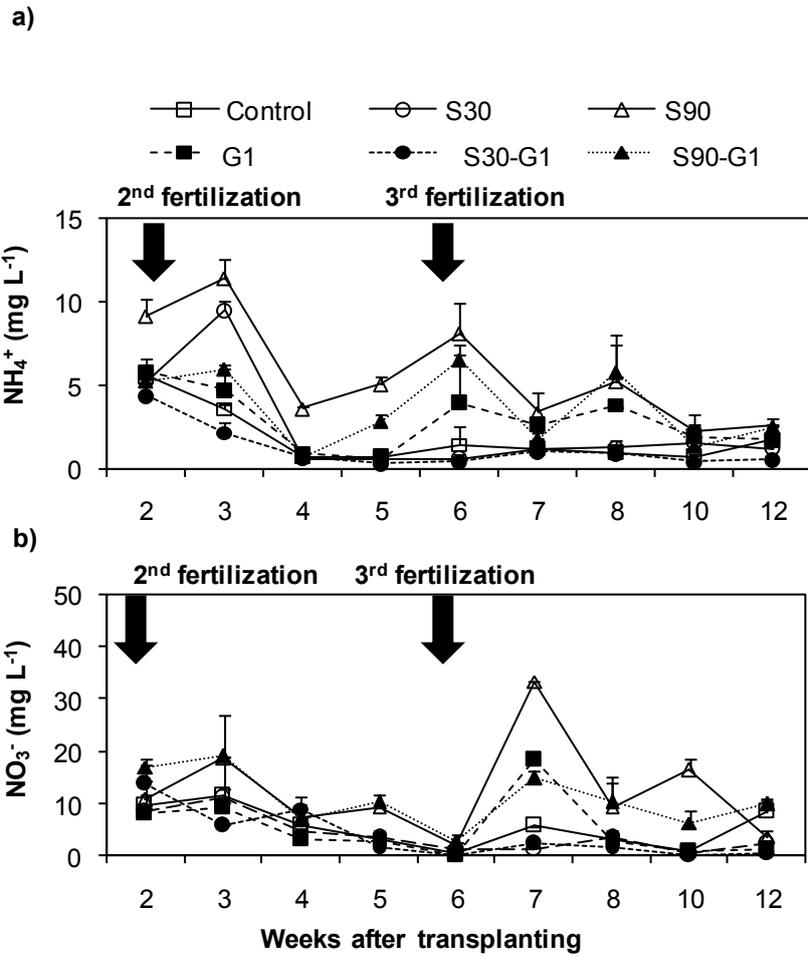
**Table 2.5** Spearman rank order correlation analysis

Pot experiment							
Parameters	Gypsum application	pH	EC	Soil Temp (°C)	Maximum Tillers (numbers pot <sup>-1</sup> )	Total numbers of dead leaves (numbers pot <sup>-1</sup> )	Above biomass yield (g pot <sup>-1</sup> )
Methane emission per crop season (kg CH <sub>4</sub> ha <sup>-1</sup> )	-0.332 <sup>ns</sup>	-0.278 <sup>ns</sup>	0 <sup>**</sup>	0.614 <sup>**</sup>	0.819 <sup>**</sup>	0.0836 <sup>ns</sup>	0.704 <sup>**</sup>

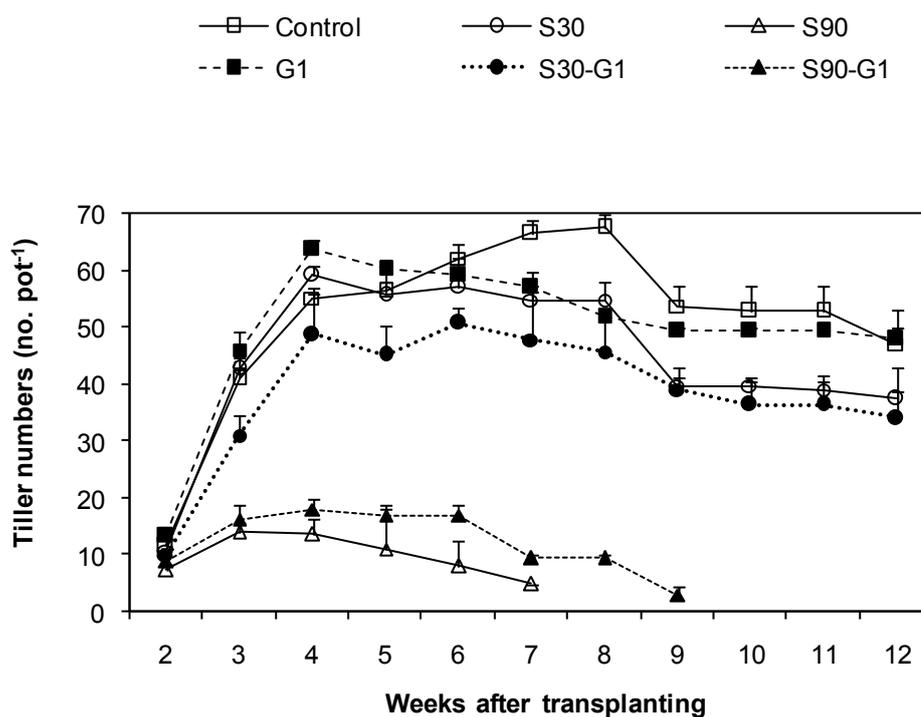
<sup>\*\*</sup>, <sup>\*</sup> and <sup>ns</sup> stand for significance at 1%, 5% and non-significance respectively.



**Fig.2.1** Distribution of rainfall and changes of soil environment under different salinity levels with or without addition of gypsum fertilizer. Error bars indicate standard deviation (n=3 replications).

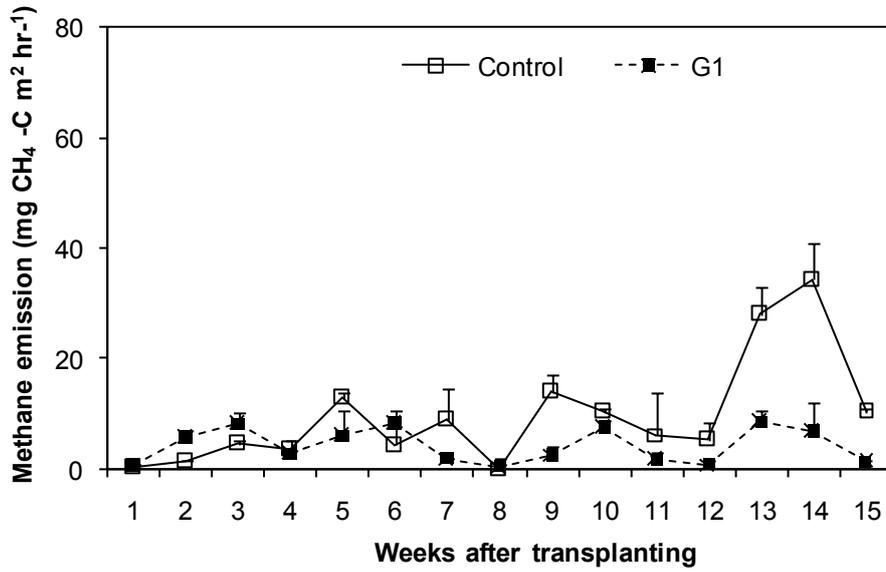


**Fig.2.2** Changes of  $\text{NH}_4^+$  and  $\text{NO}_3^-$  concentration under different salinity levels with or without addition of gypsum fertilizer. Error bars indicate standard deviation (n=3 replications).

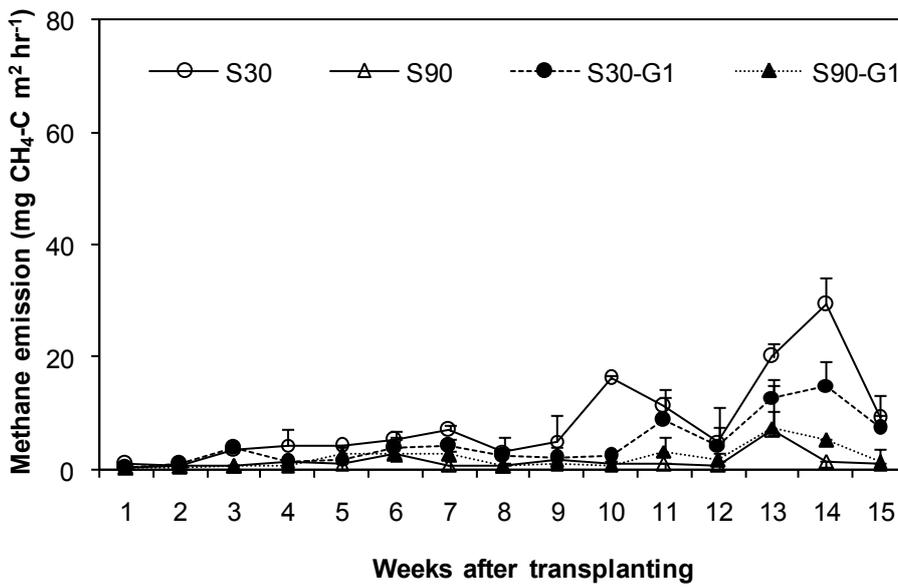


**Fig.2.3** Tillering pattern under different salinity levels with or without addition of gypsum fertilizer. Error bars indicate standard deviation (n=3 replications).

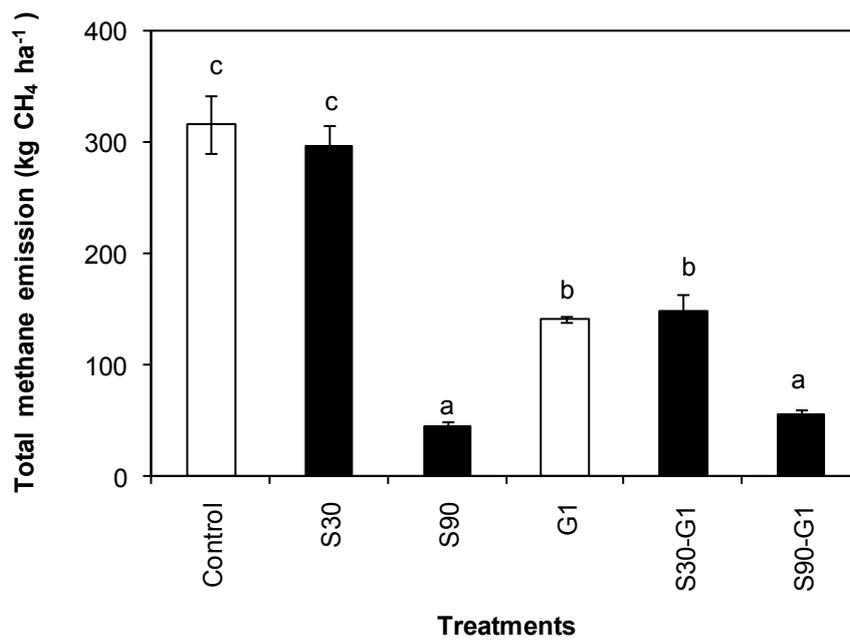
a) non-saline condition



b) saline condition



**Fig.2.4** Methane emission pattern under (a) non-saline and (b) saline condition. Error bars indicate standard deviation (n=3 replications).



**Fig.2.5** Total CH<sub>4</sub> emission under different salinity levels with or without addition of gypsum

fertilizer. Error bars indicate standard deviation (n=3 replications).

## **Chapter 3**

### **Influence of different rates of gypsum fertilizer addition on methane emission in paddy rice soil affected by saline irrigation water**

#### **The objectives of this study are;**

1. To evaluate CH<sub>4</sub> emission due to the application of gypsum fertilizer to salt affected soil related with rice growth
2. To compare the effect of different rates of gypsum fertilizer on CH<sub>4</sub> emission between saline condition and non-saline condition

### **3.1 Materials and methods**

#### **3.1.1 Experimental site and Treatments**

To evaluate the different rates of gypsum application upon CH<sub>4</sub> emission under Na-salinized paddy soil, pot experiment was conducted with (3) different gypsum application rates either in non-saline or saline condition. The chemical composition of gypsum fertilizer is shown in Table 3.1. The treatments were assigned according to a randomized complete block design with 3 replications and comprised 1) control (tap water), 2) 25 mmol L<sup>-1</sup> NaCl (S25), 3) gypsum 1 ton ha<sup>-1</sup> (G1), 4) gypsum 2.5 ton ha<sup>-1</sup> (G2.5), 5) gypsum 5 ton ha<sup>-1</sup> (G5), 6) 25 mmol L<sup>-1</sup> NaCl plus gypsum 1 ton ha<sup>-1</sup> (S25-G1), 7) 25 mmol L<sup>-1</sup> NaCl plus gypsum 2.5 ton ha<sup>-1</sup> (S25-G2.5), and 8) 25 mmol L<sup>-1</sup> NaCl plus gypsum 5 ton ha<sup>-1</sup> (S25-G5). The layout of experiment was as shown in Fig.3.1. It was conducted at Field Museum, Honmachi farm, Field Science Center, Tokyo University of Agriculture and Technology from 21<sup>st</sup> June, 2011 and to 23<sup>rd</sup> September, 2011.

#### **3.1.2 Preparation of soil and cultivation of rice in pot experiment**

The soils used in this experiment were also collected from paddy fields in FM Fuchu Honmachi, Field Science Center of the Tokyo University of Agriculture and Technology (see for detail Chapter 2). Its texture sandy loam composing with 16% clay, 33% silt, 51% sand. The pH and EC of soil at soil water ratio (1:2.5) is 6 and 0.231. A salt tolerant Indica rice variety, Dorfak cultivar which is originated from Iran, was cultivated in this study.

About 8 kg of soil was placed in 30 × 20 cm plastic pots. About one week before transplanting, puddling was conducted by irrigating the pots twice on alternate days with specific salinity level 25 mmol L<sup>-1</sup> NaCl solutions or tap water (control). The chemical fertilizers such as 40 kg P ha<sup>-1</sup> and 70 kg K ha<sup>-1</sup> were applied one day before transplanting. 70 kg N ha<sup>-1</sup> was applied in 3 equal splits at basal (one day before transplanting), active tillering stage (2 weeks after transplanting) and panicle initiation stage (4 weeks after transplanting). Urea, ammonium phosphate and potassium chloride were used as a source of N, P and K, respectively. Twenty-one-day-old seedlings were transplanted with 2 seedlings per pot. A water level of about 2–3 cm was maintained in pots at each irrigation time. Irrigation was performed when the soil water reached the saturation level (i.e., no standing water) as intermittent irrigation.

### **3.1.3 Measured parameters and analytical methods**

The agronomic characters such as tiller numbers, numbers of dead leaves were collected at weekly interval. Soil environmental data such as soil pH, EC, NH<sub>4</sub><sup>+</sup> concentration, NO<sub>3</sub><sup>-</sup> concentration in flooded water were also monitored at weekly interval. Methane gas samples were collected by using the closed chamber method weekly. The analytical methods for all the above parameters are conducted similar as the pot experiment in chapter (2). Total organic carbon content flooded water was also checked (2) times at 6 weeks and 10 weeks after transplanting by using Total Organic Carbon analyser (TOC-VCPH, Shimadzu Corp., Japan).

### 3.1.4 Statistical analysis

All the data were evaluated by an analysis of variance (ANOVA) by using CropStat 7.0 statistical software. Comparison of treatment means was performed using least significant differences (LSD) at  $p=0.05$ . Correlation and regression analysis was performed using Sigma Plot 11.0 statistical software.

## 3.2 Results

### 3.2.1 Soil environment during the rice-growing season

Soil temperature throughout the growing season ranged from 27 to 34°C (Fig. 1b). There was no large difference in temperature between the treatments. However, a sudden drop of soil temperature at 4 weeks after transplanting was observed following a period of rain on that sampling day. The intermittent application of saline water resulted in higher EC levels during the latter growth period from 9 weeks after transplanting. The range of flood water' EC value under saline condition was observed from 1.11- 2.58  $\text{dSm}^{-1}$  in the vegetative stage of rice growth (until 7 weeks after transplanting) and from 1.13 - 4.8  $\text{dSm}^{-1}$  in the reproductive stage of rice growth. Gypsum application under saline conditions showed higher EC values compared to those of non-saline conditions (Fig. 1h). The pH was in the range of 6.7–8.4 throughout the growing season (Fig. 3.1d).

The 2 highest peaks of  $\text{NH}_4^+$  and  $\text{NO}_3^-$  concentration were observed at and immediately after fertilizer addition. The  $\text{NH}_4^+$  and  $\text{NO}_3^-$  concentration in flooded water was 0.02–18.7  $\text{mg L}^{-1}$  (Fig. 3.2a) and 0.02–18.5  $\text{mg L}^{-1}$  (Fig. 3.2b), respectively. The

$\text{NH}_4^+$  concentrations in the non-saline control and S25 were lower than those in other treatments at the time of the second fertilization. There was no large difference in  $\text{NH}_4^+$  concentration during the latter growth period (from 4 weeks after transplanting), except for S25-G5. In the early growth period (until 5 weeks after transplanting), the concentration of  $\text{NO}_3^-$  was higher in S25 than in other treatments.

### **3.2.2 Impact of different rates of gypsum fertilizer addition on rice growth**

The significant reduction of tiller number was observed in S25 and S25-G5 over that of non-saline control at maximum tillering stage (Table 3.2). Moreover, salinity suppressed rice growth in terms of the number of dead leaves (Table 3.2). There were no significant differences in above-ground plant biomass yield under different rates of gypsum application in non-saline or saline conditions. However, significant reduction of above-ground biomass yield was observed in the addition of gypsum G5 to saline condition (Table 3.2).

### **3.2.3 Impact of different rates of gypsum fertilizer addition on methane emission**

Methane emission was slightly higher during the vegetative growth stage (Fig. 3.4a, b). Total  $\text{CH}_4$  emission under intermittent irrigation using saline water with salinity level S25 did not differ significantly from that of the non-saline control (Fig. 3.5). The addition of gypsum suppressed  $\text{CH}_4$  emission under non-saline and saline conditions. Under non-saline conditions,  $\text{CH}_4$  emission in G1 was about 22% lower than that of the non-saline control, although the difference was not significant. Methane emission in G2.5 and G5 was

significantly lower than that of the non-saline control. Methane emissions were reduced in G2.5 and G5 compared with the non-saline control by 52% and 73%, respectively. Under saline conditions, reductions of CH<sub>4</sub> emission due to the addition of gypsum were 23%, 27% and 61% in G1, G2.5 and G5, respectively. The results show that a higher rate of gypsum addition produces a higher reduction of CH<sub>4</sub> emission. Although the above-ground plant biomass yield, which is known as a major CH<sub>4</sub> transport conduit, did not differ significantly except for S25-G5, the reduction of CH<sub>4</sub> emission due to the addition of gypsum was lower under saline conditions than non-saline conditions. Our findings also demonstrated a negative relationship between gypsum application rates and CH<sub>4</sub> emission ( $p < 0.001$ ) based on the result of regression analysis (Fig 3.6).

### **3.3 Discussion**

#### **3.3.1 Impact of different rates of gypsum fertilizer addition on rice growth**

At maximum tillering stage, the lower tiller numbers in S25 compared with non-saline control might relate with the mild osmotic stress in vegetative stage of rice growth. Homaei (2002) pointed that as the salinity and osmotic pressure increase, this process leads to less water and nutrient absorb including N and potassium by plants. Similar as this statement, the higher NO<sub>3</sub><sup>-</sup> concentration was also observed in S25 over that of non-saline control in the present study, although there was no significant difference in NH<sub>4</sub><sup>+</sup> concentration in the flood water. As ammonium phosphate fertilizer was applied as nitrogen source in this study, this higher NO<sub>3</sub><sup>-</sup> concentration in S25 compared with non-saline control

treatment might be converted from  $\text{NH}_4^+$  which was less absorbed by rice plants due to osmotic stress and the antagonistic effect of  $\text{NH}_4^+$  and  $\text{Na}^+$  ions in S25. Furthermore, Abdelgadir et al. (2005) also observed less  $\text{NO}_3^-$  N uptake of shoots and roots of rice plants due to antagonistic effect of  $\text{Na}^+$  and  $\text{Cl}^-$  ion under NaCl salinity stress. The higher numbers of dead leaves in S25 over that of control might be due to the salt (NaCl) load exceeding the ability of the cells to compartmentalize salts in the vacuole. The salt taken up by the plant concentrates in the old leaves; continued transport of salt into transpiring leaves over a long period of time eventually results in very high  $\text{Na}^+$  and  $\text{Cl}^-$  concentrations, and the leaves died.

In non-saline condition, there was no significant difference in maximum tiller numbers, numbers of dead leaves, and above-ground plant biomass yield under different rate of gypsum application such as gypsum  $1 \text{ ton ha}^{-1}$  (G1),  $2.5 \text{ ton ha}^{-1}$  (G2) and  $5 \text{ ton ha}^{-1}$  (G3). It was indicating that  $\text{Ca}^{2+}$  concentration in normal soil is enough for plant growth. Even though the  $\text{Ca}^{2+}$  concentration is exceeded over plant's requirement due to the addition of gypsum fertilizer, it could not disturb the plant growth in non-saline condition.

In saline condition, the addition of gypsum  $5 \text{ ton ha}^{-1}$  (G5) significantly suppressed maximum tiller numbers and above-ground biomass yield. The growth reduction in S25-G5 might be due to the highest EC level under the addition of G5 during the vegetative growth stage. Furthermore, The higher  $\text{Ca}^{2+}$  concentration in G5 in saline condition might not allow much absorption of  $\text{NH}_4^+$  ion as it is indicated by significantly higher  $\text{NH}_4^+$  concentration of

flood water in S25-G5 in the reproductive stage of rice growth due to antagonistic effect of  $\text{NH}_4^+$  and  $\text{Ca}^{2+}$  (Fig. 3.2b).

### **3.3.2 Impact of different rates of gypsum fertilizer addition on methane emission**

There was also no statistically significant difference in total  $\text{CH}_4$  emission between S25 and the non-saline control. However, the extent of  $\text{CH}_4$  emission in S25 was numerically lower than that in non-saline control (Fig. 3.5b), although the aboveground plant biomass yields were slightly higher in S25 compared with the non-saline control (Table 3.3). This lower value of  $\text{CH}_4$  emission in S25 might be due to the existence of a higher  $\text{NO}_3^-$  concentration in S25 in the early growth stage (Fig. 3.2d). Furthermore, Dubey (2005) mentioned that soils containing greater amounts of readily decomposable organic substrates (acetate, formate, methanol, methylated amines, etc.) and low amounts of electron acceptors such as ferric ion, manganese,  $\text{NO}_3^-$  and  $\text{SO}_4^{2-}$  are likely to show high production of  $\text{CH}_4$ . It can therefore be concluded that the extent of  $\text{CH}_4$  emission under saline conditions is influenced not only by the above-ground plant biomass yield, but also by the existence of electron acceptors such as  $\text{NO}_3^-$  that accumulate in rice-growing environments due to the osmotic stress of rice plants.

The observed reduction of  $\text{CH}_4$  emission under gypsum application either in saline and non-saline condition might be due to the result of substrate competition between sulphate reducers and methanogens as stated by Van der Gon and Neue (1994). The higher value of  $\text{CH}_4$  reduction under higher rates of gypsum addition might be related with the

amount of  $\text{SO}_4^{2-}$  added to the soil. According to Van der gon et al. (2001), one mole of  $\text{SO}_4^{2-}$  is needed to reduce one mole of  $\text{CH}_4$  produced. These considerations support the notion of a direct and proportional correlation between the amount of  $\text{SO}_4^{2-}$  added and the reduction of  $\text{CH}_4$  production. However, the amount of  $\text{SO}_4^{2-}$  added to rice cultivation in pot culture might be lost from the soil through plant uptake. It may be one possible factor that the level of  $\text{CH}_4$  reduction was not directly proportional to the amount of  $\text{SO}_4^{2-}$  added in the present study.

In addition, Chin and Conrad (1995) and Rothfuss and Conrad (1993) reported that 20–30% of  $\text{CH}_4$  production in rice fields is derived from the conversion of  $\text{H}_2$  or  $\text{CO}_2$  and 70–80% is derived from  $\text{CH}_3\text{COO}^-$ . Although sulfate-reducing bacteria can outcompete methanogens for  $\text{H}_2$  or  $\text{CO}_2$ , it is impossible to completely outcompete them for  $\text{CH}_3\text{COO}^-$  that was derived from a native soil organic C source and dead plant materials. Pangala et al. (2010) also indicated that level of  $\text{CH}_4$  suppression in the wetland is influenced by the ratio of  $\text{SO}_4^{2-}$  and organic matter. Therefore, the lower reduction of  $\text{CH}_4$  emission under saline conditions in this study was due to the higher availability of organic C sources from leaf senescence under these conditions and their contribution as a source of nutrients for methanogens.

### **3.3.3 Relationship between rice growth and methane emission**

The result of spearman rank order correlation analysis showed that the soil temperature positively correlated with total  $\text{CH}_4$  emission ( $p=0.04^*$ ) (Table 3.3). Methane is

the final product of organic matter decomposition by soil microorganism under strict anaerobic condition. The increase in soil temperature may accelerate the decomposition rate of soil organic matter and enhance carbon availability for methanogenic bacteria to produce more CH<sub>4</sub>. However, there was no significant relationship between total CH<sub>4</sub> emission and rice growth such as maximum tiller numbers per pot, total numbers of dead leaves, above-ground biomass yield because the gypsum application was the most prominent impacting factor upon CH<sub>4</sub> emission rather than the rice growth as shown by significant negative relationship between total CH<sub>4</sub> emission and amount of gypsum fertilizer added ( $p = -0.754^{**}$ ) (Table 3).

### **3.4 Conclusion**

As there was no significant reduction of above-plant biomass yield by intermittent irrigating with saline water in this study, saline water with salinity level 25 mmol L<sup>-1</sup> NaCl can be used in rice cultivation if no fresh water is available. Under non-saline condition, gypsum fertilizer can be used as CH<sub>4</sub> mitigation option without disturbing the rice growth. In case of saline condition, addition of gypsum fertilizer up to 2.5 ton ha<sup>-1</sup> can be recommended as CH<sub>4</sub> mitigation option in paddy rice fields. Significant growth reduction in gypsum 5 ton ha<sup>-1</sup> in saline condition might not only due to excessive concentration of Ca<sup>2+</sup> ion alone but also due to the total cations concentration such as Na<sup>+</sup> concentration plus Ca<sup>2+</sup> which might led to reduction of essential nutrient's uptake NH<sub>4</sub>- N. Although the addition of SO<sub>4</sub><sup>2-</sup> containing

fertilizer can suppress  $\text{CH}_4$  emission, the level of suppress depend not only on the amount of  $\text{SO}_4^{2-}$  added but also on carbon availability in the soil.

**Table 3.1** Chemical composition of gypsum fertilizer

<b>Purity percentage of gypsum fertilizer</b>	<b>Chemical composition of gypsum</b>			
	<b>Calcium</b>	<b>Hydrogen</b>	<b>Sulphur</b>	<b>Oxygen</b>
98.5% of (CaSO <sub>4</sub> .2H <sub>2</sub> O)	23.80%	2.34%	18.62%	55.76%

**Table 3.2** Effect of irrigation water salinity and different rates of gypsum fertilizer addition on rice growth

Treatments	Maximum Tiller numbers (numbers pot <sup>-1</sup> )	Numbers of dead leaves (Numbers pot <sup>-1</sup> )	Above-ground plant biomass yield(g pot <sup>-1</sup> )
1. Control	54 ± 2.5 a	62 ± 6.4 b	95.9 ± 10.6 a
2. S25	48 ± 9.5 bc	86 ± 5.9 a	101.3 ± 19.7a
3. G1	54 ± 3.8 a	58 ± 3.5 b	106.4 ± 1.1 a
4. G2.5	52 ± 5.0 ab	61 ± 5.3 b	95.6 ± 22.1 a
5. G5	55 ± 3.1 a	63 ± 5.1 b	103.2 ± 16.8 a
6. S25-G1	52 ± 3.5 ab	99 ± 21.8 a	90.6 ± 27.2 a
7. S25-G2.5	51 ± 4.2 ab	101 ± 10.0 a	104.8 ± 10.4 a
8. S25-G5	46 ± 7.0c	95 ± 4.6 a	65.0 ± 14.6 b

Values are the means ± standard deviation (n= 3 replications). In each column, means followed by a common letter are not significantly different by using least significant difference (LSD) at p=0.05.

**Table 3.3** Results of ANOVA test for CH<sub>4</sub> emission and crop growth

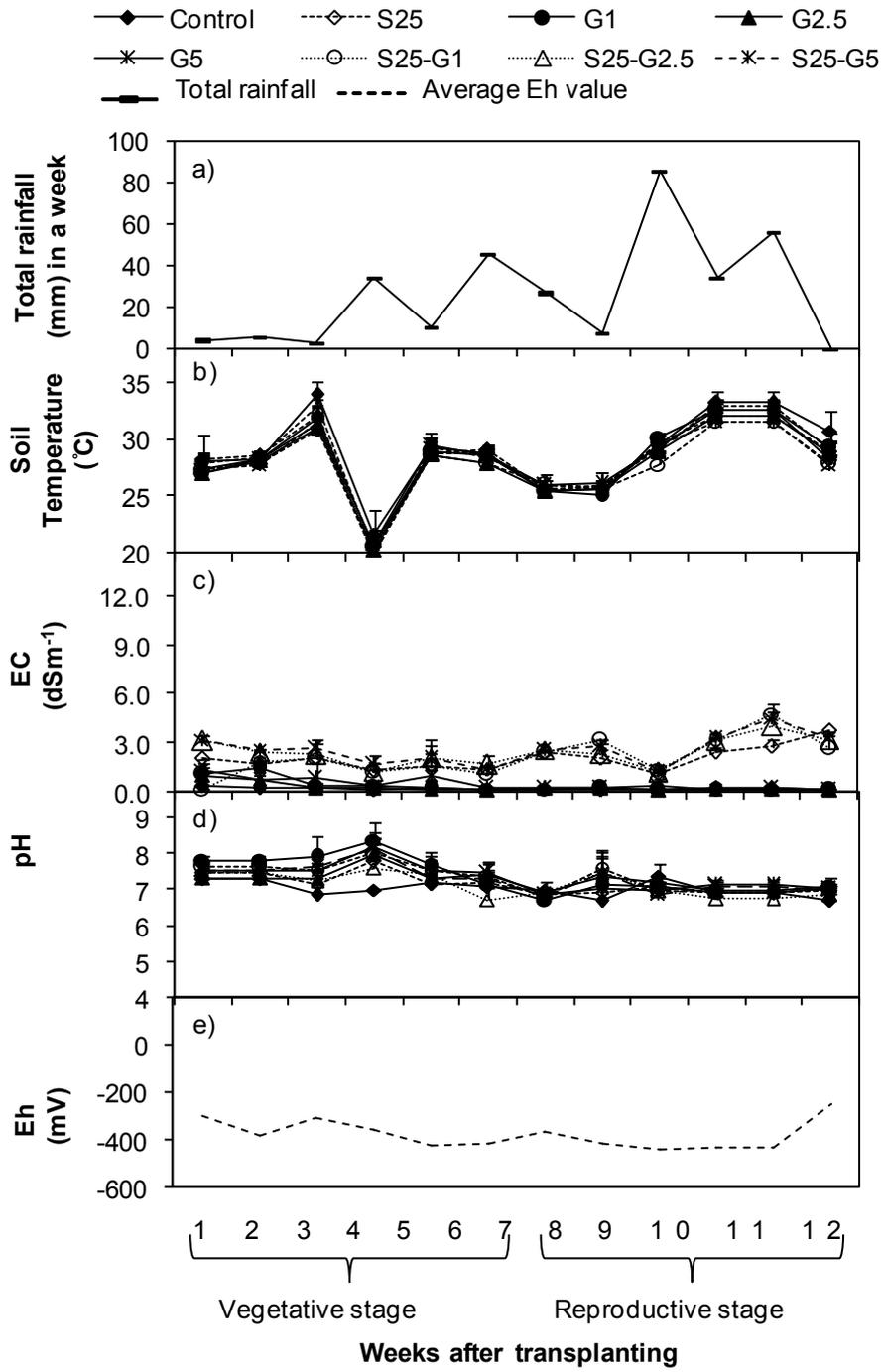
Source of variation	Maximum tiller numbers (no. pot <sup>-1</sup> )	Total numbers of dead leaves (no. pot <sup>-1</sup> )	Above biomass yield (g pot <sup>-1</sup> )	CH <sub>4</sub> emission in one crop season (kg CH <sub>4</sub> ha <sup>-1</sup> )
Salinity	**	**	ns	ns
Gypsum application rate	ns	ns	ns	**
Salinity* Gypsum application rates	ns	ns	*	*

\*\* , \* and ns stand for significance at 1%, 5% and non-significance respectively.

**Table 3.4** Spearman rank order correlation analysis

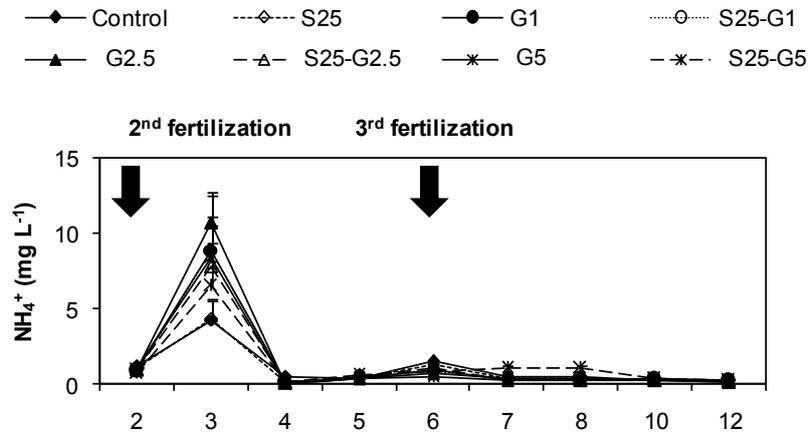
<b>Parameters</b>	<b>Gypsum application</b>	<b>pH</b>	<b>EC</b>	<b>Soil Temp ·C</b>	<b>Maximum Tillers (no. pot<sup>-1</sup>)</b>	<b>Total numbers of dead leaves (no. pot<sup>-1</sup>)</b>	<b>Above biomass yield (g pot<sup>-1</sup>)</b>
<b>Methane emission per crop season (Kg CH<sub>4</sub> ha<sup>-1</sup>)</b>	-0.754 **	-0.15 Ns	-0.074 ns	0.44 *	0.00 ns	0.084 ns	0.17 ns

\*\* , \* and ns stand for significance at 1%, 5% and non-significance respectively.

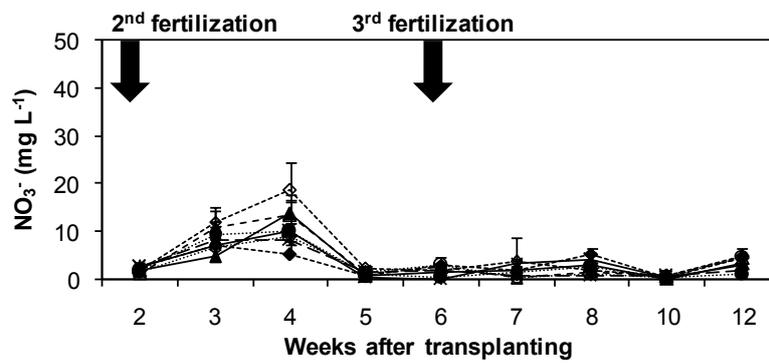


**Fig.3.1** Distribution of rainfall and changes of soil environment under different gypsum application rates. Error bars indicate standard deviation (n=3 replications).

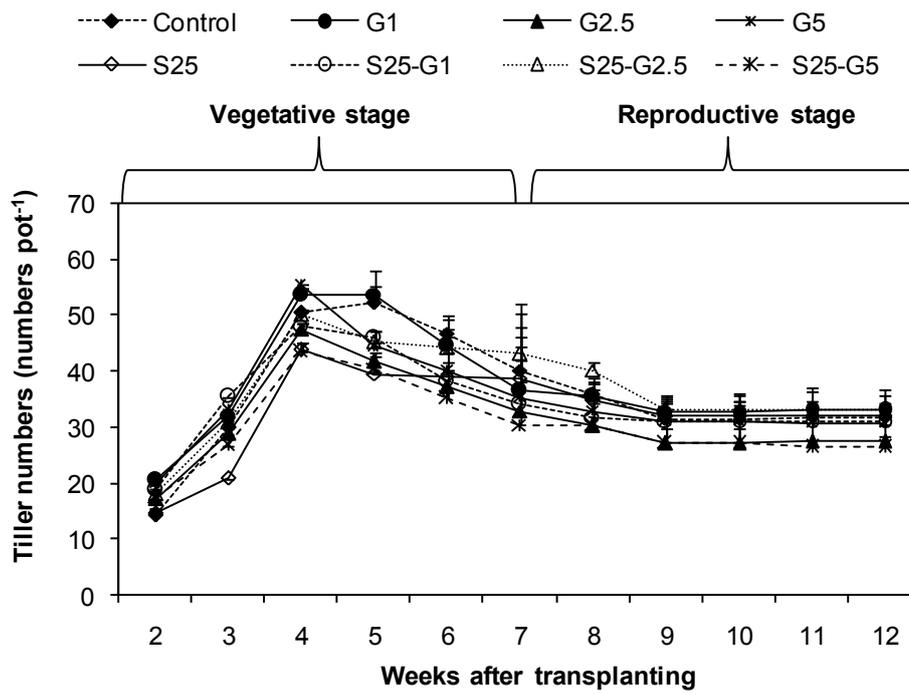
(a)



(b)

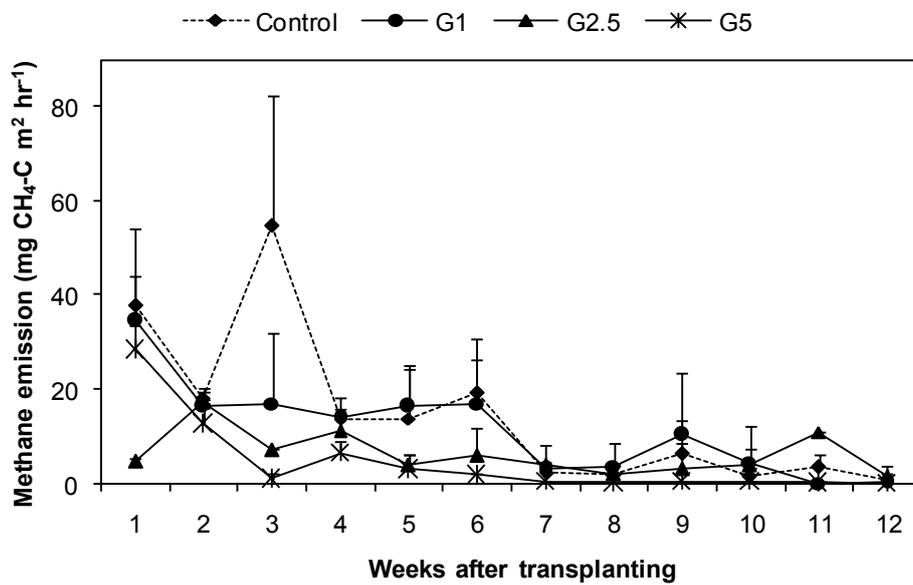


**Fig.3.2** Changes of  $\text{NH}_4^+$  and  $\text{NO}_3^-$  concentration under different rates of gypsum application to non-saline and saline condition. Error bars indicate standard deviation (n=3 replications).

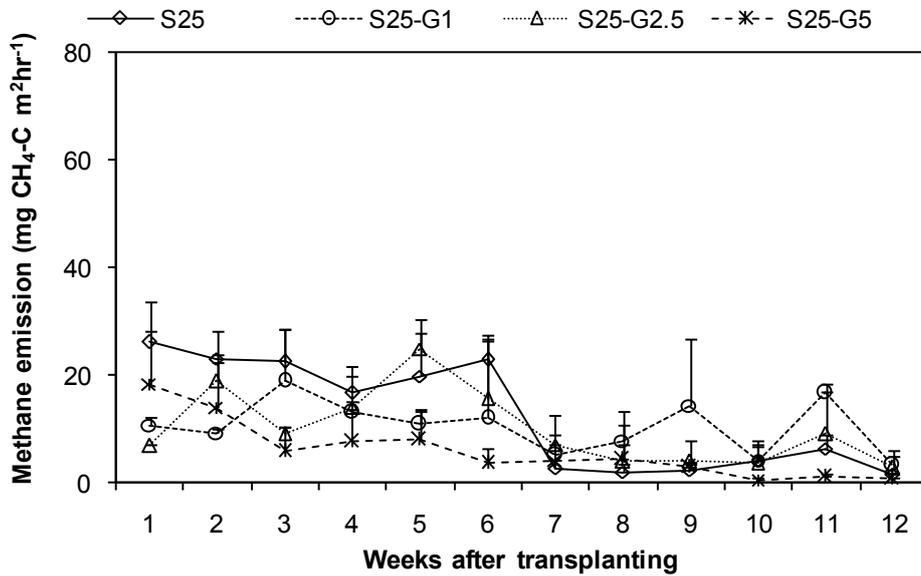


**Fig.3.3** Tiller pattern under different gypsum application rates to non-saline and saline condition. Error bars indicate standard deviation (n=3 replications).

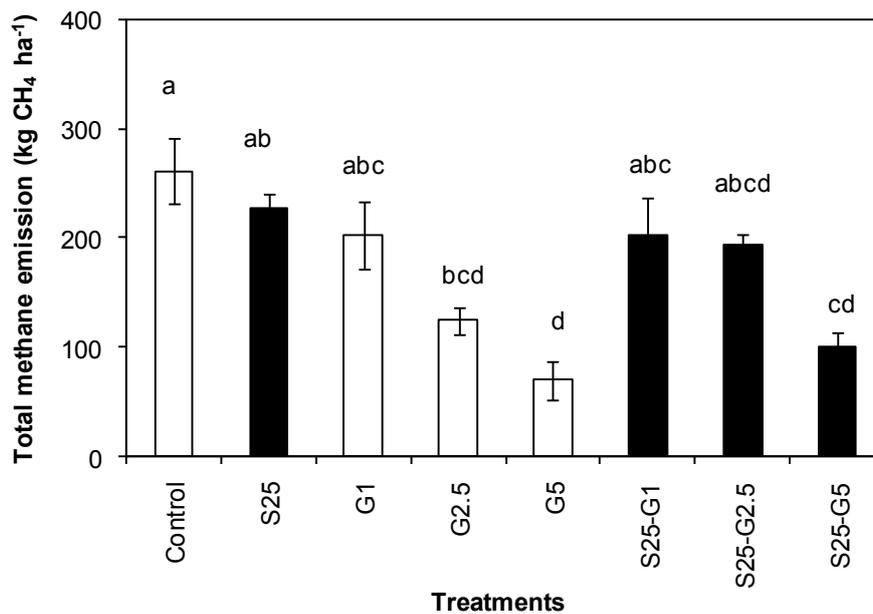
(a) Non-saline condition



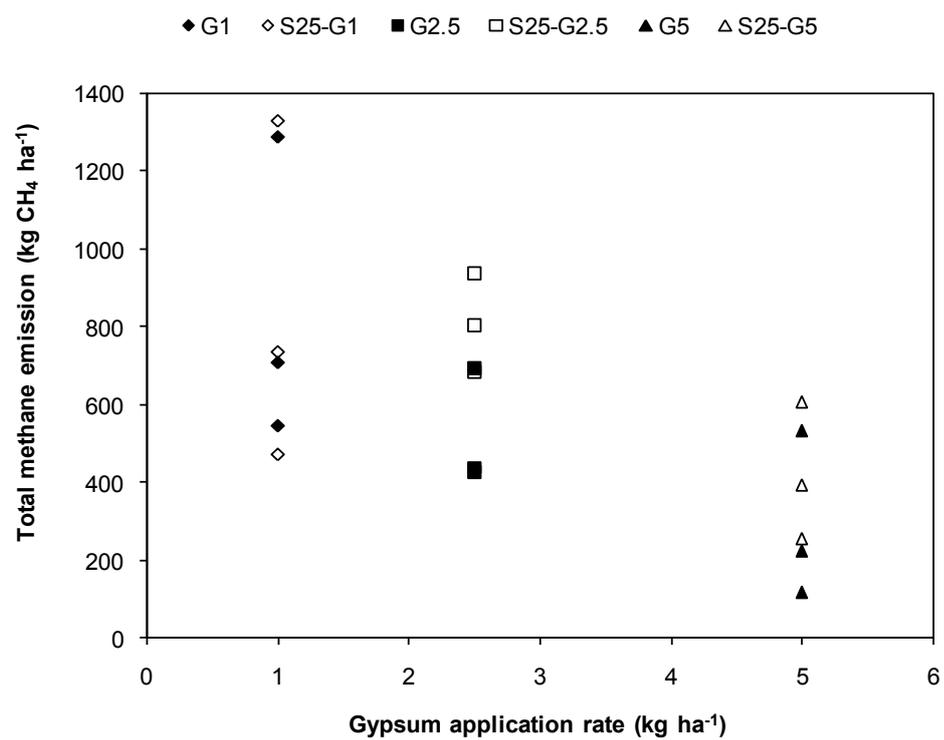
(b) Saline condition



**Fig.3.4** Methane emission pattern under different gypsum application rates to non-saline and saline condition. Error bars indicate standard deviation (n=3 replications).



**Fig.3.5** Total CH<sub>4</sub> emission under different gypsum application rates to non-saline and saline condition. Means followed by a common letter are not statistically significant different by using least significant differences (LSD) at p=0.05. Error bars indicate standard error (n=3 replications).



**Fig.3.6** Relationship between total CH<sub>4</sub> emission and gypsum fertilizer application rates

## **Chapter 4**

### **Evaluation of methane emission in coastal saline paddy soil with different Na/Ca ratios by adding gypsum fertilizer**

#### **The objectives of this study are;**

1. To evaluate rice growth in coastal saline soil with different Na/Ca ratios by adding gypsum fertilizer
2. To identify the different rates of gypsum fertilizer application on CH<sub>4</sub> emission in coastal saline soil related with rice growth

## **4.1 Materials and methods**

### **4.1.1 Experimental site**

The pot experiment was conducted inside a Phytotron at Tokyo University of Agriculture and Technology, Fuchu, Tokyo, Japan. Natural daylight was used. The day and night temperature inside the Phytotron was maintained as 25° C and 30°C respectively.

### **4.1.2 Preparation of soil and cultivation of rice**

The saline soils used in this experiment were collected on 28<sup>th</sup> July 2012 from the upper 5cm depth of tsunami-flooded paddy field in Sendai, in Japan (38.23°N Latitude and 140.96°E Longitude). The soil texture was loam, the pH was 6.6, EC 5.87, CEC 23.51 meq 100g soil<sup>-1</sup>, where 4.3 meq 100g soil<sup>-1</sup> was exchangeable sodium, 0.8 meq 100g soil<sup>-1</sup> was exchangeable calcium. The total nitrogen content was 51.6 g kg<sup>-1</sup> and total carbon 5.6g kg<sup>-1</sup>. The sodium adsorption ratio (SAR) value was 16.

About 3.5 kg of soil was placed into Wagner pots with the area of 0.02 m<sup>2</sup>. Puddling was done by irrigating the pots with tap water at about one week before transplanting. Chemical fertilizers were applied one day before transplanting on 30<sup>th</sup> October 2013 at the rate of 35 kg N ha<sup>-1</sup>, 40 kg P ha<sup>-1</sup> and 70 kg K ha<sup>-1</sup>. Urea, ammonium phosphate and potassium sulfate were used as a source of N, P and K, respectively. As it is required to maintain the appropriate Na/Ca and roots supplied with elevated levels of external Ca<sup>2+</sup> (5-

10 mM) are often to maintain their  $K^+$  reported by Aslam et al. (2001) , different rates of gypsum fertilizer were added (Table 4.1).

There were four treatments; no gypsum (control), gypsum 0.5 ton ha<sup>-1</sup> (G0.5), gypsum 1 ton ha<sup>-1</sup> (G1) and gypsum 2 ton ha<sup>-1</sup> (G2). All the treatments were laid out in a completely randomized design with 3 replications. The gypsum fertilizer treatments were given on 30<sup>th</sup> October 2013. Thirty days old seedlings of a salt tolerant Indica rice variety, Dorfak were transplanted with one seedling per pot on 31<sup>st</sup> October, 2013 and harvested at maximum tillering stage on 30<sup>th</sup> November, 2013. A water level of about 2–3 cm was maintained in pots throughout the growing seasons by irrigating regularly with tap water.

#### **4.1.3 Measured parameters**

Soil redox potential value (Eh), total organic carbon (TOC), ammonium (NH<sub>4</sub><sup>+</sup>) ion concentration, nitrate ion concentration (NO<sub>3</sub><sup>-</sup>) in flooded water, and CH<sub>4</sub> gas sampling were collected at ten days interval. At the end of experiment, shoot weight, root weight, soil pH and soil EC value were recorded. The root length and total numbers of root tips were also measured by scanning with an Epson perfection V700 photo scanner and Win RHIZO Pro v.2009c software. Sodium and potassium ion concentrations in the plants were also detected after harvesting rice plants.

#### **4.1.4 Analytical method for soil environment data**

The Eh value of soil was monitored by platinum electrodes, which were inserted at 5 cm depth in each plot throughout the rice growth (SWC-201RP, Sanyo, Japan). For TOC,  $\text{NH}_4^+$ -N and  $\text{NO}_3^-$ -N analysis, water samples were filtered with 0.45- $\mu\text{m}$  filter paper at first. Total organic carbon content in flooded water was detected by Total Organic Carbon analyzer (TOC-VCPH, Shimadzu Corp., Japan). Both  $\text{NH}_4^+$ -N and  $\text{NO}_3^-$ -N concentrations were determined by using a UV spectrophotometer (UV-VI Mini 1240, Shimadzu Corporation, Kyoto, Japan), by Indophenol method at 630 nm and absorption at 230 nm, respectively. At the end of experiment, the pH and EC of the soil was measured by portable meters (Beckman,  $\Phi$  260 pH/ Temp/ mV meter, and ES-51 COND METER, Horiba, Japan, respectively). To analyse the  $\text{Na}^+$  and  $\text{K}^+$  ion concentration in plants, the above ground plant samples were dried at 60-70 °C. About 5 g of each grounded plant samples were digested by wet digestion method (Jones and Case, 1990). In the digest,  $\text{Na}^+$  and  $\text{K}^+$  were determined by Hitachi Z-5010 polarized Zeeman atomic absorption spectrophotometer.

#### **4.1.5 Gas sampling, analysis and calculation**

Gas sampling was taken at 10-days interval by closed-chamber methods mentioned in chapter (2).

All the data were evaluated by an analysis of variance (ANOVA) by using CropStat 7.0 statistical software. Comparison of treatment means was performed using least significant differences (LSD) at  $p=0.05$ .

## **4.2 Results**

### **4.2.1 Soil environment during the experiment**

The average  $\text{NH}_4^+$ -N concentration ranged from 2.8 to 3.9 mg N L<sup>-1</sup>, the average  $\text{NO}_3^-$ -N concentration from 0.6 to 1.0 mg N L<sup>-1</sup> (Table 4.2). Total organic carbon concentration of flooded water in all treatments becomes higher at 30 days after transplanting. At 30 days after transplanting, TOC in control was significantly lower than that of G0.5 and G2 but not significantly different with that of G1. The average redox potential (Eh) value lied between -365.5 to -434.0 mV during the experiment. At the end of experiment, the soil pH values were observed as 6.5, 6.0, 6.0, and 5.8 in control, G0.5, G1, and G2 respectively. The soil EC values were observed as 4.9, 6.5, 7.5 and 7.7 in control, G0.5, G1 and G2, respectively.

### **4.2.2 Sodium and potassium concentration of rice plants**

The range of  $\text{Na}^+$  concentrations in plants was observed from 0.2 to 0.4 mg g dry plant sample<sup>-1</sup> (Fig. 4.1a). The values of  $\text{Na}^+$  concentration of plants were lower in gypsum treatments than that of control. The significant lower  $\text{Na}^+$  concentration was observed only in G2 treatment. The range of  $\text{K}^+$  concentration in plants was observed from 1.2 to 1.5 mg g dry plant sample<sup>-1</sup> (Fig. 4.1b). The higher  $\text{K}^+$  ion concentration was observed in plants of gypsum treatments except G2 compared with control.

### **4.2.3 Rice growth in coastal saline with different Na/Ca ratios**

The range of shoot dry weight was 5.4 to 6.4 g pot<sup>-1</sup> (Fig. 4.2a). The range of root dry weight as observed from 1.4 to 1.7 g pot<sup>-1</sup> (Fig. 4.2b). Although the roots dry weight in G2 was not significantly different with that of control, the lowest shoot dry weight was observed in G2. There was no improvement in shoot dry weight by adding gypsum fertilizer to saline soil. However, the total numbers of root length and root tips in G0.5 was significantly higher than of other treatments (Fig. 4.3a and b).

### **4.3.3 Methane emission in coastal saline soil**

The CH<sub>4</sub> emission was monitored during vegetative stage of rice growth. The CH<sub>4</sub> flux was suppressed in all treatments until 20 days after transplanting (Fig. 4.4a). The highest CH<sub>4</sub> flux was observed at 30 days after transplanting (Fig. 4.4a). The total CH<sub>4</sub> emission was ranged from -18 to 150 mg m<sup>-2</sup> over 30 days growth period (Fig. 4.4b). The significant higher total CH<sub>4</sub> emission was observed in G0.5 and G1 compared with control and G2. The lowest and negative value of total CH<sub>4</sub> emission was observed in G2.

## **4.3 Discussions**

### **4.3.1 Ion concentration of plants and plant growth**

The addition of gypsum as a source of Ca<sup>2+</sup> decrease Na<sup>+</sup> ion concentrations in plants under saline condition (Fig.4.1a), because addition of Ca<sup>2+</sup> protect the integrity and permeability of plasma membranes against Na<sup>+</sup> toxicity (Tester and Davenport, 2003).

Moreover, the existence of high  $\text{Na}^+$  level in the experimental soil seems to interfere with  $\text{K}^+$  metabolism in control treatment and the addition of  $\text{Ca}^{2+}$  also reduced the interference of  $\text{Na}^+$  in G0.5 and G1. While  $\text{K}^+$  concentration was higher in G0.5 and G1 compared to control, the lowest value of  $\text{K}^+$  concentration was observed in G2. The excessive concentration of  $\text{Ca}^{2+}$  in G2 seems to cause a competition between  $\text{Ca}^{2+}$  and  $\text{K}^+$  uptake by rice plants.

Even though the ion balance was improved in G0.5 and G1 compared to control, there was no significant improvement in shoot dry weight in gypsum fertilizer addition as  $\text{Ca}^{2+}$  source (Fig.4.2a). No significant changes of chlorophyll content among all treatments (Fig.4.1c) indicated that available nitrogen contents ( $\text{NH}_4^+$  -N and  $\text{NO}_3^-$  -N) are not limited for plant growth. Thus, the growth reduction under gypsum fertilizer addition was due to the higher EC values under gypsum fertilizer addition.

The value of SAR of experimental soil was 16 (Section 4.1.2) which was higher than 13, a limit for saline-sodic or sodic soil described by US Salinity Lab Staff, Agriculture hand book (Richard, 1954). Calcium concentration of experimental soil was about 0.80 meq 100g soil<sup>-1</sup> (4.1mM) (Section 4.1.2). The  $\text{Ca}^{2+}$  level should be maintained at 5-10 mM in the external solution of saline condition (Aslam et al., 2001). High  $\text{Ca}^{2+}$  concentrations can reduce the permeability of plasma membrane to  $\text{Na}^+$ . The reduction in membrane permeability to  $\text{Na}^+$  by  $\text{Ca}^{2+}$  reduces the accumulation of  $\text{Na}^+$  by passive influx (Cramer et al., 1985). Thus, the gypsum fertilizer was applied as  $\text{Ca}^{2+}$  source. The  $\text{Ca}^{2+}$  ions in the gypsum fertilizer will replace  $\text{Na}^+$  ions that adsorbed to the soil particles and  $\text{NaSO}_4$  will be leached

away from the root zone. In case of pot experiment, there was limited space to wash away  $\text{Na}^+$  along with water, and thus, the addition of gypsum led to increase in soil EC value. Flowers and Flowers (2005) stated that salinity has three potential effects on plants such as (a) lowering of the water potential, (b) direct toxicity of any Na and Cl absorbed and (c) interference with the uptake of essential nutrients. No improvement of rice growth under G0.5 and G1 might not exactly relate with ion toxicity and deficiency because of the lower  $\text{Na}^+$  and higher  $\text{K}^+$  concentration of plants as shown in Fig. 4.1a and b and  $\text{NH}_4^+$ -N and  $\text{NO}_3^-$ -N of flooded water (Table. 4.2). Furthermore, the higher values of root length and root tips were observed in G0.5 and G1 although the total root dry weight was not significant among the treatments. Probably, these higher values of root length and root tips in G0.5 and G1 might be due to the formation fine roots under higher EC values of gypsum treatments compared to the control because Neumann and Römheld, (2002) found that root growth stimulation particularly of fine root structures is frequently induced by moderate limitations of P, N, Fe, and water. In case of G2, the reduction in the growth of rice plants under gypsum treatments especially in G2 treatment might more relate with osmotic effect that may reduce the ability of plants to take up water (Rad et al., 2012).

#### **4.3.2 Methane emission**

The higher total  $\text{CH}_4$  emission in gypsum fertilizer treatments; G0.5 and G1 might be due to its higher organic carbon content of flooded water that might be contributed either from native soil organic matter or root exudates. In the presence of  $\text{SO}_4^{2-}$ , sulphate reducers

are stronger competitors than methanogens for carbon substrate leading to a reduced production of CH<sub>4</sub> in the rice soil (Van der gon and Neue, 1994). However, the observed significant highest CH<sub>4</sub> emission in G0.5 might also relate with the amount of added SO<sub>4</sub><sup>2-</sup> ion that did not reach the threshold limit necessary for a successful competition of carbon substrates between sulphate reducers with methanogens. Furthermore, methanogens and SO<sub>4</sub><sup>2-</sup> reducing bacteria can use more than one substrate and the affinities for these substrates differ. As a result, higher CH<sub>4</sub> emission was observed in G0.5 and G1 compared to control. The optimum pH of paddy soils required for methanogenic bacteria to produce CH<sub>4</sub> is around 6 to 6.6 (Setyanto et al., 2002). The lowest CH<sub>4</sub> emission in G2 might relate with its lowest pH value (about 5.82), highest input of SO<sub>4</sub><sup>2-</sup> concentration from gypsum fertilizer and lowest above dry matter yield.

#### **4.4 Conclusion**

It can be concluded that the Na/Ca ratio around 6 is the optimum ratio in coastal saline soil which contains excess Na<sup>+</sup> concentration. Methane emission in coastal saline soil was closely related with the root growth and total carbon content of flooded water. Moreover, this study proved that although sulphate containing fertilizers are frequently recommended as CH<sub>4</sub> mitigation option, the extent of CH<sub>4</sub> emission was mainly depend on the extent of SO<sub>4</sub><sup>2-</sup> ions added and carbon availability of those soils. Moreover, as the higher TOC concentrations were observed in gypsum treated soils, it might relate with carbon contribution of the root exudates which is the important carbon source for methanogens.

Therefore, further research should more emphasize on carbon contribution of root exudates under those gypsum treated soils.

**Table4.1** Different Na<sup>+</sup>/Ca<sup>2+</sup> ratios of coastal saline soil after adding gypsum fertilizer

Ion conc:	Initial soil	Final soil (After adding gypsum)		
		Gypsum (0.5 tonha <sup>-1</sup> )	Gypsum (1 tonha <sup>-1</sup> )	Gypsum (2 tonha <sup>-1</sup> )
Na(mM)	33	33	33	33
Ca(mM)	4	5.5	7	10
<b>Na/Ca</b>	<b>8</b>	<b>6</b>	<b>4</b>	<b>3</b>

**Table 4.2** Soil environmental data under different rates of gypsum fertilizer addition

Treatment	Average NH <sub>4</sub> <sup>+</sup> -N (mg L <sup>-1</sup> )	Average NO <sub>3</sub> <sup>-</sup> -N (mg L <sup>-1</sup> )	TOC (mg L <sup>-1</sup> )		Soil Eh	Soil pH	Soil EC (dSm <sup>-1</sup> )
			20 DAT	30 DAT			
Control	3.9 ± 0.3a	0.6 ± 0.2a	7 ± 2a	11 ± 3 b	-385 ± 26a	6.5± 0.1a	4.9 ± 0.7a
G0.5	3.8 ± 0.2a	0.8 ± 0.1a	9 ± 1a	26 ± 11a	-356 ± 63a	6.0± 0.1b	6.3 ± 1.0a
G1	2.8 ± 0.2a	1.0 ± 0.2a	4 ± 1b	17 ± 2 ab	-425 ± 26a	5.9 ± 0bc	7.5 ± 0.2a
G2	4.5 ± 0.2a	0.7 ± 0.3a	8 ± 0a	25 ± 6 a	-451 ± 14a	5.8 ± 0c	7.7 ± 0.3a

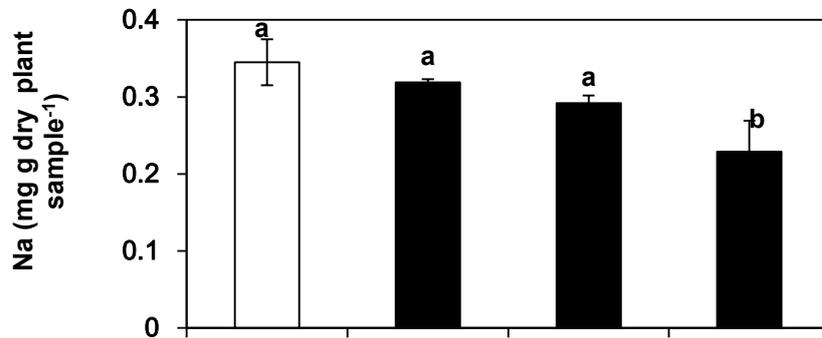
Values are the means ± standard deviation (n=3 replications). In each column, means followed by a common letter are not statistically significant different by using least significant difference (LSD) at p= 0.05.

**Table 4.3** Pearson product moment correlation analysis among measured parameters

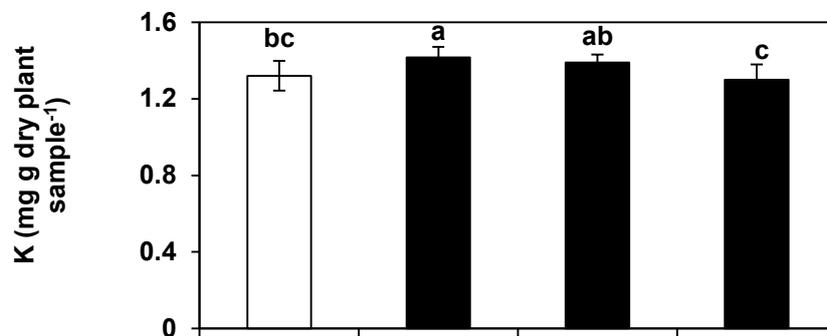
Parameters	pH	EC	Shoot dry weight (g pot <sup>-1</sup> )	TOC in flooded water		Total CH <sub>4</sub> emission for 30 days (mg CH <sub>4</sub> m <sup>2</sup> )
				at 20 DAT (mg C L <sup>-1</sup> )	at 30 DAT (mg C L <sup>-1</sup> )	
Amount of SO <sub>4</sub> <sup>2-</sup> added in each treatments	<b>-0.83</b> *	<b>0.782</b> **	<b>-0.704</b> **	-0.0446 Ns	0.463 ns	<b>-0.346</b> ns
pH		<b>-0.815</b> **	0.552 ns	0.14 Ns	-0.549 ns	-0.155 ns
EC			-0.39 ns	-0.275 Ns	0.324 ns	-0.072 ns
Shoot dry weight (g pot <sup>-1</sup> )				-0.04 Ns	<b>-0.655</b> *	0.09 ns
TOC in flooded water at 20 DAT (mg C L <sup>-1</sup> )					0.282 ns	0.19 ns
TOC in flooded water at 30 DAT (mg C L <sup>-1</sup> )						0.382 ns

\*\* , \* and ns stand for significance at 1%, 5% and non-significance respectively.

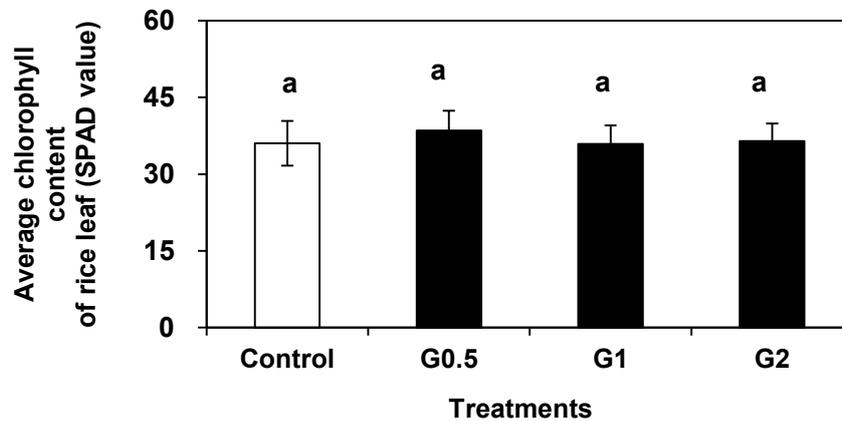
(a)



(b)

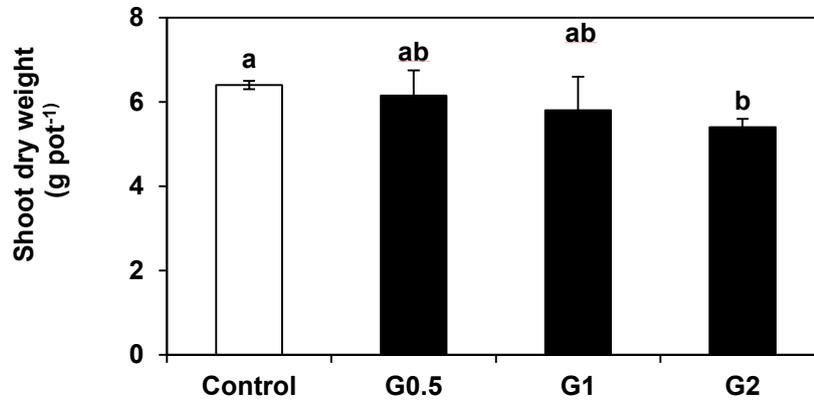


(c)

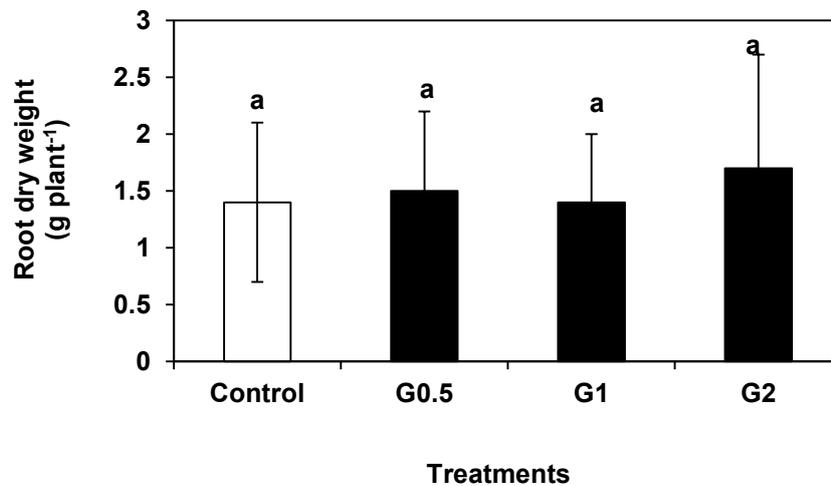


**Fig.4.1** Effect of different rates of gypsum fertilizer application on (a) Na<sup>+</sup>, (b) K<sup>+</sup> concentration in plant samples and (c) average chlorophyll content of leaf. Mean followed by a common letter are not statistically significant different by using least significant difference (LSD) at p=0.05. Error bars indicate standard deviation, n=3.

(a)

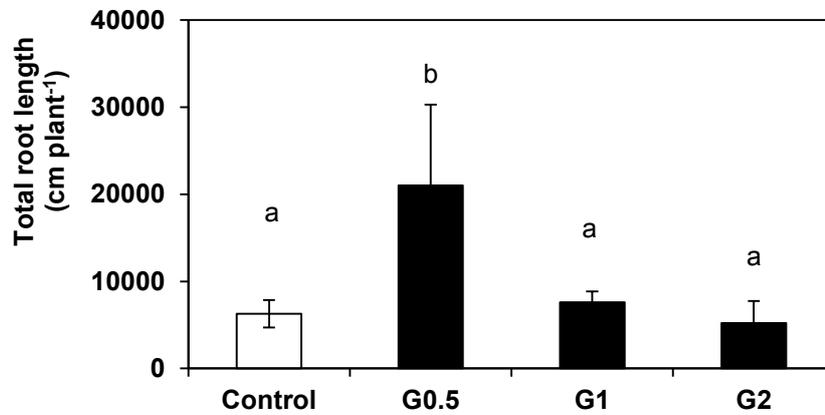


(b)

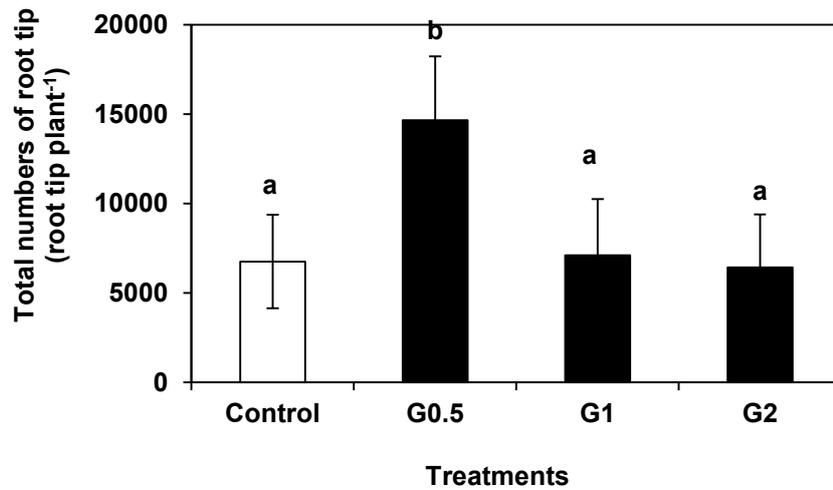


**Fig.4.2** Effect of different rates of gypsum fertilizer application on (a) shoot dry weight, and (b) root dry weight. Means followed by a common letter are not statistically significant different by using least significant difference (LSD) at  $p=0.05$ . Error bars indicate standard deviation,  $n=3$ .

(a)

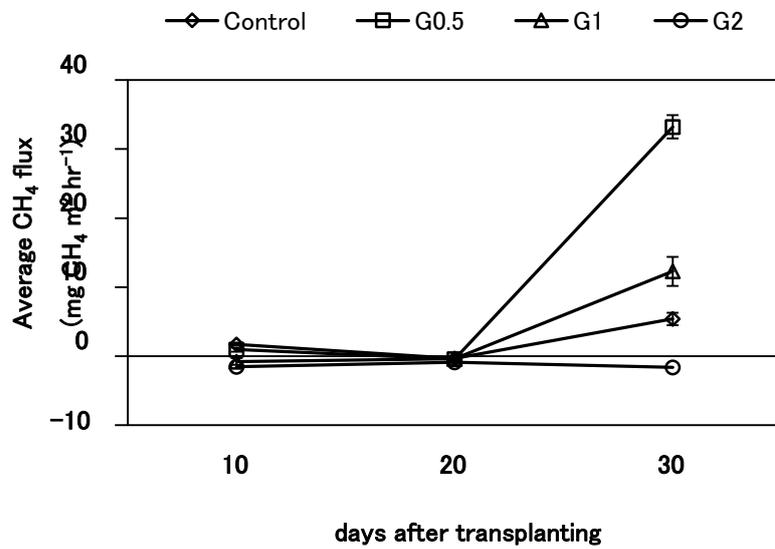


(b)

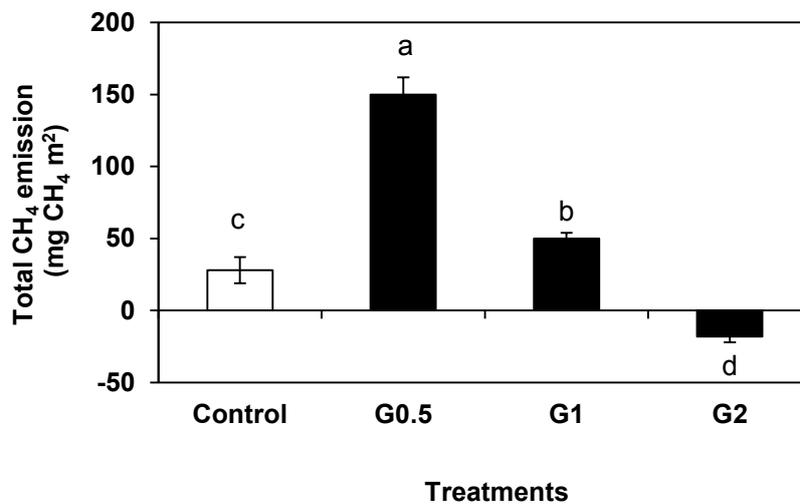


**Fig.4.3** Effect of different rates of gypsum fertilizer application on (a) total root length, (b) total numbers of root tips. Means followed by a common letter are not statistically significant difference by using least significant difference (LSD) at  $p=0.05$ . Error bars indicate standard deviation,  $n=3$ .

(a)



(b)



**Fig.4.4** Effect of different rates of gypsum fertilizer application on total CH<sub>4</sub> emission; (a)

average CH<sub>4</sub> flux pattern, and (b) total CH<sub>4</sub> emission for 30 days. Means followed by

a common letter are not statistically significant different by using least significant

difference (LSD) at p=0.05. Error bars indicate standard deviation, n=3.

## **Chapter 5**

### **Study on organic acids exudation by rice plants in coastal saline soil with different Na and Ca ratios by adding gypsum fertilizer**

The objective of this study is;

To identify the quantity and quality of root exudation of rice plants grown in coastal saline soils with different Na/Ca ratios

## **5.1 Materials and methods**

### **5.1.1 Root exudates collection method**

This study was conducted at the end of rice cultivation experiment with pot in chapter (4). The rice plants from pot experiment of chapter (4) were taken out with the intact soil pot and immediately used for the root exudates collection. Firstly, the soils around the roots were removed by shaking gently inside the water in a plastic box. Then, the remaining soils attached to the roots were washed gently with water. The soil washing process was conducted carefully to avoid damage to the roots. The root exudates of rice plants under each treatment were collected by using filter paper method which is slightly modified to the method of Neumann (1999). The rice plants were taken out with the intact soil from the pot. Subsequently, ten not damaged root tips were selected to collect the root exudates. The tip of the root was sandwiched between two pieces of 7 mm diameter filter paper (4A, ADVANTEC). After that, the filter papers with root were wetted by putting one drop of milli-Q water. After 30 minutes, the total 20 pieces of filter papers were put into centrifugation tube fitted with 0.45  $\mu\text{m}$  filter paper (Centricut ultra-mini W-50, 50000MW, Kurabo, Japan). About (0.2 ml) of distilled water was added to the ultracentrifugation tube and the diluted root exudates was centrifuged for 10 minutes at 12000 rpm. The centrifuge tubes containing the filtrate were kept in the freezer until analysing the organic acids.

### 5.1.2 Root exudates analysis and calculation method

High performance liquid chromatography (HPLC) was used for determination of low molecular weight organic acids in soil (Van Hees et al., 1999). HPLC was performed on a Shimadzu Organic Acid Analysis System LC-10AD (Non-suppressor, post column type) equipped with electric conductivity detector CDD-6A and two Shim-pack 102H columns at a temperature of 40 °C. A 5 mM *p*-toluenesulfonic acid (pH 2.8) was used as a carrier solution at a flow rate of 0.8mL min<sup>-1</sup>. The analysis of organic acids was carried out by injecting a 100 µl of solution in the system. Individual organic acids were identified and calibrated by comparing retention times with those of standards prepared with the known amounts of Citric, Tartaric, Malic, Succinic, Lactic, Formic, Acetic, and Propionic acids. The concentration of each organic acid under each treatment was calculated by dividing the area of each organic acid with the value of recovery rate. To know the recovery rate, one sample was randomly selected and 4 solutions were prepared such as standard plus milli-Q water (1:1), sample plus milli-Q water (1:1) and standard plus sample (1:1). The recovery rate of each organic acid in a selected sample was calculated as follows.

$$F = 100 * (D-B)/A \quad (1)$$

F = recovery rate (%)

A= the area of standard plus milli-Q water

B= the area of selected sample plus milli-Q water

D= the area of standard plus selected sample

## 5.2 Results and discussion

### 5.2.1 Organic acids exudation from coastal saline soil with different Na/Ca ratios

Six organic acids species such as Citric, Tartaric, Malic, Lactic, Formic and Acetic acids were observed in the root exudates of control, G0.5, and G1 (Fig. 5.1). Only four organic acids species, Citric, Tartaric, Malic, and Lactic acids were observed in G2 (Fig. 5.2). Among the 6 organic acid species, Lactic acid exudation was the highest and the range of lactic acid exudation was observed as 3.9 to 7.9  $\mu\text{mol ml}^{-1}$  root tip<sup>-1</sup>day<sup>-1</sup>. The minimum amount of exudation was observed in Acetic acid and the range of its exudation was found as 0.08 to 0.4  $\mu\text{mol ml}^{-1}$  root tip<sup>-1</sup>day<sup>-1</sup>. The order of organic acid exudation in root exudates was observed as Lactic acid > Malic acid > Tartaric acid > Formic acid > Citric acid > Acetic acid. Furthermore, the higher values of organic acid exudation potential were observed in gypsum fertilizer treatments over that control (Fig. 5.3).

The higher concentration of total organic acids exudation (Fig. 5.3) might relate with higher EC values in G0.5 and G1 (Table 4.2). Under the salt stress, most of the plants accumulate low molecular weight organic solutes inside (Jouyban, 2012), because plants need to maintain internal water potential below that of soil and maintain turgor and water uptake for growth (Tester and Davenport, 2003). Organic acids can serve as an active metabolic solute, regulating osmotic pressure and balancing excess cation (Sun et al., 2002). In order to maintain their charge balance, roots release protons whenever they take up more

cations than anions, and take up protons when the opposite occurs (Hinsinger et al., 2003). Lo'pez-Bucio et al. (2000) also pointed that plants would activate more metabolites and some unknown substances under environmental stress. The quantities and qualities of root exudates can be determined by plant species, the ages of individual plants, and external factors such as biotic and abiotic stressors (Badri and Vivanco, 2009). Many studies have shown that plants' roots under Aluminium stress exude higher concentration of Malic acid in wheat crop (Delhaize et al., 1993b, Basu et al., 1994). In our study, higher concentration of Malic acid and Tartaric acid was observed in gypsum fertilizer treatments that showed higher EC values, which might be due to the metabolic regulation of organic acids under salinity stress.

Plants discharge about 5 to 21% of photosynthetic carbon as root exudates (Kumar et al., 2006) and root exudation clearly represents a significant carbon cost to the plant (Walker et al., 2014). Therefore, higher amount of carbon contribution from root exudates under gypsum fertilizer addition as shown in Fig.5.3 may induce numerically lower shoot dry weight in G0.5 and G1, and statistically lower shoot dry weight in G2 compared to control. The lowest concentrations and less species of organic acid in root exudates of G2 compared with G0.5 and G1 might also relate with its lower shoot dry weight because the significant reduction of shoot dry weight in G2 may reduce the extent of photosynthetic carbon which may result in less species and lower concentrations of organic acids in G2.

## **5.1 Conclusion**

It can be concluded that if the soil is more saline, the rice plants will secrete more organic acids via root exudation. Under saline condition, the amount of organic acids exuded by rice plants to the soil may depend on shoot and root growth especially root tips of rice plants. Although the highest amount of Malic acid and Tartaric acids was observed in gypsum fertilizer treatments, the cellular mechanism of those acids in rice plants was unknown yet.

**Table 5.1** Different Na<sup>+</sup>/Ca<sup>2+</sup> ratios of coastal saline soil after adding different rates of gypsum fertilizer

Ion conc:	Initial soil	Final soil (After adding gypsum)		
		Gypsum (0.5 tonha <sup>-1</sup> )	Gypsum (1 tonha <sup>-1</sup> )	Gypsum (2 tonha <sup>-1</sup> )
Na(mM)	33	33	33	33
Ca(mM)	4	5.5	7	10
<b>Na/Ca</b>	<b>8</b>	<b>6</b>	<b>4</b>	<b>3</b>

**Table 5.2.** Soil environment under different Na/Ca ratios

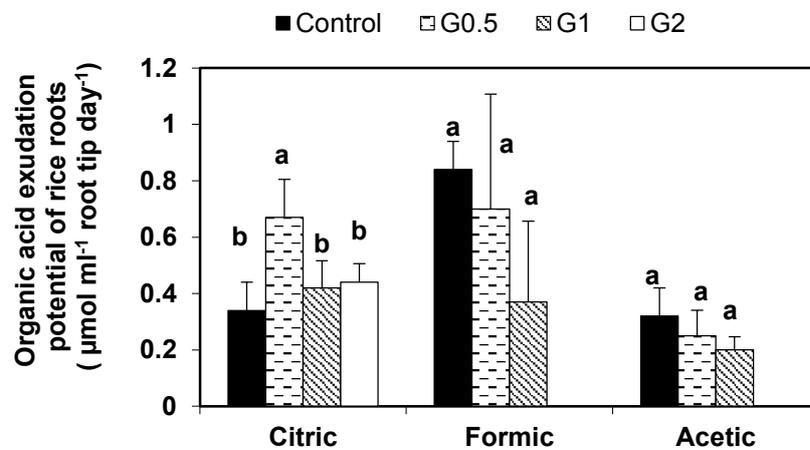
<b>Treatments</b>	<b>Soil pH</b>	<b>Soil EC (dSm<sup>-1</sup>)</b>
Na/Ca = 8 (Control )	6.5 ± 0.1a	4.9 ± 0.7b
Na/Ca= 6 (G0.5)	6.0 ± 0.1b	5.3 ± 1ab
Na/Ca=4 (G1)	5.9 ± 0.0bc	7.5 ± 0.2a
Na/Ca=3 (G2)	5.8 ± 0.0c	7.7 ± 0.3a

**ANOVA test for soil pH and EC**

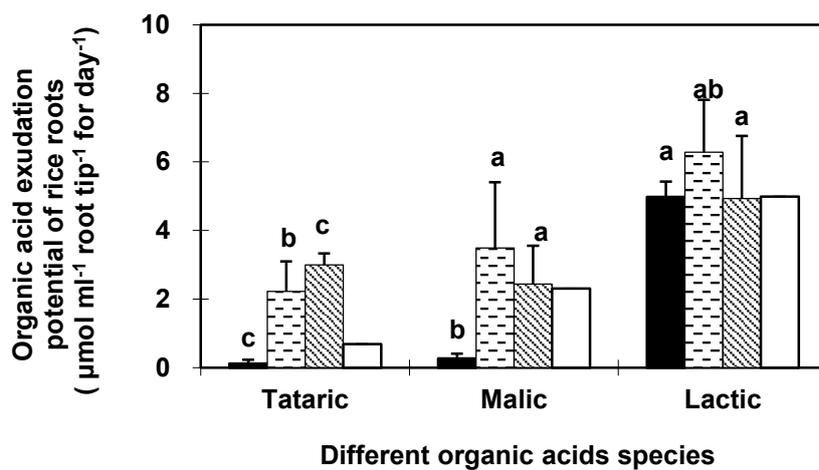
<b>Source of variation</b>	<b>Soil pH</b>	<b>Soil EC (dSm<sup>-1</sup>)</b>
Treatments	**	**

\*\* , \* and ns stand for significance at 1%, 5% and non-significance respectively.

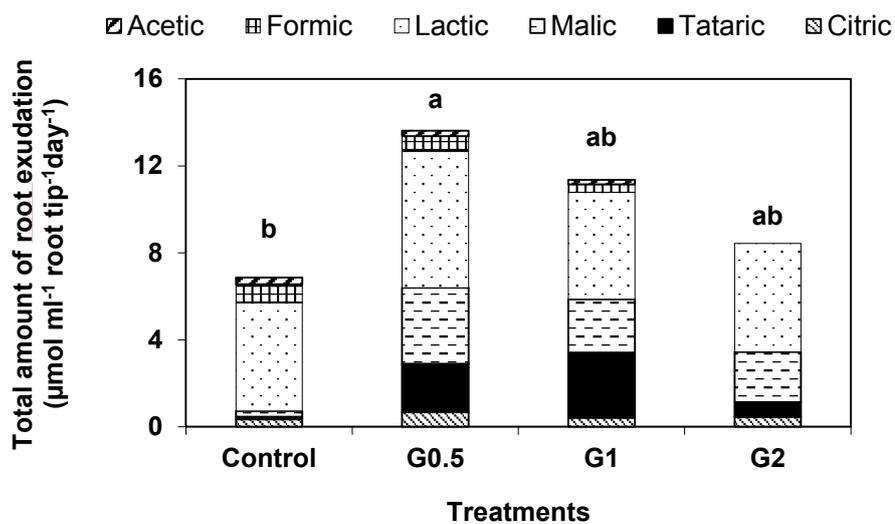
(a)



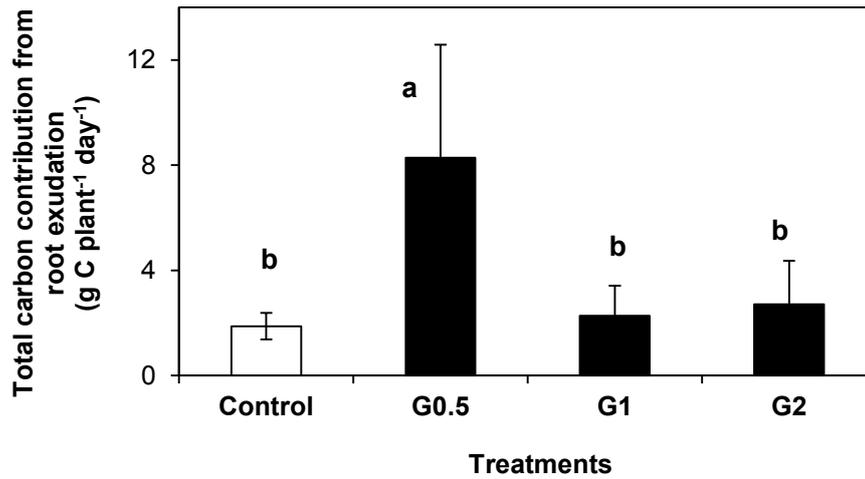
(b)



**Fig.5.1** Organic acids exudation potential of rice under different Na/Ca ratios. Means followed by a common letter within each acids are not statistically significant different by using least significant difference (LSD) at  $p=0.05$ . Error bars indicate standard deviation,  $n=3$ .



**Fig.5.2** Effect of different rates of gypsum fertilizer addition on total amount of organic acids exudation potential per root tip per day. Means comparison is done for the total organic acid exudation of rice plants under each treatment. Means followed by a common letter are not statistically significant different by using least significant difference (LSD) at  $p=0.05$ .



**Fig. 5.3** Effect of different rates of gypsum fertilizer addition on total carbon contribution from rhizosphere exudation of rice plants. Means followed by a common letter are not statistically significant different by using least significant difference (LSD) at  $p=0.05$ .

## **Chapter 6**

### **General discussion and overall conclusion**

## **6.1 Methane emission in paddy rice soil affected by saline irrigation water**

Paddy rice fields are identified as one of the major sources of global warming  $\text{CH}_4$  gas (IPCC, 2007). The extent of  $\text{CH}_4$  emission is largely influenced by the soil condition to which rice is planted and management practices associated with rice cultivation. Rice is widely grown in coastal line of humid tropic (Reddy, 2014). In those regions, rice fields are affected by salinity due to periodic sea water intrusion during the rainy season and irrigation water salinity during the dry season. The most prevalent salt in the soil solution of coastal saline soil is NaCl which show the most detrimental effect on rice growth. Therefore,  $\text{CH}_4$  emission was evaluated in paddy rice soil affected by saline irrigation water with NaCl in chapter 2 and 3.

Methane emission from paddy field results from a balance of  $\text{CH}_4$  production by methanogenic bacteria, methane oxidation by methanotrophic bacteria and transportation by rice plant. (Win et al., 2012). The complexity of the influencing factors of  $\text{CH}_4$  emission under different irrigation water salinity levels from the current experiment is summarized in Fig. 6.1.

Under saline conditions (S25, S30 and S90), rice growth are decreased as shown in Table 2.3 and Table 3.2. The significant higher dead leaves were observed in all salinity levels such as S25 and S30 and S90. Wassmann and Aulakh (2000) reported that  $\text{CH}_4$  production in rice fields largely depends on plant-borne material that can be either decaying tissue or root exudates. Thus, the increase in total numbers of dead leaves under saline condition may contribute the carbon source for methanogenic bacteria. In another side,

salinity significantly suppressed the above-ground biomass yield in S30 and S90. As rice plants are the major transporters of CH<sub>4</sub> gas from soil to atmosphere, the decrease in above-ground biomass yield in S30 and S90 may reduce CH<sub>4</sub> transport capacity of rice plants.

According to the results of laboratory incubation experiment, CH<sub>4</sub> production was increased up to salinity 30 mmol L<sup>-1</sup> NaCl (Table 2.2). This increase in CH<sub>4</sub> production in S30 might be due to the Na requirement of methanogens. According to the report of Jarrell and Kalmokoff (1988), sodium is required for amino acid transport, growth, methanogenesis, and internal pH regulation in methanogenic bacteria. In case of CH<sub>4</sub> emission in paddy rice cultivation, CH<sub>4</sub> emission in S30 was not significant different with CH<sub>4</sub> emission in control. Although CH<sub>4</sub> production is enhanced in S30, no difference of CH<sub>4</sub> emission value between S30 and control might be due to lower above-ground biomass yield of S30. As rice plants serve as major CH<sub>4</sub> transporters from soil to atmosphere, the reduced above ground biomass yield may limit the CH<sub>4</sub> transport capacity by rice plants. The addition of S25 to paddy rice cultivation could not suppress above-ground biomass yield. Thus, CH<sub>4</sub> transportation factor is not limited in S25. Although enhancing effect of CH<sub>4</sub> production was observed up to salinity level S30 in laboratory incubation experiment, no difference of CH<sub>4</sub> emission between S25 and control might be due to the existence of a higher NO<sub>3</sub><sup>-</sup> concentration in S25 in the early growth stage (Fig. 3.2). At the highest salinity level (S90), CH<sub>4</sub> production was decreased. It might be due to Cl<sup>-</sup> toxicity under higher additions of NaCl. The lowest CH<sub>4</sub> emission in paddy rice cultivation was also observed in S90. Thus, this

lowest CH<sub>4</sub> emission in S90 might be due its lowest CH<sub>4</sub> production potential, the shortest growth duration period and the lowest above ground biomass yield.

To improve the rice growth in Na<sup>+</sup> saturated soil/water, gypsum fertilizer is commonly applied as Ca<sup>2+</sup> source. Therefore, the effect of gypsum fertilizer addition to paddy rice soil on rice growth and CH<sub>4</sub> emission was also studied in non-saline and saline condition. In chapter 1, the gypsum 1 ton ha<sup>-1</sup> was applied to both non-saline and saline conditions (S30 and S90). The addition of gypsum fertilizer significantly reduced in CH<sub>4</sub> emission in both non-saline and saline condition (Fig. 2.5). In the presence of SO<sub>4</sub><sup>2-</sup>, SO<sub>4</sub><sup>2-</sup>-reducing bacteria will out compete methanogens for the same substrates such as hydrogen (H<sub>2</sub>) or carbon dioxide (CO<sub>2</sub>) or Acetate (CH<sub>3</sub>COO<sup>-</sup>) that methanogens use in methane production (Lindau et al., 1993; Van der gon and Neue, 1994; Epule et al., 2011). Therefore, the reduction of CH<sub>4</sub> emission following the addition of gypsum (Fig. 2.5) might be due to the competition between SO<sub>4</sub><sup>2-</sup>-reducing bacteria and CH<sub>4</sub>-producing bacteria for substrates as stated by Lindau et al. (1993), Van der gon and Neue (1994) and Epule et al. (2011). In chapter 3, three different rates of gypsum fertilizer are applied to the paddy field under non-saline and saline condition with salinity level S25. The higher application gypsum resulted higher reduction in CH<sub>4</sub> emission. However, the amount of reduction due to gypsum fertilizer addition is lower in saline condition than that of non-saline condition. It might be due to the availability of C source from the dead plant materials under saline condition.

## 6.2 Methane emission in coastal saline soil

Although NaCl is the most abundant ion in most saline soils, the soil solution of saline soils is composed of a range of dissolved salts, such as NaCl, Na<sub>2</sub>SO<sub>4</sub>, MgSO<sub>4</sub>, CaSO<sub>4</sub>, MgCl<sub>2</sub>, KCl, and Na<sub>2</sub>CO<sub>3</sub>. Thus, CH<sub>4</sub> emission was evaluated in coastal saline soil in chapter 4 and 5. To provide appropriate Na/Ca ratios for rice growth, different rates of gypsum fertilizer such as 0.5, 1 and 2 ton ha<sup>-1</sup> were added to the coastal saline soil. After adding gypsum fertilizers treatment, the estimated Na/Ca ratios were 8, 6, 4, and 3 in control, G0.5, G1 and G2, respectively.

The addition of gypsum fertilizers (CaSO<sub>4</sub>.2H<sub>2</sub>O) resulted to decrease soil pH and increase in soil EC. At the same time, it will also increase in soil SO<sub>4</sub><sup>2-</sup> concentration. Furthermore, The K<sup>+</sup> ion concentration of plants was increased and Na<sup>+</sup> ion concentration of plant was decreased in G0.5 and G1 (Fig. 4.1). However, no improvement in ion balance and the significant reduction of shoot dry weight was observed in G2. It might be due to the excess Ca<sup>2+</sup>/SO<sub>4</sub><sup>2-</sup> ion concentration in G2. Although gypsum fertilizer is known to inhibit CH<sub>4</sub> emission, the higher total CH<sub>4</sub> emission in G0.5 was observed over 30 days study. The lowest CH<sub>4</sub> emission was observed in G2. Thus, the possible mechanism of CH<sub>4</sub> emission in coastal saline soil will be explained as shown in Fig. 6.2.

Under the salt stress, most of the plants accumulate low molecular weight organic solutes (Jouyban, 2012). Organic acids can serve as an active metabolic solute, regulating osmotic pressure and balancing excess cation (Sun et al., 2002). In order to maintain their

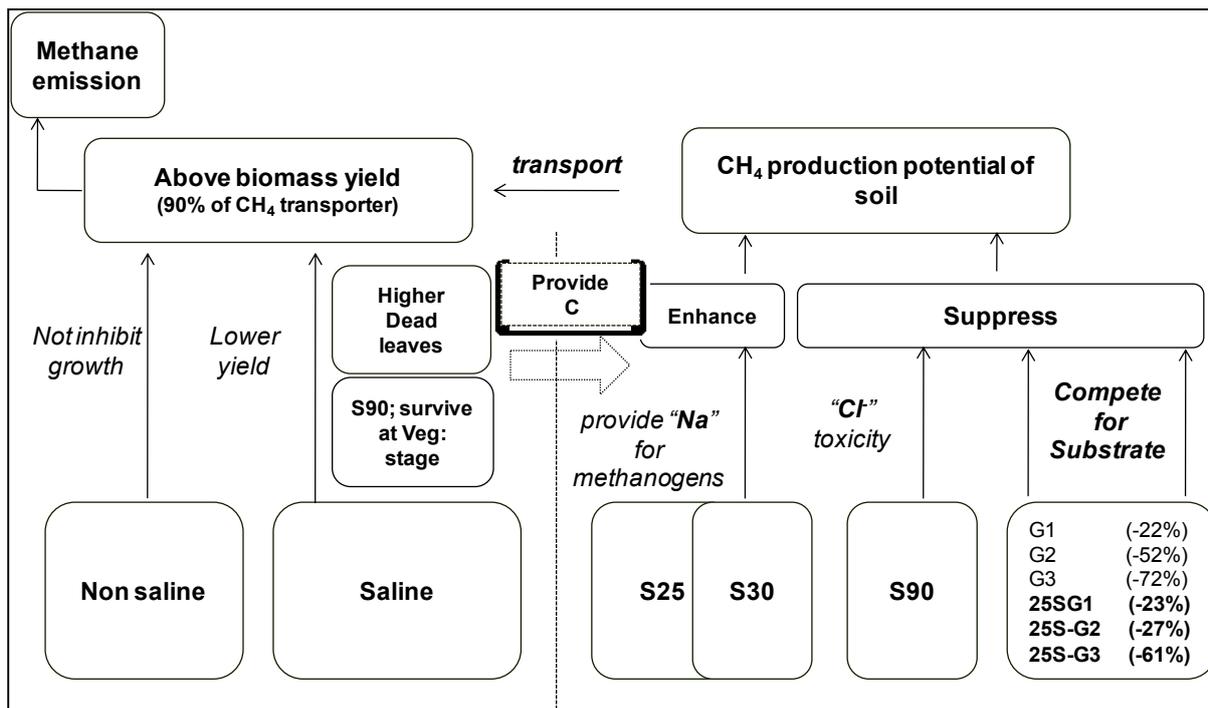
charge balance, roots release protons whenever they take up more cations than anions, and take up protons when the opposite occurs (Hinsinger et al., 2003). As a result, higher organic acids exudation was observed in gypsum fertilizer treatments, which showed higher EC values, compared to that of control (Fig. 5.2). Among the gypsum fertilizer treatments, the significant highest organic acids exudation potential was observed in G0.5. The less organic acids species and lowest organic acids exudation potential was observed in G2. It might be due to the lowest shoot dry weight which may turn to reduce photosynthetic carbon of plant because plants discharge about 5 to 21% of photosynthetic carbon as root exudates (Kumar et al., 2006). Thus, the highest CH<sub>4</sub> emission in G0.5 might be due to the level of SO<sub>4</sub><sup>2-</sup> input was not enough to inhibit CH<sub>4</sub> production, the highest C availability from root exudation. The lowest CH<sub>4</sub> emission in G2 might relate with its lowest shoot dry weight, pH value, and highest level of SO<sub>4</sub><sup>2-</sup> input.

### **6.3 Overall conclusion**

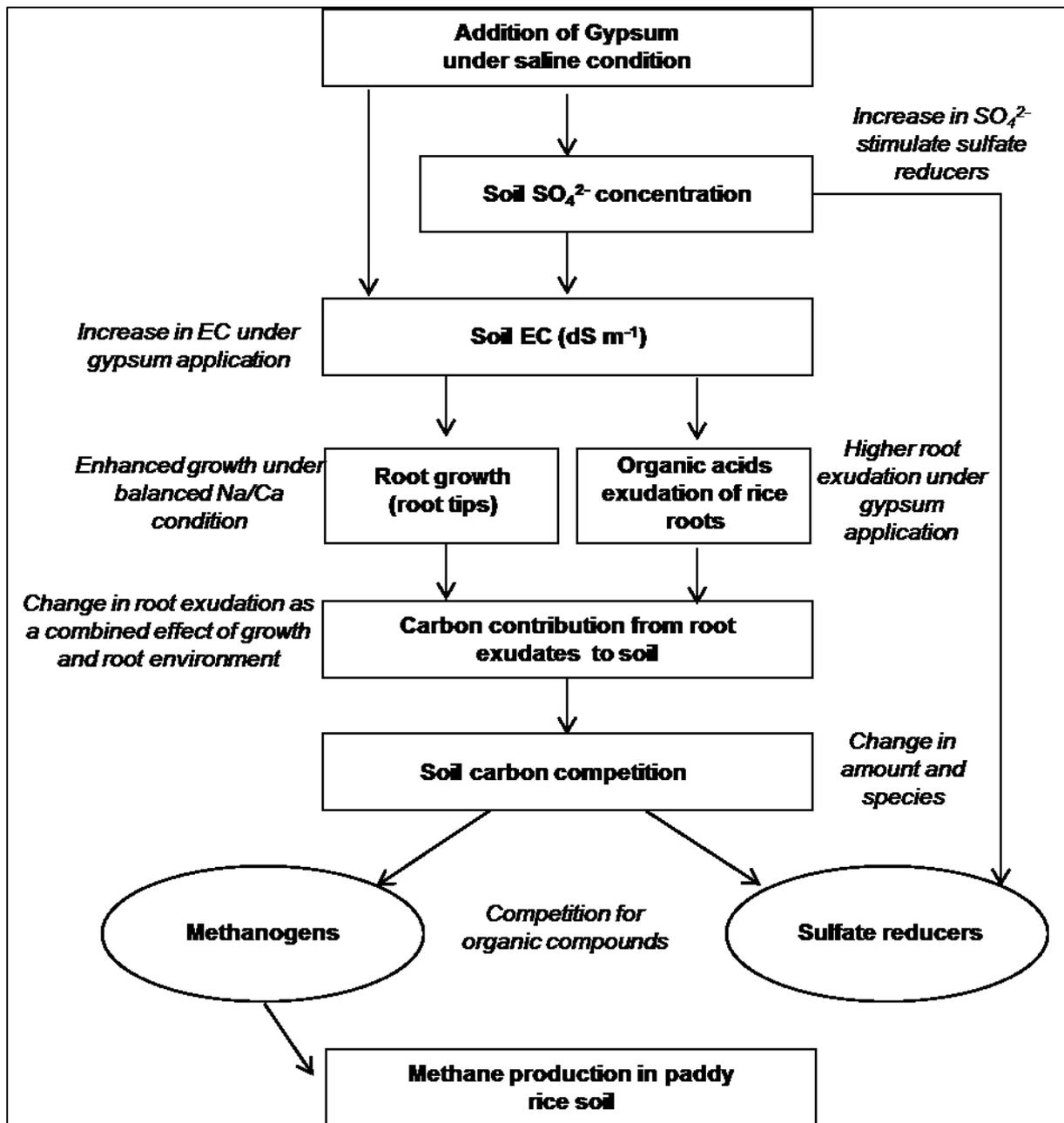
The continuous irrigation of saline water can severely affects the rice growth. To reduce the irrigation water salinity problem in rice cultivation, irrigation water with salinity level up to 25 mmol L<sup>-1</sup> NaCl can be used intermittently if no fresh water is available for rice cultivation. To improve the ion balance of rice plants which are grown in saline soil, the appropriate Na/Ca ratio around 6 should be maintained.

Although salinity is generally known to suppress CH<sub>4</sub> emission, the addition of water with salinity level up to 30 mmol L<sup>-1</sup> NaCl favours CH<sub>4</sub> production and could not suppress

CH<sub>4</sub> emission in rice cultivation. In the paddy rice cultivation, Although gypsum fertilizer is known to reduce CH<sub>4</sub> emission from paddy rice soil, the level of CH<sub>4</sub> emission under gypsum fertilizer addition depends on both the level of SO<sub>4</sub><sup>2-</sup> added and the carbon availability of the soil. As rice plants excrete more organic acids to adjust the osmotic stress under saline condition, the organic acids of root exudates clearly represent the main carbon pool for CH<sub>4</sub> emission in saline condition. However, the extent of organic acids exudation may depend on the shoot and root growth of rice plants.



**Fig.6.1** A schematic diagram for chapter 2 and 3 explaining the mechanism of CH<sub>4</sub> emission relation C and SO<sub>4</sub><sup>2-</sup> availability under different levels of irrigation water salinity and gypsum application rates



**Fig.6.2** A schematic diagram for chapter 4 and 5 explaining the mechanism of CH<sub>4</sub> emission in relation to C and SO<sub>4</sub><sup>2-</sup> availability under different rates of gypsum fertilizer application

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